

Lanthanide Complexes with an Azo-dye Chromophore Ligand: Syntheses, Crystal Structures, and Near-infrared Luminescence by Long-wavelength Excitation

Yun-long Chen^{a,b}, Min Feng^{b,c*}, Xiaofei Zhu^{a*}, Zhiping Zheng^{b,c*}

a School of Chemistry and Life Science, Changchun University of Technology, Changchun, 130012, China

b Department of Chemistry, Southern University of Science and Technology, Shenzhen, 518055, China

c Key University Laboratory of Rare Earth Chemistry of Guangdong, Southern University of Science and Technology, Shenzhen, Guangdong 518055.

E-mail addresses: fengm@sustech.edu.cn (M. Feng), zhuxiaofei@ccut.edu.cn (X. Zhu), zhengzp@sustech.edu.cn (Z. Zheng)

Contents

Table S1. X-ray Crystallographic Data for 1-3.	3
Table S2. X-ray Crystallographic Data for 4-6.	4
Table S3. Selected bonds lengths [Å] for 1-3.5	
Table S4. Selected bonds lengths [Å] for 4-6.5	
Table S5. Coordination geometry of complexes 1-6.	5
Table S6. TD-DFT calculated excitation energies and main compositions of the low-lying electronic transitions for H ₂ C2R ²⁻ .	6
Table S7. TD-DFT calculated excitation energies and main compositions of the low-lying electronic transitions for [Lu ₂ (HC2R) ₂ (DMSO) ₁₀].	7
Figure S1. The molecular structures of complexes 1 (a) and 9 (b) with the main planes of their ligands. All H atoms are omitted except the H on N. Color code: Cyan (Er), Red (O), Blue (N), Yellow (S), Gray (C), White (H).	8
Figure S2. Powder X-ray diffraction patterns of complexes 1-3 and 4-6 (b).	8
Figure S3. The molecular orbitals of H ₂ C2R ²⁻ .	9
Figure S4. The S1, S3, S4, S8, S10, S11 ($f > 0.03$) hole and electron distributions of H ₂ C2R ²⁻ . Pink: hole, cyan: electron.	10
Figure S5. The molecular orbitals of [Lu ₂ (HC2R) ₂ (DMSO) ₁₀].	11
Figure S6. The S1, S5, S11, S19, S21, S34 ($f > 0.01$) Hole and electron distributions of [Lu ₂ (HC2R) ₂ (DMSO) ₁₀]. Pink: hole, cyan: electron.	12
Figure S7. UV-vis absorption spectra of ligand Na ₂ H ₂ C2R and complexes 1-6 in solid state.	13
Figure S8. Energy level diagram for Ln complexes sensitized via a ligand-centered triplet excited state.	13
Figure S9. Emission spectra of 1 and 4.	14
Figure S10. NIR luminescence decay curve of complex 4 at 980 nm.	14
Figure S11. NIR emission (a) and excitation spectra (b) of 2 and 5.	15
Figure S12. Emission spectra of 1 under $\lambda_{ex} = 465\text{--}751$ nm.	15
Figure S13. Emission spectra of 1 under $\lambda_{ex} = 346\text{--}746$ nm.	16
Figure S14. Phosphorescent spectrum of complex 3 at room temperature.	16
Figure S15. The absolute QY measurement of 4 ($\lambda_{ex} = 659$ nm). Results via FLS1000 – Integrating Sphere	17

Table S1. X-ray Crystallographic Data for **1-3**.

complex	1	2	3
Empirical formula	$C_{32}H_{54}Yb_2N_4O_{34}S_4$	$C_{32}H_{54}Er_2N_4O_{34}S_4$	$C_{32}H_{54}Gd_2N_4O_{34}S_4$
Formula weight	1513.11	1501.55	1481.53
Temperature/K	100.02	100.05	99.99
Crystal system	monoclinic	monoclinic	monoclinic
Space group	$P2_1/n$	$P2_1/n$	$P2_1/n$
a/ \AA	15.1083(9)	15.1450(6)	15.2177(10)
b/ \AA	7.4003(5)	7.4186(3)	7.4551(4)
c/ \AA	23.3340(14)	23.3610(8)	23.3788(13)
$\alpha/^\circ$	90	90	90
$\beta/^\circ$	107.382(3)	107.3690(10)	107.394(2)
$\gamma/^\circ$	90	90	90
Volume/ \AA^3	2489.7(3)	2505.04(17)	2531.0(3)
Z	2	2	2
$\rho_{\text{calc}} \text{g/cm}^3$	2.018	1.991	1.944
μ/mm^{-1}	4.009	3.602	2.869
F(000)	1500	1492	1476
Radiation	MoK α ($\lambda = 0.71073$)	MoK α ($\lambda = 0.71073$)	MoK α ($\lambda = 0.71073$)
Reflections collected	25613	30812	20817
Independent reflections	6421 [$R_{\text{int}} = 0.0953$, $R_{\text{sigma}} = 0.0920$]	5775 [$R_{\text{int}} = 0.0638$, $R_{\text{sigma}} = 0.0452$]	5213 [$R_{\text{int}} = 0.0747$, $R_{\text{sigma}} = 0.0665$]
Data/restraints/parameters	6421/1/360	5775/0/360	5213/1/360
Goodness-of-fit on F^2	1.007	1.04	1.042
Final R indexes [$ I >= 2\sigma(I)$]	$R_I = 0.0455$, $wR_2 = 0.0794$	$R_I = 0.0301$, $wR_2 = 0.0663$	$R_I = 0.0395$, $wR_2 = 0.0864$
Final R indexes [all data]	$R_I = 0.0850$, $wR_2 = 0.0936$	$R_I = 0.0301$, $wR_2 = 0.0663$	$R_I = 0.0538$, $wR_2 = 0.0929$
Largest diff. peak/hole / e \AA^{-3}	1.25/-1.55	1.19/-1.16	1.34/-1.06
CCDC number	2303845	2303847	2303852

Table S2. X-ray Crystallographic Data for **4-6**.

