

## Supporting Information

### **Immobilization of a chiral rhodium catalyst on carbon nanotubes via non-covalent interactions for heterogeneous asymmetric hydrogenation**

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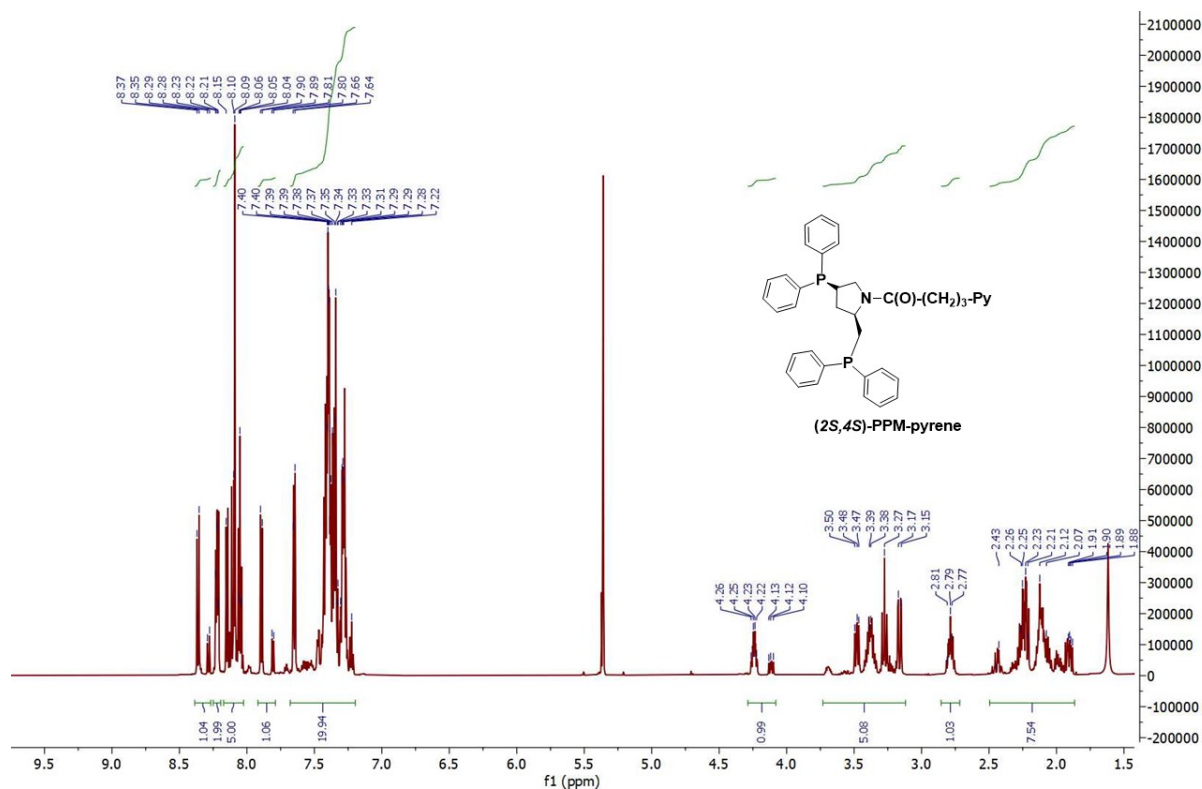
Chemical structure of (2S,4S)-PPM-pyrene is shown. The structure consists of a pyrene moiety attached to a 1,3-bis(phenylthio)propane backbone. The pyrene moiety is attached to the 1,3-bis(phenylthio)propane backbone via a pyridine ring. The pyrene moiety is attached to the 1,3-bis(phenylthio)propane backbone via a pyridine ring. The pyrene moiety is attached to the 1,3-bis(phenylthio)propane backbone via a pyridine ring.

$^1\text{H}$  NMR spectrum of (2S,4S)-PPM-pyrene in  $\text{CDCl}_3$ . The spectrum shows peaks for the pyrene moiety (7.05, 9.58 ppm), the PPM backbone (22.93, 23.41 ppm), and the pyridine ring (7.2-8.5 ppm). The chemical structure of (2S,4S)-PPM-pyrene is shown, featuring a pyrene group attached to a 1,3-bis(phenylthio)propane backbone. The x-axis is labeled f1 (ppm) and ranges from 5 to -30. The y-axis is labeled intensity. The peaks are assigned to the major conformer (blue stars) and the minor conformer (red circles).

