## Supporting Information

# Reactions of Gas-phase Uranyl Formate/Acetate Anions: Intra-complex Hydride Attack to Convert Carboxylates to Aldehydes 

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## LIST OF FIGURES

Figure S . Electrospray ionization spectrum generated from mix of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CH}_{3}\right)_{2}\right]$ and $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)_{2}\right]$ in $50: 50 \mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$. Asterisks identify peaks for which acetate is replaced by nitrate, which is a residual species in the capillary of the electrospray ionization source.

Figure S2. Product ion spectra derived from MS ${ }^{n} \mathrm{CID}$ of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)_{2}\right]$ : (a) CID (MS/MS stage) of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)_{2}\right]^{-}$at $\mathrm{m} / \mathrm{z} 439$, (b) CID ( $\mathrm{MS}^{3}$ stage) of dissociation product ion at $m / z$ 395, (c) CID (MS ${ }^{4}$ stage) of dissociation product ion at $m / z$ 348, and (d) CID (MS ${ }^{5}$ stage) of dissociation product ion at $m / z$ 301. In the spectra, the circles and arrows illustrate the $\mathrm{MS}^{n}$ pathway. In each spectrum, the bold peak label indicates the precursor selected for CID while labels in italics represent the products from dissociation or ion-molecule reactions as indicated in the text.

Figure S3. Product ion spectra derived from MSn ${ }^{n}$ CID of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)_{2}\right]^{-}$continued: (a) CID (MS ${ }^{3}$ stage) of product ion at $m / z 392$ generated by initial CID of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)_{2}\right]^{-}$at $m / z 439$, (b) CID (MS ${ }^{4}$ stage) of dissociation product ion at $m / z 348$. The bold peak labels indicate the precursor selected for CID while labels in italics represent the products from dissociation or ion-molecule reactions as indicated in the text.

Figure S4. Product ion spectra generated by isolation of ion at $m / z$ 389, without imposed collisional activation, in the ion trap for reaction with $\mathrm{H}_{2} \mathrm{O}$ (ca. $1 \times 10^{-6}$ torr): (a) 1 ms , (b) 10 ms , (c) 100 ms , (d) 1 s and (e) CID of $m / z 405$ reaction product.

Figure S5. Product ion spectra generated by isolation of ion at $\mathrm{m} / \mathrm{z} 301$ (derived from CID of unlabeled precursor), without imposed collisional activation, in the ion trap for reaction with $\mathrm{H}_{2} \mathrm{O}$ (ca. $1 \times 10^{-6}$ torr): (a) 1 ms , (b) 10 ms , (c) 100 ms and (d) 1 s .

Figure S6. Product ion spectra generated by isolation of ion at $\mathrm{m} / \mathrm{z} 304$ (derived from CID of complex containing $\mathrm{CD}_{3} \mathrm{CO}_{2} \mathrm{H}$ ), without imposed collisional activation, in the ion trap for reaction with $\mathrm{H}_{2} \mathrm{O}$ (ca. $1 \times 10^{-6}$ torr): (a) 1 ms , (b) 10 ms , (c) 100 ms and (d) 1 s .

Figure S7. Product ion spectra derived from $\mathrm{MS}^{n} \mathrm{CID}$ of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)\right]$ : (a) CID (MS/MS stage) of $\left[\mathrm{UO}_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)_{2}\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CD}_{3}\right)\right]^{-}$at $\mathrm{m} / \mathrm{z} 422$, (b) $\mathrm{CID}\left(\mathrm{MS}^{3}\right.$ stage) of dissociation product ion at $m / z 378$ and (c) CID (MS ${ }^{4}$ stage) of dissociation product ion at $m / z 348$. In the spectra, the
circles and arrows illustrate the $\mathrm{MS}^{n}$ pathway. In each spectrum, the bold peak label indicates the precursor selected for CID while labels in italics represent the products from dissociation or ion-molecule reactions as indicated in the text.

Figure S8. Product ion spectra generated by isolation of ion at $m / z 375$, without imposed collisional activation, in the ion trap for reaction with $\mathrm{H}_{2} \mathrm{O}$ (ca. $1 \times 10^{-6}$ torr): (a) 1 ms , (b) 10 ms , (c) 100 ms , (d) 1 s and (e) CID of $m / z 391$ reaction product.