complex	4	5	6
Empirical formula	C ₅₂ H ₉₀ Yb ₂ N ₄ O ₂₈ S ₁₆	C ₅₂ H ₉₀ Er ₂ N ₄ O ₂₈ S ₁₆	C ₅₂ H ₉₀ Gd ₂ N ₄ O ₂₈ S ₁₆
Formula weight	2126.35	2114.79	2094.77
Temperature/K	100	100	100.02
Crystal system	triclinic	triclinic	triclinic
Space group	P-1	P-1	P-1
a/Å	11.2546(8)	11.2494(6)	11.2789(8)
b/Å	14.4042(9)	14.5051(7)	14.5397(11)
c/Å	15.0132(9)	15.0460(8)	15.0454(11)
α/°	113.027(3)	113.056(3)	113.012(3)
β/°	90.449(4)	90.302(3)	90.218(3)
γ/°	111.806(4)	111.853(3)	111.884(2)
Volume/Å ³	2046.0(2)	2063.62(19)	2074.8(3)
Z	1	1	1
ρ _{calc} g/cm ³	1.726	1.702	1.677
μ/mm ⁻¹	10.4	9.526	2.063
F(000)	1074	1070	1062
Radiation	GaKα (λ = 1.34139)	GaKα (λ = 1.34139)	MoKα (λ = 0.71073)
Reflections collected	51458 9428	24195 7841	26624 9543
Independent reflections	[R _{int} = 0.0857, R _{sigma} = 0.0624]	[R _{int} = 0.0747, R _{sigma} = 0.0895]	[R _{int} = 0.0622, R _{sigma} = 0.0755]
Data/restraints/parameters	9428/0/452	7841/0/452	9543/0/452
Goodness-of-fit on F ²	1.105	1.064	1.016
Final R indexes [I>=2σ (I)]	R ₁ = 0.0425, wR ₂ = 0.1046	R ₁ = 0.0504, wR ₂ = 0.1158	R ₁ = 0.0393, wR ₂ = 0.0772
Final R indexes [all data]	R ₁ = 0.0533, wR ₂ = 0.1078	R ₁ = 0.0668, wR ₂ = 0.1211	R ₁ = 0.0572, wR ₂ = 0.0824
Largest diff. peak/hole / e Å ⁻³	0.79/-1.53	0.78/-1.41	1.81/-0.96
CCDC number	2303867	23038468	2303869

Table S3. Selected bonds lengths [Å] for **1-3**.

	1	2	3
Atom	Length	Length	Length
Ln1-O1	2.242(4)	2.269(3)	2.313(3)
Ln1-O2	2.207(4)	2.232(3)	2.274(3)
Ln1-O4 ¹	2.427(4)	2.426(3)	2.454(3)
Ln1-O1W	2.337(4)	2.362(3)	2.416(4)
Ln1-O2W	2.302(4)	2.320(3)	2.367(4)
Ln1-O3W	2.397(4)	2.417(3)	2.459(3)
Ln1-O4W	2.293(4)	2.320(3)	2.377(4)
Ln1-O5W	2.393(4)	2.421(3)	2.459(4)

O4¹ is the oxygen on an adjacent ligand (HC2R).**Table S4.** Selected bonds lengths [Å] for **4-6**.

	4	5	6
Atom	Length	Length	Length
Ln-O1	2.270(3)	2.301(4)	2.353(3)
Ln-O2	2.285(3)	2.318(4)	2.360(3)
Ln-O8 ¹	2.324(3)	2.352(4)	2.391(3)
Ln-O9	2.279(3)	2.315(4)	2.363(3)
Ln-O10	2.423(3)	2.390(3)	2.467(3)
Ln-O11	2.374(3)	2.434(4)	2.423(3)
Ln-O12	2.269(3)	2.306(4)	2.342(3)
Ln-O13	2.282(3)	2.281(4)	2.352(3)

O8¹ is the oxygen on an adjacent ligand (HC2R).**Table S5.** Coordination geometry of complexes **1-6**.

Eight-Coordinated Metal centers

	O _h CU	D _{4d} SAPR	D _{2d} TDD	C _{2v} JBTPR	C _{2v} BTTPR	D _{2d} JSD	T _d TT
1	9.735	2.402	0.376	2.337	2.139	2.485	10.438
2	9.82	2.461	0.413	2.422	2.188	2.558	10.525
3	9.799	2.534	0.447	2.56	2.277	2.697	10.546
4	10.213	3.066	0.374	2.75	2.176	2.288	10.838
5	10.161	3.002	0.36	2.804	2.139	2.321	10.803
6	10.058	2.977	0.375	2.877	2.143	2.484	10.754

CU: Cube; SAPR: Square antiprism; TDD: Triangular dodecahedron; JBTPR: Biaugmented trigonal prism J50; BTTPR: Biaugmented trigonal prism; JSD: Snub diphenoïd J84; TT: Triakis tetrahedron.

HPY: Hexagonal pyramid; PBPY: Pentagonal bipyramid; COC: Capped octahedron; CTPR: Capped trigonal prism; JPBPY: Johnson pentagonal bipyramid J13; JETPY: Johnson elongated triangular pyramid J7.

Table S6. TD-DFT calculated excitation energies and main compositions of the low-lying electronic transitions for H₂C2R²⁻.

Excited						
	E (eV)	E (nm)	f	Tansition	Contribution	Assignment
state						
1	2.5834	480	0.6944	HOMO → LUMO	98.70%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO-1 → LUMO	72.70%	$\pi \rightarrow \pi^*$
3	3.3932	365	0.3519	HOMO-3 → LUMO	14.80%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO-2 → LUMO	9.50%	$\pi \rightarrow \pi^*$
4	3.4636	358	0.0348	HOMO-2 → LUMO	89.60%	$n \rightarrow \pi^*$
				HOMO-1 → LUMO	6.40%	$\pi \rightarrow \pi^* n \rightarrow \pi^*$
8	4.2243	294	0.1751	HOMO → LUMO+1	80.30%	$\pi \rightarrow \pi^*$
				HOMO-7 → LUMO	13.70%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO-7 → LUMO	71.40%	$\pi \rightarrow \pi^*$
10	4.4906	276	0.1389	HOMO → LUMO+1	9.10%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO → LUMO+2	6.60%	$\pi \rightarrow \pi^*$
11	4.5563	272	0.1485	HOMO → LUMO+2	76.50%	$\pi \rightarrow \pi^*$
				HOMO-7 → LUMO	5.30%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO → LUMO+3	62.50%	$\pi \rightarrow \pi^*$
14	4.788	259	0.1068	HOMO-10 → LUMO	11.70%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO → LUMO+2	5.00%	$\pi \rightarrow \pi^*$
				HOMO-1 → LUMO+1	49.90%	$\pi \rightarrow \pi^*$
23	5.4006	230	0.5475	HOMO-3 → LUMO+1	26.10%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
				HOMO → LUMO+3	6.80%	$\pi \rightarrow \pi^*$

Table S7. TD-DFT calculated excitation energies and main compositions of the low-lying electronic transitions for $[\text{Lu}_2(\text{HC}2\text{R})_2(\text{DMSO})_{10}]$.

