### **Supporting Information**

## Host-guest interaction induced room-temperature phosphorescence enhancement of organic dyes: a computational study

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**Table S4.** Calculated transition orbitals assignment and composition (%) of atoms in frontier molecule orbitals of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] in S<sub>1</sub> state and T<sub>1</sub> state calculated by NAO method.

**Table S5.** Calculated vertical transition energy (eV), oscillator strength (f), assignments and transition nature in S<sub>n</sub> of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7].

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at S<sub>1</sub> ( ${}^{\mu_{S_1}-s_0}$ ) and T<sub>1</sub> ( ${}^{\mu_{T_1}-s_0}$ ), energy gap between the S<sub>1</sub> and T<sub>1</sub> ( ${}^{\Delta E_{T_1}-s_0}$ ), spin orbit coupling constant ( ${}^{\xi_{T_1}-s_1}$ ) between the S<sub>1</sub> and T<sub>1</sub>, and the ratio of  ${}^{\xi_{T_1}-s_1}$  to  ${}^{\Delta E_{T_1}-s_0}$  ( ${}^{\xi_{T_1}-s_1/\Delta E_{T_1}-s_1}$ ).

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**Table S8.** Calculated phosphorescence radiation rate ( $k_P$ ) of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] at BMK/6-31G\*\* level in crystal.

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## III. The optimized geometries and the corresponding coordinates of the QM region of the PYCl, PYBr and their complexes QM/MM models at both $S_0$ and $T_1$ states.

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**Fig. S1** The optimized unit cell structure of four host-guest complexes: (a) PYCl/CB[6], (b) PYBr/CB[6], (c) PYCl/CB[7], (d) PYBr/CB[7].



Fig. S2 The QM/MM model of (a) PYCl, (b) PYCl/CB[6] and (c) PYCl/CB[7] with the representative distance between the plane of  $PY^+$  and the counterion labeled.



**Fig. S3** The wavelength of Phosphorescence at  $T_1$ -geometry calculated by different functionals, including TPSSH, B3LYP, BMK and M06-2X coupled with 6-31G\*\* basis set for PYBr.



**Fig. S4** Single-crystal X-ray diffraction analysis of (a) PYCl, (b) PYBr. The dotted line shows the  $\pi$ - $\pi$  interaction force between two adjacent molecules and hydrogen bond. Here, H-bond with the distance between the donor (D) and acceptor (A) atoms is less than 3.5 Å, and the angle of D-H-A is greater than 135°.



C-H…O (d1, d2, d3, d4) : 1.476, 1.632, 1.790, 2.407 C-H…N (d5, d6) : 2.538, 2.801

PYCI/CB[6]





 $\begin{array}{l} \textbf{C-H} \cdots \textbf{O} \ (\textbf{d1}, \textbf{d2}, \textbf{d3}) : 1.743, 1.940, 2.542 \\ \textbf{C-H} \cdots \textbf{N} \ (\textbf{d4}, \textbf{d5}, \textbf{d6}) : 2.057, 2.443, 2.813 \end{array}$ 

PYBr/CB[6]







C-H…O (d1, d2, d3) : 1.491, 2.211, 2.231 C-H…N (d4) : 3.028

#### PYBr/CB[7]

Fig. S5 Intermolecular interactions of (a) PYCl/CB[6], (b) PYBr/CB[6], (c) PYCl/CB[7], (d) PYBr/CB[7].

b



**Fig. S6** The structural superposition and RMSD values of (a) PYCl, (b) PYBr, (c) PYCl/CB[6], (d) PYBr/CB[6], (e) PYCl/CB[7], (f) PYBr/CB[7] in the optimized S<sub>0</sub> and T<sub>1</sub> geometries in crystals.



Fig. S7 (a) The frontier molecular orbitals at  $S_1$ -geometry of (a) PYCl/CB[7], PYBr/CB[7]. (b) The energy levels of key transition orbitals and the corresponding transition properties of the PYBr, PYBr/CB[6] and PYBr/CB[7] at  $S_1$ .



**Fig. S8** The transition orbitals for the excited singlet state  $S_n$  (n $\leq$ 4) of (a) PYCl, (b) PYBr, (c) PYCl/CB[6], (d) PYBr/CB[6], (e) PYCl/CB[7], (f) PYBr/CB[7].



**Fig. S9** The frontier natural molecule transition orbitals for the T<sub>1</sub> of (a) PYCl, (b) PYBr, (c) PYCl/CB[6], (d) PYBr/CB[6], (e) PYCl/CB[7], (f) PYBr/CB[7].



