

Supporting Information

Ruthenium doped cobalt sulphide electrocatalyst derived from ruthenium-cobalt Prussian blue analogue (RuCo-PBA) for enhanced hydrogen evolution reaction (HER)

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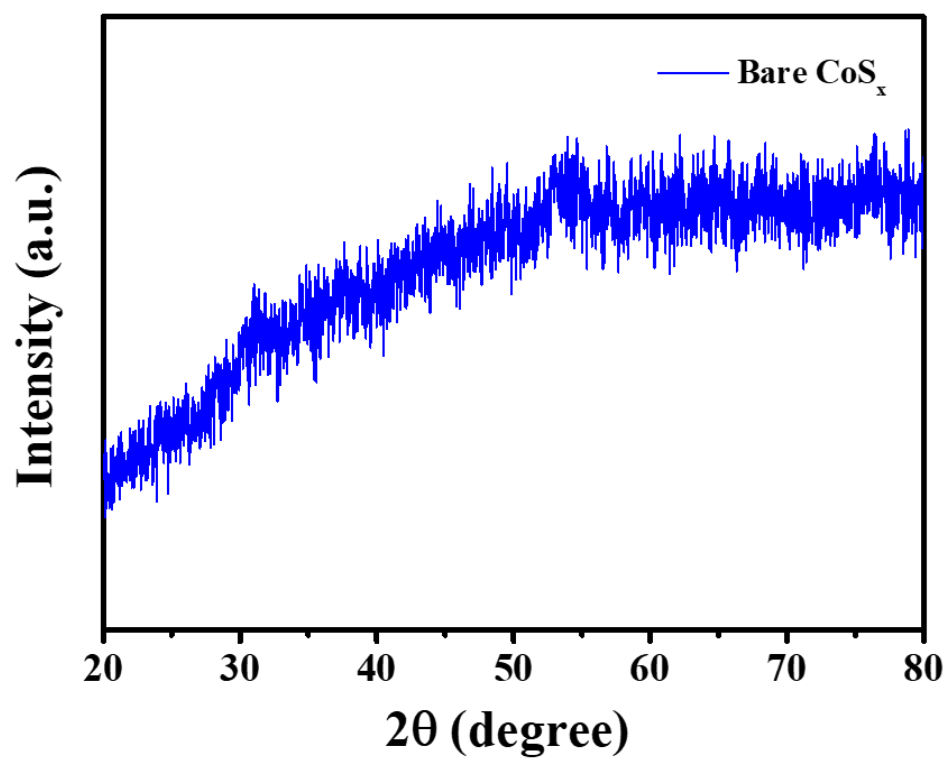


Fig. S1: Powder X-ray diffraction study of bare CoS_x.

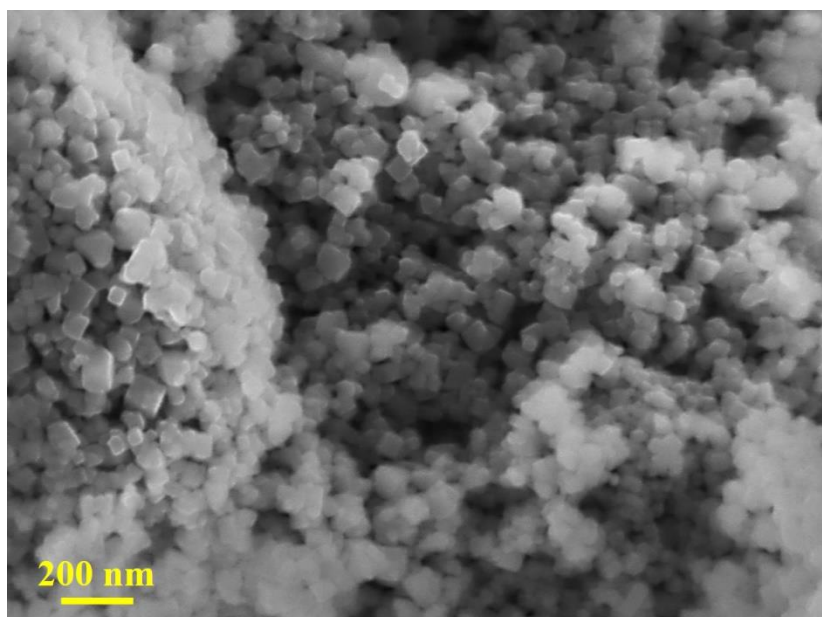


Fig. S2: FESEM image of RuCo-PBA.