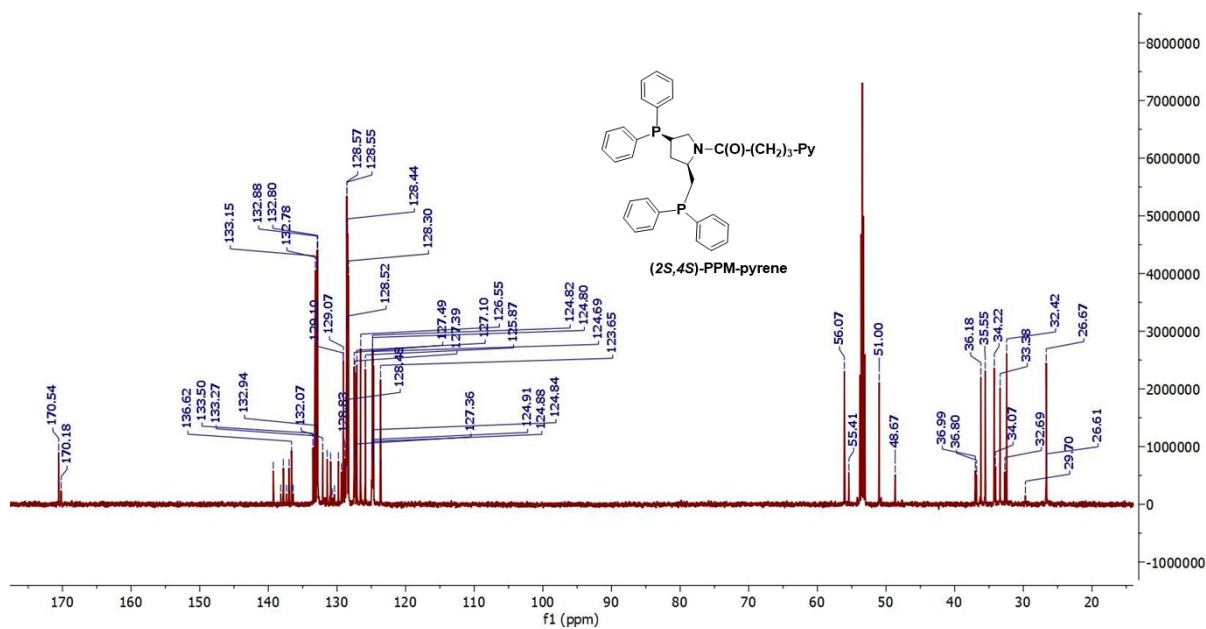
Major conformer

Minor conformer

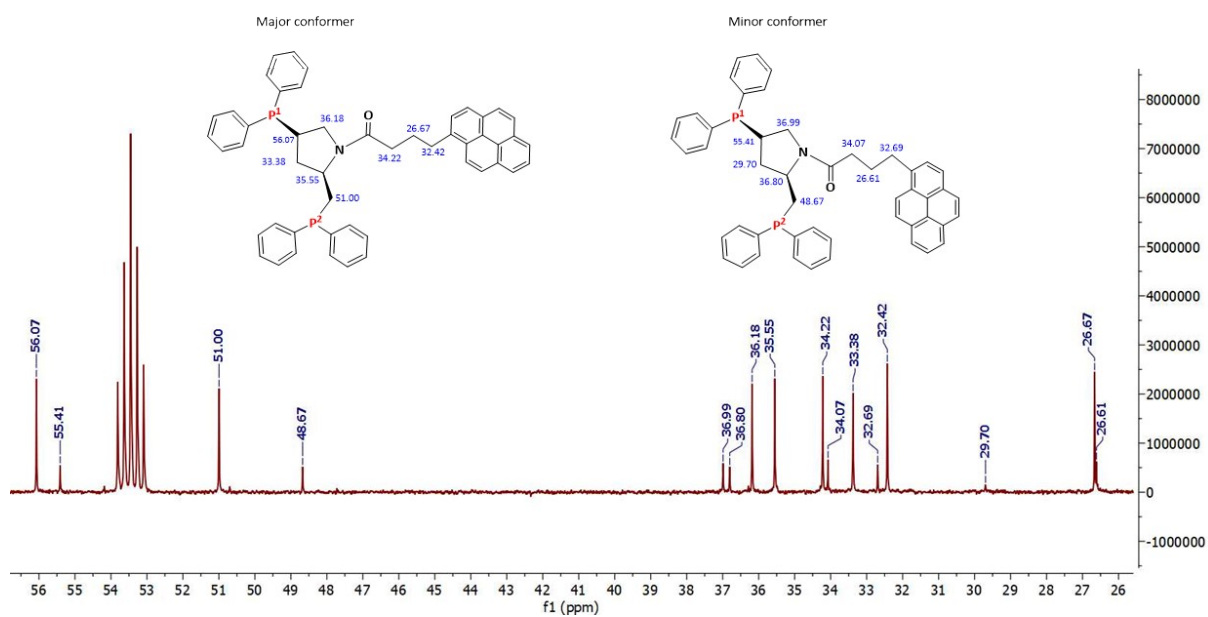
**Figure S2:**  $^1\text{H}$ ,  $^{31}\text{P}$ , and  $^{13}\text{C}$ -chemical shifts of the (2*S*,4*S*)-PPM-pyrene ligand. Attribution made by  $^1\text{H}$ ,  $^1\text{H}\{^{31}\text{P}\}$  select and broadband,  $^{13}\text{C}\{^1\text{H}\}$ ,  $^{13}\text{C}\{^1\text{H}\}$ ,  $\{^{31}\text{P}\}$  COSY, HSQC, HMBC ROESY, HMQC  $^1\text{H}$ - $^{31}\text{P}$  on a Bruker Avance NEO 600. The majority of the protons and carbons of each conformer, including those of the pyrene group were assigned except the protons and carbons of the phenyl groups which are too overlapping to be unambiguously assigned.



