

Electronic Supplementary Information

Direct reduction of NO into N₂ catalyzed by fullerene-supported rhodium clusters

Ruomeng Li, Ya-Ke Li*, Jianzhi Xu, Gao-Lei Hou*

MOE Key Laboratory for Non-Equilibrium Synthesis and Modulation of Condensed Matter, School of Physics, Xi'an Jiaotong University, Xi'an, 710049 Shaanxi, China

E-mail: yake.li@xjtu.edu.cn; gaolei.hou@xjtu.edu.cn

Table of Content

Figure S1. Calculated isomers of C ₆₀ Rh ₄ ⁺	S2
Figure S2. Calculated isomers of Rh ₄ ⁺	S2
Figure S3. Calculated isomers of C ₆₀ Rh ⁺	S2
Figure S4. Comparison between theoretical and experimental infrared spectra of C ₆₀ Rh ₄ ⁺ and IRMPD spectra.....	S3
Figure S5. Reaction pathway of NO on C ₆₀ Rh ⁺	S4
Figure S6. Reaction pathway of NO on square Rh ₄ ⁺	S5
Figure S7. Calculated NBO partial charges (<i>e</i>) for the reaction of pyramid Rh ₄ ⁺ and NO.....	S6
Figure S8. Calculated NBO partial charges (<i>e</i>) for the reaction of square Rh ₄ ⁺ and NO.....	S7
Figure S9. Calculated NBO partial charges (<i>e</i>) for the reaction of C ₆₀ Rh ⁺ and NO.....	S8
Cartesian coordinates of C ₆₀ Rh ₄ ⁺ , C ₆₀ Rh ⁺ , and Rh ₄ ⁺	S9
References.....	S11

Confirmation of the ground state structures of Rh_4^+ , $\text{C}_{60}\text{Rh}_4^+$, and C_{60}Rh^+

Figures S1, S2, and S3 respectively show the energy differences between the isomers of Rh_4^+ , $\text{C}_{60}\text{Rh}_4^+$, and C_{60}Rh^+ . Calculations show that for Rh_4^+ the pyramidal structure is more stable than the square structure, with the doublet states being slightly more stable than the quartet states. For $\text{C}_{60}\text{Rh}_4^+$, the structure with a square Rh_4^+ moiety possessing a quartet state is most stable and its simulated infrared spectrum gives the largest cosine similarity score (see Figure S4). For C_{60}Rh^+ , the η^{6-6} structure with a triplet state is the most stable one, consistent with previous results.¹

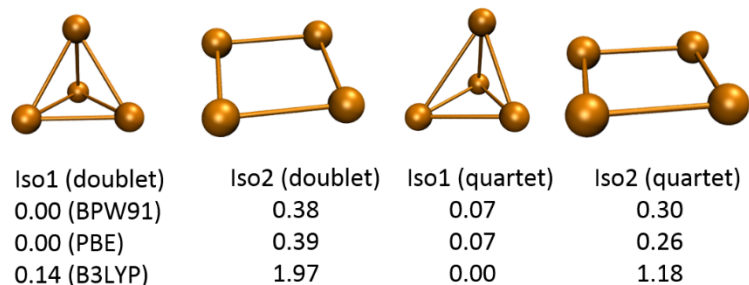


Figure S1. Energy differences (in eV) between different isomers of Rh_4^+ at the BPW91, PBE, and B3LYP levels.

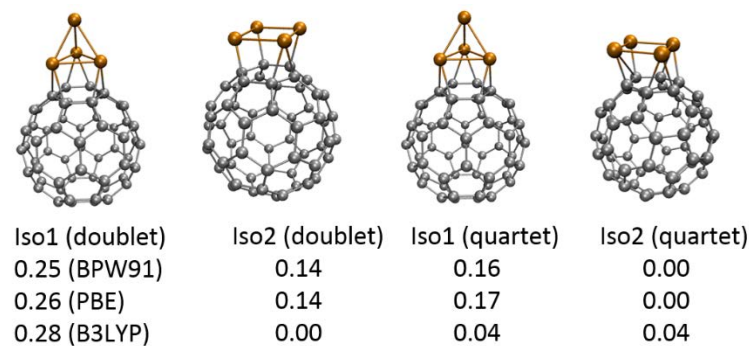


Figure S2. Energy differences (in eV) between different isomers of $\text{C}_{60}\text{Rh}_4^+$ at the BPW91, PBE, and B3LYP levels.