PYCIPYBrPYCI/CB[6]PYBr/CB[6]PYCI/CB[7]PYBr/CB[7]Fig. S10 Energy level of HOMO, LUMO, and the corresponding energy gap based on T1 geometries of the studied molecules. "H" means the "HOMO", "L" means the "LUMO".



Fig. S11 Transition orbitals of the high-lying triplet states of (a) PYCl and (b) PYBr.



b



Fig. S12 Transition orbitals of the high-lying triplet states of (a) PYCl/CB[6] and (b) PYBr/CB[6].



PYCI/CB[7]

PYBr/CB[7]

Fig. S13 Transition orbitals of the high-lying triplet states of (a) PYCl/CB[7] and (b) PYBr/CB[7].

## II. Supplementary Tables

<b>Table S1.</b> Table of unit cell parameters and cell volume for the host-guest complexes:	PYCl/CB[6], PYBr/CB[6],
PYC1/CB[7], PYBr/CB[7].	

Complexes	<i>a</i> (Å)	<i>b</i> (Å)	c (Å)	α	β	γ	Volume (Å <sup>3</sup> )
PYCl/CB[6]	20.74	13.97	10.68	90.00	120.97	90.00	2652.88
PYBr/CB[6]	20.86	13.99	10.88	90.11	121.42	89.87	2709.62
PYCl/CB[7]	12.85	20.13	31.66	90.00	92.59	90.00	8185.58
PYBr/CB[7]	12.95	19.91	32.27	89.97	91.26	89.96	8319.44

**Table S2.** Selected structural parameters of PYCl and PYBr at both the  $S_0$  and  $T_1$  geometry, the geometrical changes  $|\Delta(T_1-S_0)|$  in crystal, respectively.

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			PYCl			PYBr	
		$\mathbf{S}_0$	$T_1$	$\Delta  T_1\text{-}S_0 $	$S_0$	$T_1$	$\Delta  T_1 - S_0 $
	$Br-X^-$	3.00	2.99	0.01	3.26	3.23	0.03
	C1-Br	2.00	1.96	0.04	2.01	1.98	0.03
	C4-C7	1.48	1.41	0.07	1.48	1.41	0.07
	C2-C1-Br	120.12	120.76	0.64	119.27	119.59	0.32
	C11-C7-C4	122.13	122.61	0.48	120.17	120.07	0.10
$D_1$	C5-C4-C17- C11	0.70	1.08	0.38	15.91	8.13	7.78

**Table S3.** Selected structural parameters of PYCl/CB[6], PYBr/CB[6], PYCl/CB[7], PYBr/CB[7] at both the  $S_0$  and  $T_1$  geometry, the geometrical changes  $|\Delta(T_1-S_0)|$  in crystal, respectively.



		PYCl/CB[6]		F	PYBr/CB[6]		PYCl/CB[7]		PYBr/CB[7]				
		$S_0$	$T_1$	$\Delta  T_1 - S_0 $	$S_0$	$T_1$	$\Delta  T_1 - S_0 $	$S_0$	$T_1$	$\Delta  T_1 - S_0 $	$S_0$	$T_1$	$\Delta  T_1 - S_0 $
	C1-Br	2.05	2.00	0.05	2.13	2.10	0.03	2.02	1.99	0.03	2.02	1.99	0.03
	C4-C7	1.44	1.38	0.06	1.47	1.39	0.08	1.48	1.41	0.07	1.48	1.41	0.07
	C2-C1-Br	114.72	114.59	0.13	126.32	127.27	0.95	118.31	118.61	0.30	118.21	118.75	0.74
	C11-C7-C4	120.65	120.67	0.02	121.89	122.92	1.03	120.78	121.99	1.21	120.99	122.72	1.73
$D_1$	C5-C4-C17- C11	1.24	1.11	0.13	1.08	2.20	1.12	34.58	1.16	33.42	29.48	6.67	22.81

