How non-aqueous media direct the reaction of Ca(OH)₂ with CO₂ to different forms of CaCO₃: operando midinfrared and X-ray absorption spectroscopy studies

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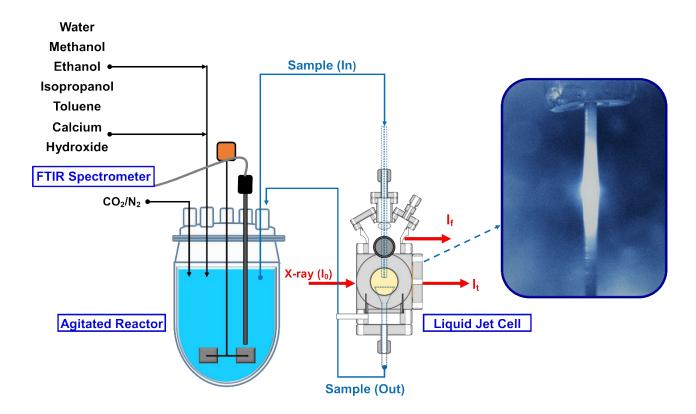


Figure S1. Simplified schematic of the continuous-flow liquid-jet PAT experimental setup for simultaneous *operando* mid-IR and XAS measurements.^{1, 2}

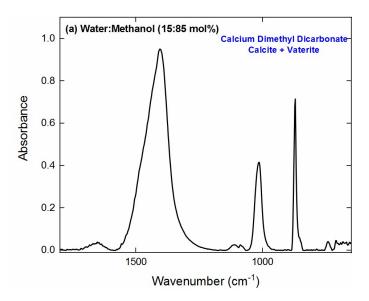


Figure S2. Mid-IR of the aged post-carbonation product for the water-methanol system.

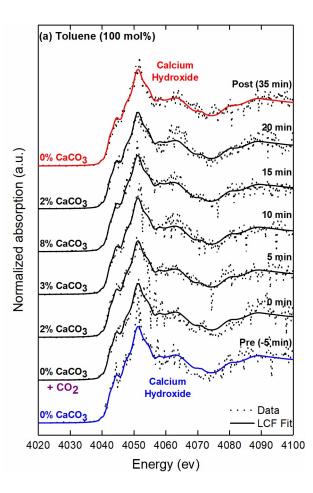


Figure S3. Time-resolved Ca K-edge XANES of the 20-minute carbonation of $Ca(OH)_2$ in pure toluene – no conversion observed.

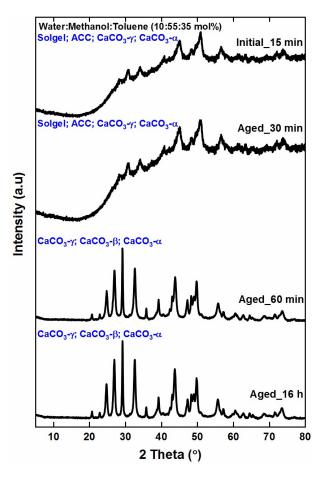


Figure S4. Time-resolved XRD following the aging of the post-carbonation product (15 mins to 16 hours) precipitated in a ternary water-methanol-toluene (10:55:35 mol%) system.

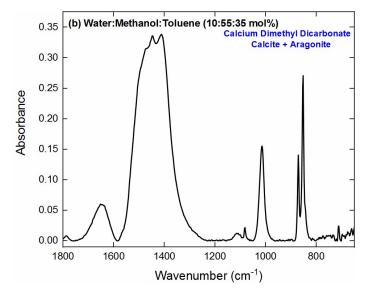


Figure S5. Mid-IR spectrum of the aged (60 hours) post-carbonation products for the watermethanol-toluene system. shows a mixture of calcite and aragonite.

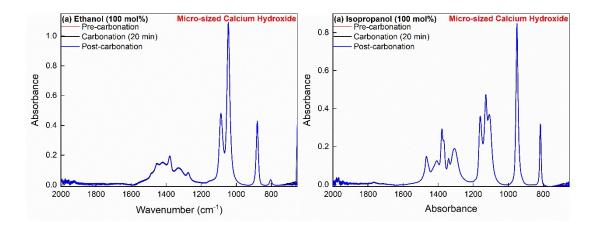


Figure S6. Mid-IR of micro-sized $Ca(OH)_2$ carbonation in pure (a) ethanol (product – no conversion) and (b) isopropanol (product – no conversion).

REFERENCES

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