```
Supplementary Material
 1
 2
 3 Whole-cell catalytic production of ethylene glycol from C1 compounds using
   engineered glycolaldehyde synthase
 4
 5
   Xinyu Tian<sup>1,2,#</sup>, Jianyu Long<sup>1,2,#</sup>, Jia Yin<sup>1,2</sup>, Mingfang Zhou<sup>1,2</sup>, Chaofeng Shao<sup>1,2</sup>,
 6
 7 Bigiang Chen<sup>1,2*</sup>, Tianwei Tan<sup>1,2,3*</sup>
 8
   1. Beijing Key Lab of Bioprocess, National Energy R&D Center for Biorefinery, Beijing
 9
10 100029, PR China.
11 2. College of Life Science and Technology, Beijing University of Chemical Technology,
12 Beijing 100029, PR China.
13 3. Lead contact
14
15
16
```

*Co-corresponding author.

[#]These authors contributed equally to this work.

E-mail address: Biqiang Chen: chenbq@mail.buct.edu.cn; Tianwei Tan: twtan@mail.buct.edu.cn

17 Supporting methods

18 Molecular dynamics simulation

Molecular dynamics simulations were performed using GROMACS-2022 19 software¹ and GPU acceleration was used. The initial all-atom coordinates were derived 20 from the crystal structure (PDB ID: 6a50)². The amber99sb-ildn³ all-atom force field 21 was used to model GALS-M3 and WT. After ThDP binds to the enzyme backbone, 22 there will be proton transfer and conformational changes (Figure S1b), so the charge 23 distribution is significantly different from that of ThDP in the AP state. When Mg²⁺ 24 ions cooperate with the coenzyme ThDP, the charge distribution of each atom of ThDP 25 will be affected. Gaussian 16 combined with Ambertools 23 was used to fit the 26 electrostatic potential charge of the Mg²⁺-ThDP complex and the single ThDP. The 27 functional used b3lyp and the basis set used 6-311+g(d,p). In the modeling process of 28 Mg2+-ThDP complex and single ThDP, ThDP adopts the enamine-carbon anion 29 conformation (Figure S12). The ThDP conformation in this form is the near reaction 30 conformation of the aldehyde condensation reaction⁴. 31

32 Periodic boundary conditions were applied and the explicit water model TIP3P⁵ was used to solvate the entire system. TIP3P water filled the entire dodecahedral water 33 box (the protein was located in the center of the box and the minimum distance from 34 the boundary was 10Å). Na+ and Cl- were used to balance the system charge to make 35 the net charge of the system 0. The cutoff distance for short-range non-bonded 36 interactions (short-range van der Waals forces and short-range Coulomb forces) was 37 set to 14Å, and the long-range electrostatic interactions were treated using the PME 38 method⁶. The grid spacing of the FFT was set to 1.6Å. 39

The steepest descent algorithm and the conjugate gradient method were used in combination to fully relax the entire system, with a maximum force of 1 kJ/mol/nm. The temperature was increased and the pressure was equilibrated by canonical ensemble (NVT) simulations and isothermal isobaric ensemble (NPT) simulations. During the NVT, the system temperature was uniformly increased to 310 K by simulated annealing, and the temperature was controlled using a V-rescale thermostat⁷. During the NPT, the temperature was controlled at 310 K and the system pressure was

equilibrated at 1 bar using a Berendsen barostat⁸. The equilibrium simulation time was 47 100 ps. After the NVT and NPT, a formal MD simulation was performed for 20 ns, at 48 which the system temperature was controlled at 310 K using a Nose-Hoover thermostat⁹ 49 and the pressure was controlled at 1 bar using a Parrinello-Rahman barostat¹⁰. The 50 formal MD simulation was performed three times in parallel, each with a different 51 random initial velocity. The Newtonian equations of motion were integrated using the 52 leap-frog algorithm with an integration step of 2 fs, and the bonds were constrained 53 using the LINCS algorithm¹¹. During the simulation, both the protein and the ligand 54 were restrained by a force constant of 1000 kJ/mol·nm². 55