Excited state	E (eV)	E (nm) f	Transition	Contribution	Assignment
1	1.4963 829	0.0172	HOMO → LUMO	50.70%	n → π^*
			HOMO-1 → LUMO+1	48.70%	n → π^*
5	2.0984 591	0.3951	HOMO-2 → LUMO	47.40%	$\pi \rightarrow \pi^*$
			HOMO-3 → LUMO+1	44.10%	$\pi \rightarrow \pi^*$
11	2.6485 468	0.5621	HOMO-4 → LUMO	35.20%	$\pi \rightarrow \pi^*$
			HOMO-5 → LUMO+1	34.70%	$\pi \rightarrow \pi^*$
19	3.3305 372	0.0164	HOMO-6 → LUMO	11.30%	$\pi, \sigma \rightarrow \pi^* \pi \rightarrow \pi^*$
			HOMO-7 → LUMO+1	9.60%	$\pi, \sigma \rightarrow \pi^*$
21	3.357 369	0.0711	HOMO-14 → LUMO	24.60%	$\sigma, \pi \rightarrow \pi^*$
			HOMO-15 → LUMO+1	22.10%	$\sigma, \pi \rightarrow \pi^*$
34	3.6237 342	0.3428	HOMO-12 → LUMO	12.80%	$\pi \rightarrow \pi^* \sigma, \pi \rightarrow \pi^*$
			HOMO-10 → LUMO	8.50%	$\pi \rightarrow \pi^*$
21	3.357 369	0.0711	HOMO-19 → LUMO+1	8.00%	$\sigma, \pi \rightarrow \pi^*$
			HOMO-11 → LUMO+1	7.00%	$\pi \rightarrow \pi^*$
34	3.6237 342	0.3428	HOMO-10 → LUMO	41.70%	$\pi \rightarrow \pi^*$
			HOMO-11 → LUMO+1	38.10%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
34	3.6237 342	0.3428	HOMO-12 → LUMO	6.50%	$\pi \rightarrow \pi^*$
			HOMO-13 → LUMO+1	39.40%	$\pi \rightarrow \pi^*$
34	3.6237 342	0.3428	HOMO-12 → LUMO	26.60%	$\pi \rightarrow \pi^* \pi \rightarrow \pi^*$
			HOMO-14 → LUMO	16.30%	$\sigma, \pi \rightarrow \pi^*$

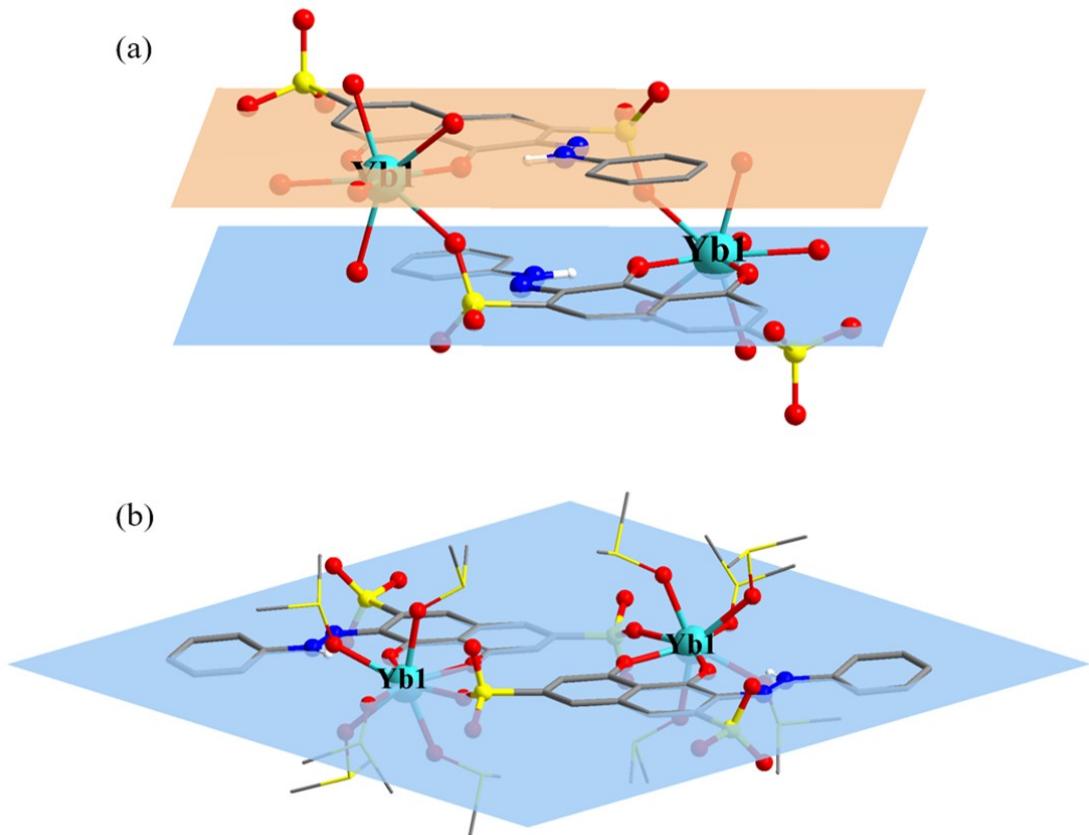


Figure S1. The molecular structures of complexes **1** (a) and **9** (b) with the main planes of their ligands. All H atoms are omitted except the H on N. Color code: Cyan (Er), Red (O), Blue (N), Yellow (S), Gray (C), White (H).

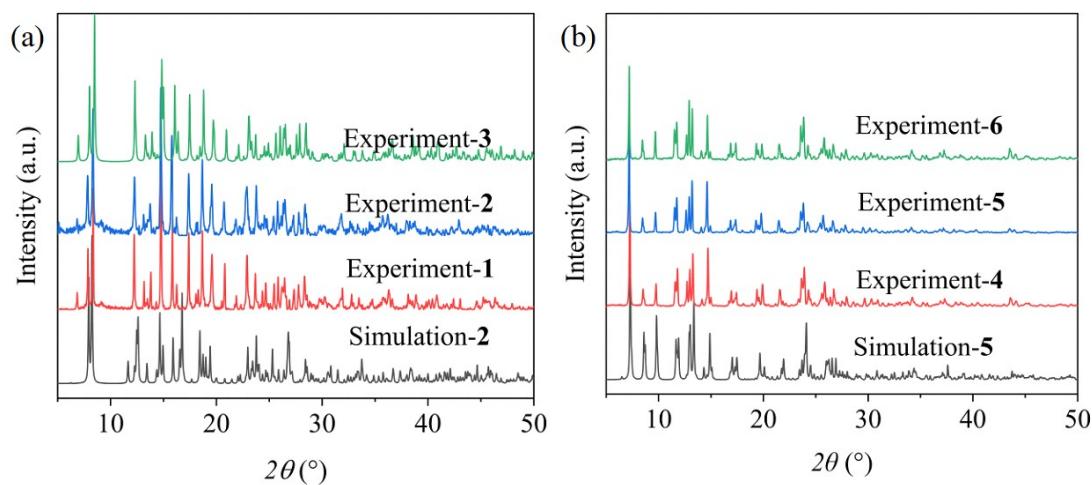


Figure S2. Powder X-ray diffraction patterns of complexes **1-3** and **4-6** (b).