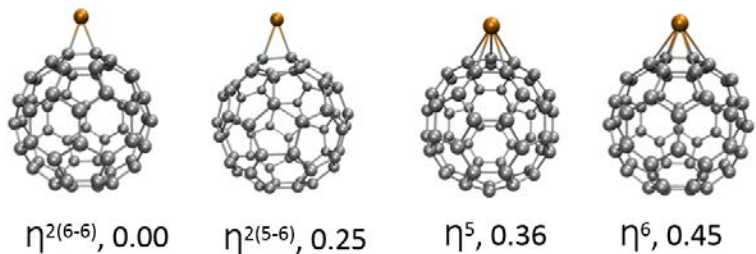


Figure S3. Energy differences (in eV) between different isomers of C_{60}Rh^+ at the BPW91 level.

Cosine Similarity Score

Cosine similarity score was employed to quantitatively evaluate the agreement between calculated and experimental infrared spectra.²⁻⁵ Figure S4 displays the calculated cosine similarity scores of the theoretical spectra and experimental one for $C_{60}Rh_4^+$. It can be seen that the cosine similarity score for the Iso2 quartet is largest, suggesting Iso2 $C_{60}Rh_4^+$ with a quartet state the most probable structure exist in the experiment.

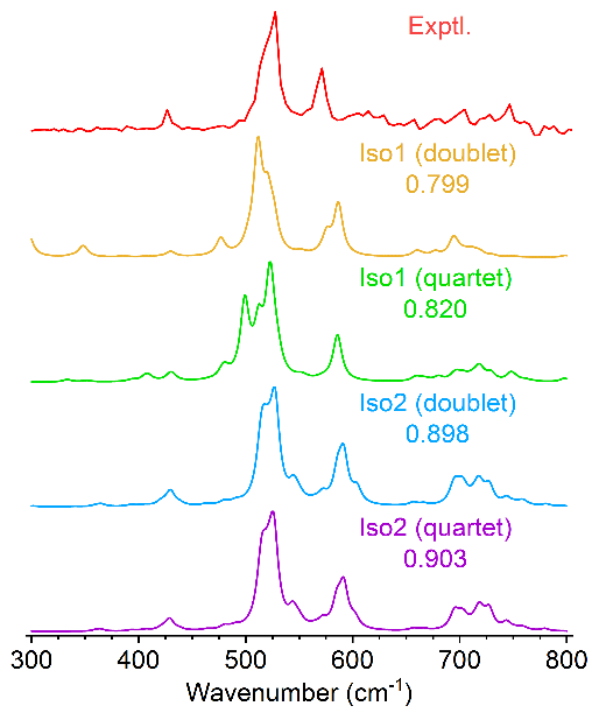


Figure S4. The theoretical infrared spectra for the four isomers of $C_{60}Rh_4^+$ and their comparison with experimental spectrum. Cosine similarity scores are provided.

