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Supporting Information

for

Designing One-compartment H₂O₂ Fuel Cell Using Electroactive Phenalenyl-based

[Fe₂(hnmh-PLY)₃] Complex as the Cathode Material

Nisha Kamboj^[a], Ayan Dey^[b], Sunita Birara^[a], Moumita Majumder^{*[c]}, Srijan Sengupta^{*[b]},

Ramesh K. Metre*[a]

^[a] Department of Chemistry, Indian Institute of Technology Jodhpur, Rajasthan- 342030, India

^[b] Department of Metallurgical and Materials Engineering, Indian Institute of Technology Jodhpur, Rajasthan- 342030, India

^[c] Department of Chemistry, School of Science and Environmental Studies, Dr. Vishwanath Karad MIT World Peace University, Pune, Maharashtra-411038, India

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CORRESPONDING AUTHOR FOOTNOTE: * To whom correspondence should be addressed. E-mail: <u>srijansengupta@iitj.ac.in</u>, <u>moumita83iitd@gmail.com</u>, <u>rkmetre@iitj.ac.in</u>. Phone: (+91) 291-280-1309.

1. Crystallographic data for the hnmh-PLYH₂ ligand and complex 1.

Identification code	hnmh-PLYH ₂	Complex 1
Empirical formula	$C_{96}H_{64}N_8O_8$	$C_{72}H_{42}Fe_2N_6O_6$
Formula weight	1457.55	1198.82
Temperature/K	273	100
Crystal system	orthorhombic	triclinic
Space group	P2 ₁ 2 ₁ 2 ₁	P-1
a/Å	4.6410(8)	13.2008(16)
b/Å	13.252(3)	15.706(2)
c/Å	27.377(5)	16.0595(17)
α/°	90	81.353(4)
β/°	90	77.826(4)
γ/°	90	67.397(4)
Volume/Å ³	1683.8(5)	2995.9(6)
Ζ	1	2
$\rho_{calc}g/cm^3$	1.437	1.329
µ/mm ⁻¹	0.093	0.544
F(000)	760.0	1232.0
Crystal size/mm ³	0.2 imes 0.2 imes 0.2	0.2 imes 0.2 imes 0.2
Radiation	MoKα (λ = 0.71073)	MoK α ($\lambda = 0.71073$)
2Θ range for data collection/°	4.278 to 56.796	3.98 to 57.118
Index ranges	$-6 \le h \le 6, -17 \le k \le 17, -36 \le l \le 36$	$\begin{array}{l} \textbf{-16} \leq h \leq 17, \textbf{-21} \leq k \leq 21, \textbf{-20} \leq 1 \\ \leq 21 \end{array}$
Reflections collected	23273	48330
Independent reflections	4229 [R _{int} = 0.1155, R _{sigma} = 0.0708]	15275 [$R_{int} = 0.0518$, $R_{sigma} = 0.0615$]
Data/restraints/parameters	4229/0/254	15275/0/765
Goodness-of-fit on F ²	1.140	1.221
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0653, wR_2 = 0.1632$	$R_1 = 0.0773, wR_2 = 0.1880$
Final R indexes [all data]	$R_1 = 0.0966, wR_2 = 0.1897$	$R_1 = 0.1027, wR_2 = 0.2004$
Largest diff. peak/hole / e Å ⁻³	0.36/-0.42	2.04/-0.55

Table S1. Crystal data and structure refinement parameters for the ligand, $hnmh-PLYH_{2}$, and complex 1.

2. General characterization of hnmh-PLYH₂ ligand and complex 1.



Figure S1. (a) ¹H and (b) ¹³C NMR spectra of hnmh-PLYH₂ ligand recorded in CDCl₃ solvent.



Figure S2. (a) HRMS spectrum of hnmh-PLYH₂ ligand. HRMS (CH₃CN, positive ionization): calcd. for $C_{24}H_{17}N_2O_2$ m/z = 365.1290 [M+H]⁺, found 365.1286 [M+H]⁺. (b) HRMS spectrum of complex 1. HRMS (CH₃CN, positive ionization): calcd. for $C_{72}H_{42}Fe_2N_6O_6$ m/z = 1199.1943 [M+H]⁺, found 1199.1984 [M+H]⁺.



Figure S3. FT-IR spectrum of (a) hnmh-PLYH₂ and (b) complex 1.



Figure S4. UV-vis spectrum of (a) hnmh-PLYH₂ and (b) complex 1 recorded in CH_2Cl_2 solvent.



Figure S5. (a) XPS, (b) solid state X-band EPR, and (c) elemental analysis of complex 1.

 H₂O₂ fuel cell performance test and CV experiments for mechanistic insights into H₂O₂ reduction by GC-1 electrode.



Figure S6. Electrocatalytic reduction of H_2O_2 with bare glassy carbon (GC) electrode in acetate buffer (pH 3) containing Ag/AgCl (3.0 M KCl) as the reference, and Pt-wire as the counter electrode at a scan rate of 100 mV s⁻¹.