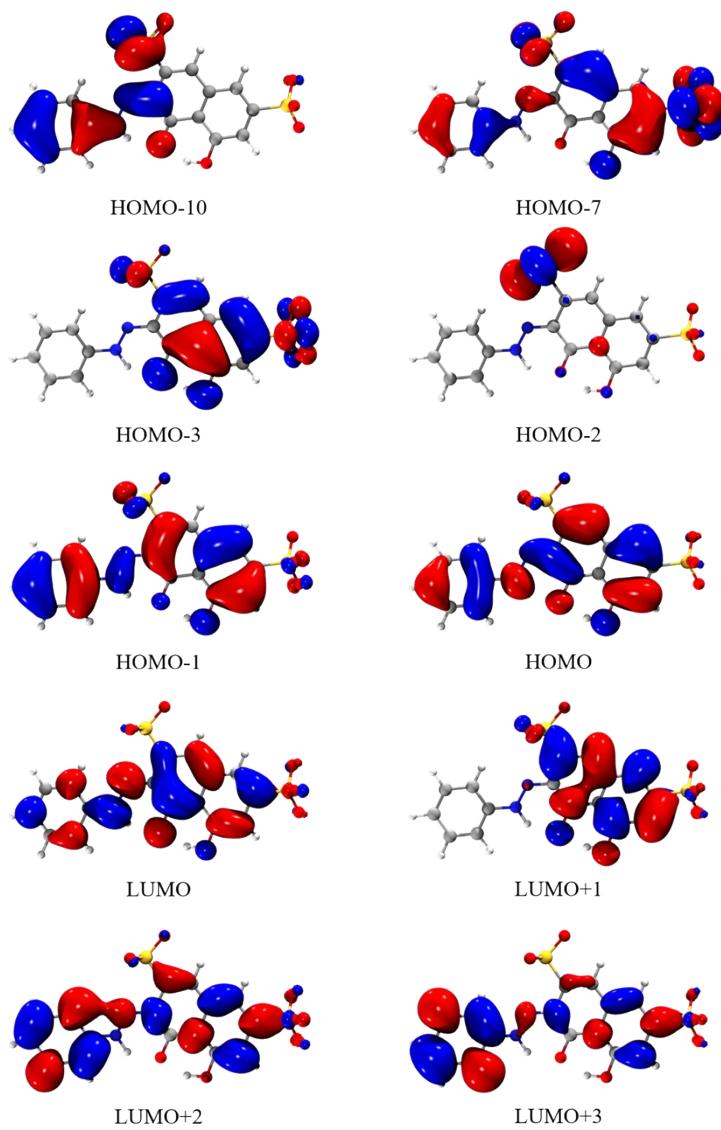


Figure S3. The molecular orbitals of $\text{H}_2\text{C}_2\text{R}^{2-}$.

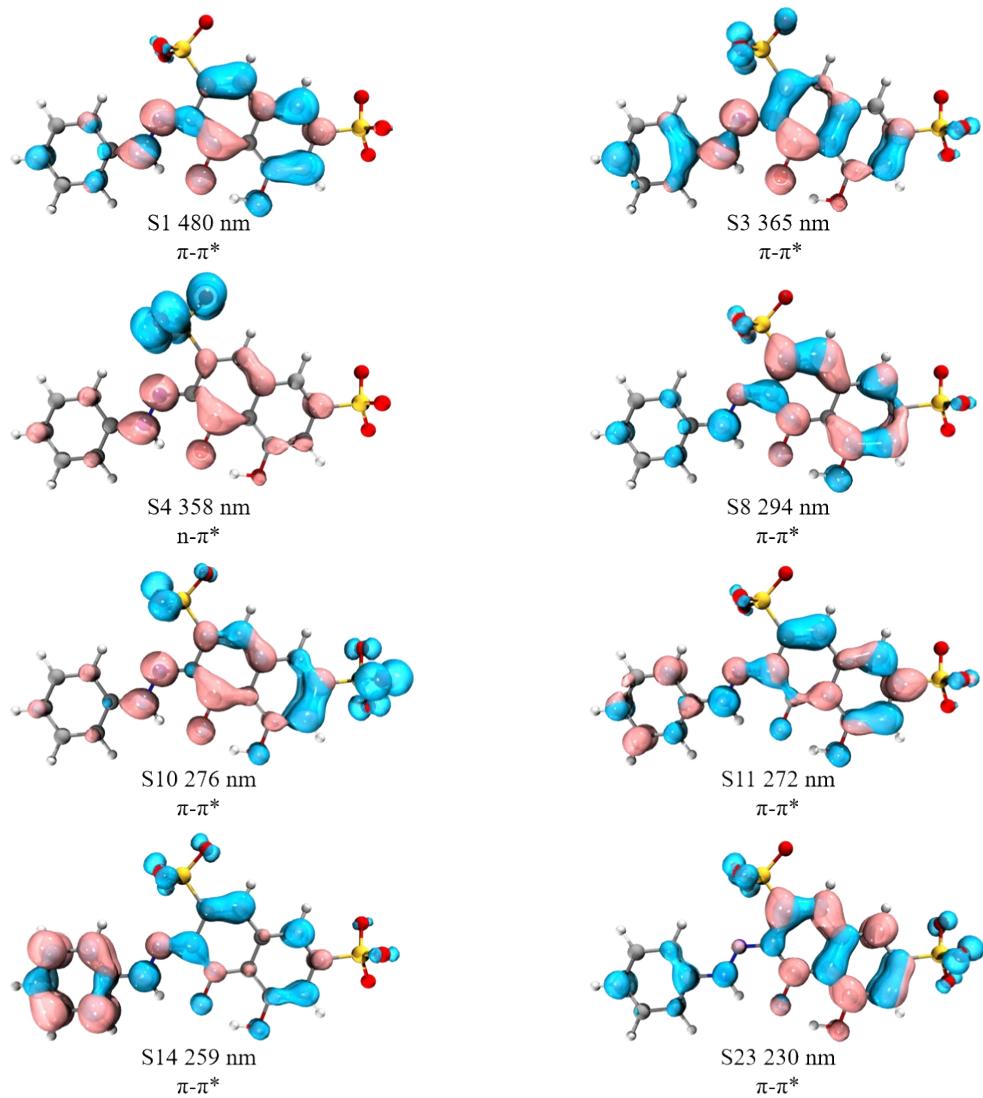


Figure S4. The S1, S3, S4, S8, S10, S11 ($f > 0.03$) hole and electron distributions of $\text{H}_2\text{C}_2\text{R}^{2-}$. Pink: hole, cyan: electron.