Compounds		$S_1$ $T_1$						
	Assignments	n	π	nature	Assignment	n	π	nature
DVC1		91 700/	16 100/	(12. 7*)	HOMO-3→LUMO (42%)	47.60%	51.30%	(π, π*
PICI	$HOMO \rightarrow LOMO (97\%)$	81.70%	16.10%	$(n, \pi^*)$	HOMO→LUMO (36%)	89.20%	5.70%	(n, π*)
PYBr	HOMO→LUMO (98%)	79.90%	18.30%	$(n, \pi^*)$	HOMO-3→LUMO (89%)	42.00%	62.10%	(π, π*
PYCl/CB[6]	HOMO→LUMO (97%)	30.10%	64.50%	$(\pi, \pi^*)$	HOMO→LUMO (89%)	25.60%	68.90%	(π, π*
PYBr/CB[6]	HOMO→LUMO (98%)	28.10%	67.80%	$(\pi, \pi^*)$	HOMO→LUMO (93%)	24.50%	71.10%	(π, π*]
PYCl/CB[7]	HOMO→LUMO (98%)	31.60%	66.10%	$(\pi, \pi^*)$	HOMO→LUMO (94%)	29.50%	67.80%	(π, π*)
PYBr/CB[7]	HOMO→LUMO (98%)	32.10%	65.30%	$(\pi, \pi^*)$	HOMO→LUMO (94%)	29.40%	67.70%	(π, π*)

**Table S4.** Calculated transition orbitals assignment and composition (%) of atoms in frontier molecule orbitals of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] in S<sub>1</sub> state and T<sub>1</sub> state calculated by NAO method.

Table S5. Calculated vertical transition energy (eV), oscillator strength (f), assignments and transition nature in	n S <sub>n</sub>
of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7].	

	States	Energy (eV)	f	Assignments	nature
	$\mathbf{S}_1$	3.06	0.0131	HOMO→LUMO (98%)	$(n, \pi^*)$
NACI	$S_2$	3.46	0.0001	HOMO-1→LUMO (99%)	(n, <b>π</b> *)
PYCI	$S_3$	3.48	0.0370	HOMO-2→LUMO (99%)	$(n, \pi^*)$
	$S_4$	4.14	0.0100	HOMO-3→LUMO (95%)	$(\pi, \pi^*)$
	$\mathbf{S}_1$	2.52	0.0065	HOMO→LUMO (98%)	$(n, \pi^*)$
<b>N</b> 40	$S_2$	2.60	0.0014	HOMO-1→LUMO (99%)	$(n, \pi^*)$
PYBr	$S_3$	2.66	0.0008	HOMO-2→LUMO (99%)	$(n, \pi^*)$
	$S_4$	4.23	0.0040	HOMO→LUMO+1 (64%) HOMO-3→LUMO (25%)	$(\pi,\pi^*)$
	$\mathbf{S}_1$	3.84	0.8343	HOMO→LUMO (95%)	$(\pi, \pi^*)$
PYCI/CB[6]	$S_2$	4.44	0.0149	HOMO-2→LUMO (58%) HOMO-1→LUMO (29%)	(n, π*)
	$\mathbf{S}_1$	4.20	0.8619	HOMO→LUMO (95%)	$(\pi, \pi^*)$
PYBr/CB[6]	$S_2$	4.64	0.0024	HOMO-1→LUMO (75%) HOMO→LUMO+1 (10%)	$(\pi,\pi^*)$
	$S_3$	4.76	0.0020	HOMO-2→LUMO+1 (53%) HOMO→LUMO+1 (20%)	$(n, \pi^*)$
	$\mathbf{S}_1$	4.00	0.6218	HOMO→LUMO (96%)	$(\pi,\pi^*)$
PYCl/CB[7]	$S_2$	4.35	0.0184	HOMO-1→LUMO (93%)	$(\pi,\pi^*)$
	$S_3$	5.02	0.0006	HOMO-1→LUMO+1 (5%) HOMO-2→LUMO+1 (85%)	$(n, \pi^*)$
	$\mathbf{S}_1$	4.18	0.8000	HOMO→LUMO (96%)	$(\pi,\pi^*)$
PYBr/CB[7]	$S_2$	4.47	0.0237	HOMO-1→LUMO (88%)	$(\pi, \pi^*)$
	$S_3$	5.10	0.0086	HOMO→LUMO+1 (23%) HOMO-2→LUMO (63%)	$(n, \pi^*)$

**Table S6.** The calculated properties of the excited states of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7], including oscillator strength of the S<sub>1</sub> ( $f_{S_1}-s_0$ ) and the T<sub>1</sub> ( $f_{T_1}-s_0$ ), transition dipole moment both at S<sub>1</sub> ( $\mu_{T_1}-s_0$ ) and T<sub>1</sub> ( $\mu_{T_1}-s_0$ ), energy gap between the S<sub>1</sub> and T<sub>1</sub> ( $\mu_{T_1}-s_0$ ), spin orbit coupling constant ( $\xi_{T_1}-s_1$ ) between the S<sub>1</sub> and T<sub>1</sub>, and the ratio of  $\xi_{T_1}-s_1$  to  $\mu_{T_1}-s_0$  ( $\xi_{T_1}-s_1$ ).

