Luminescence Properties and Judd-Ofelt Analysis of Tb³⁺ Doped

Sr₂YTaO₆ Double Perovskite Phosphors for White LED

Applications

Li Wang^{1, 2}, Yuhang Zhang^{1, *}, Duan Gao¹, Xuezhu Sha¹, Xin Chen¹, Yanqiu Zhang¹, Jinsu Zhang¹, Xizhen Zhang¹, Yongze Cao¹, Yichao Wang¹, Xiangping Li¹, Sai Xu¹, Hongquan Yu¹, Baojiu Chen^{1, *}

School of Science, Dalian Maritime University, Dalian, 116026, Liaoning, PR China
 College of Medical Laboratory, Dalian Medical University, Dalian, 116044, Liaoning, PR China

Support Information

1. J-O calculation process based on emission spectra and radiative transition lifetime

The relative integrated PL intensity for ${}^{5}D_{4} \rightarrow {}^{7}F_{J}$ of Tb³⁺ ions can be firstly calculated from the PL spectrum of SYT: 5 mol% Tb³⁺ phosphor. Meanwhile, the sum of relative integrated PL intensity $I = \sum_{I=0-6} I_{7_{F_{J}}}$ of ${}^{5}D_{4}$ energy level of Tb³⁺ ions ($I = \sum_{I=0-6} I_{7_{F_{J}}}$) is usually proportional to the total radiative transition rate J = 0-6 [1] [2], and the relationship can be represented as follows: $I = a \sum_{I=0-6} A_{7_{F_{I}}}$ [1] [2], and the relationship can be represented as follows:

$$= a \sum_{J=0-6}^{A_{7}} A_{7_{F_{J}}}$$
(S1)

where $\int_{0}^{2} = 0 - 6 A_{7_{F_j}}$ is equal to the reciprocal of the radiative transition lifetime τ_0 , and the expression is written as

$$\sum_{J=0-6} A_{7_{F_J}} = \frac{1}{\tau_0}$$
(S2)

^{*} Corresponding Authors: YHZhang email: <u>yuhangzhang@dlmu.edu.cn</u>; BJChen email: <u>bjchen@dlmu.edu.cn</u> Fax and Tel.: 86-411-8472-8909

According to the above analysis, substitute $\tau_0 = 3.63$ ms into Eq. (S2) and combine it with Eq. (S1), the coefficient *a* can be confirmed as 9.74×10^{-4} . And then, the radiative transition rate from 5D_4 to 7F_1 were gotten by using Eq. (S3).

$$I_{7_{F_{J}}} = a \times A_{7_{F_{J}}}$$
(S3)

Furthermore, the radiative transition rates for a certain transition from state J to state J' can be expressed as

$$A_{J-J}^{md} = \frac{64\pi^4 v^3}{3h(2J+1)} n^3 \left(\frac{eh}{4\pi m_e c}\right)^2 \langle \Psi J \| L + 2S \| \Psi' J' \rangle^2$$
(S4)

$$A_{J-J}^{ed} = \frac{64\pi^4 e^2 v^3 n(n^2+2)^2}{3h(2J+1)} \sum_{\substack{\beta = 2,4,6}} \Omega_{\beta} \langle \Psi J \| U^{(\lambda)} \| \Psi J' \rangle^2$$
(S5)

where A^{md} and A^{ed} are the contribution from magnetic dipole (MD) to the transition rate and the contribution from electric dipole (ED), respectively.v is the central wavenumber for $J \rightarrow J'$ transition. *h* is Planck's constant ($h = 6.626 \times 10^{27} \text{ erg} \cdot \text{s}$), 2J+1 is the degeneracy of the initial state, *n* is the refractive index of the host (here *n* is 1.98 for SYT host), and $\langle \Psi J \| L + 2S \| \Psi' J' \rangle^2$ is the squared matrix element for MD line strength.[3] *e* is the charge of an electron ($e = 4.80 \times 10^{-10} \text{ esu}$) and $\langle \Psi J \| U^{(\lambda)} \| \Psi' J' \rangle$ is the reduced matrix element for the J \rightarrow J' transition.[3] Based on the above analysis, by applying the least squares method to the equations corresponding to all the transitions from ⁵D₄ to ⁷F_J, the parameters Ω_{λ} ($\lambda = 2, 4, 6$) were calculated.

Additionally, the fluorescence branch ratios for the ${}^{5}D_{4}$ energy level under 318 nm excitation can be derived from the following equation:

$$\beta = \frac{A_{J-J}}{\sum_{J=0-6} A_{J-J}}$$
 (S6)

[1] Tian, B.; Chen, B.; Tian, Y.; Li, X.; Zhang, J.; Sun, J., et al., Excitation pathway and temperature dependent luminescence in color tunable $Ba_5Gd_8Zn_4O_{21}$:Eu³⁺ phosphors. *J Mater Chem C* **2013**, *1* (12), 2338.

[2] Tian, Y.; Chen, B.; Hua, R.; Sun, J.; Cheng, L.; Zhong, H., et al., Optical transition, electron-phonon coupling and fluorescent quenching of La₂(MoO₄)₃:Eu³⁺ phosphor. *J Appl Phys* 2011, *109* (5), 053511.
[3] Loiko, P.; Mateos, X.; Dunina, E.; Kornienko, A.; Volokitina, A.; Vilejshikova, E., et al., Judd-Ofelt modelling and stimulated-emission cross-sections for Tb³⁺ ions in monoclinic KYb(WO₄)₂ crystal. *J Lumin* 2017, *190*, 37-44.

2. Calculation for CCT and CRI

Based on the obtained color coordinates, the CCT value can be calculated by using the McCamy's approximate[1]

$$CCT = -437n^3 + 3601n^2 - 6861n + 5514.31$$
 (S7)

where $n = \frac{x - x_e}{y - y_e}$. (x, y) is the chromaticity coordinates. (x_e, y_e) is the coordinate of the epicenter and

its value is (0.332, 0.186).

Moreover, general color rending index (CRI) is the average of eight special color rending indices.[2]

$$CRI = \frac{1}{8} \sum_{i=1}^{8} R_i = \frac{1}{8} \sum_{i=1}^{8} (100 - 4.6\Delta E_i)$$
(S8)

where R_i is the special color rendering index of a certain color sample *i*, ΔE_i is the corresponding color difference between the sample under the reference light source and the test light source.

[1] McCamy, C. S., Correlated color temperature as an explicit function of chromaticity coordinates. *Color Res Appl* **1992**, *17* (2), 142-144.

[2] Jin, X.; Xie, Y.; Tang, R.; Geng, X.; Che, J.; Lou, L., et al., Novel double perovskite Sr₃WO₆: Sm³⁺, Na⁺ orange-red emitting phosphors with high thermal stability for white LEDs. *J Alloy Compd* **2022**, *899*, 162739.