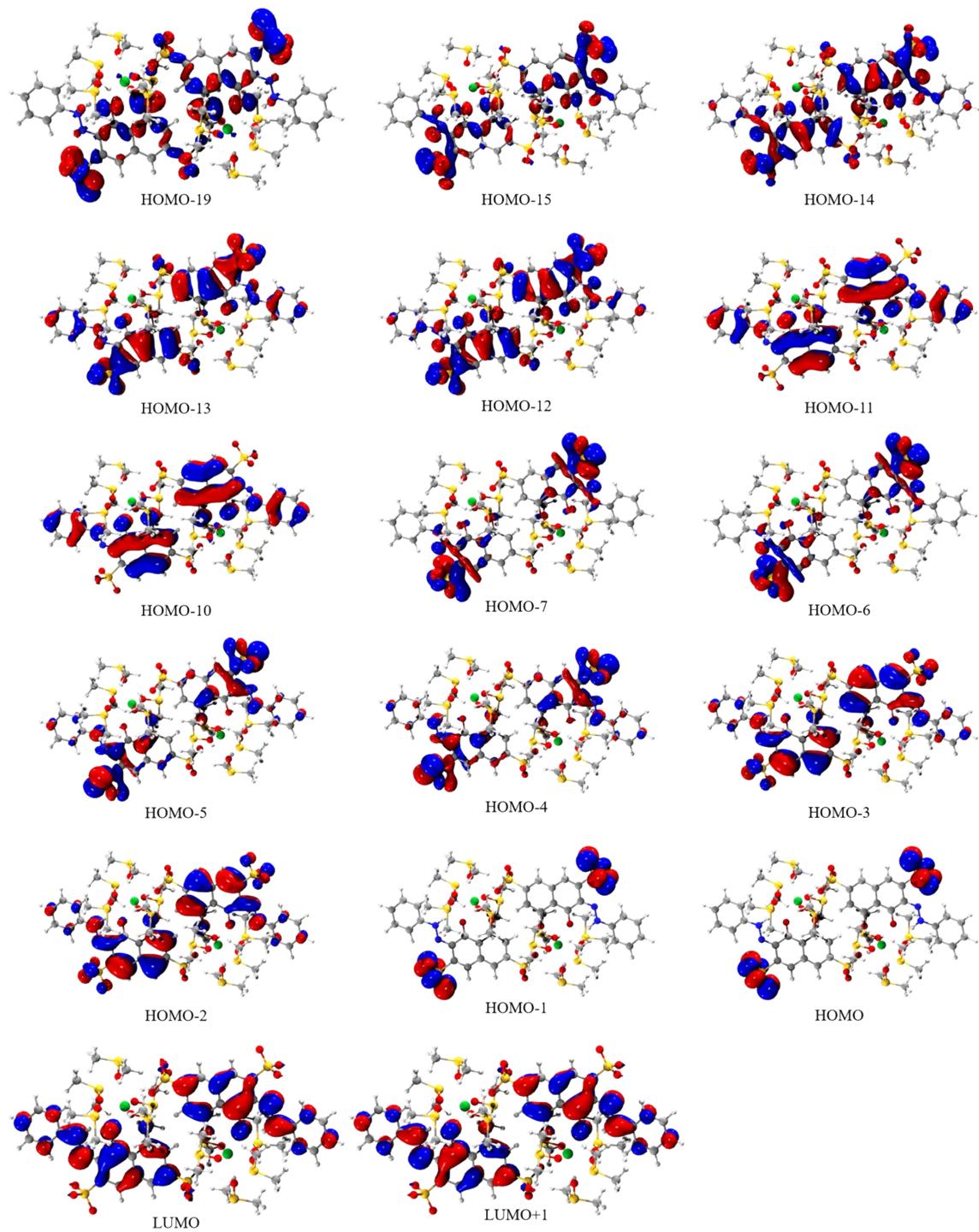


Figure S5. The molecular orbitals of $[\text{Lu}_2(\text{HC}_2\text{R})_2(\text{DMSO})_{10}]$.

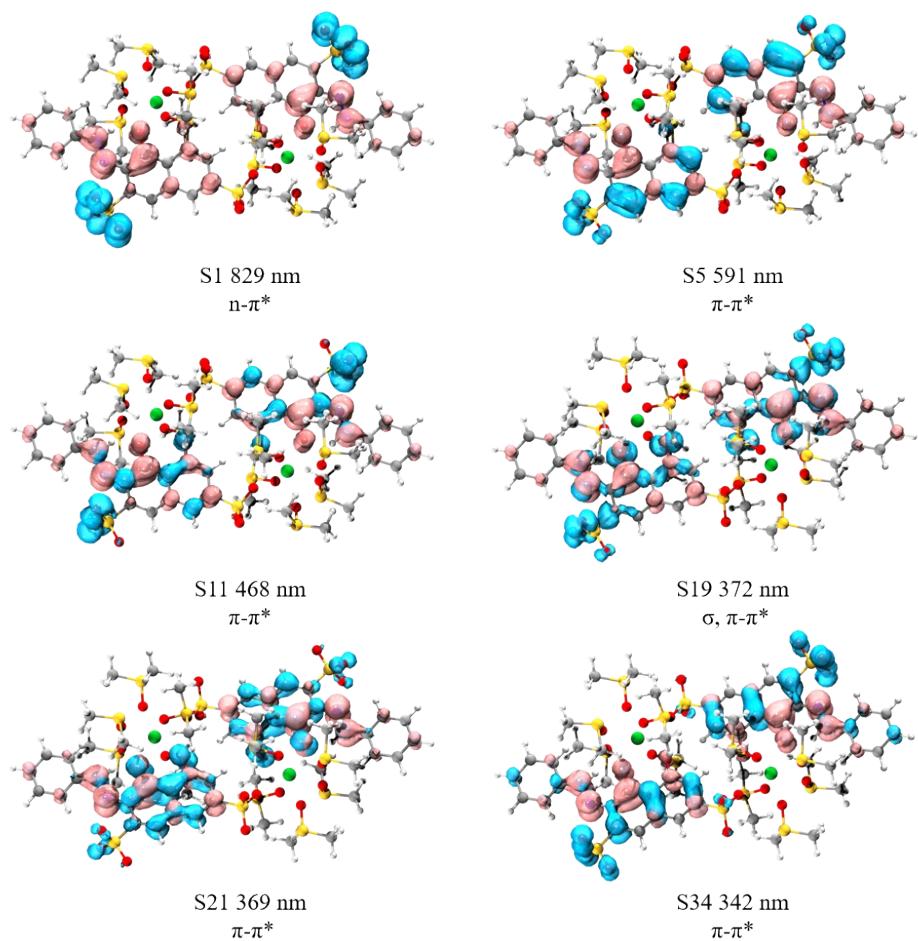


Figure S6. The S1, S5, S11, S19, S21, S34 ($f > 0.01$) Hole and electron distributions of $[\text{Lu}_2(\text{HC}_2\text{R})_2(\text{DMSO})_{10}]$. Pink: hole, cyan: electron.