Samples	$f_{S_1-S_2}$	$f_{T_1 - S_0}$	$\mu_{T_1 - S_0}$	$\mu_{S_1-S_0}$	$\Delta E_{T_1-S_1}$	$\xi_{T_1-S_1}$	$\frac{\xi_{T_1-S_{1/2}}}{\Lambda E_{T_1-S_{1/2}}}$
Samples	1 50	(10-7)	(Debye)	(Debye)	(eV)	(cm <sup>-1</sup> )	$\frac{2m_{T_1} - S_1}{(\text{cm}^{-1}/\text{eV})}$
PYCl	0.1051	0.96	0.0035	3.02	0.52	2.68	5.15
PYBr	0.0058	0.98	0.0036	0.78	0.10	1.62	16.2
PYCl/CB[6]	0.6402	6.72	0.0095	6.78	1.17	7.27	6.21
PYBr/CB[6]	0.7605	4.47	0.0080	7.04	1.51	2.29	1.51
PYCl/CB[7]	0.5049	1.14	0.0039	5.86	0.98	0.77	0.79
PYBr/CB[7]	0.7345	0.99	0.0036	6.97	1.32	0.30	0.23

**Table S7.** Calculated oscillator strength of the  $T_1$  state  $\begin{pmatrix} f_{T_1} - S_0 \end{pmatrix}$ , vertical excitation energy of  $T_1$ -S<sub>0</sub>, phosphorescence wavelength of  $T_1$  calculated at BMK/6-31G\*\* level of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] in crystal.

Compounds	$f_{T_1-S_0}$ (10-7)	$\frac{\Delta E_{T_1 - S_0}}{\text{(eV)}}$	Emission (Cal.) (nm)	Emission (Exp.) (nm)
PYC1	0.96	2.59	478	426
PYBr	0.98	2.44	509	470
PYCl/CB[6]	6.72	2.24	554	500
PYBr/CB[6]	4.47	2.26	549	500
PYCl/CB[7]	1.14	2.25	550	482
PYBr/CB[7]	0.99	2.30	539	/

**Table S8**. Calculated phosphorescence radiation rate ( $k_P$ ) of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] at BMK/6-31G\*\* level in crystal.

	PYCl	PYBr	PYCl/CB[6]	PYBr/CB[6]	PYCl/CB[7]	PYBr/CB[7]
$k_{\rm p}({\rm s}^{-1})$	2.80×10 <sup>1</sup>	2.53×10 <sup>1</sup>	$2.08 \times 10^{2}$	1.49×10 <sup>2</sup>	2.50×10 <sup>1</sup>	$2.28 \times 10^{1}$

**Table S9.** Calculated reorganization energies ( $\lambda$ ) of PYCl, PYBr, PYCl/CB[6], PYBr/CB[6], PYCl/CB[7] and PYBr/CB[7] in crystal at BMK/6-31G(d,p) <u>level by AP method.</u>

	$\lambda(meV)$
PYCl	712
PYBr	691
PYCl/CB[6]	629
PYBr/CB[6]	651
PYCl/CB[7]	932
PYBr/CB[7]	865

# III. The optimized geometries and the corresponding coordinates of the QM region of the PYCl, PYBr and their complexes at both $S_0$ and $T_1$ states.

**Structure S1.** The optimized geometry and the corresponding coordinate of the QM region of PYCl crystal at  $S_0$  state

