Supporting information

Self-supported Porous NiMo Electrocatalyst to Boost the Catalytic Activity of Hydrogen Evolution Reaction

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Fig. S1. Optical images of (a) NF array (b) Ni/NF array, and NiMo/Ni/NF array. The NiMo/Ni/NF array is darker than the NF array.



Fig. S2. EDS analysis spectra of NiMo/Ni/NF.



Fig. S3. SEM images of NiMo/NF.



Fig. S4. Surface wettability of the NiMo/Ni/NF (a,b) and the NF (c,d).



Fig. S5. The HER LSV curves of NiMo/Ni/NF in different (a) concentration of Ni and Mo (mM) (b) electrodeposition current density, (c) electrodeposition temperature time, (d) electrodeposition temperature, and (e) electrodeposition pH.



Fig. S6. Cyclic voltammograms (CVs) at different scan rates of the prepared samples. a NiMo/Ni/NF, b NiMo/NF, c Ni/NF, d NF.



Fig. S7. HER polarization curves for NiMo/NF, NiMo/NF, and NF, normalized by electrical surface area (ECSA).



Fig. S9. Faradaic efficiency measurement of NiMo/Ni/NF.



Fig. S11. Stability test plots.



Fig. S12. Comparison of the voltage @ 10 mA/cm² of recently reported in various

electrocatalysts.



Fig. S13. Corresponding EDX mapping images of the NiMo/Ni/NF after the HER stability test for 70 h.



Fig. S14. XRD spectra comparison of fresh NiMo/Ni/NF electrocatalyst and NiMo/Ni/NF electrocatalysts after the HER stability tests.

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Catalyst	Electrolyte	Overpotential	Tafel slope	Reference
		(mV)	(mV	
		at 10 mA cm-2		
			dec-1)	
NiMo/Ni/NF	1.0 M KOH	42	44	This work
N-NiMoS		68	86	1
Ni3S ₂ @MoS ₂ /FeOOH		95	85	2
MoS ₂ /NiOOH		73	75	3
NiMo-EDA		72	89	4
MoS ₂ /Ni ₃ S ₂		110	83	5
NiMoV		49	48.3	6
A-MoS ₂ -Ni ₃ S ₂ -NF		95	107.5	7
Mo-W-S-2@Ni ₃ S ₂		98	92	8
Ni-Mo NGTS		65	67	9
NiCo ₂ S ₄ /Ni		65	84.5	10

 Table S1. Comparison of the potentials at 10 mA cm⁻² with recently reported HER catalysts

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