

Supporting information

Factors Governing Protonation of Keggin-type Polyoxometalates: Influence of Core Structure in Clusters

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This file includes:

Figs. S1 to S12

Tables S1 to S5

Other Supplementary Materials for this manuscript include the following:

.xyz and .chk files (ZIP)

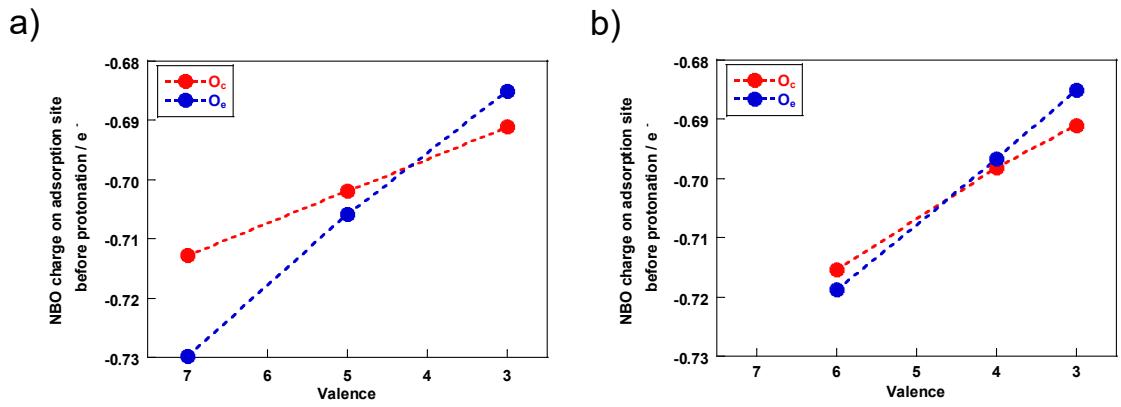


Figure S1. NBO charge on each adsorption site before protonation in α -PW12.

a) restricted DFT b) unrestricted DFT

The same trend of amount of charge change as Fig. 2 was observed in not only close shell systems but also open shell systems, indicates this trend of change of O_c and O_e occurs regardless of occupation.

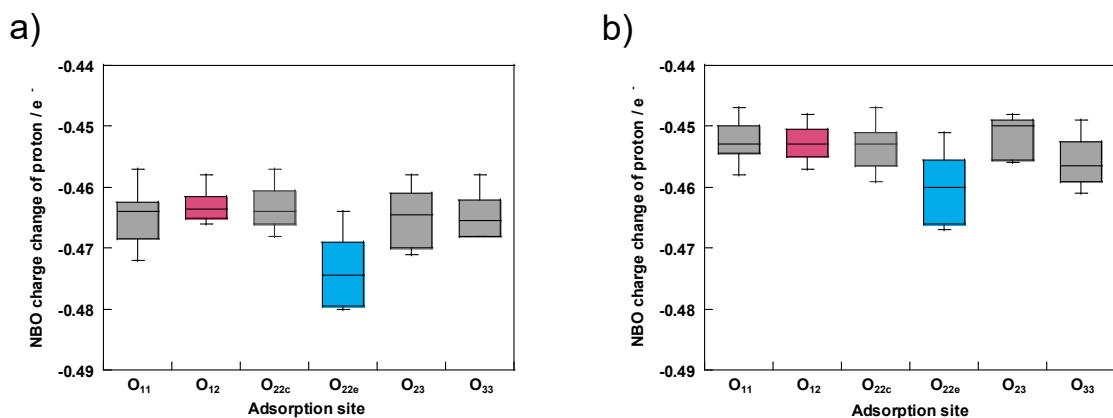


Figure S2 4 Box plot of NBO charge change of proton upon protonation in a) β -XMo12 and b) β -XW12. The upper and lower whiskers are set as the maximum and minimum values. The plots outside the whisker are outliers.

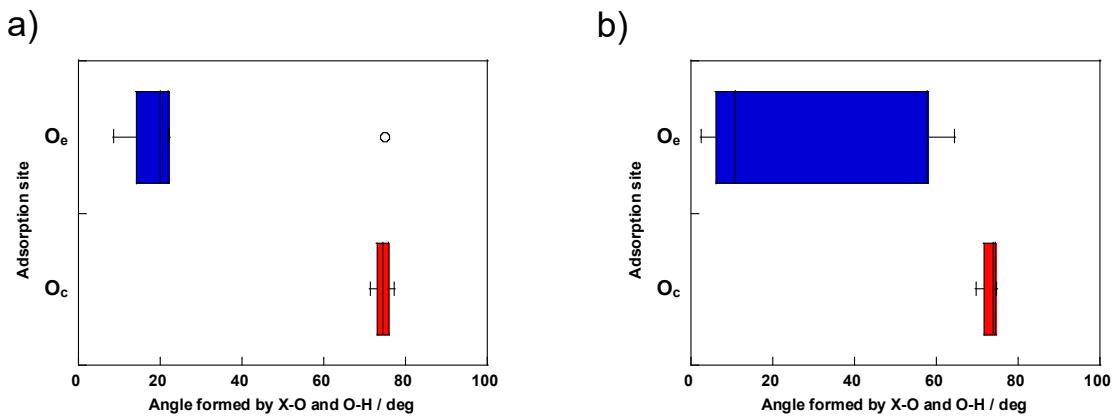


Figure S3 Box plot illustrating the relationship between adsorption site and protonation angles, angles between X-O and O-H, in a) α -XM₁₂ and b) α -XW₁₂. The upper and lower whiskers represent the maximum and minimum values. Outlying points outside the whisker are considered as outliers.