-0.51394900	1.83157300	0.51570800
-8.80106400	-1.69135800	-0.55256700
-10.15028600	-2.26242800	-0.61452100
-10.16166700	-3.10556600	-1.30402300
-10.85068300	-1.50848200	-0.97796900
-10.41008900	-2.57309600	0.39914000
-7.72804400	-2.45489100	-0.82811900
-7.92757400	-3.44575200	-1.22010100
-6.44380100	-1.98511000	-0.62886300
-5.62654800	-2.66242400	-0.85032400
-6.23244800	-0.67419600	-0.14608000
-7.39217200	0.09938700	0.10034700
-7.34662000	1.12499100	0.44330400
-8.64779300	-0.43237700	-0.09811500
-9.55326700	0.13317100	0.10139400
-4.87776500	-0.12124600	0.06266100
-4.71474900	1.19194100	0.54691000
-5.56652200	1.79513800	0.83165900
-3.44836600	1.74471700	0.70632900
-3.35150100	2.74819900	1.11125000
-2.32051300	0.98859100	0.37231000
-2.45482900	-0.32952700	-0.07521800
-1.57718200	-0.91715100	-0.32888300
-3.72611900	-0.88117600	-0.22787000
-3.80042100	-1.90602200	-0.57456300
2.10299900	3.26829700	0.77721300
	$\begin{array}{r} -0.51394900\\ -8.80106400\\ -10.15028600\\ -10.16166700\\ -10.85068300\\ -10.41008900\\ -7.72804400\\ -7.92757400\\ -6.44380100\\ -5.62654800\\ -6.23244800\\ -7.39217200\\ -7.34662000\\ -8.64779300\\ -9.55326700\\ -4.87776500\\ -4.71474900\\ -5.56652200\\ -3.44836600\\ -3.35150100\\ -2.32051300\\ -2.45482900\\ -1.57718200\\ -3.72611900\\ -3.80042100\\ 2.10299900\\ \end{array}$	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$



**Structure S2.** The optimized geometry and the corresponding coordinate of the QM region of PYCl crystal at  $T_1$  state

Br	-0 53061100	1 84840200	0 48098200
N	-8.78272800	-1.69066500	-0.56062800
С	-10.12098000	-2.26565900	-0.60323400
H	-10.13854000	-3.11202700	-1.28948400
Н	-10.83248500	-1.51710900	-0.95932200
Н	-10.37923700	-2.57768700	0.41303700
С	-7.69314000	-2.46511300	-0.83899700
Н	-7.89992500	-3.45344000	-1.23114300
С	-6.42081000	-2.00146400	-0.64712800
Н	-5.60497100	-2.67804100	-0.87446800
С	-6.19472000	-0.65864700	-0.14927900
С	-7.39136400	0.12139200	0.10424300
Н	-7.34999600	1.14757000	0.44622100
С	-8.63062100	-0.41015800	-0.09874400
Н	-9.54214500	0.14693900	0.09052600
С	-4.90184500	-0.13454500	0.06496300
С	-4.71375300	1.22535600	0.58402600
Н	-5.56894800	1.80217700	0.90525400
С	-3.46810900	1.76521200	0.72339300
Н	-3.35998900	2.76086800	1.14617800
С	-2.30708900	1.00876200	0.36025100
С	-2.44861600	-0.34698100	-0.08328700
Н	-1.56034700	-0.92232500	-0.33023800
С	-3.68747600	-0.90927000	-0.21944200
Н	-3.76647000	-1.93942600	-0.54357300
Cl	2.08806800	3.26122000	0.77870000



Structure S3. The optimized geometry and the corresponding coordinate of the QM region of PYBr crystal at  $S_0$  state

Br	1.06563000	-0.48400800	0.01054200
Ν	-5.01001000	0.32720900	-6.70680100
С	-5.99738200	0.39529800	-7.79511100
Н	-6.32192800	-0.61953700	-8.04311100
Н	-6.85391200	0.98777100	-7.46774700
Н	-5.54056100	0.87413300	-8.66438500
С	-5.40860400	0.42748800	-5.42368100
Н	-6.46278500	0.61867800	-5.24430800
С	-4.49737800	0.30961300	-4.38993500
Н	-4.87355100	0.43563300	-3.38198900
С	-3.13155600	0.07202700	-4.65649300
С	-2.75941900	-0.03480700	-6.01709600
Н	-1.73638600	-0.23261200	-6.32102700
С	-3.71452900	0.09599900	-7.00494300
Н	-3.47541400	0.02639700	-8.05989800
С	-2.14048500	-0.05141800	-3.56428200
С	-2.56304700	-0.26866000	-2.23572600
Н	-3.61399600	-0.38119600	-1.98702000
С	-1.63467100	-0.39443300	-1.20189400
Н	-1.98770700	-0.58425500	-0.19179400
С	-0.26762100	-0.31697400	-1.48974500
С	0.17220400	-0.11338700	-2.80309100
Н	1.23485200	-0.06402300	-3.02686000
С	-0.75768800	0.02716000	-3.83497700
Н	-0.39536200	0.21219900	-4.84309400
Br	3.15359000	-0.67239900	2.50457100



**Structure S4.** The optimized geometry and the corresponding coordinate of the QM region of PYBr crystal at  $T_1$  state