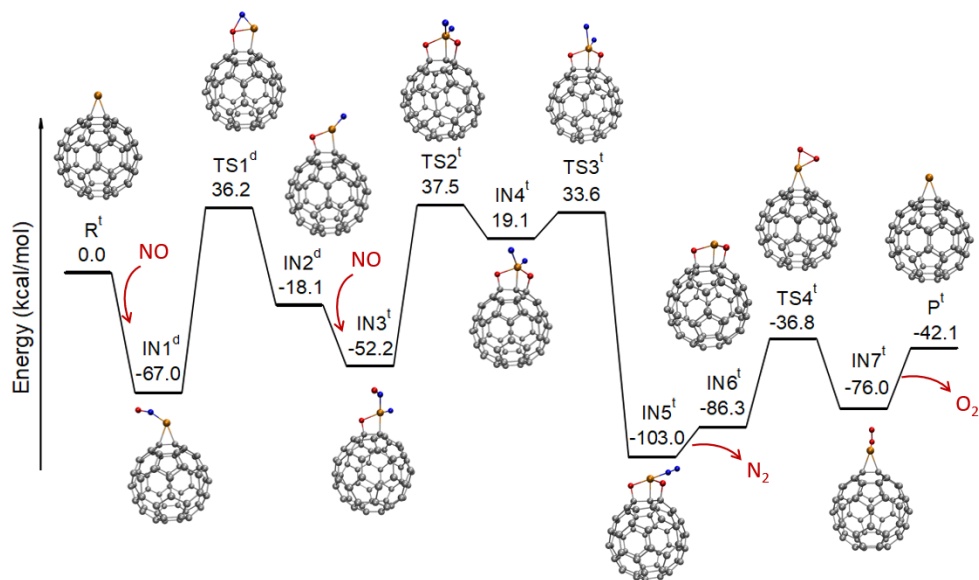


Figure S5. Reaction pathway of NO reduction on C₆₀Rh⁺ calculated at the BPW91/6-31G(d)&SDD level of theory. R represents for reactants, IN for intermediates, TS for transition states, and P for products. All energies are provided relative to the energies of the reactants, i.e., C₆₀Rh⁺ and two NO molecules. Calculations show that N–O dissociations post high energy barriers of 103.2 (TS1) and 89.7 (TS2) kcal/mol, respectively.

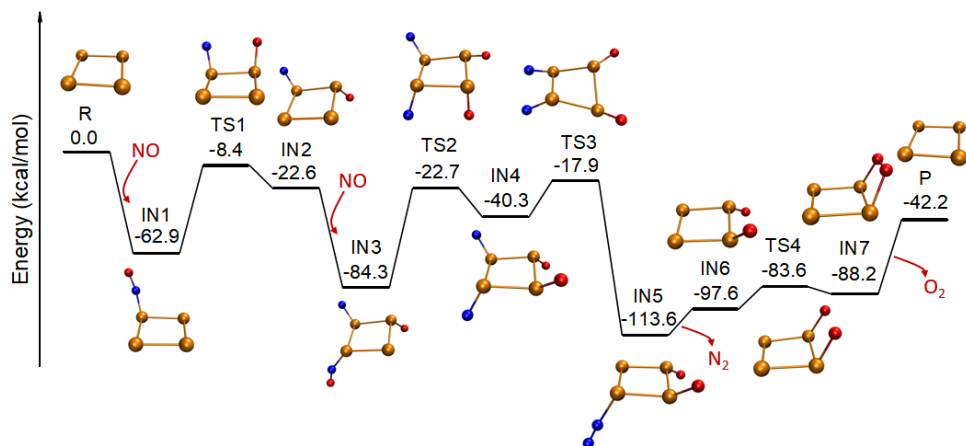


Figure S6. Reaction pathway of NO reduction on square Rh₄⁺ calculated at the BPW91/6-31G(d)&SDD level of theory. R represents for reactants, IN for intermediates, TS for transition states, and P for products. All energies are provided relative to the energies of the reactants, i.e., Rh₄⁺ and two NO molecules.

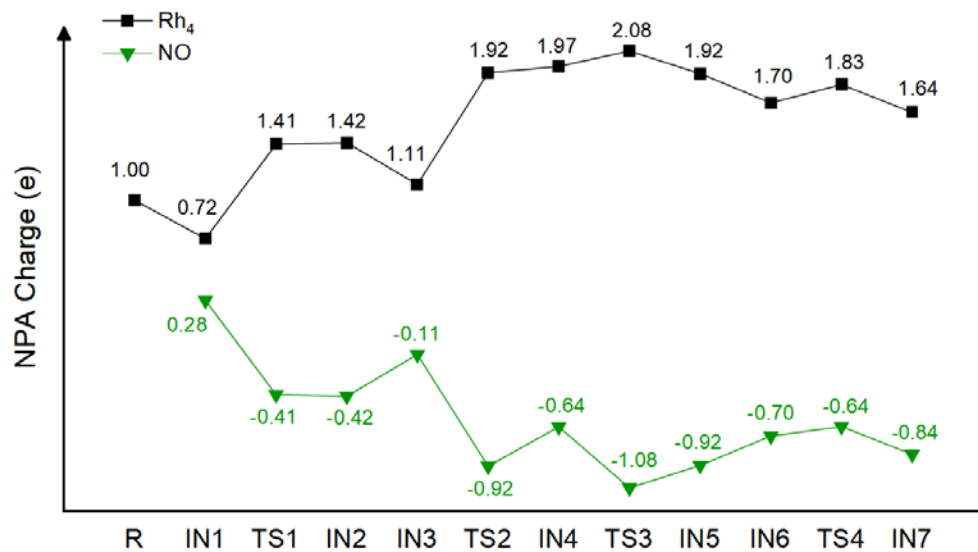


Figure S7. Calculated NBO partial charges (e) of pyramid Rh₄⁺ cluster (black square) of NO reduction catalyzed by Rh₄⁺