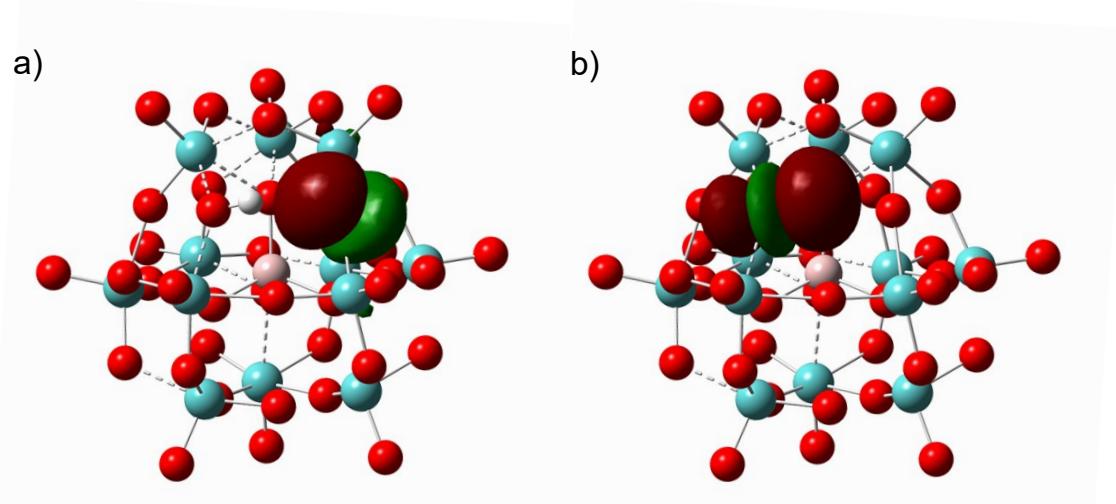


Figure S4 NBO orbitals of a) donor level and b) acceptor level for proton adsorption on O_{12} site in β -Al Mo_{12} .

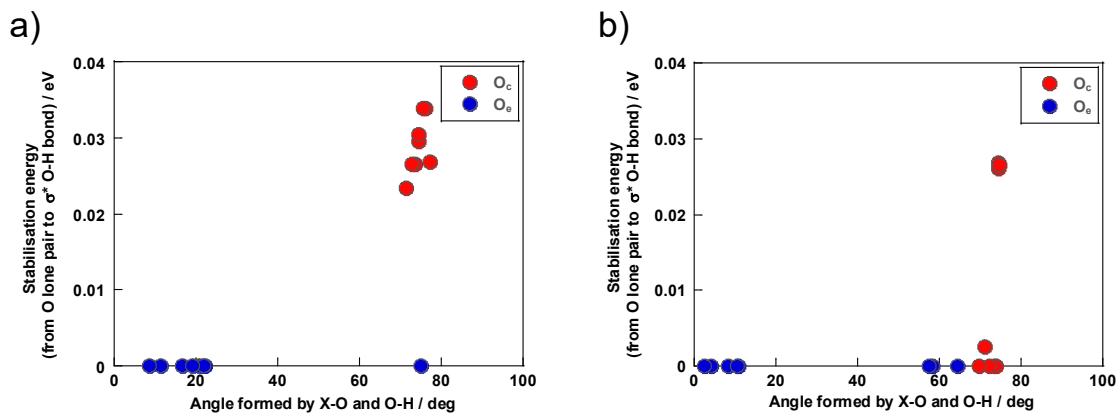
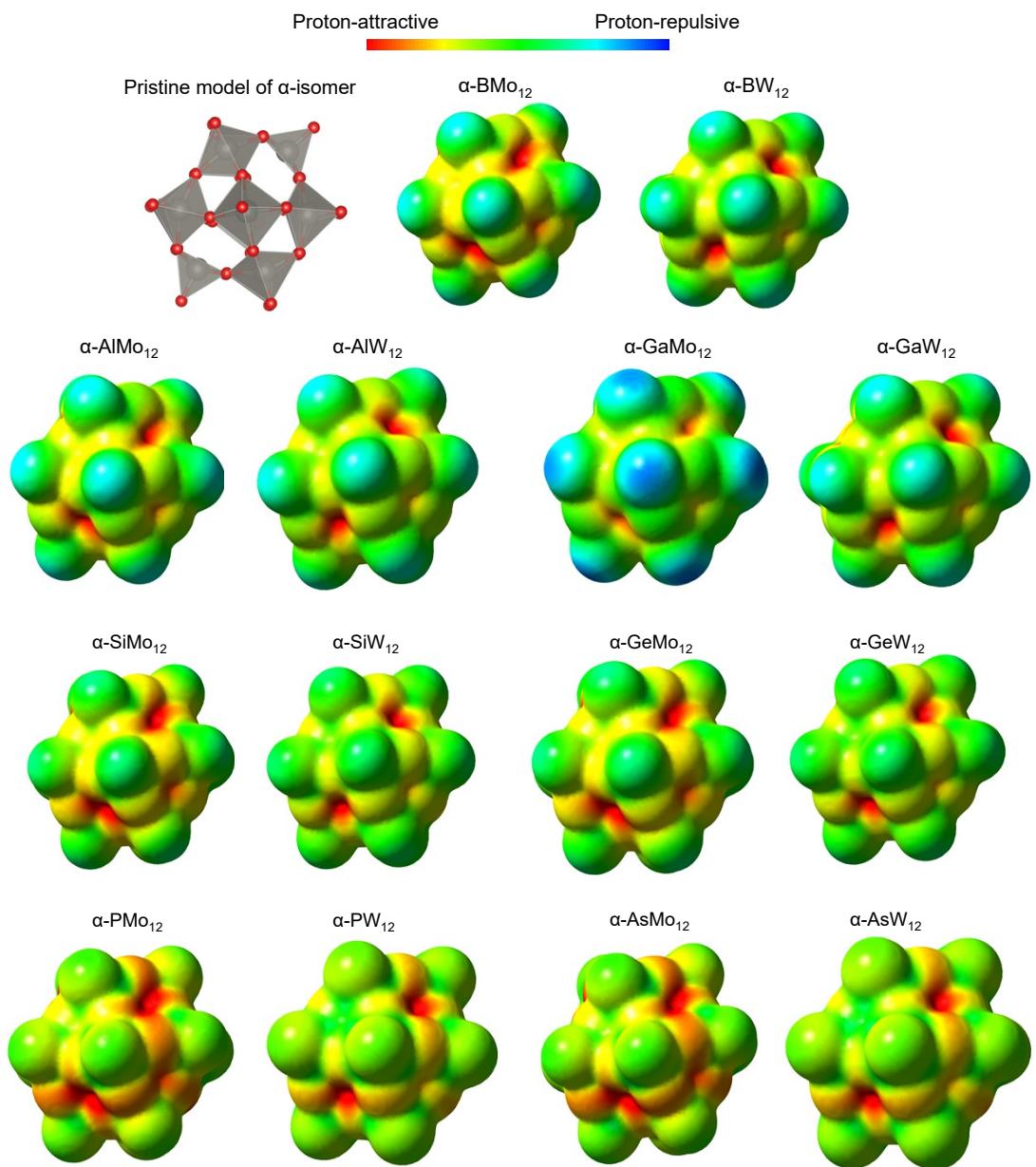
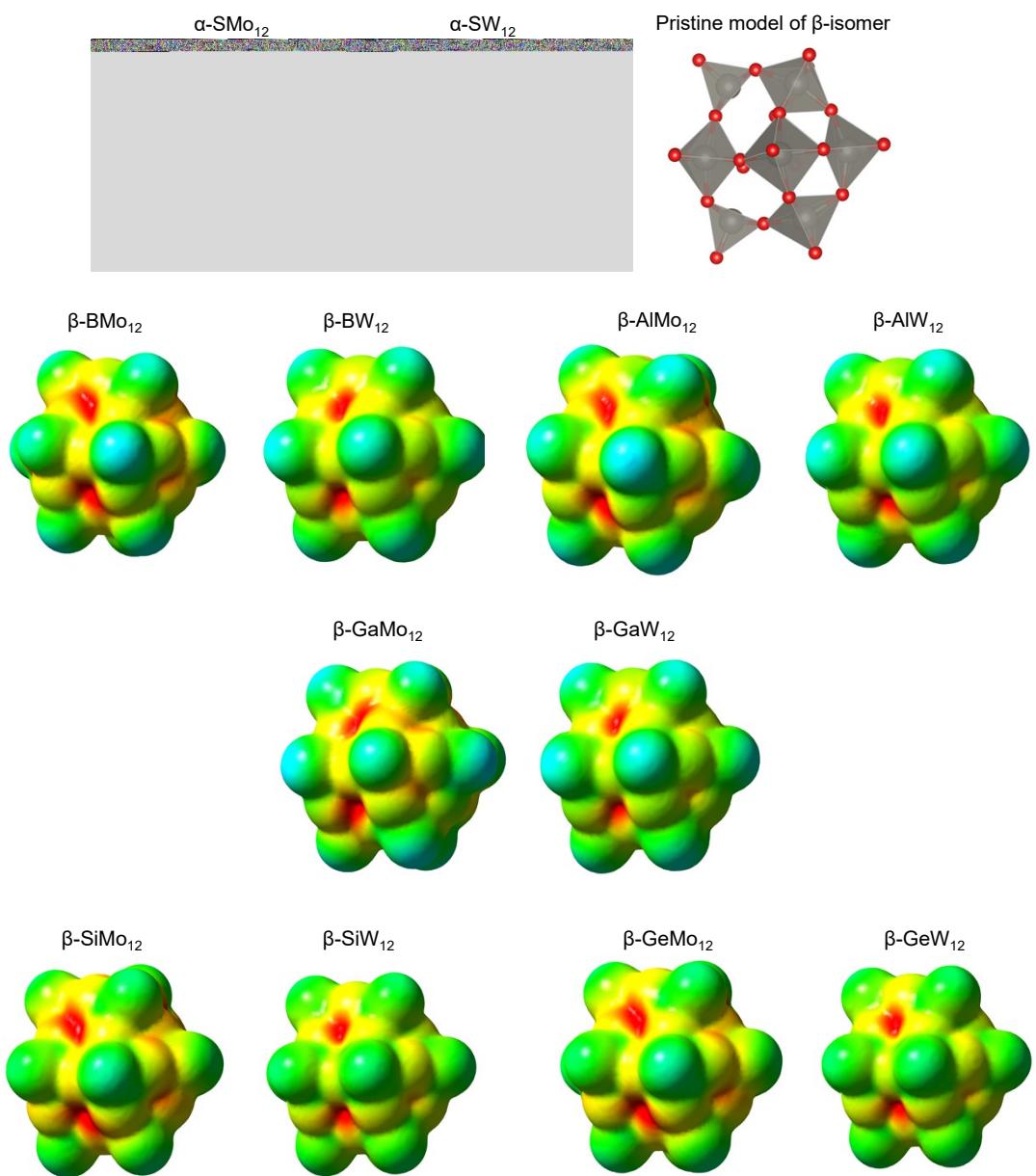