Br	1.08126500	-0.48213700	0.02659500
Ν	-4.98407200	0.31358100	-6.68232900
С	-5.96613200	0.37580900	-7.76289000
Н	-6.28812200	-0.63916200	-8.02297500
Н	-6.83072300	0.95746900	-7.43445300
Н	-5.51628000	0.86131800	-8.63295600
С	-5.39562500	0.35959100	-5.37891400
Н	-6.45566500	0.51584300	-5.20747500
С	-4.50060600	0.22595500	-4.35264700
Н	-4.89644900	0.29504400	-3.34740400
С	-3.08363100	0.05742600	-4.61212800
С	-2.70823700	0.01969200	-6.01275100
Н	-1.68044000	-0.12148300	-6.32891800
С	-3.66074200	0.12488100	-6.98602500
Н	-3.42600400	0.07624000	-8.04184100
С	-2.13819600	-0.07038700	-3.56952500
С	-2.55492900	-0.22233900	-2.16819000
Н	-3.60568300	-0.29139900	-1.91281400
С	-1.63490500	-0.35009200	-1.16386800
Н	-1.97827700	-0.50695000	-0.14409500
С	-0.23073600	-0.32910500	-1.45580500
С	0.21177200	-0.19002700	-2.81269300
Н	1.27772300	-0.17713500	-3.02849800
С	-0.68980800	-0.06674900	-3.83566900
Н	-0.32914600	0.05334400	-4.85215800
Br	3.14658800	-0.66739700	2.49715700



**Structure S5.** The optimized geometry and the corresponding coordinate of the QM region of PYCl/CB[6] crystal at  $S_0$  state

Ν	-4.18580300	-0.96227200	0.90556300
С	-5.36663300	-1.08582200	1.72655400
С	-3.92885400	0.24027100	0.40499100
С	-2.61561200	0.59008500	0.20278300
С	-1.63763700	-0.37913500	0.19758100
С	-2.04113200	-1.70248000	0.33129700
С	-3.28117300	-1.94852900	0.84013100
С	-0.23892300	-0.04313100	0.10498400
С	0.67599900	-1.07964300	0.18231500
С	2.00219500	-0.80921100	0.02007200
С	2.52655300	0.43603700	-0.24109600
С	1.62148700	1.49573000	-0.09978800
С	0.25629600	1.26320800	0.03844500
Br	4.46365200	0.42701400	-0.91035400
Н	-5.20676500	-0.57221000	2.67918500
Н	-5.50100700	-2.14563100	1.89128300
Н	-6.20712300	-0.61548900	1.23014400
Н	-4.80343700	0.87131800	0.30421500
Н	-2.28424200	1.60073800	0.07877900
Н	-1.34438400	-2.48346100	0.14079500
Н	-3.60883300	-2.89029200	1.25179100
Н	0.40227800	-2.11003700	0.34157800
Н	2.63110500	-1.63053900	0.10035100
Н	1.92047600	2.51195500	-0.03062100
Н	-0.40652400	2.12249700	0.07970900



**Structure S6.** The optimized geometry and the corresponding coordinate of the QM region of PYCl/CB[6] crystal at T<sub>1</sub> state

N	-4.17375300	-0.97749000	0.90893900
С	-5.33152200	-1.09993300	1.75047400
С	-3.92178200	0.24350900	0.39604500
С	-2.62275700	0.60288400	0.20055100
С	-1.61228100	-0.36763400	0.19197100
С	-2.02371300	-1.71813000	0.33702200
С	-3.25663500	-1.97096500	0.82928800
С	-0.26539000	-0.04035400	0.13710800
С	0.69480800	-1.09872800	0.22000400
С	1.99172400	-0.82118200	0.04452500
С	2.53755900	0.45467100	-0.22524900
С	1.60208100	1.53610600	-0.05881300
С	0.26729600	1.31219200	0.10457100
Br	4.42433000	0.42819800	-0.91303600
Н	-5.18323400	-0.55050500	2.68788500
Н	-5.43889100	-2.15905300	1.94858500
Н	-6.19538300	-0.66794000	1.25592300
Н	-4.80450700	0.86178700	0.29372900
Н	-2.29345400	1.61357000	0.07245400
Н	-1.32962000	-2.49455900	0.12285000
Н	-3.59896300	-2.92142300	1.20437500
Н	0.41694300	-2.12241100	0.39944600
Н	2.63655400	-1.63102100	0.10813900
Н	1.91500200	2.54744400	0.03372100
Н	-0.39616900	2.16022800	0.23435400