After the simulation, the rms and rmsf modules of GROMACS were used to 56 extract the RMSD of the protein skeleton and ThDP and the RMSF of each amino acid 57 of the protein; the cluster module was used to perform protein conformation clustering 58 analysis and extract the stable state protein conformation. The clustering method was 59 $gromos^{12}$, and the RMSD cut-off was set to 0.1nm. Caver2.0¹³ was used to analyze the 60 tunneling situation during the simulation, and the stable protein conformation (15ns-61 20ns) was taken for tunneling scanning calculation. Starting from the 15ns (including 62 the 15ns), the trajectory file was sampled every 10ps, and finally a protein conformation 63 sample containing 501 frames of continuous frames was obtained for tunneling 64 analysis. 65

66

68 Supporting Tables

69

70 Table S1. Primers used in this study.

Primers	Sequence
W463X-F	ctgcgtnnkttcgctggtgttctggaagc
W463X-R	cagcgaamnnacgcagcataccgtaggtacc
T87X-F	gctcgtnnktctcactctccgctgatcgt
T87X-R	gagtgagamnnacgagcgttagacagagcaccc
G401X-F	ctgctnnkggtctgggtttcgctctgcc
G401X-R	ccagaccmnnagcagcgcagaagtagtaagaacc
C398X-F	ctacttcnnkgctgctggtggtctgggt
C398X-R	caccagcagcmnngaagtagtaagaacccgg
F397X-F	ctacnnktgcgctgctggtggtctgggtttc
F397X-R	cagcgcamnngtagtaagaacccgggttacgc

72 3. Supporting Figures

73



74

75 Figure S1. Structure of thiamine pyrophosphate (ThDP) and thiamine (VB1). a.

76 ThDP. b. Structural changes of ThDP during the reaction. ThDP is mostly in the form 77 of AP when not bound to the enzyme. It is activated to IP by proton transfer when bound 78 to the enzyme, and then forms a ylide in the near-reaction state and an enamine-79 carbanion form by proton transfer.





82 Figure S2. Coefficient of variation (CV) determination of TTC method. WT were

83 cultured in 96-well plates and the response of all WT at 510 nm was measured when

84 derivatized with TTC.



Figure S3. Catalytic activity screening of active clones in the saturation mutant
library. The Y-axis corresponds to the relative activity of the variants compared to the
WT. The X-axis corresponds to the number of clones in descending order. The clones
with an activity 1.18 times higher than that of the WT were selected for sequencing
analysis. a. W463X library; b. T87X library; c. G401X library; d. C398X library; e.
F397X library.



94

95 Figure S4. Michaelis-Menten plots of variants catalyzing FALD condensation to 96 form GALD. a. Variant A416T; b. Variant T87A/A416T; c. Variant A416T/W463I; d. 97 Variant T87A/A416T/W463I. The enzyme addition amount was 0.1 mg/mL and the 98 reaction temperature was 37°C. The reactions were performed independently at pH 8. 99 The data in a-d represent means \pm s.d., as determined from n = 3 independent 100 experiments.



Figure S5. Changes of RMSD of WT and M3 protein backbone over time during
three parallel 20 ns MD simulations. a. Changes of RMSD of WT protein backbone
over time during three parallel 20 ns MD simulations. b. Changes of RMSD of M3
protein backbone over time during three parallel 20 ns MD simulations.



108

109 Figure S6. The change of gyration radius of WT and M3 proteins over time during

110 three parallel 20 ns MD simulations. a. The change of gyration radius of WT protein

111 over time during three parallel 20 ns MD simulations. b. The change of gyration radius

- 112 of M3 protein over time during three parallel 20 ns MD simulations.
- 113



114

Figure S7. Changes and distributions of Curvature, Length, Min radius, and 115 Radius (Average Radius) of Tunnel A in 15 ns-20 ns WT and M3 proteins during 116 the simulation. a. Changes and distributions of Min_radius of Tunnel A in 15 ns-20 ns 117 WT and M3 proteins during the simulation. b. Changes and distributions of Radius 118 (Average Radius) of Tunnel A in 15 ns-20 ns WT and M3 proteins during the 119 simulation. c. Changes and distributions of Curvature of Tunnel A in 15 ns-20 ns WT 120 and M3 proteins during the simulation. d. Changes and distributions of Length of 121 Tunnel A in 15 ns-20 ns WT and M3 proteins during the simulation. 122





125 Figure S8. Relative positions and bonding diagrams of residues 416 and L403 in

126 WT and M3. Residues 416, L403, and ThDP are represented as stick models. Gray

127 represents WT; orange represents M3; and green lines represent hydrogen bonds.