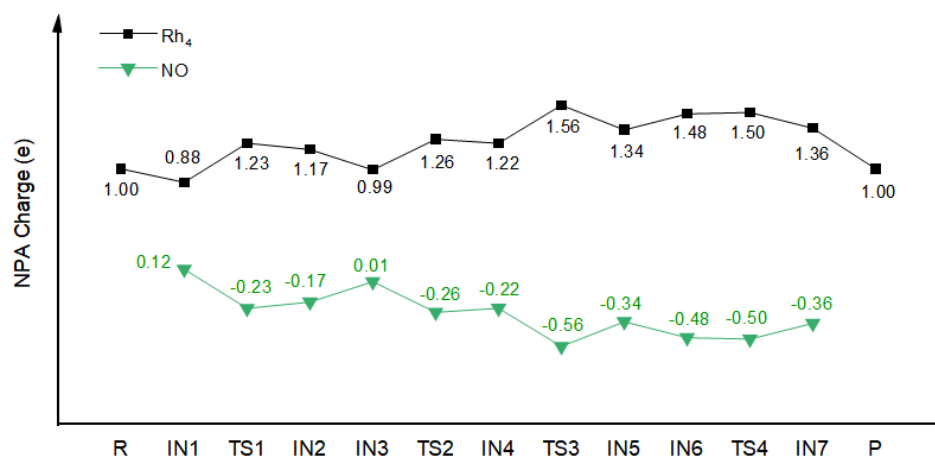


Figure S8. Calculated NBO partial charges (e) of square Rh_4^+ cluster (black square) of NO reduction catalyzed by Rh_4^+

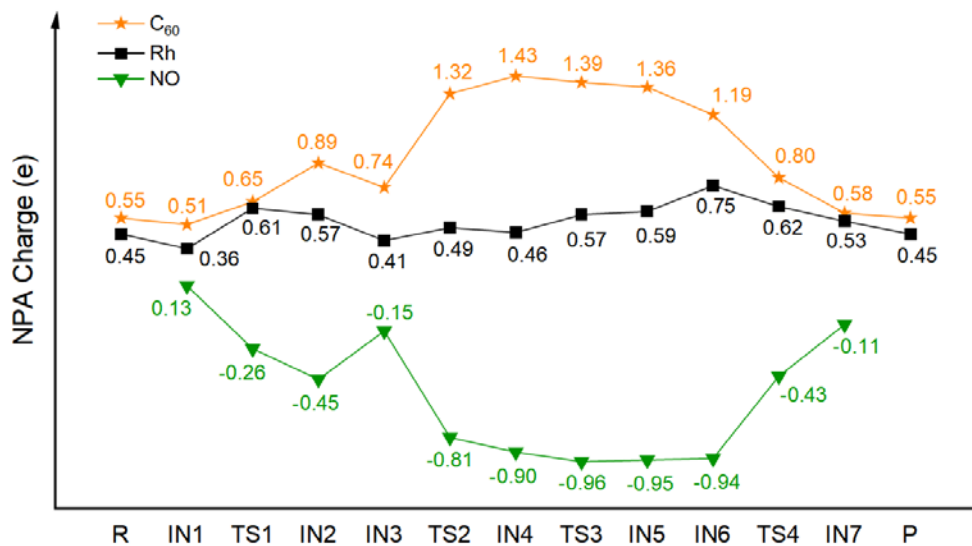


Figure S9. Calculated NBO partial charges (e) of Rh⁺ (black square) and C₆₀ (orange star) of NO reduction catalyzed by C₆₀Rh⁺

Cartesian coordinates of the most stable C₆₀Rh₄⁺, C₆₀Rh⁺, and Rh₄⁺ at the BPW91/6-31G(d)&SDD level of theory