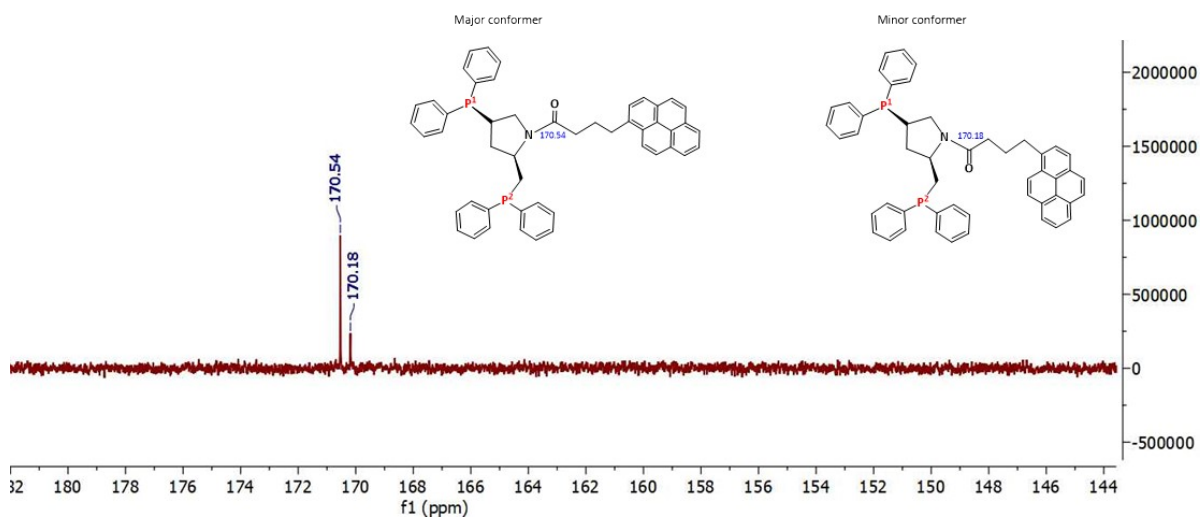
**Figure S3:** <sup>1</sup>H NMR spectrum of the (2*S*,4*S*)-PPM-pyrene ligand.