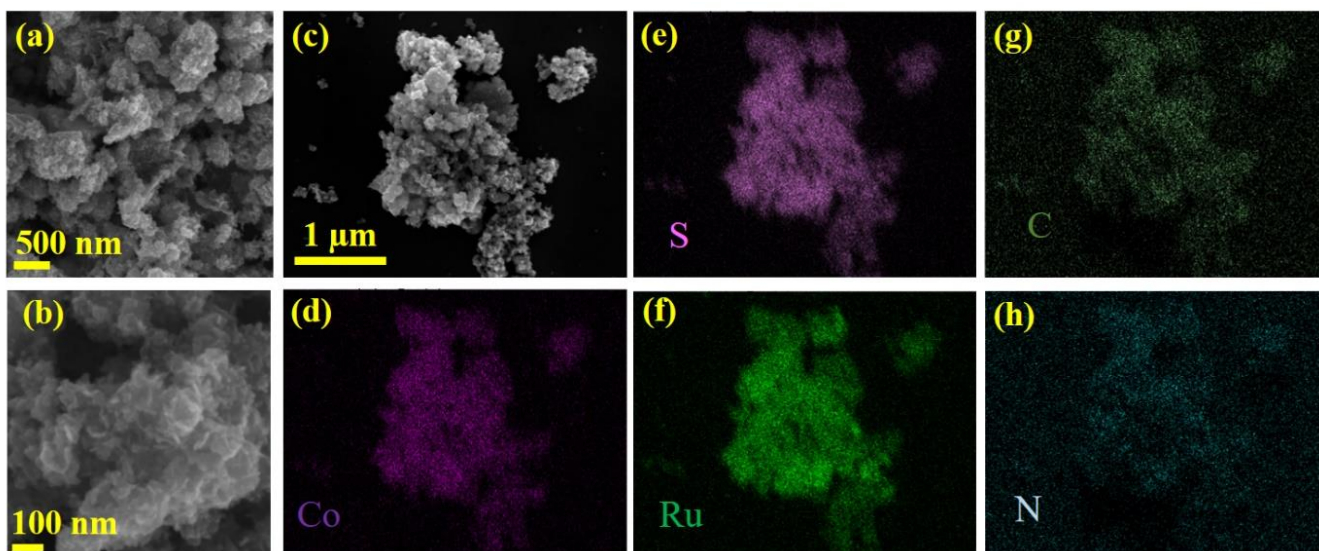


Fig. S3: FESEM images of $\text{Co}_9\text{S}_8/\text{Ru}@24\text{H}$ at different scales (a-c) and elemental mapping of $\text{Co}_9\text{S}_8/\text{Ru}@24\text{H}$ (d-h).