C ₆₀ Rh ₄ ⁺				C ₆₀ Rh ⁺			
C	8.5931280	6.7238380	7.5364780	C	2.1089020	-2.2498480	-1.7938820
C	9.4582700	6.3148530	6.4189110	C	1.2936900	-1.7040880	-2.8645880
C	7.6709300	5.7625170	8.1271720	C	1.5306790	-3.1045930	-0.8334570
C	9.4215980	4.9973880	5.9426630	C	-0.0736710	-2.0290860	-2.9412830
C	9.5093960	4.7352090	4.5172900	C	-1.0468060	-1.0016560	-3.2677620
C	9.5499350	7.4220430	5.4823380	C	1.7343240	-0.3355590	-3.1147160
C	9.6425580	7.1739790	4.1007210	C	0.7935160	0.6573180	-3.4285360
C	9.6196220	5.8052170	3.6088130	C	-0.6222170	0.3191680	-3.5029720
C	5.7909210	7.5697550	8.2658840	C	2.8005500	-1.9815320	0.9633160
C	6.7707700	8.5068530	7.5719080	C	3.3902040	-1.0904080	-0.0216480
C	6.2584510	6.1412450	8.5552880	C	1.8781760	-2.9609770	0.5663220
C	8.1379890	8.1144760	7.2558410	C	3.0568370	-1.2249490	-1.3801770
C	8.7359940	8.5205190	5.9741390	C	2.8252750	-0.0395980	-2.1958290
C	6.0179650	9.2887850	6.5752740	C	3.5031190	0.2363400	0.5714810
C	6.6254480	9.6984810	5.3796710	C	3.2784060	1.3783010	-0.2151970
C	8.0074400	9.3094750	5.0732810	C	2.9346670	1.2382860	-1.6242060
C	4.1458960	5.2608700	7.4880580	C	0.4109740	-2.4768940	2.5271650
C	3.6945830	6.6399060	7.2092580	C	1.3854870	-1.4768630	3.0151700
C	5.3987670	5.0272240	8.0625690	C	0.6658770	-3.2127410	1.3660390
C	4.5127330	7.7309570	7.5163260	C	2.6021020	-1.2240370	2.2186540
C	4.6460890	8.8174830	6.5635910	C	2.9850160	0.1752160	1.9279470
C	2.9462430	6.6170390	5.9681970	C	0.5955320	-0.3205580	3.4933450
C	3.0396650	7.6903380	5.0592540	C	1.0251710	0.9914180	3.2421870
C	3.9066050	8.8129520	5.3638820	C	2.2369800	1.2424090	2.4473640
C	5.7733260	3.1234480	6.5109520	C	-1.7180280	-3.0432630	0.7126580
C	4.4620310	3.3530230	5.9335860	C	-1.9792860	-2.2689850	1.9175710
C	6.2384240	3.9631920	7.5434330	C	-0.4151820	-3.5080950	0.4412940
C	3.6684530	4.4150260	6.4118360	C	-0.9277800	-1.9852910	2.8049620
C	2.9255780	5.2465090	5.4736790	C	-0.8141890	-0.6587800	3.3987490
C	4.5480680	3.0961350	4.5044910	C	-2.9563950	-1.2384430	1.5907750
C	3.8332220	3.9015700	3.6014490	C	-2.8470230	0.0394270	2.1624040
C	3.0034160	4.9943910	4.0963770	C	-1.7558090	0.3344050	3.0823560
C	8.4864770	4.0261300	6.5232510	C	-0.6721620	-2.9159070	-1.9496360
C	8.0109330	3.1702840	5.4512760	C	-2.0185940	-2.4317050	-1.6670950
C	7.6209670	4.4024890	7.5606810	C	0.1143590	-3.4420150	-0.9156060
C	6.6760010	2.7249400	5.4439950	C	-2.5321840	-2.4970520	-0.3584160
C	5.9206730	2.7046210	4.2004330	C	-3.2990930	-1.3792340	0.1823070
C	8.6395440	3.6071970	4.2105250	C	-2.2492840	-1.2506720	-2.4792130
C	7.9098930	3.5903750	3.0128560	C	-2.9837820	-0.1705900	-1.9560370
C	6.5250560	3.1325130	3.0070840	C	-3.5218610	-0.2373010	-0.6026550
C	4.0139730	5.7026320	2.0915070	C	-2.1267230	2.2499100	1.7662820
C	4.4592200	4.3422500	2.3648160	C	-3.0775030	1.2247200	1.3476100
C	5.7832380	3.9646580	2.0719980	C	-3.4092500	1.0887880	-0.0098450
C	4.9071860	6.6367740	1.5347490	C	-1.5450930	3.1036880	0.8102490
C	4.0527750	8.3973480	3.0533370	C	0.6581630	2.9072740	1.9255780
C	3.1271430	7.4284570	3.6314430	C	0.0579010	2.0205330	2.9153830
C	3.1114160	6.1055890	3.1577710	C	-1.3123050	1.6988090	2.8348790
C	4.9263140	8.0062730	2.0253270	C	-0.1278080	3.4392770	0.8941820
C	6.7767260	9.2855940	3.0655990	C	1.6999560	3.0433300	-0.7400670
C	5.8733570	9.6850240	4.1380050	C	2.5138970	2.4918670	0.3284670
C	4.5345850	9.2510000	4.1286330	C	2.0035030	2.4233300	1.6406410
C	6.3123030	8.4607000	2.0303800	C	0.4000330	3.5070000	-0.4639500
C	8.4171530	7.1494480	2.0918030	C	-0.4387310	2.4609330	-2.5439500
C	8.8985090	8.0038510	3.1659640	C	0.9072190	1.9833030	-2.8354140
C	8.0932470	9.0539310	3.6464920	C	1.9579580	2.2686390	-1.9496900
C	7.1462980	7.3712500	1.5357840	C	-0.6864930	3.2133110	-1.3812530
C	6.7064930	4.9323730	1.4959050	C	-2.8013840	1.9715480	-0.9982620
C	8.0217830	4.7009170	2.0744670	C	-2.5431610	1.1944870	-2.2015440
C	8.8631890	5.7870760	2.3669930	C	-1.3829340	1.4351390	-2.9610060
C	6.2764040	6.2448440	1.2302910	C	-1.8895240	2.9635890	-0.5933490
Rh	5.8390550	8.4378690	10.1048570	Rh	3.0854410	-2.0816780	4.0530090
Rh	8.8743760	6.6334690	9.6459080				
Rh	8.0705250	9.0842430	9.1508800				
Rh	6.5818390	6.1751120	10.5638460				