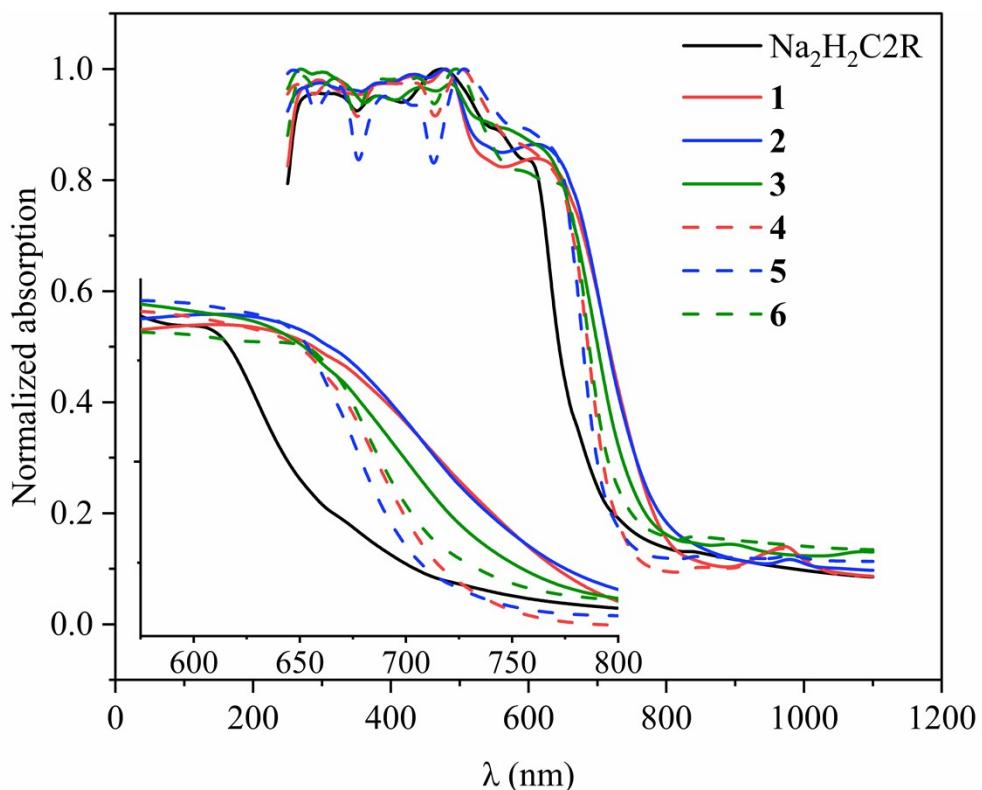


Figure S7. UV-vis absorption spectra of ligand $\text{Na}_2\text{H}_2\text{C}_2\text{R}$ and complexes **1**-**6** in solid state.

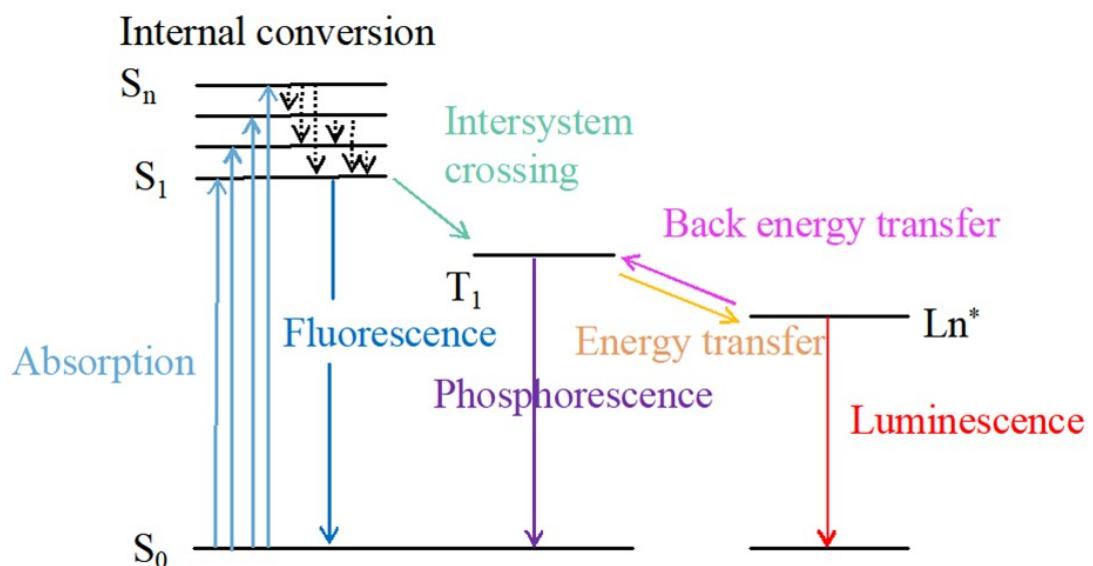


Figure S8. Energy level diagram for Ln complexes sensitized via a ligand-centered triplet excited state.

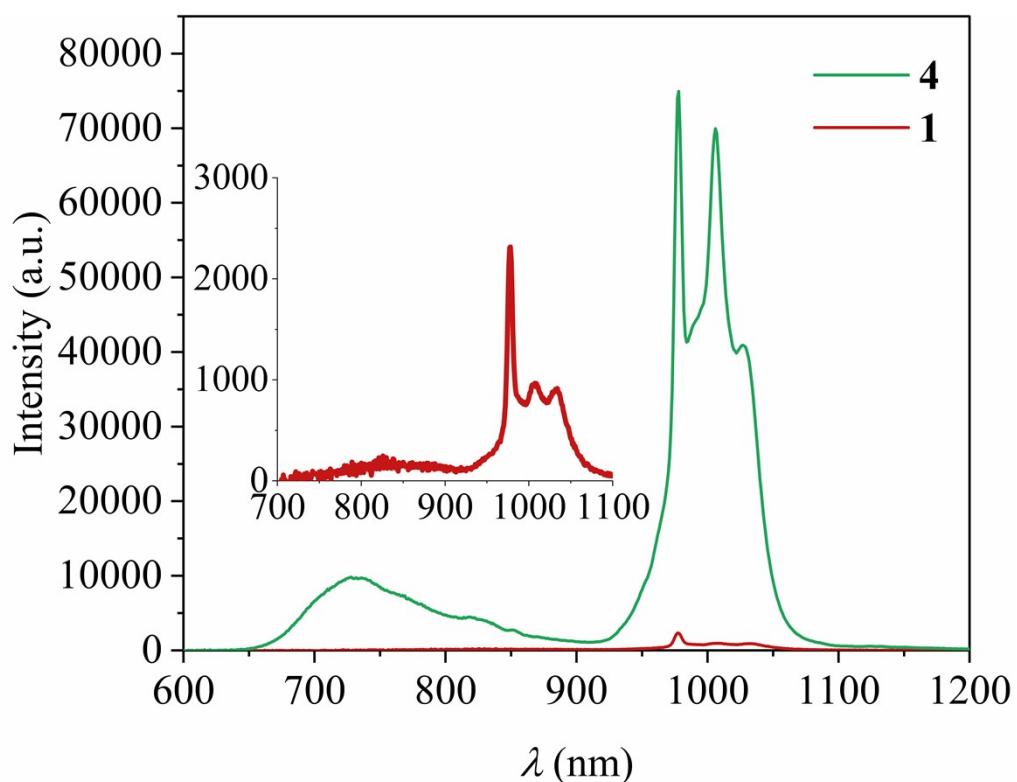


Figure S9. Emission spectra of **1** and **4**.