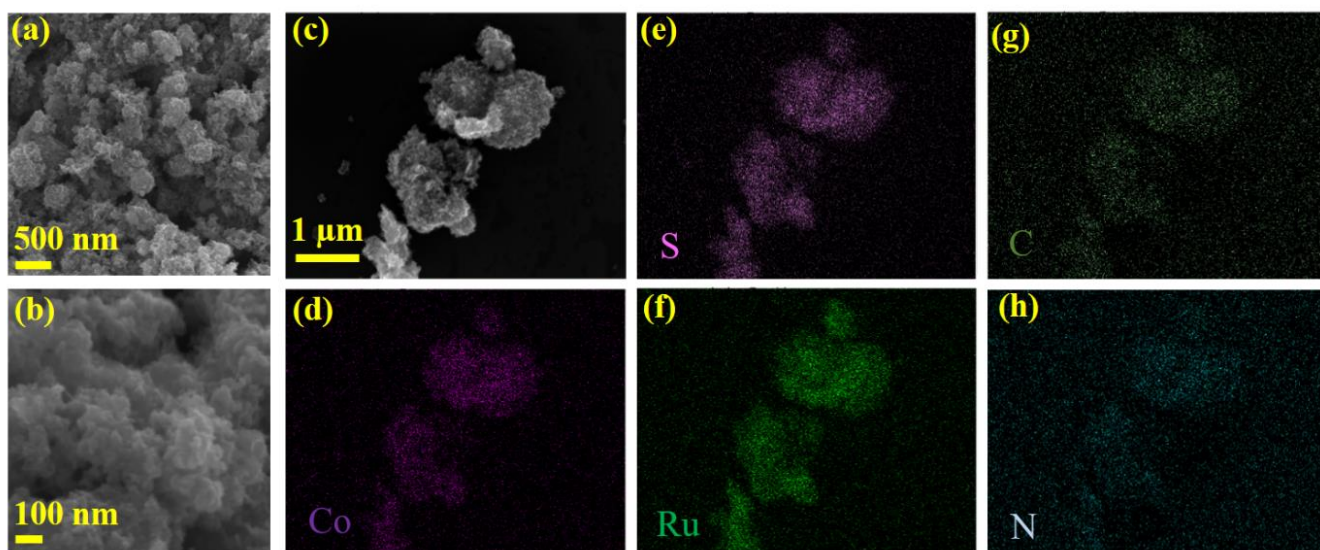


Fig. S4: FESEM images of $\text{Co}_9\text{S}_8/\text{Ru}@72\text{H}$ at different scales (a-c) and elemental mapping of $\text{Co}_9\text{S}_8/\text{Ru}@72\text{H}$ (d-h).

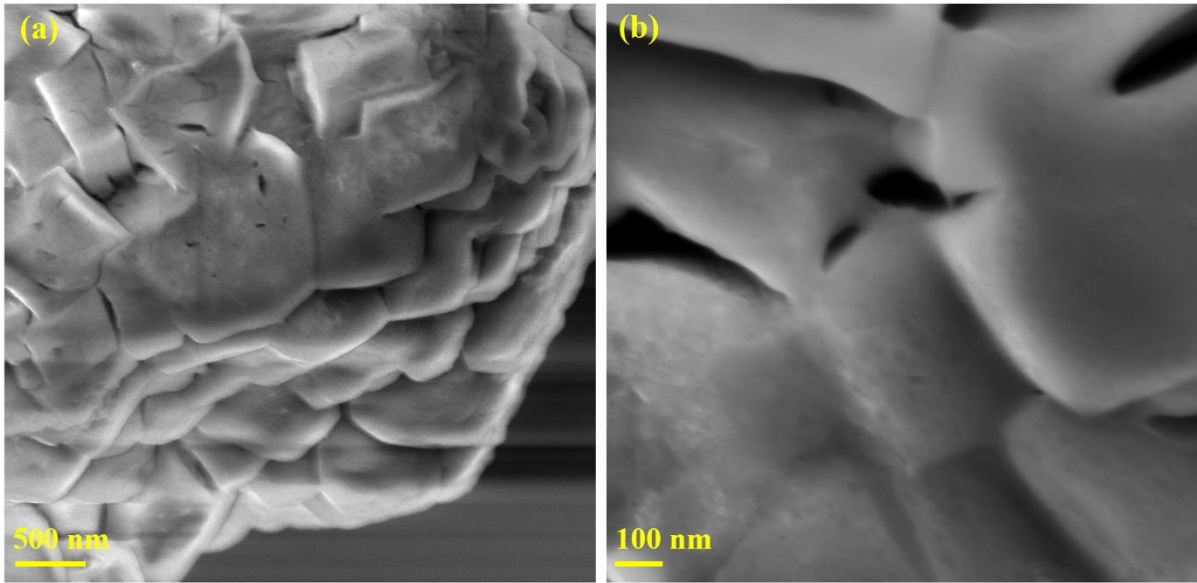


Fig. S5: FESEM images of CoS_x at different scales (a & b).