130 Figure S9. RMSF of all amino acids for WT and M3 during three parallel 20 ns

131 MD simulations.



134 Figure S10. Probability of hydrogen bonding between residue L403 and ThDP in

135 WT and M3.



138 Figure S11. Optimization of whole-cell catalytic reaction conditions. a. Optimizing

- 139 reaction pH. b. Optimizing reaction temperature.
- 140



143 Figure S12. Initial ThDP modeling conformation and its atomic nomenclature. The

144 molecule is shown as a stick model, with C atoms in gray, S atoms in yellow, N atoms

145 in dark blue, P atoms in orange, and O atoms in red.



Figure S13. SDS-PAGE of WT and activity-enhanced variants. All sample volumes
were kept the same to facilitate observation of the effect of mutations on enzyme
expression.

153 Reference

- M. J. Abraham, T. Murtola, R. Schulz, S. Páll, J. C. Smith, B. Hess and E.
 Lindahl, SoftwareX, 2015, 1-2, Medium: ED; Size: p. 19-25.
- 156 2. X. Lu, Y. Liu, Y. Yang, S. Wang, Q. Wang, X. Wang, Z. Yan, J. Cheng, C. Liu,
- 157 X. Yang, H. Luo, S. Yang, J. Gou, L. Ye, L. Lu, Z. Zhang, Y. Guo, Y. Nie, J.
- Lin, S. Li, C. Tian, T. Cai, B. Zhuo, H. Ma, W. Wang, Y. Ma, Y. Liu, Y. Li and
- 159 H. Jiang, Nat Commun, 2019, 10, 1378.
- 160 3. K. Lindorff-Larsen, S. Piana, K. Palmo, P. Maragakis, J. L. Klepeis, R. O. Dror
- and D. E. Shaw, Proteins: Structure, Function, and Bioinformatics, 2010, 78,
 1950-1958.
- 163 4. F. Planas, M. J. McLeish and F. Himo, ACS Catalysis, 2019, 9, 5657-5667.
- 164 5. F. Duarte, P. Bauer, A. Barrozo, B. A. Amrein, M. Purg, J. Åqvist and S. C. L.
- 165 Kamerlin, The Journal of Physical Chemistry B, 2014, 118, 4351-4362.
- 166 6. T. A. Darden, D. M. York and L. G. Pedersen, Journal of Chemical Physics,
 167 1993, 98, 10089-10092.
- 168 7. G. Bussi, D. Donadio and M. Parrinello, The Journal of Chemical Physics, 2007,
 169 126.
- 170 8. H. J. C. Berendsen, J. P. M. Postma, W. F. van Gunsteren, A. DiNola and J. R.
- 171 Haak, The Journal of Chemical Physics, 1984, 81, 3684-3690.
- D. J. Evans and B. L. Holian, The Journal of Chemical Physics, 1985, 83, 40694074.
- 174 10. M. Parrinello and A. Rahman, Journal of Applied Physics, 1981, 52, 7182-7190.

175 11.	B. Hess, Journal of Chemical Theory and Computation, 2008, 4, 116-122.
176 12.	X. Daura, K. Gademann, B. Jaun, D. Seebach, W. F. van Gunsteren and A. E.
177	Mark, Angewandte Chemie International Edition, 1999, 38, 236-240.
178 13.	A. Jurcik, D. Bednar, J. Byska, S. M. Marques, K. Furmanova, L. Daniel, P.
179	Kokkonen, J. Brezovsky, O. Strnad, J. Stourac, A. Pavelka, M. Manak, J.
180	Damborsky and B. Kozlikova, Bioinformatics, 2018, 34, 3586-3588.
181	