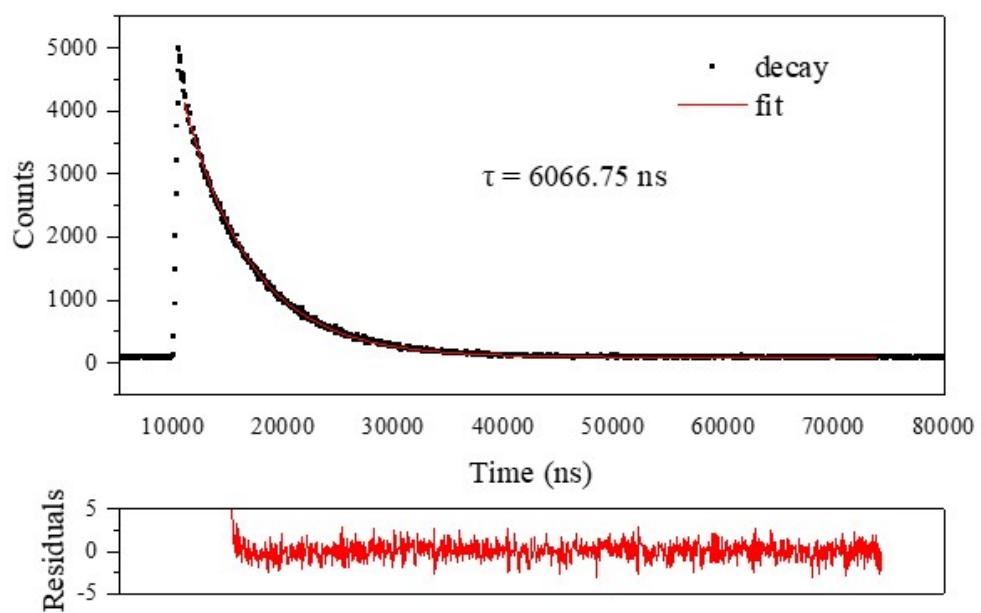


Figure S10. NIR luminescence decay curve of complex **4** at 980 nm.

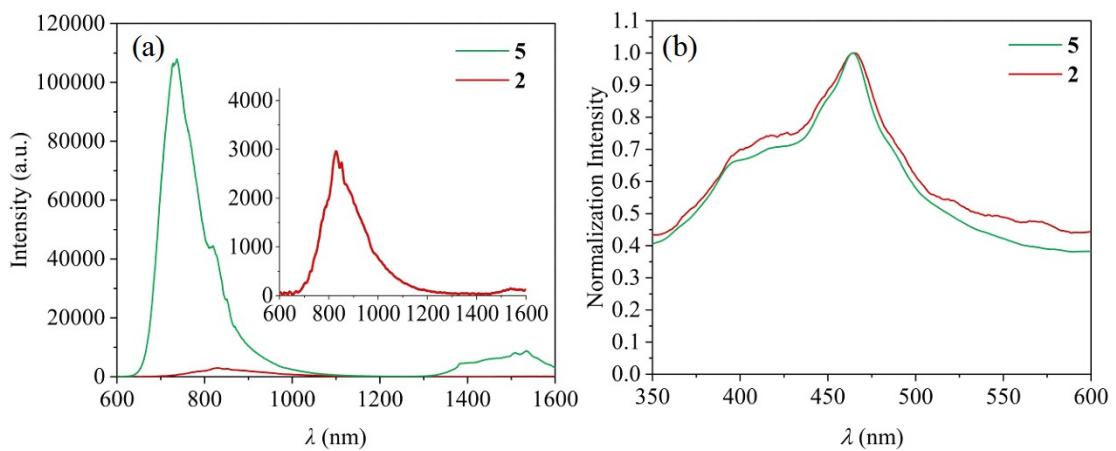


Figure S11. NIR emission (a) and excitation spectra (b) of **2** and **5**.

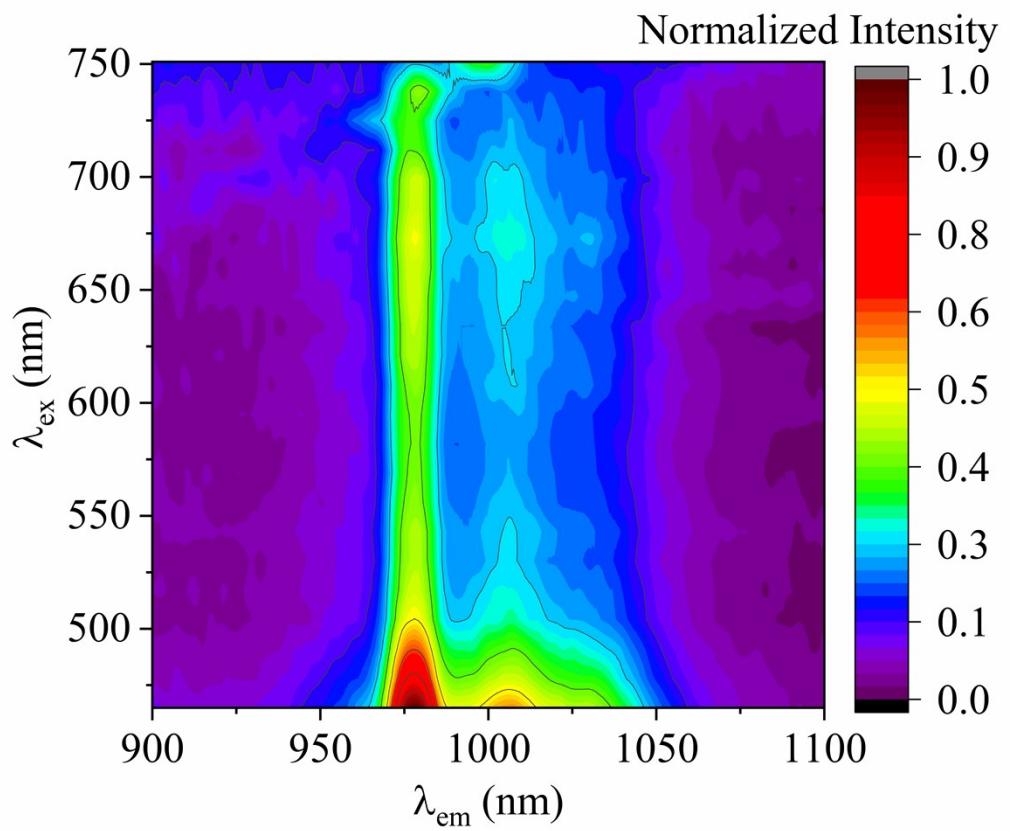


Figure S12. Emission spectra of **1** under $\lambda_{\text{ex}} = 465\text{--}751$ nm.