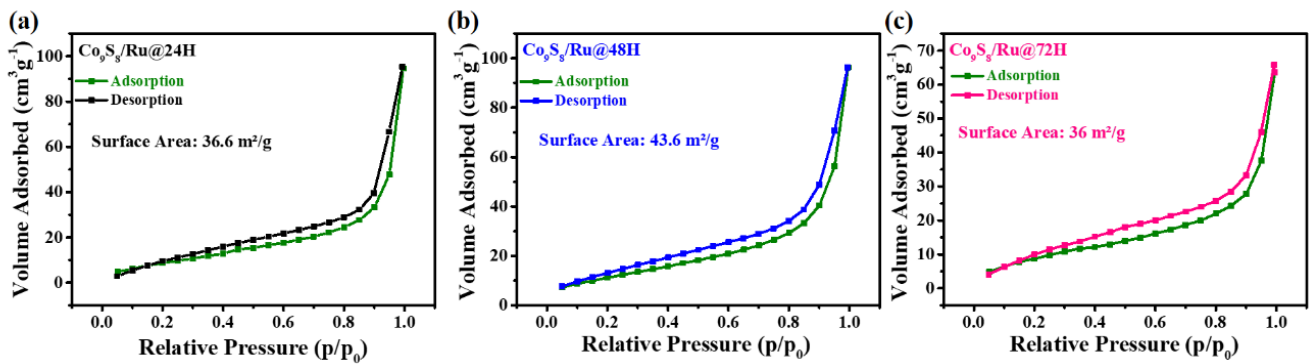


Fig. S6: Nitrogen adsorption-desorption isotherm for $\text{Co}_9\text{S}_8/\text{Ru}@24\text{H}$ (a), $\text{Co}_9\text{S}_8/\text{Ru}@48\text{H}$ (b) and $\text{Co}_9\text{S}_8/\text{Ru}@72\text{H}$ (c).

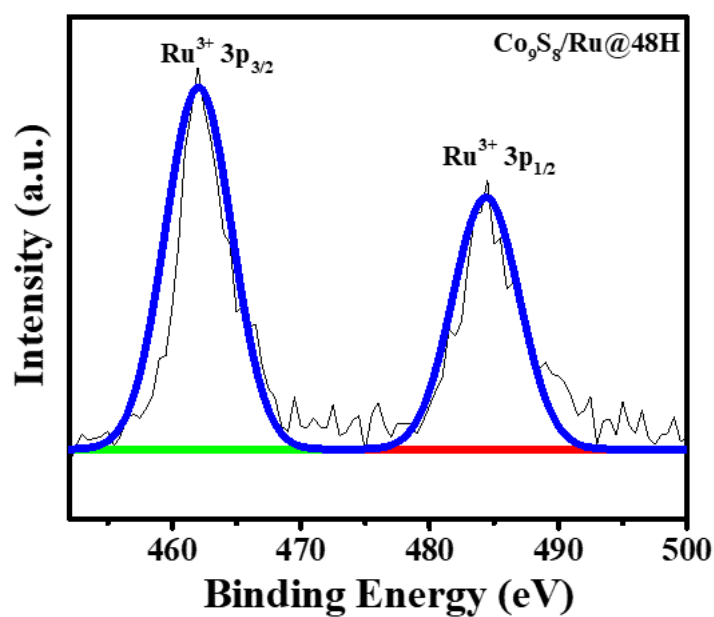


Fig. S7: XPS spectrum of Ru 3p in $\text{Co}_9\text{S}_8/\text{Ru}@48\text{H}$.