Figure S9. Relevant minima and transition state structures for the dissociation of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)_{2}\right]^{-}$.
Figure S10. The relevant minima and transition state structures for the dissociation of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}\right.\right.$ -$\left.\left.\mathrm{CH}_{3}\right)_{2}\right]^{-}$.

Figure S11. A reaction energy diagram for CID of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CH}_{3}\right)_{2}\right]^{-}$.
Figure S12. The relevant minima and transition state structures for the dissociation of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}\right.\right.$ $\left.\left.\mathrm{CH}_{3}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\right]^{-}$.

Figure S13. Relevant minima and transition state structures for the reactions of $\left[\mathrm{UO}_{2}(\mathrm{O})\left(\mathrm{CH}_{3}\right)\right]^{-}$with $\mathrm{H}_{2} \mathrm{O}$.

## LIST OF TABLES

Table S1. The zero-point corrected electronic energies and free energies for minima and transition states for intra-complex hydride attack during CID of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)_{2}\right]^{-}$.

Table S2. The zero-point corrected electronic energies and free energies for minima and transition states for intra-complex hydride attack during CID of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{CH}_{3}\right)_{2}\right]$

Table S3. The zero-point corrected electronic energies and free energies for minima and transition states for intra-complex hydride attack during CID of $\left[\mathrm{UO}_{2}(\mathrm{H})\left(\mathrm{O}_{2} \mathrm{C}\right.\right.$ $\left.\left.\mathrm{CH}_{3}\right)\left(\mathrm{O}_{2} \mathrm{C}-\mathrm{H}\right)\right]^{-}$.

Table S4. The zero-point corrected electronic energies and free energies for minima and transition states for reaction of $\left[\mathrm{UO}_{2}(\mathrm{O})\left(\mathrm{CH}_{3}\right)\right]^{-}$with $\mathrm{H}_{2} \mathrm{O}$.


Figure S1.


Figure S2.


Figure S3.


Figure 54.


Figure 55.


Figure 56.


Figure 57.


Figure 88.


TSII $\rightarrow$ III


Figure S9.






Figure S10.


Figure S11.


Figure S12


Figure S13.

Table S1.

| Structure | Zero-point corrected electronic energy | Thermally corrected free energy |
| :---: | :---: | :---: |
| I | -1004.615232 | -1004.654507 |
| II | -1004.604059 | -1004.644649 |
| TS II->III | -1004.565179 | -1004.604376 |
| III | -1004.574499 | -1004.615464 |
| IV | -890.057297 | -890.092941 |
| formaldehyde | -114.502636 | -114.524283 |

Table S2.

| Structure |  |
| :---: | :---: |
| l-ac | Zero-point corrected electronic energy |
| -1083.221641 |  |

I-ac
II-ac
TS II-ac->III-ac
III-ac
IV -ac
acetaldehyde
Structure
V
VI
TS VI->VII
VII
VIII
acetaldehyde
IX
TS IX->X
X
XI
formaldehyde

Zero-point corrected electronic energy
Table S3.

| -1043.918513 | -1043.961161 |
| :--- | :--- |
| -1043.906032 | -1043.947874 |
| -1043.864826 | -1043.905718 |
| -1043.884324 | -1043.929082 |
| -890.057297 | -890.092941 |
| -153.810131 | -153.835063 |
| -1043.907696 |  |
| -1043.868536 | -1043.950918 |
| -1043.877684 | -1043.910748 |
| -929.359691 | -1043.922421 |
| -114.502636 | -929.397871 |
|  | -114.524283 |

Table S4.

| Structure | Zero-point corrected electronic energy | Thermally corrected free energy |
| :---: | :---: | :---: |
| XII | -740.656324 | -740.690521 |
| $\mathrm{H}_{2} \mathrm{O}$ | -76.423266 | -76.440903 |
| XIII | -817.101013 | -817.138227 |
| TSXI II-> XIV | -817.100974 | -817.137112 |
| XIV | -817.145404 | -817.184582 |
| V | -776.660695 | -776.694344 |
| CH $_{4}$ | -40.480523 | -40.497825 |
|  |  |  |
| XVI | -817.101055 | -817.138367 |
| TSXVI->XVII | -817.094361 | -817.130389 |
| XVII | -817.143981 | -817.180702 |