Figure S5 Relationship between stabilization energy and the protonation angles, angles between X-O and O-H, in a) $\alpha\text{-XMo}_{12}$ and b) $\alpha\text{-XW}_{12}$.





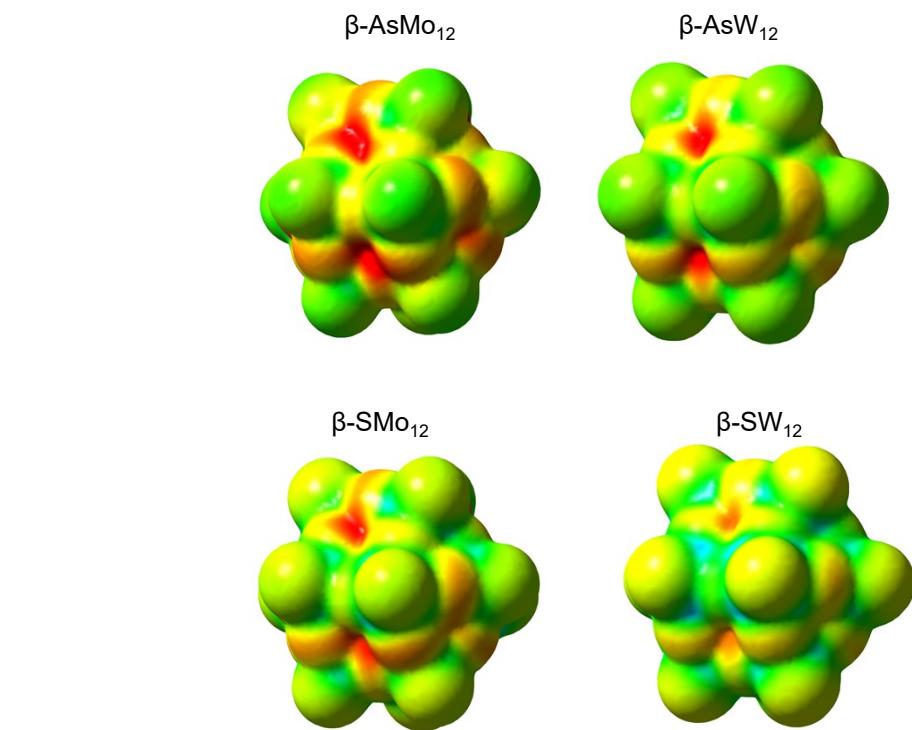


Figure S6 MEP mappings of α - and β - XM₁₂ except for β -PM₁₂.