Rh_4^+				
Rh	0.2128110	1.2404850	5.4326590	
Rh	2.4844410	0.5876130	4.9540880	
Rh	1.7865360	2.9451640	4.4989710	
Rh	2.0651730	2.2369160	6.7864860	

References

- (1) German, E.; Hou, G.-L.; Vanbuel, J.; Bakker, J. M.; Alonso, J. A.; Janssens, E.; López, M. J. Infrared spectra and structures of $C_{60}Rh_n^+$ complexes. *Carbon* **2022**, *197*, 535-543.
- (2) Fu, W.; Hopkins, W. S. Applying machine learning to vibrational spectroscopy. *J. Phys. Chem. A* **2018**, *122*, 167-171.
- (3) Kempkes, L. J.; Martens, J.; Berden, G.; Houthuijs, K. J.; Oomens, J. Investigation of the position of the radical in z_3 -ions resulting from electron transfer dissociation using infrared ion spectroscopy. *Faraday Discuss.* **2019**, *217*, 434-452.
- (4) Müller, F.; Stückrath, J. B.; Bischoff, F. A.; Gagliardi, L.; Sauer, J.; Debnath, S.; Jorewitz, M.; Asmis, K. R. Valence and Structure Isomerism of $Al_2FeO_4^+$: Synergy of Spectroscopy and Quantum Chemistry. *J. Am. Chem. Soc.* **2020**, *142*, 18050-18059.
- (5) Xu, J.; Bakker, J. M.; Lushchikova, O. V.; Lievens, P.; Janssens, E.; Hou, G.-L. Pentagon, Hexagon, or Bridge? Identifying the Location of a Single Vanadium Cation on Buckminsterfullerene Surface. *J. Am. Chem. Soc.* **2023**, *145*, 22243-22251.