Figure S7. *I-V* (red) and *I-P* (blue) curves of one-compartment H₂O₂ fuel cell with Ni anode and GC-1 cathode. The performance tests were conducted in an acetate buffer (pH 3), containing $0.3 \text{ M H}_2\text{O}_2$.



Figure S8. (a) Variation in *I-V* (red) and *I-P* (blue) curves for the one-compartment H₂O₂ fuel cell designed with **GC-1** cathode, and Ni foam anode with the change in concentration of aqueous HCl electrolyte. The concentration of H₂O₂ was maintained as 0.3 M. (b) Polarization curves for **GC-1** and **GC-1C** as a cathode in 0.1 M HCl containing 0.3 M H₂O₂, and Ni foam as the anode in one compartment H₂O₂ fuel cell. Current density is normalized by the geometrical area of the glassy carbon electrode.





Figure S9. (a) UV-vis spectra of complex 1 and the recovered catalyst from the GC-1 electrode after the one-compartment H_2O_2 fuel cell experiments. (b) PXRD pattern of complex 1, 1 + 10% wt./wt. C, and recovered catalyst from GC-1 after the one-compartment H_2O_2 fuel cell experiments.





Figure S10. Cyclic voltammogram of (a) $FeCl_3$ modified GCE, GC-Fe; hnmh-PLYH₂ ligand modified GCE, GC-L; and (b) GC-1 recorded in 0.1 M HCl at a scan rate of 100 mV sec⁻¹ (Inset shows the change in cyclic voltammogram on the addition of H₂O₂). (c) A generalized mechanism for H₂O₂ reduction by complex 1.



Figure S11. Solid-state CV of the complexes, (a) 1 and (b) [Fe₂(hmbh-PLY)₃] immobilized over GCE in 0.1 M HCl electrolyte.

4. H₂O₂ Fuel Cell Demonstration Test

A demonstration setup was designed to test the application of complex 1 modified cathode **GC-1C** in the one-compartment H_2O_2 fuel cell. Four beakers containing 0.3 M H_2O_2 in 0.1 M HCl electrolyte were connected in a series setup using the conducting copper wire through the Ni foam as anode and **GC-1C** as the cathode. The OCP for the three setups in series was measured as ~2.5 V. As shown in Figure S11, the current flow in the designed H_2O_2 fuel cell setup has successfully glowed a red LED bulb (light emitting diode, 2 V), as a consequence of the reactions at the cathode and anode's surface.



Figure S12. Demonstration setup of the one-compartment H_2O_2 fuel cell employing GC-1 cathode and Ni foam as an anode.

5. DFT optimized atomic Cartesian coordinates of metal complex 1

Fe	13.40278321563452	12.40663045007605	3.38381576982838
Fe	15.36040268156612	9.50506586262461	4.90776043474558
0	15.50434839121535	7.78577397209173	3.99909515517464
0	12.41916932925645	12.13616207683466	1.72088177234550
0	11.94929017030265	13.22455577788315	4.37364359289031
0	14.29280548591336	14.05211316199941	2.81273896590917
0	14.85322159050611	8.84847146897547	6.66674541203073
0	17.27328835983996	9.66702812089816	5.28500449352787
N	14.79112682311064	11.00784028299603	2.42050020621097
N	12.65422421455948	10.59834096455075	4.38736594687717
N	14.78209371334551	11.41825589277245	5.80609619884536
N	15.92554698246721	10.66383798395257	3.16301655162129
N	14.89035638137707	12.51395370566529	4.94483334196206
N	13.34425285625267	9.41443411225680	4.10190328823627
С	14.80211933032047	7.05225092160531	3.21583124361553
С	14.26566499851428	9.27632982909758	7.73263282405410
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С	13.47243971372434	7.37337085152113	2.78831108687530
С	14.15900422501033	11.62092377366071	6.92919413015459
Н	13.84994390919306	12.65299830690347	7.12443110412251

С	16.39069595339570	14.15896943165396	3.96631325301131
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Н	10.51893599285684	12.10199593987149	0.04389374536359
С	14.53903066537930	10.24109805577673	1.39843427252784
Н	15.29226060630758	9.47487614700169	1.18650964292634
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Η	14.37839901133780	7.31083095884408	8.62111106101258
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Н	17.44574718201773	16.75079287609725	1.18553163975609
С	13.38839763447412	8.69962028461891	9.93803697512900
Н	13.21885983795477	7.96416860174163	10.73009085128906
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Н	22.85550945981217	13.35199774349191	1.97923671939378
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С	10.87805410186391	7.99987669957710	1.79979691180602

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Η	16.78886247620951	12.14485186582850	6.75576661410733
С	18.28096857597649	13.42178285078127	5.95936328296476
Η	19.01558845371859	13.14010553112224	6.71810323300915

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