**Structure S7.** The optimized geometry and the corresponding coordinate of the QM region of PYBr/CB[6] crystal at  $S_0$  state

N	3.88865300	1.75911300	0.69310000
С	4.94591800	2.36432100	1.50470200
С	2.92951500	2.56368800	0.23928500
С	1.65513500	2.05532200	0.06416800
С	1.42453100	0.69927700	0.12482400
С	2.54883000	-0.11422700	0.24899900
С	3.77067800	0.42732100	0.58799000
С	0.06738400	0.13570700	0.02649800
С	-1.13238700	0.86368900	-0.11408100
С	-2.36572400	0.20512700	-0.27396200
С	-2.41816300	-1.18454400	-0.38680300
С	-1.27165400	-1.84365100	-0.01444400
С	-0.05389900	-1.23983200	0.17059600
Н	5.37342300	3.24158800	1.01188900
Η	5.69358200	1.59068000	1.67424900
Η	4.51130100	2.67741100	2.45507400
Η	3.23815800	3.59412600	0.09943900
Н	0.79744100	2.68013000	-0.07583200
Η	2.43436900	-1.16882600	0.15538900
Η	4.66455600	-0.15041400	0.79547500
Η	-1.13642000	1.95082900	-0.08479400
Η	-3.28285100	0.74446600	-0.18001300
Н	-1.31683700	-2.85438800	0.21965100
Н	0.77691200	-1.87759500	0.43639900
Br	-4.13095900	-2.34410300	-0.87503400

**Structure S8.** The optimized geometry and the corresponding coordinate of the QM region of PYBr/CB[6] crystal at T<sub>1</sub> state

Ν	3.87539500	1.72736700	0.68087100
С	4.91246800	2.32416300	1.51147100
С	2.90449500	2.55216700	0.21706400
С	1.63622700	2.06809500	0.07138600
С	1.36785200	0.68451700	0.16026200
С	2.52967900	-0.14879700	0.25469600
С	3.75554200	0.38222200	0.54009100
С	0.07916300	0.15937700	0.11513100
С	-1.19635800	0.90733600	0.01802900
С	-2.38888000	0.25844900	-0.20887400
С	-2.43273200	-1.16051900	-0.42224400
С	-1.27262800	-1.84005200	0.01736200
С	-0.08379200	-1.27020400	0.27948400
Н	5.35587300	3.19817400	1.02287500
Н	5.65485700	1.55086700	1.70738600
Н	4.46606300	2.66222300	2.44978800
Н	3.22918800	3.57547300	0.06363400
Н	0.77951000	2.69087800	-0.08245100
Н	2.39849800	-1.20054400	0.14220000
Н	4.66162900	-0.19812700	0.67833200
Н	-1.20337900	1.98135000	0.18205100
Н	-3.31608400	0.77991000	-0.09697200
Н	-1.36034100	-2.85153300	0.23212700
Н	0.71968300	-1.89568800	0.63817000
Br	-4.12165700	-2.32846700	-0.88122700

**Structure S9.** The optimized geometry and the corresponding coordinate of the QM region of PYCl/CB[7],  $S_0$  state

Ν	-6.12123700	1.52645000	0.20749200
С	-5.27423700	1.27414300	-0.97942700
С	-5.67823500	2.33116500	1.19532300
С	-6.46501700	2.63061300	2.29572100
С	-7.77749200	2.11771600	2.38687100
С	-8.17866100	1.23711800	1.36260300
С	-7.33843300	0.95488400	0.30335600
С	-8.73047900	2.54411500	3.43501900
С	-10.08134500	2.71903700	3.08304800
С	-10.99634300	3.24098000	3.99790300
С	-10.53757600	3.57163000	5.27185000
С	-9.21769200	3.35417600	5.67270400
С	-8.30931300	2.83905900	4.74343100
Н	-4.71715400	2.18703600	-1.20444300
Н	-5.92064600	1.01141600	-1.81946200
Н	-4.57826900	0.45248800	-0.79342700
Н	-4.69003800	2.74431200	1.04351400
Н	-6.08052500	3.30438400	3.05606700
Н	-9.14146200	0.74755600	1.42175000
Н	-7.61170900	0.26711300	-0.48607400
Н	-10.43011100	2.48134800	2.08490800
Н	-12.02839900	3.38780700	3.69700600
Н	-8.88739800	3.57847800	6.68212800
Н	-7.29312000	2.62739000	5.05972700
Br	-11.83808300	4.41348600	6.57084000