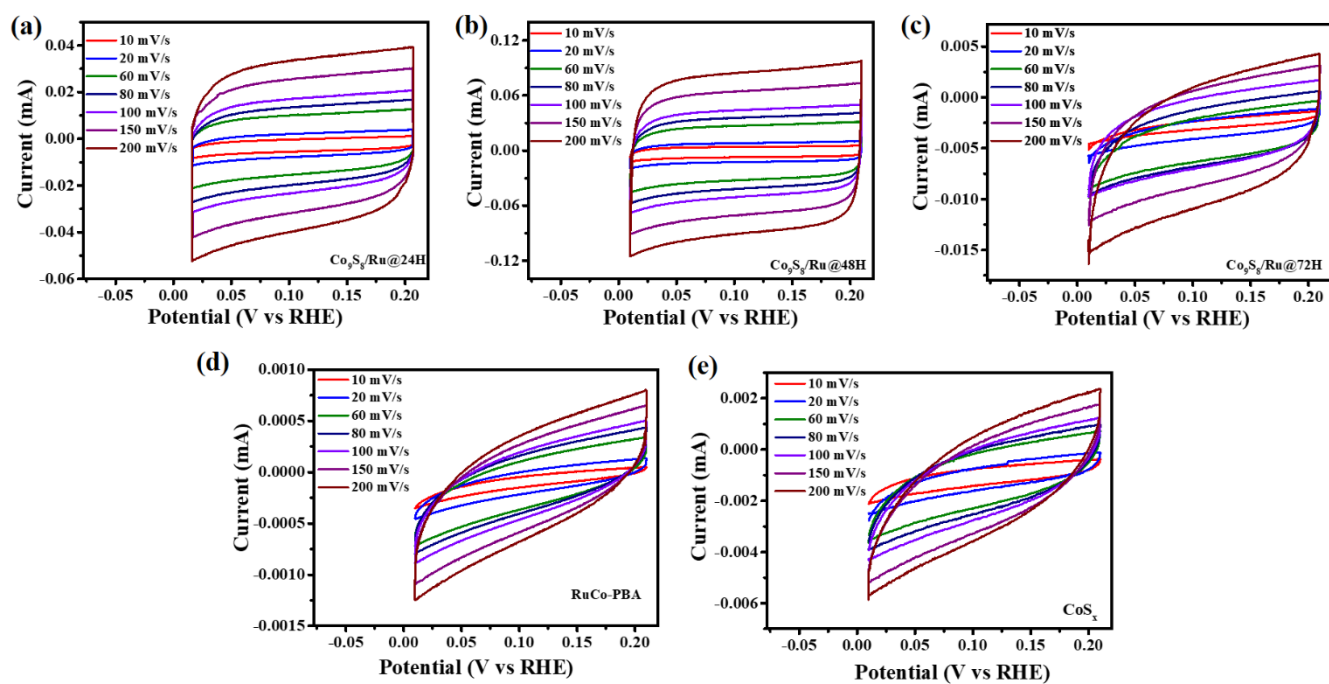


Fig. S8: Cyclic voltammograms (CVs) of $\text{Co}_9\text{S}_8/\text{Ru}@24\text{H}$ (a), $\text{Co}_9\text{S}_8/\text{Ru}@48\text{H}$ (b), $\text{Co}_9\text{S}_8/\text{Ru}@72\text{H}$ (c), RuCo-PBA (d) and bare CoS_x (e).

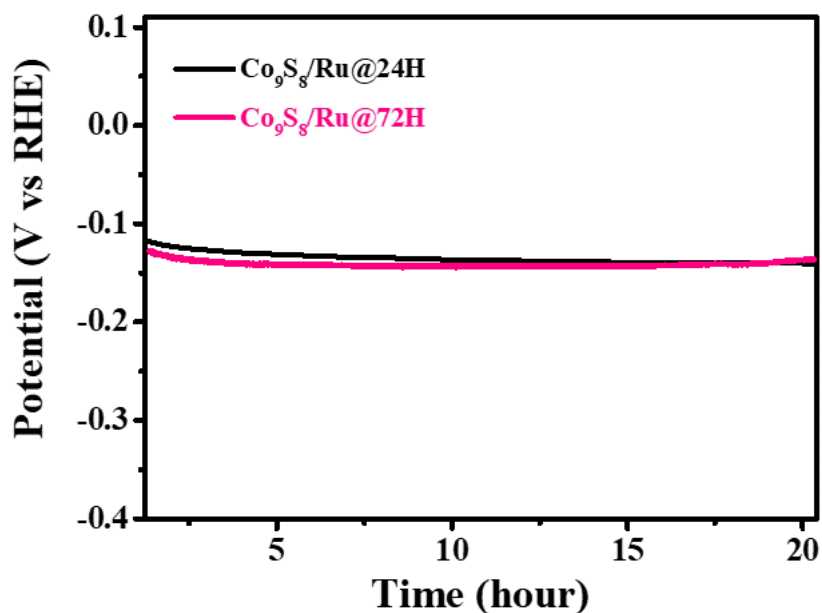


Fig. S9: Chronopotentiometry test of Co₉S₈/Ru@24H and Co₉S₈/Ru@72H in 0.5 M H₂SO₄ acidic solution.

Table S1: Comparison of metal sulphide-based electrocatalysts for HER.

	Overpotential at 10 mA cm ⁻² (mV)	Tafel slope (mV dec ⁻¹)	Electrolyte	References
Co ₉ S ₈ -30@MoS _x /CC	98	64.8	0.5 M H ₂ SO ₄	1
Co ₉ S ₈ -NDCL	149@20mA cm ⁻²	70	0.5 M H ₂ SO ₄	2
Co ₉ S ₈ /NC@MoS ₂	117	68.8	0.5 M H ₂ SO ₄	3
MoS ₂ /Co ₉ S ₈ /MoC@CNT-N	174.2	84.7	0.5 M H ₂ SO ₄	4
CoS ₂ nanowires	145	51.6	0.5 M H ₂ SO ₄	5
Co _x S _y	188	96	0.5 M H ₂ SO ₄	6
Co ₉ S ₈ @MoS ₂	106	51.8	0.5 M H ₂ SO ₄	7
Zn-Co ₉ S ₈ @CF-(1-1)	278	85.2	0.5 M H ₂ SO ₄	8
Co ₉ S ₈ /Ru@48H	94	84	0.5 M H ₂ SO ₄	This work

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