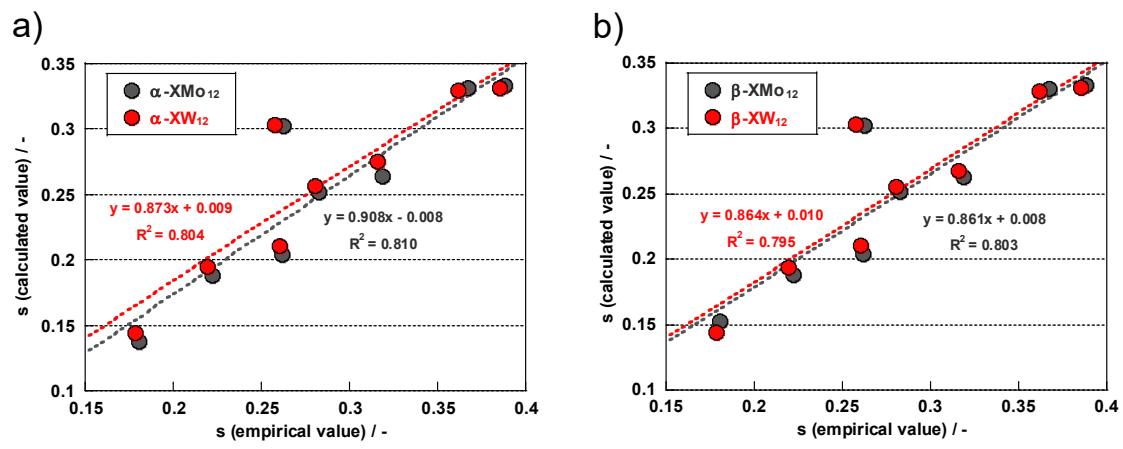


Figure S7 Correlation between empirical and calculated bond strength values on a) α - XM_{12} and b) β - XM_{12} .

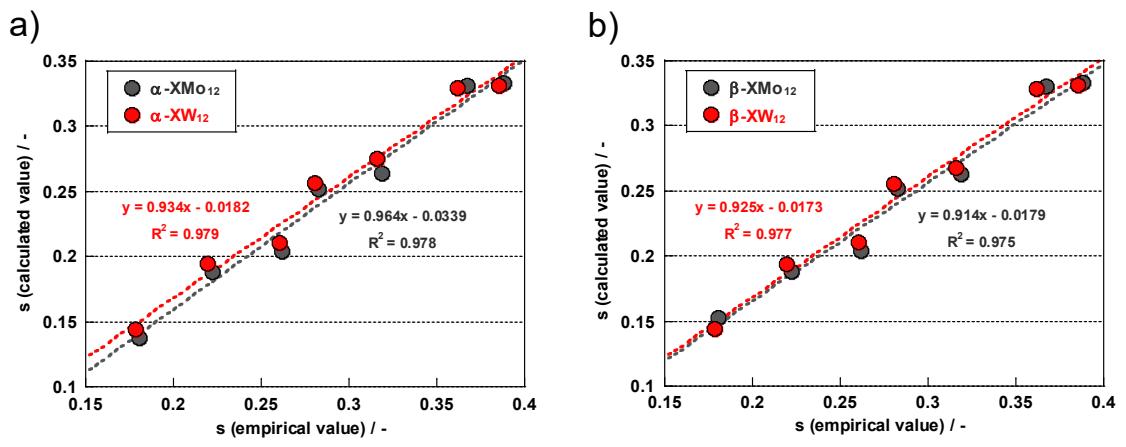


Figure S8 Correlation between empirical and calculated bond strength values except X=B a) on α -XM₁₂ and b) on β -XM₁₂.

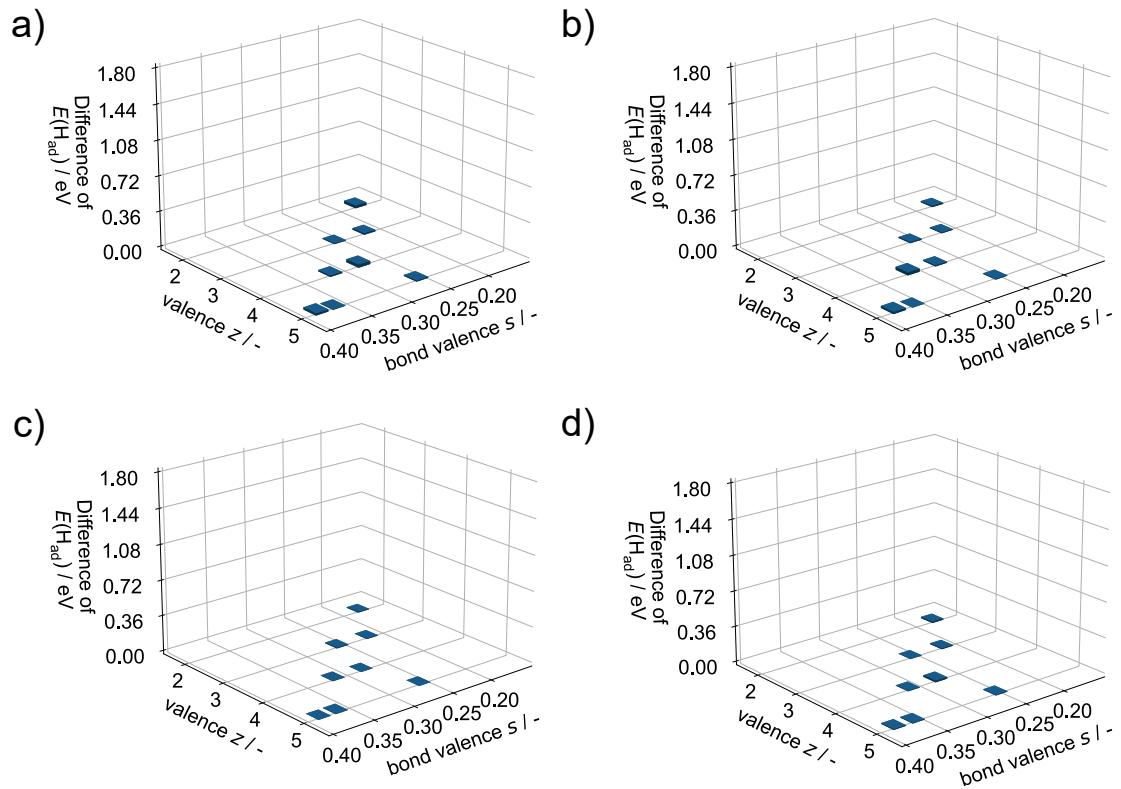


Figure S9 Absolute error of predicted protonation energy in a) $\alpha\text{-XMo}_{12}$, b) $\alpha\text{-XW}_{12}$, c) $\beta\text{-XMo}_{12}$, and d) $\beta\text{-XW}_{12}$. The width of axis corresponding to absolute error (from 0.0 to 1.8) is adjusted to the same as the drawn width of $E(H_{ad})$ in Fig. 8 (from -10.6 to -12.4).