**Structure S10.** The optimized geometry and the corresponding coordinate of the QM region of PYCl/CB[7],  $T_1$  state

N	-6.23385300	1.55240100	0.15580200
С	-5.34850400	1.27741500	-0.98188100
С	-5.70449900	1.74434300	1.40696600
С	-6.49830200	2.02178500	2.48717800
С	-7.93662900	2.14109400	2.34914900
С	-8.43476600	1.93346400	1.00478500
С	-7.58438700	1.65630100	-0.02881100
С	-8.79232600	2.49182300	3.41994600
С	-10.23540800	2.68766500	3.20801600
С	-11.01956500	3.27899600	4.15739900
С	-10.44130900	3.66321700	5.40486400
С	-9.08113800	3.35343300	5.72026000
С	-8.28681300	2.75815400	4.77688700
Н	-4.81687400	2.19746400	-1.25415600
Н	-5.94795300	0.91911500	-1.82091300
Н	-4.62344700	0.50139400	-0.72544300
Н	-4.62487200	1.67625800	1.46760900
Н	-6.01887700	2.15238200	3.45060700
Н	-9.49253800	1.99143900	0.78452100
Н	-7.92709800	1.51692000	-1.04481300
Н	-10.70407400	2.36855100	2.29041100
Н	-12.07386200	3.43795900	3.95569200
Н	-8.67712500	3.57444100	6.70378400
Н	-7.27645200	2.48229900	5.04867200
Br	-11.59445400	4.57198900	6.75111400



**Structure S11.** The optimized geometry and the corresponding coordinate of the QM region of PYBr/CB[7],  $S_0$  state

N	-6.06168000	1.48056800	0.27588500
С	-5.21892000	1.27696800	-0.92354800
С	-5.56928000	2.13398300	1.34710700
С	-6.36857300	2.45494200	2.43222100
С	-7.73798200	2.11351000	2.42834300
С	-8.18301800	1.33684100	1.33840400
С	-7.33442000	1.03866800	0.29052600
С	-8.68939800	2.61271000	3.44790900
С	-8.28298100	2.90889700	4.76232300
С	-9.18126800	3.49100900	5.66215400
С	-10.47756300	3.77487700	5.22861000
С	-10.92205500	3.45889700	3.94620300
С	-10.01775500	2.87284500	3.06002500
Н	-5.87541000	1.20402500	-1.79345900
Н	-4.62675600	0.36373300	-0.84050500
Н	-4.55267800	2.13731100	-1.02354500
Н	-4.52561300	2.40728500	1.27722900
Н	-5.93882700	3.01446800	3.25658700
Н	-9.19097500	0.94536000	1.32773300
Н	-7.64162900	0.44217700	-0.55808300
Н	-7.28422700	2.66169200	5.10958500
Н	-8.85967600	3.71618300	6.67429800
Н	-11.93475100	3.66874900	3.61795500
Н	-10.35776600	2.65450100	2.05381100
Br	-11.76388100	4.69977800	6.48556800



**Structure S12.** The optimized geometry and the corresponding coordinate of the QM region of PYBr/CB[7],  $T_1$  state

Ν	-5.94344300	1.62875300	0.33481400
С	-4.99118200	1.45073500	-0.76814400
С	-5.50119900	1.93212200	1.59364100
С	-6.36738800	2.17618900	2.62408200
С	-7.79995900	2.15686700	2.41075500
С	-8.20774700	1.73160800	1.08427300
С	-7.28539900	1.47598600	0.10862900
С	-8.72129900	2.58801000	3.39149800
С	-8.30926700	2.91740500	4.76837600
С	-9.17876700	3.52166300	5.63894800
С	-10.51487400	3.80478000	5.22161900
С	-10.99220100	3.41365000	3.93303200
С	-10.13401200	2.81173400	3.05554900
Н	-5.45549700	1.82691800	-1.68414100
Н	-4.73228700	0.39506900	-0.89926200
Н	-4.07994600	2.01149700	-0.54548900
Н	-4.42802600	1.96536200	1.70868000
Н	-5.95253700	2.40764300	3.59743500
Н	-9.25086300	1.57476500	0.84767200
Н	-7.54920500	1.14474600	-0.88723000
Н	-7.31702800	2.65911100	5.11936200
Н	-8.85072900	3.76890300	6.64444100
Н	-12.02111600	3.59371900	3.63757900
Н	-10.50777600	2.53917300	2.08025200
Br	-11.76305800	4.74756500	6.46255400