**Figure S4:** <sup>13</sup>C NMR spectrum of the (2*S*,4*S*)-PPM-pyrene ligand.

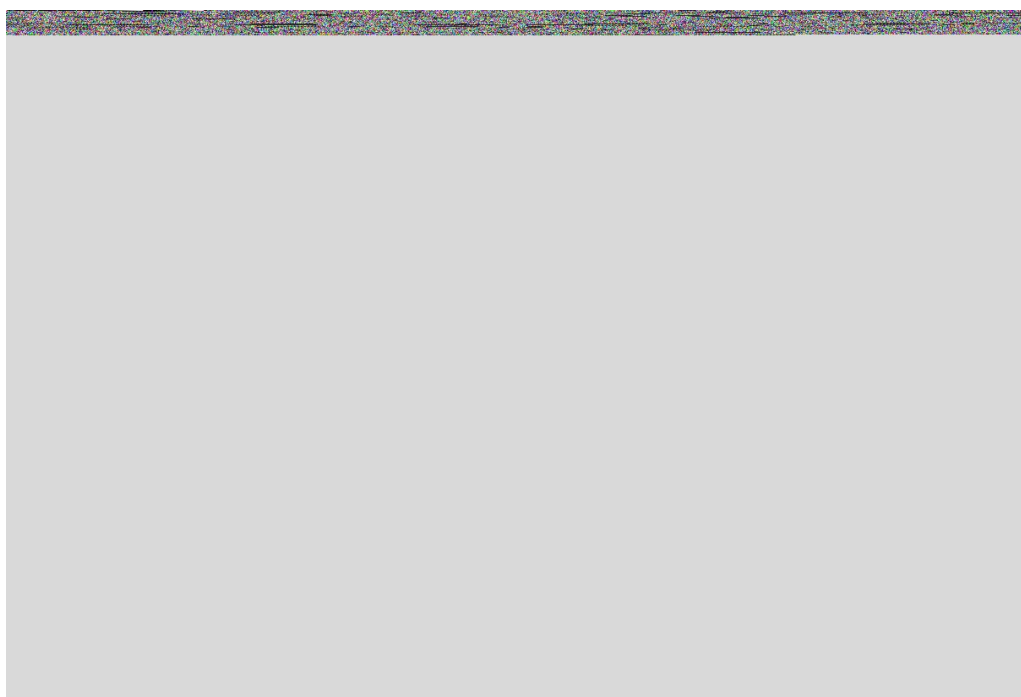


**Figure S5:** Zoom spectrum of aliphatic carbons of the  $(2S,4S)$ -PPM-pyrene ligand.

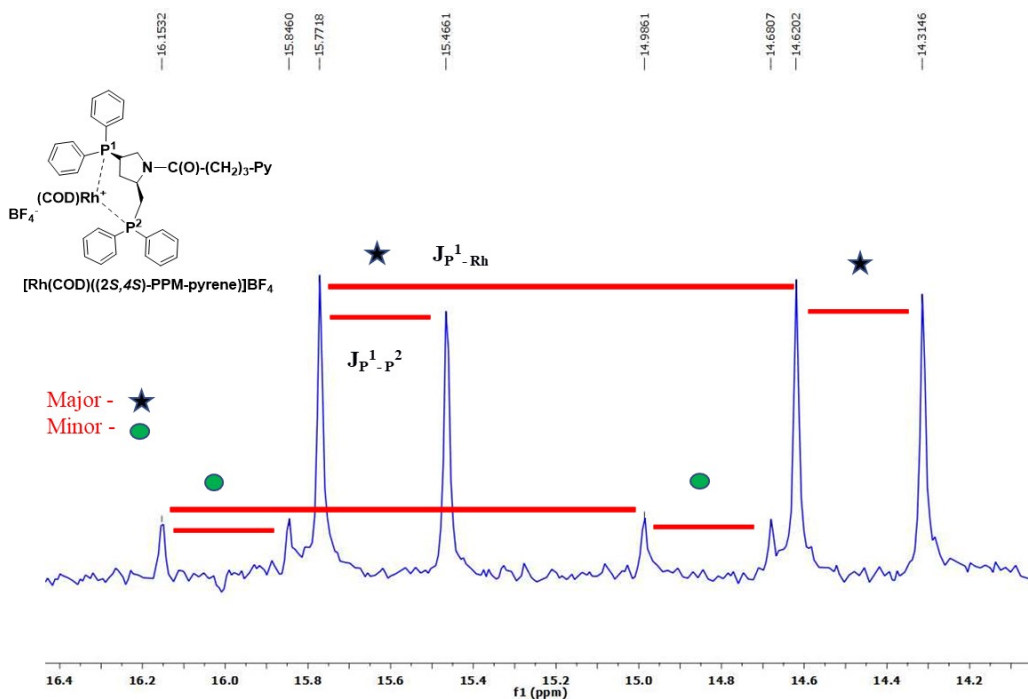


**Figure S6:** Zoom spectrum of carbonyl carbons of the  $(2S,4S)$ -PPM-pyrene ligand.

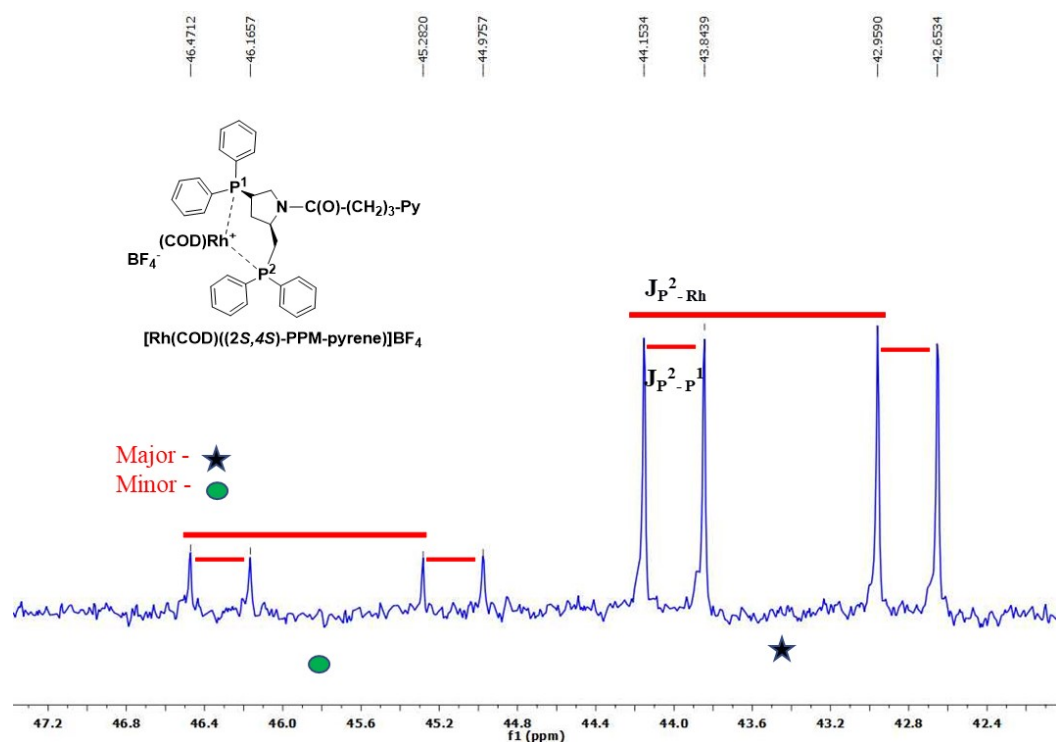
2. Comparison of  $^{31}\text{P}\{^1\text{H}\}$  NMR data of (2*S*,4*S*)-PPM-pyrene ligand and  $[\text{Rh}(\text{COD})((2*S*,4*S*)\text{-PPM-pyrene})]\text{BF}_4$  complex.



**Figure S7:**  $^{31}\text{P}\{^1\text{H}\}$  NMR spectra of (2*S*,4*S*)-PPM-pyrene ligand (bottom, red) and  $[\text{Rh}(\text{COD})((2*S*,4*S*)\text{-PPM-pyrene})]\text{BF}_4$  complex (top, blue).