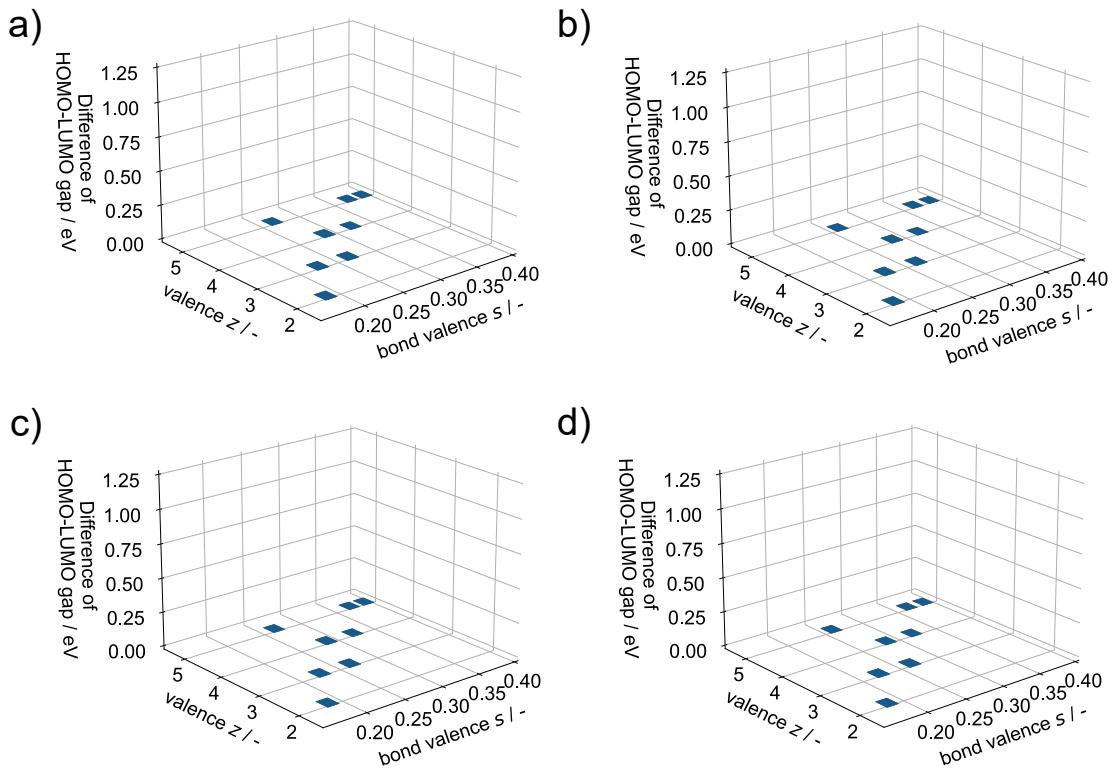


Figure S10 Absolute error of predicted HOMO-LUMO gap in a) $\alpha\text{-XMo}_{12}$, b) $\alpha\text{-XW}_{12}$, c) $\beta\text{-XMo}_{12}$, and d) $\beta\text{-XW}_{12}$. The width of axis corresponding to absolute error (from 0.0 to 1.25) is adjusted to the same as the drawn width of HOMO-LUMO gap in Fig. 9 (from 3.75 to 5.00).