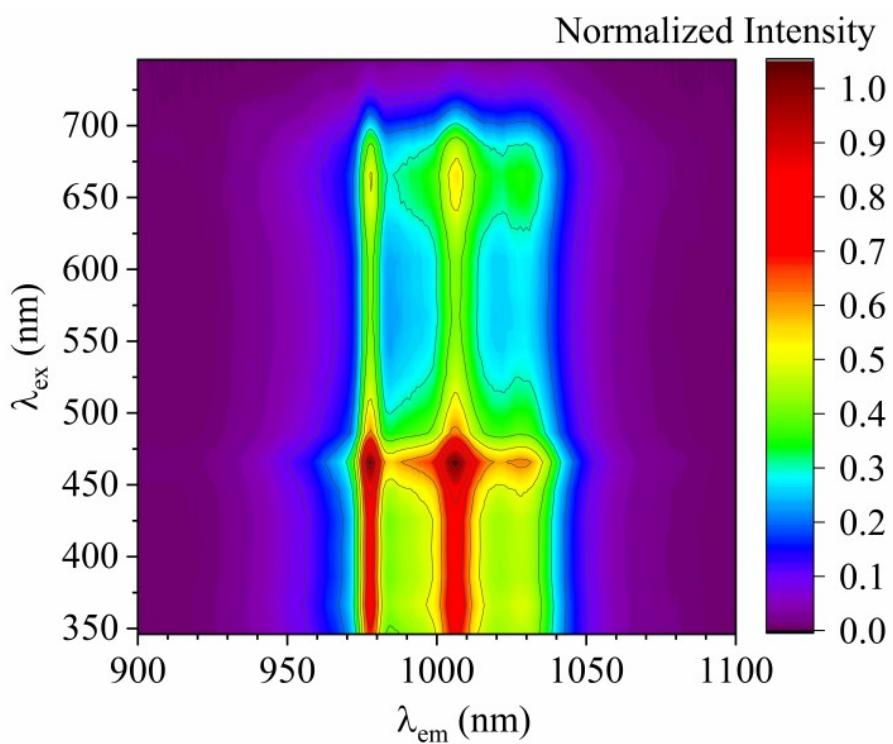


Figure S13. Emission spectra of **1** under $\lambda_{\text{ex}} = 346$ –746 nm.

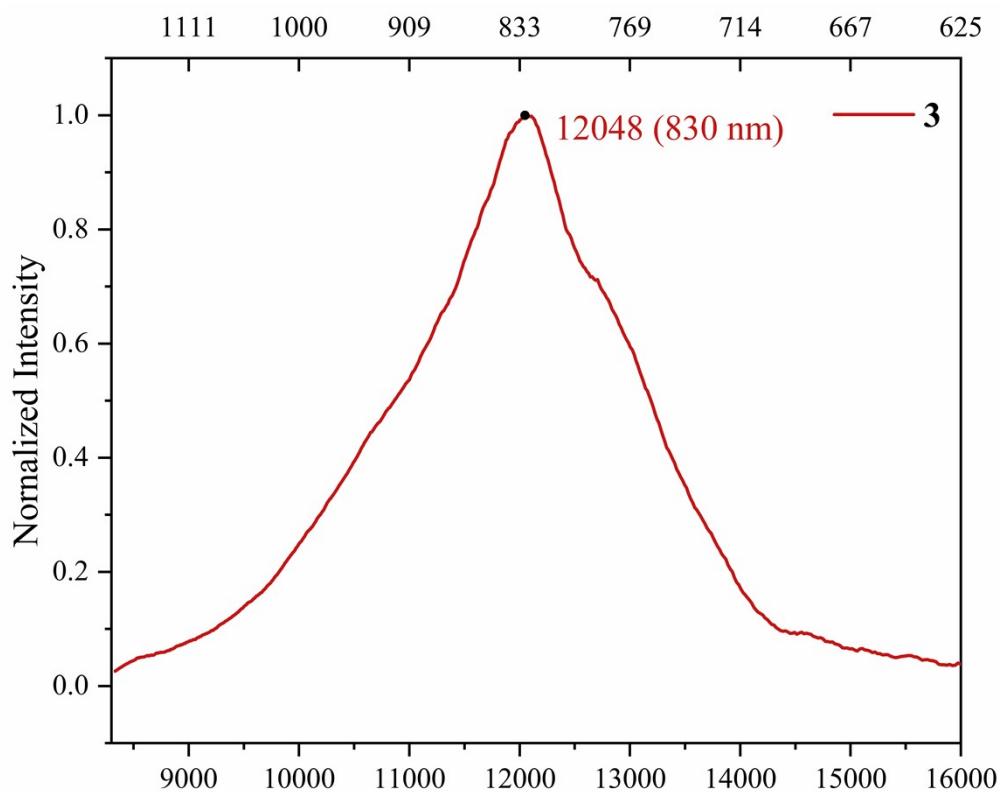


Figure S14. Phosphorescent spectrum of complex **3** at room temperature.

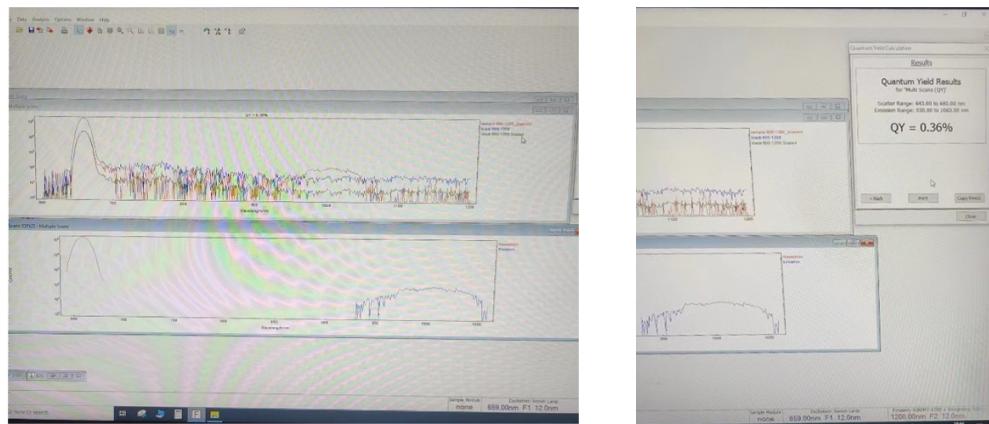


Figure S15. The absolute QY measurement of **4** ($\lambda_{\text{ex}} = 659 \text{ nm}$). Results via FLS1000 – Integrating Sphere