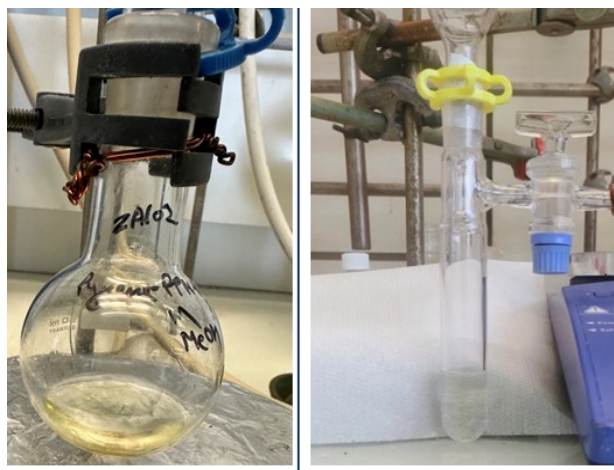


**Figure S8:** Zoom of  $^{31}\text{P}\{^1\text{H}\}$  spectrum of the  $[\text{Rh}(\text{COD})((2*S*,4*S*)\text{-PPM-pyrene})]\text{BF}_4$  complex showing the couplings of  $\text{P}^1$  with Rh and  $\text{P}^2$  in the major and minor isomers.



**Figure S9:** Zoom of  $^{31}\text{P}\{^1\text{H}\}$  spectrum of the  $[\text{Rh}(\text{COD})((2S,4S)\text{-PPM-pyrene})]\text{BF}_4$  complex showing the couplings  $\text{P}^2$  with Rh and  $\text{P}^1$  in the major and minor isomers.

### 3. Immobilization of $[\text{Rh}(\text{COD})((2S,4S)\text{-PPM-pyrene})]\text{BF}_4$ via non-covalent $\pi\text{-}\pi$ interaction



**Figure S10:** The adsorption of chiral **complex Rh** onto CNTs in MeOH: homogeneous complex solution (left), and the filtrate after immobilization of complex on CNTs (right).

#### 4. FTIR spectra of CNTs and CNT@complex Rh.

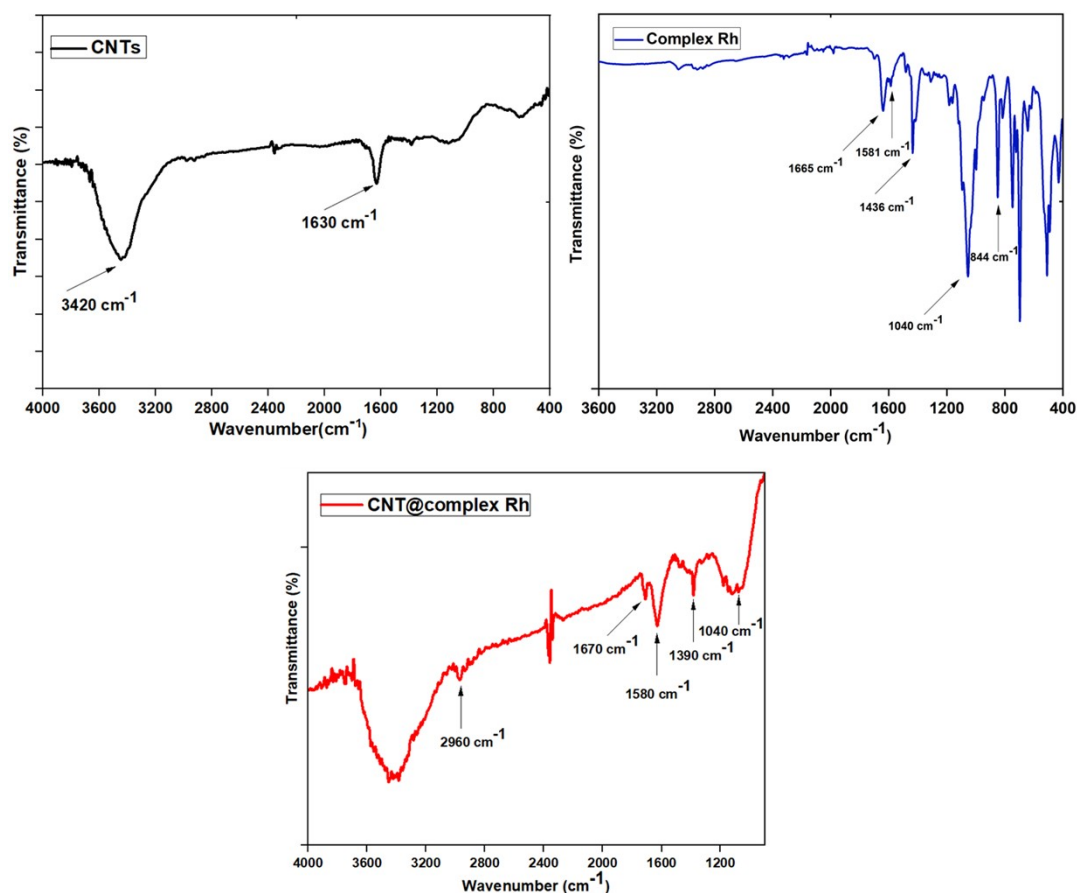


Figure S11: FTIR spectra of the CNTs and CNT@complex Rh.