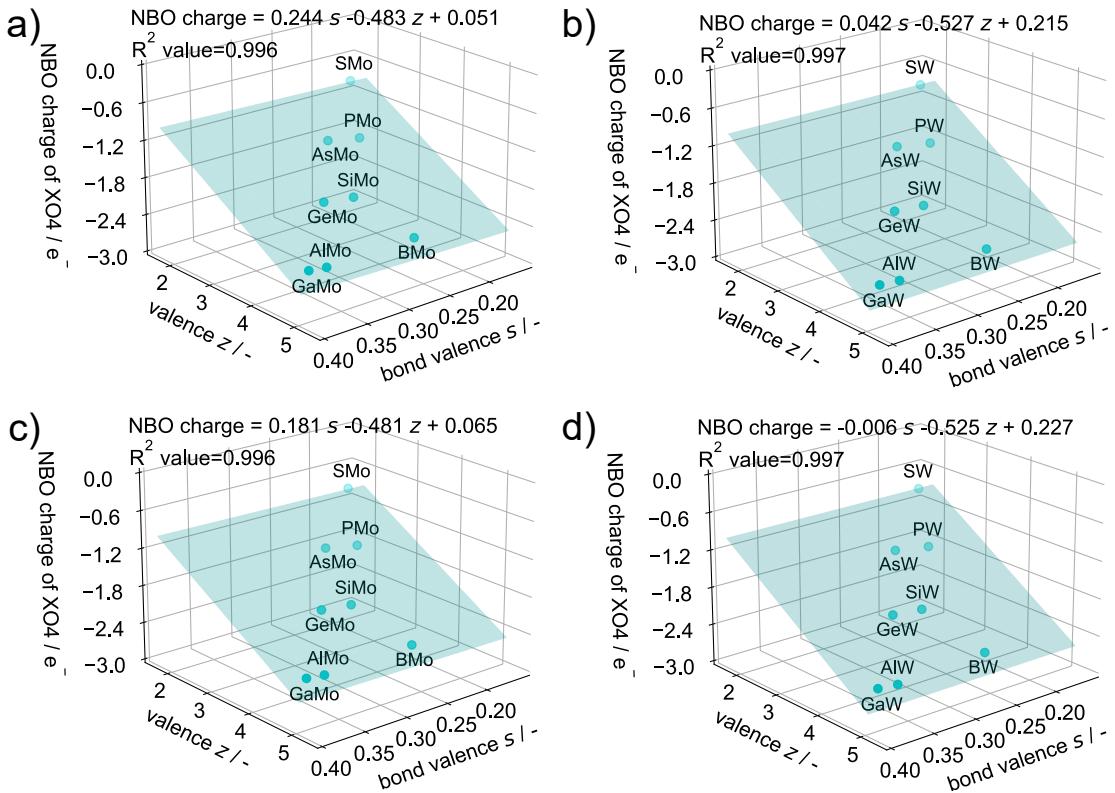


Figure S11 Regression of NBO charges of XO_4 core structure in a) $\alpha\text{-XM}_{12}$, b) $\alpha\text{-XW}_{12}$, c) $\beta\text{-XM}_{12}$, and d) $\beta\text{-XW}_{12}$.

We discuss that the valences of POM are derived from charges of XO_4 structure in main text.

Moreover, we evaluate the NBO charges of XO_4 structures. The clear relationship between the NBO charges and governing factors for various responses suggests that the governing factors strongly involve degree of electron transfer from core structure to shell structure.

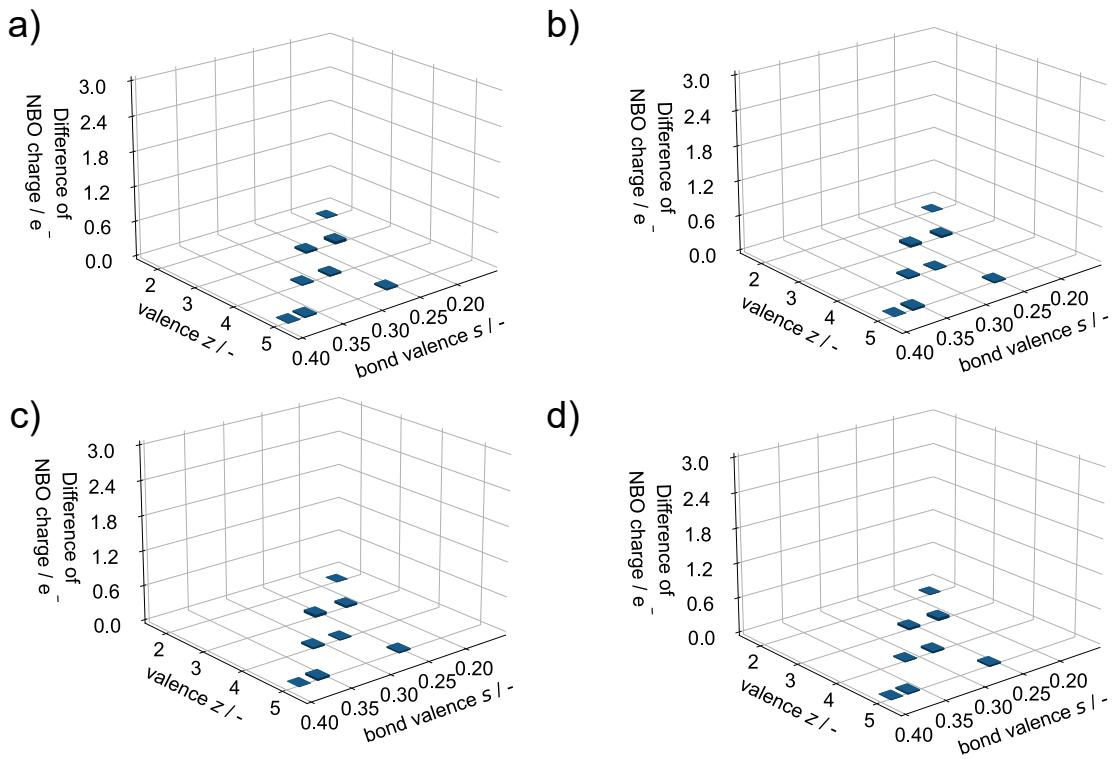


Figure S12 Absolute error of NBO charges of XO_4 core structure in a) $\alpha\text{-XMo}_{12}$, b) $\alpha\text{-XW}_{12}$, c) $\beta\text{-XMo}_{12}$, and d) $\beta\text{-XW}_{12}$. The width of axis corresponding to absolute error (from 0.0 to 3.00) is adjusted to the same as the drawn width of NBO charge of XO_4 in Fig. S11 (from -3.00 to 0.00).

Table S1. Adsorption energies of each adsorption site A) M=Mo, B) M=W.

A)

X	z	α-isomer (eV)		β-isomer (eV)					
		H ⁺ on O _c	H ⁺ on O _e	H ⁺ on O ₁₁	H ⁺ on O ₁₂	H ⁺ on O _{22e}	H ⁺ on O _{22c}	H ⁺ on O ₂₃	H ⁺ on O ₃₃
B	5	-12.074	-12.134	-12.136	-12.213	-12.212	-11.978	-12.180	-12.144
Al	5	-12.151	-12.186	-12.192	-12.267	-12.219	-12.059	-12.232	-12.157
Ga	5	-12.178	-12.214	-12.217	-12.273	-12.226	-12.073	-12.254	-12.158
Si	4	-11.795	-11.796	-11.798	-11.927	-11.879	-11.679	-11.826	-11.847
Ge	4	-11.824	-11.834	-11.833	-11.947	-11.891	-11.708	-11.859	-11.857
P	3	-11.471	-11.423	-11.501	-11.598	-11.532	-11.340	-11.544	-11.545
As	3	-11.500	-11.465	-11.487	-11.615	-11.546	-11.370	-11.489	-11.556
S	2	-11.185	-11.128	-11.211	-11.293	-11.206	-11.034	-11.238	-11.256

B)

X	z	α-isomer (eV)		β-isomer (eV)					
		H ⁺ on O _c	H ⁺ on O _e	H ⁺ on O ₁₁	H ⁺ on O ₁₂	H ⁺ on O _{22e}	H ⁺ on O _{22c}	H ⁺ on O ₂₃	H ⁺ on O ₃₃
B	5	-11.961	-11.972	-11.958	-12.009	-12.039	-11.726	-11.955	-11.961
Al	5	-12.017	-12.056	-12.043	-12.088	-12.097	-11.832	-12.051	-12.026
Ga	5	-12.034	-12.095	-12.073	-12.102	-12.111	-11.852	-12.089	-12.037
Si	4	-11.650	-11.664	-11.649	-11.708	-11.680	-11.431	-11.635	-11.675
Ge	4	-11.671	-11.675	-11.679	-11.740	-11.726	-11.470	-11.685	-11.698
P	3	-11.303	-11.284	-11.302	-11.360	-11.320	-11.068	-11.288	-11.335
As	3	-11.357	-11.354	-11.317	-11.389	-11.349	-11.110	-11.324	-11.359
S	2	-10.979	-10.957	-10.969	-11.038	-10.991	-10.742	-10.987	-11.017

Table S2. Most favorable proton adsorption site using M06/Lanl2DZ level.