#### 5. Raman spectra of the CNTs and CNT@complex Rh.

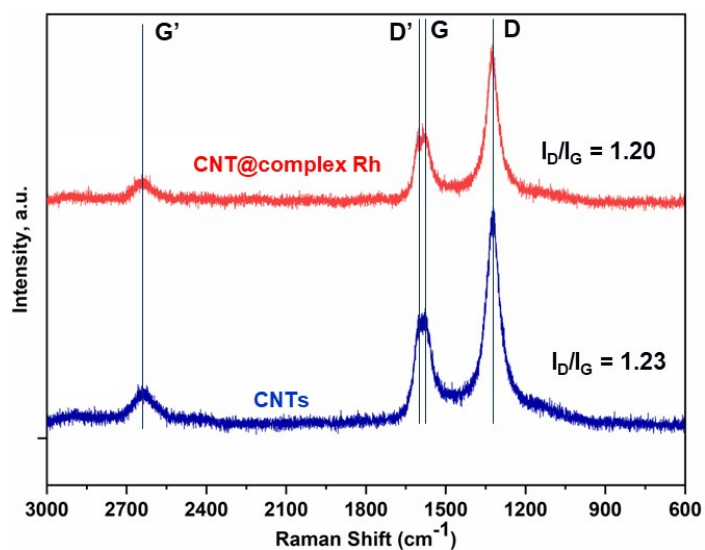


Figure S12: Raman spectra of the CNTs (blue) and CNT@complex Rh (red).

6.  $N_2$  adsorption-desorption isotherms of CNTs and CNT@complex Rh

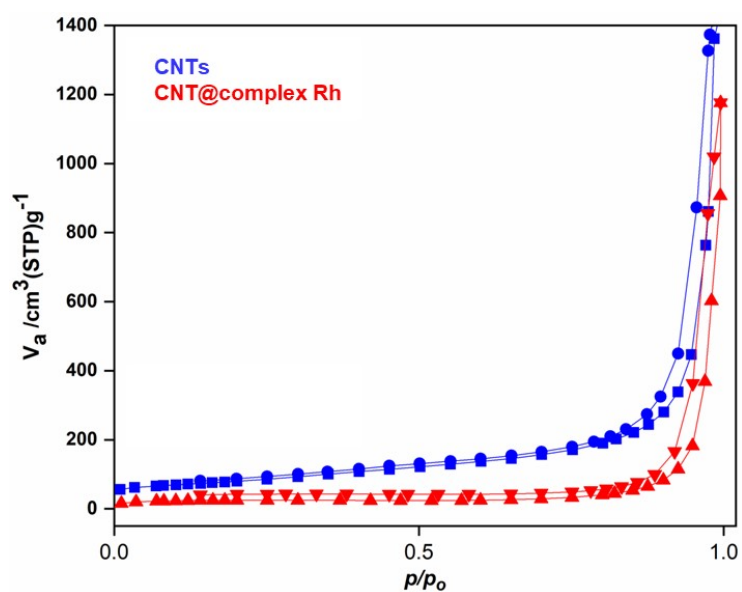


Figure S13:  $N_2$  adsorption-desorption isotherms of CNTs (blue) and CNT@complex Rh (red).

7. XPS data of the fresh CNT@complex Rh

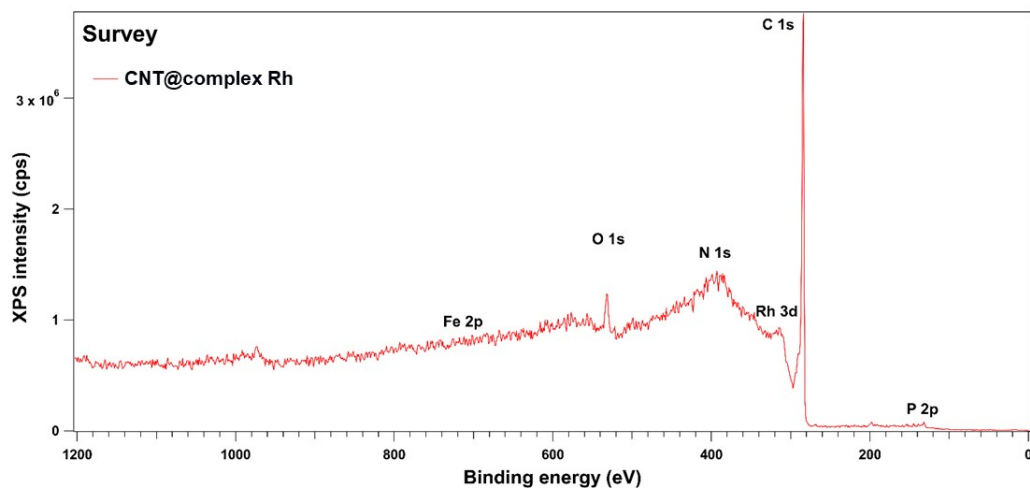
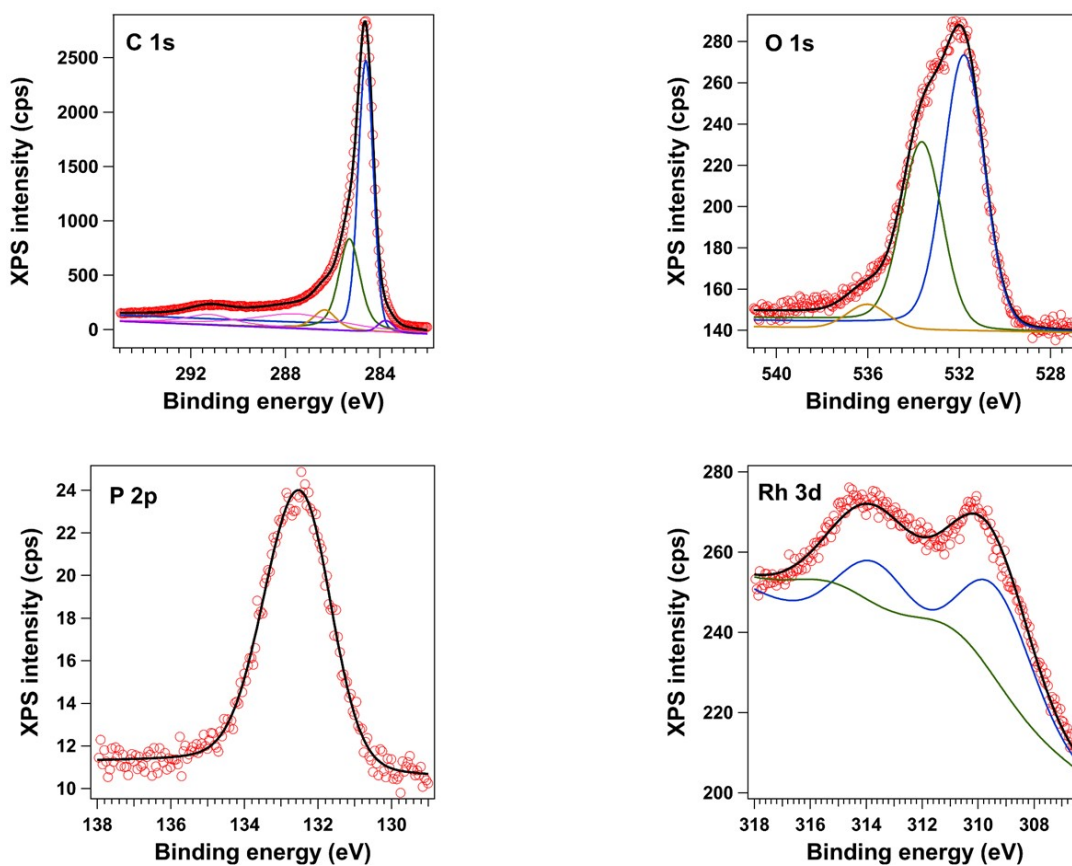