X	z	α-isomer		β-isomer	
		M=Mo	M=W	M=Mo	M=W
B	5	O _e	O _e	O _{22e}	O ₂₃
Al	5	O _e	O _e	O _{22e}	O ₂₃
Ga	5	O _e	O _e	O _{22e}	O ₂₃
Si	4	O _e	O _e	O _{22e}	O _{22e}
Ge	4	O _e	O _e	O _{22e}	O _{22e}
P	3	O _e	O _e	O _{22e}	O ₁₂
As	3	O _e	O _e	O _{22e}	O ₁₂
S	2	O _e	O _e	O _{22e}	O ₁₂

Table S3. HOMO-LUMO gap comparison of some Keggin POM.

Composition	Experimental value / eV	Calculated value / eV	Reported value / eV	reference
$\alpha\text{-PW}_{12}$	3.5	4.7	2.7	[Gao 2015] [Hiskia 2001]
$\alpha\text{-SiW}_{12}$	4.4	4.3	2.8	[Hiskia 2001]
$\alpha\text{-PMo}_{12}$	3.9	4.8	-	[Hiskia 2001]

The reported values were referred to [López 2006].

Table S4. List of calculated HOMO-LUMO gaps

Composition	α -isomer / eV	β -isomer / eV
BMo ₁₂	3.91	3.80
AlMo ₁₂	4.59	4.51
GaMo ₁₂	4.61	4.53
SiMo ₁₂	4.40	4.32
GeMo ₁₂	4.43	4.36
PMo ₁₂	4.26	4.16
AsMo ₁₂	4.28	4.20
SMo ₁₂	4.15	4.02
BW ₁₂	4.30	4.07
AlW ₁₂	4.89	4.65
GaW ₁₂	4.88	4.66
SiW ₁₂	4.79	4.51
GeW ₁₂	4.78	4.54
PW ₁₂	4.70	4.45
AsW ₁₂	4.68	4.44
SW ₁₂	4.64	4.39