Figure S14: XPS survey of the elements in fresh CNT@complex Rh.





**Figure S15:** XPS analysis of the elements present in **CNT@complex Rh**. (red: experimental data, black: fitting, blue and green: deconvolution).

**Table S1:** XPS binding energies and Rh/P atomic ratio for the **CNT@complex Rh**.

	Binding energy (eV)			
Catalyst	O1s	C1s	P2p	Rh3d
CNT@complex Rh	531.7	284.8	132.3	309.6 + 313.8
	533.3	285.3		310.5 + 315.0
	536.0	286.3		
		287		
		290.8		
Rh/P atomic ratio				
	XPS value		Theoretical value	
CNT@complex Rh	0.52		0.50	

## 8. GC analysis of catalysis by CNT@complex Rh.

**Table S2: GC analysis of catalysis result for the enantioselective hydrogenation of dimethyl itaconate by CNT-immobilized Rh complex (entry 3 of Table 1).**

Software Version	: 6.3.2.0646	Date	: 2/25/2022 10:41:45 AM
Operator	: TPROCESS	Sample Name	:
Sample Number	:	Study	:
AutoSampler	: NONE	Rack/Vial	: 0/0
Instrument Name	: CLARUS 580	Channel	: A
Instrument Serial #	: 580S12101703	A/D mV Range	: 1000
Delay Time	: 0.00 min	End Time	: 30.00 min
Sampling Rate	: 12.5000 pts/s		
Sample Volume	: 1.000000 ul	Area Reject	: 0.000000
Sample Amount	: 1.0000	Dilution Factor	: 1.00
Data Acquisition Time	: 2/23/2022 11:45:32 AM	Cycle	: 1

Raw Data File : C:\Clarus580\DATA\Zinnia\ZA79\_90.raw  
 Result File : C:\Clarus580\DATA\Zinnia\ZA79\_90.rst [Editing in Progress]  
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 Sequence File : C:\AUTOSYSTEM\XL\SEQUENCE\ZA79\_90.seq

## DEFAULT REPORT

Peak #	Time [min]	Area [μV·s]	Height [μV]	Area [%]	Norm. Area [%]	BL	Area/Height [s]
1	0.764	54071.02	46441.07	0.90	0.90	BV	1.1643
2	0.806	393901.29	106733.85	6.57	6.57	VV	3.6905
3	0.853	85904.45	79489.69	1.43	1.43	VV	1.0807
4	0.878	156542.94	84521.51	2.61	2.61	VV	1.8521
5	0.903	29653.46	61845.76	0.49	0.49	VV	0.4795
6	0.920	4861470.14	988672.15	81.11	81.11	VE	4.9172
7	1.103	4043.73	1526.98	0.07	0.07	EV	2.6482
8	1.204	189.42	208.49	0.00	0.00	VB	0.9086
9	1.314	450.87	240.97	0.01	0.01	BV	1.8710
10	1.386	109.23	73.54	0.00	0.00	VB	1.4853
11	1.705	591.21	228.39	0.01	0.01	BB	2.5885
12	1.867	240.80	88.35	0.00	0.00	BB	2.7257
13	2.051	709.10	244.46	0.01	0.01	BB	2.9006
14	2.161	3433.46	1929.70	0.06	0.06	BV	1.7793
15	2.249	162.31	76.50	0.00	0.00	VB	2.1217
16	4.656	218.19	73.41	0.00	0.00	BB	2.9722
17	7.812	603.57	127.21	0.01	0.01	BV	4.7447
18	7.945	76427.56	11037.13	1.28	1.28	VV	6.9246
19	8.334	37390.11	4998.38	0.62	0.62	VB	7.4804
20	11.544	285505.02	29549.25	4.76	4.76	BE	9.6620
21	11.941	1801.68	222.44	0.03	0.03	EB	8.0995
22	12.050	35.52	72.10	0.00	0.00	BB	0.4926
23	12.067	60.02	78.61	0.00	0.00	BB	0.7635
24	12.092	55.78	82.34	0.00	0.00	BB	0.6774
25	12.112	38.09	67.90	0.00	0.00	BB	0.5610
		5993608.95	1.42e+06	100.00	100.00		

The retention times for the produced methyl succinate were 7.9 and 8.3 min for the (*S*) and (*R*) forms, respectively, and 11.5 min for dimethyl itaconate.

$R = 0.62$

$S = 1.28$

Reactant = 4.76

$$\% \text{ Conversion} = 100 - \frac{\text{Area of Reactant}}{\text{Total area}} * 100$$

$$\% \text{ Conversion} = 100 - \frac{4.76}{4.76 + 1.28 + 0.62} * 100$$

$$\% \text{ Conversion} = 100 - \frac{4.76}{4.76 + 1.28 + 0.62} * 100$$

$$\% \text{ Conversion} = 29$$

$$R = 0.62$$

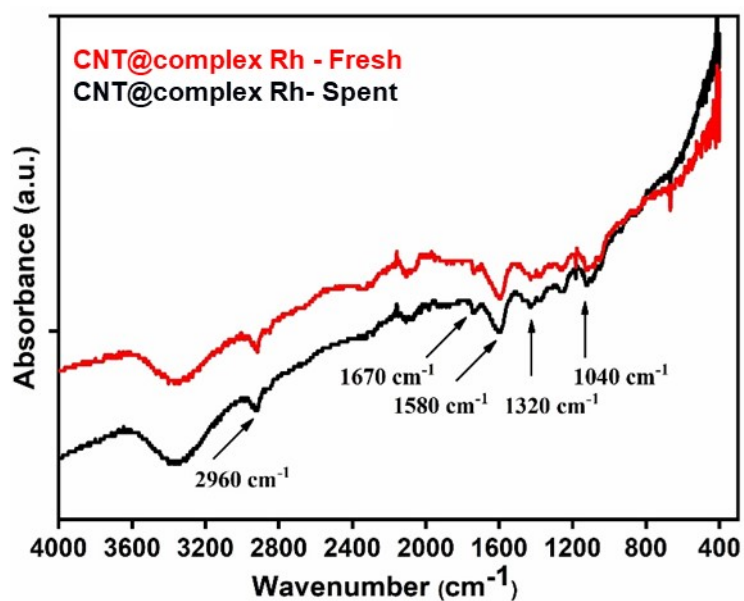
$$S = 1.28$$

$$\% ee = \frac{|R - S|}{R + S} * 100$$

$$\% ee = \frac{|0.62 - 1.28|}{0.62 + 1.28} * 100$$

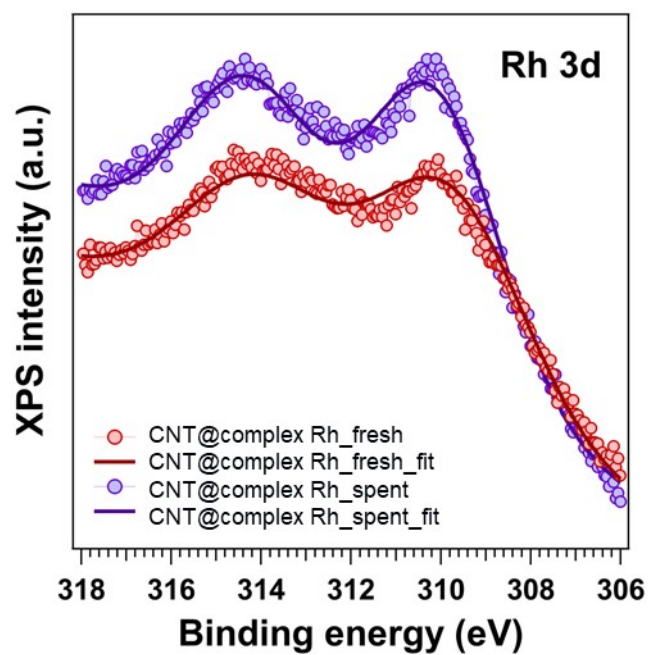
$$\% ee = 34 (S)$$

## 9. FTIR spectra of fresh and spent CNT@complex Rh catalyst.



**Figure S16:** FTIR spectra of fresh (red) and spent (black) CNT@complex Rh catalyst.

10. XPS spectra of Rh for the fresh and spent of CNT@complex Rh.



**Figure S17:** XPS spectra of Rh for the fresh (red) and spent (violet) CNT@complex Rh.