Table S5. Structural properties of α - and β -XM12 prior to proton adsorption

α -isomer												β -isomer											
Model	$d(X-O_1)$ (Å)	$d(O_1-M)$ (Å)	$d(M-O_2)$ (Å)	$d(M-O_3)$ (Å)	$d(X-O_4)$ (Å)	$d(O_4-M)$ (Å)	$d(X-O_5)$ (Å)	$d(O_5-M)$ (Å)	$d(M-O_6)$ (Å)	$d(M-O_{11})$ (Å)	$d(M-O_{12})$ (Å)	$d(M-O_{21})$ (Å)	$d(M-O_{22})$ (Å)	$d(M-O_{23})$ (Å)	$d(M-O_{24})$ (Å)	$d(M-O_{25})$ (Å)	$d(M-O_{26})$ (Å)	$d(M-O_{27})$ (Å)	$d(M-O_{28})$ (Å)	$d(M-O_{29})$ (Å)	$d(M-O_{30})$ (Å)		
BMo	2.402	1.945	1.927	1.710	1.539	2.396	1.535	2.401	1.950	1.710	1.934	1.932	1.917	1.959	1.711	1.943	1.948	1.925	1.709				
AlMo	1.779	2.257	1.966	1.949	1.712	1.783	2.282	1.839	2.263	1.975	1.711	1.984	1.951	1.945	1.952	1.712	1.963	1.966	1.948	1.711			
GaMo	1.834	2.257	1.972	1.953	1.712	1.837	2.280	1.864	2.372	1.951	1.700	1.945	1.943	1.932	1.957	1.701	1.968	1.971	1.951	1.700			
SiMo	1.860	2.374	1.951	1.938	1.700	1.665	2.380	1.754	2.337	1.754	1.701	1.951	1.950	1.940	1.966	1.702	1.957	1.960	1.944	1.700			
GeMo	1.750	2.330	1.969	1.946	1.701	1.753	2.337	1.754	2.334	1.951	1.700	1.951	1.951	1.950	1.966	1.702	1.957	1.960	1.944	1.700			
PtMo	1.563	2.464	1.943	1.934	1.680	1.688	2.472	1.588	2.458	1.571	1.686	1.943	1.941	1.940	1.936	1.680	1.943	1.944	1.933	1.680			
AsMo	1.683	2.403	1.952	1.942	1.691	1.685	2.411	1.686	2.398	1.551	1.681	1.949	1.948	1.935	1.953	1.682	1.945	1.943	1.931	1.683			
SiMo	1.503	2.540	1.941	1.933	1.682	1.504	2.549	1.506	2.536	1.538	1.682	1.939	1.937	1.929	1.936	1.683	1.945	1.943	1.931	1.683			
BW	1.528	1.918	1.934	1.918	1.727	1.534	2.414	1.534	2.413	1.934	1.727	1.925	1.921	1.910	1.936	1.728	1.935	1.937	1.917	1.727			
AlW	1.767	2.293	1.982	1.937	1.729	1.822	2.275	1.825	2.269	1.951	1.729	1.940	1.938	1.933	1.952	1.729	1.948	1.952	1.936	1.728			
GaW	1.821	2.270	1.987	1.942	1.729	1.822	2.275	1.825	2.275	1.951	1.729	1.940	1.938	1.933	1.952	1.729	1.952	1.956	1.934	1.728			
SiW	1.656	2.387	1.941	1.929	1.718	1.657	2.393	1.660	2.383	1.939	1.718	1.940	1.937	1.932	1.944	1.719	1.942	1.943	1.928	1.718			
GeW	1.742	2.343	1.946	1.934	1.719	1.644	2.350	1.747	2.340	1.945	1.719	1.939	1.937	1.931	1.944	1.720	1.947	1.949	1.932	1.719			
PtW	1.564	2.478	1.937	1.936	1.710	1.665	2.483	1.569	2.472	1.936	1.710	1.934	1.932	1.924	1.935	1.710	1.938	1.939	1.926	1.710			
AsW	1.679	2.414	1.943	1.934	1.711	1.680	2.421	1.682	2.409	1.942	1.710	1.939	1.937	1.931	1.944	1.711	1.944	1.945	1.932	1.711			
SiW	1.502	2.554	1.936	1.929	1.704	2.560	2.560	1.504	2.548	1.934	1.703	1.935	1.934	1.927	1.932	1.704	1.935	1.936	1.927	1.704			
α -isomer												β -isomer											
Model	$M(O_1)$ (deg)	$M(O_2)$ (deg)	$M(O_3)$ (deg)	$O_1(M-O_4)$ (deg)	$O_2(M-O_5)$ (deg)	$O_3(M-O_6)$ (deg)	$O_4(M-O_{11})$ (deg)	$O_5(M-O_{12})$ (deg)	$O_6(M-O_{21})$ (deg)	$O_{11}(M-O_{22})$ (deg)	$O_{12}(M-O_{23})$ (deg)	$O_{21}(M-O_{24})$ (deg)	$O_{22}(M-O_{25})$ (deg)	$O_{23}(M-O_{26})$ (deg)	$O_{24}(M-O_{27})$ (deg)	$O_{25}(M-O_{28})$ (deg)	$O_{26}(M-O_{29})$ (deg)	$O_{27}(M-O_{30})$ (deg)					
BMo	120.201	98.098	148.251	99.614	109.471	109.677	109.265	97.942	99.343	101.474	97.405	99.341	97.976	97.305	100.241	119.105	122.890	118.203	152.543	119.451	148.277		
AlMo	117.039	98.047	151.054	100.519	109.471	109.463	108.480	97.778	100.545	101.796	98.440	98.474	98.414	98.440	100.784	117.125	145.685	116.393	158.381	115.445	166.339	151.940	
GaMo	116.794	98.076	152.374	100.664	109.471	109.393	109.549	97.646	101.567	102.002	97.351	99.765	98.924	97.939	100.465	116.917	146.173	114.604	156.339	116.614	152.584		
GeMo	121.138	98.753	152.786	101.770	109.471	109.591	109.321	99.638	101.567	102.883	98.358	101.189	100.634	99.624	101.733	121.869	146.025	118.891	157.909	121.405	151.727		
PtMo	127.146	101.505	152.083	102.702	109.471	109.700	109.242	101.399	102.827	103.649	101.159	102.449	102.321	101.431	102.868	126.912	146.569	118.983	158.831	120.699	152.812		
AsMo	126.685	101.448	153.399	102.908	109.471	109.673	109.269	101.337	102.940	103.867	101.195	102.644	102.164	101.363	102.022	125.654	147.801	120.026	148.074	120.936	152.134		
SiMo	131.584	103.423	153.266	104.173	109.471	109.793	109.148	103.045	104.296	104.864	102.651	104.145	104.195	103.379	104.107	131.441	147.663	127.524	162.935	131.473	152.868		
BW	122.160	99.216	148.513	100.812	109.471	109.754	109.187	99.353	101.645	102.656	99.888	100.324	99.083	100.818	102.816	121.312	143.453	119.770	154.948	121.892	148.502		
AlW	119.107	98.537	151.991	101.526	109.471	109.651	109.140	99.195	101.624	102.355	99.512	101.019	99.005	100.651	101.651	117.384	146.123	119.394	158.759	119.400	151.877		
GaW	118.825	98.121	152.578	101.985	109.471	109.631	109.109	99.132	101.925	102.556	99.216	101.383	100.362	99.000	101.678	119.448	146.649	116.448	159.559	119.445	152.429		
SiW	123.923	100.863	151.962	102.204	109.471	109.665	109.276	100.599	102.242	103.042	101.973	101.473	100.728	102.319	102.319	123.886	146.475	121.401	159.781	123.771	151.928		
GeW	123.050	100.901	152.985	102.900	109.471	109.745	109.186	100.784	102.575	103.289	100.603	102.188	101.728	100.715	102.582	123.244	147.315	120.756	160.842	123.000	152.961		
PtW	128.638	102.414	152.397	103.217	109.471	109.708	109.234	102.287	103.355	104.201	101.805	103.215	103.289	102.447	103.340	128.447	147.388	120.340	161.518	125.100	152.308		
AsW	127.295	102.413	153.847	103.496	109.471	109.760	109.181	102.274	103.586	104.299	101.787	103.568	103.390	102.326	103.563	127.226	148.280	124.141	162.544	129.949	154.567		
SiW	133.002	104.002	153.556	104.636	109.471	109.808	109.132	103.836	104.714	105.629	102.802	104.937	104.326	103.869	104.580	132.772	148.714	128.573	163.987	132.800	153.077		

Table S6. Comparison of differential energies of protonation between experimental and calculated values.

Composition	$\log\left(\frac{[\text{HXM}_{12}]}{[\text{XM}_{12}][\text{H}^+]} \right)$	$\log\left(\frac{[\alpha\text{-HXM}_{12}]}{[\alpha\text{-XM}_{12}][\text{H}^+]} \right)$	ΔG from $\alpha\text{-PM}_{12}$ (experimental) / eV	ΔG from $\alpha\text{-PM}_{12}$ (calculated) / eV	reference
$\beta\text{-PW}_{12}$	1.176	1.398	0.013	-0.057	[Ueda 2017]
SMo_{12}	5.6	12.7	0.420	0.286	[Himeno 2005]
GeMo_{12}	21.1	12.7	-0.497	-0.363	[Himeno 2005]
GeW_{12}	14.2	5.6	-0.509	-0.372	[Himeno 2005]
BW_{12}	22.3	5.6	-0.988	-0.669	[Himeno 2005]

The protonation energies were compared by differences from that of $\alpha\text{-PM}_{12}$ since equation (1) does not include the interaction between a proton and the solvent, acetonitrile, and does not calculate correct absolute protonation energies. Considering stability and abundance of each isomer, Gibbs energies of α -isomers were used in the cases of no mention of isomer in a reference. These comparisons showed that the Gibbs energies of experiment and calculation are almost confident.

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