

**Unlocking Room-Temperature Bistable Spin Transition at the Nanoscale: Synthesis of Core@Shell  
[Fe(NH<sub>2</sub>Trz)<sub>3</sub>(NO<sub>3</sub>)<sub>2</sub>]@SiO<sub>2</sub> Nanoparticles**

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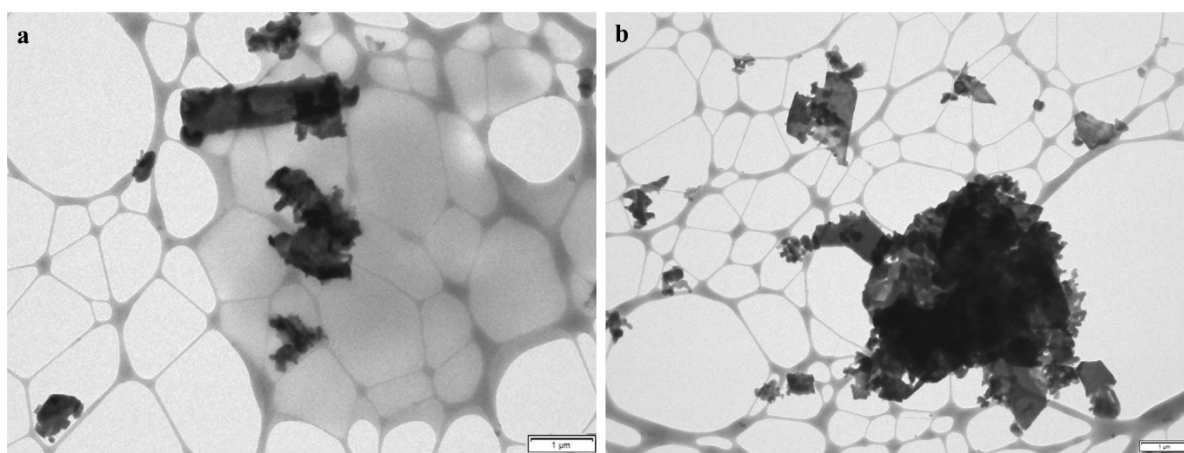


Figure S1. TEM images of two areas of sample SCO-bulk.

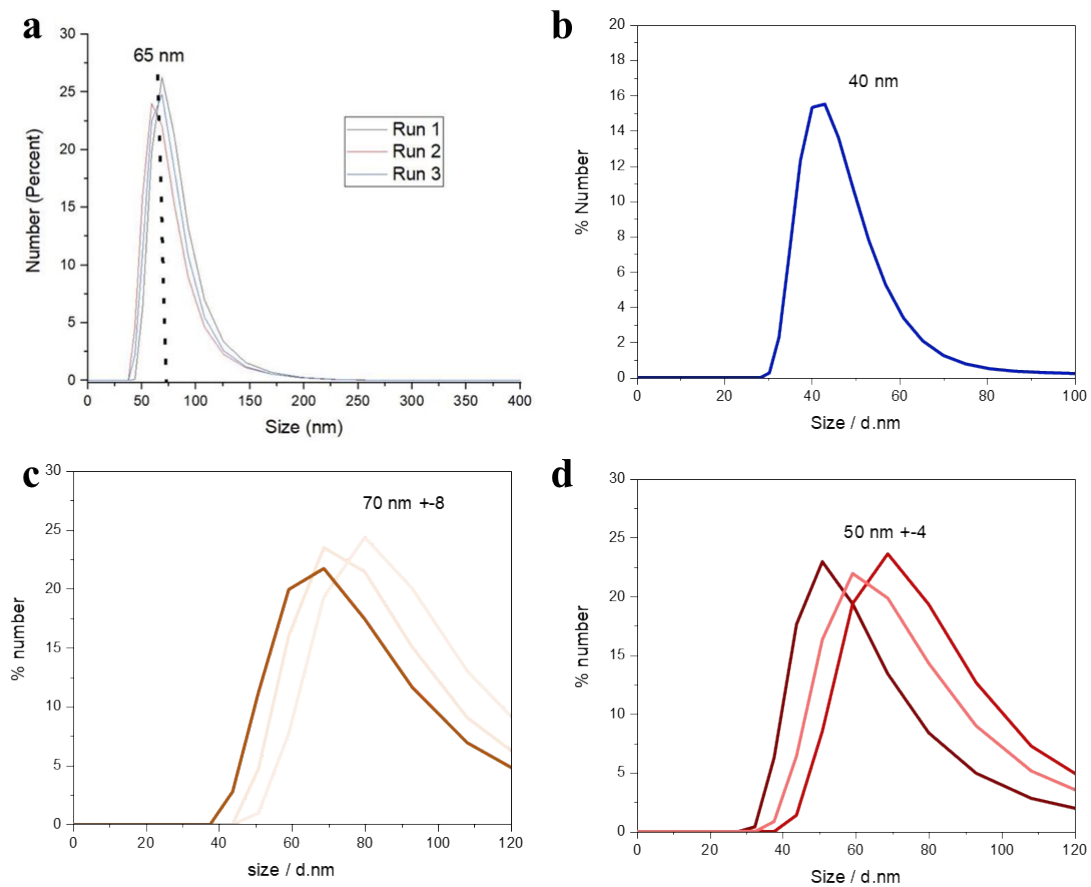


Figure S2. DLS measurements of SCO-1 (a), SCO-2 (b), SCO-SiO<sub>2</sub>-1 (c), and SCO-SiO<sub>2</sub>-2 (d).

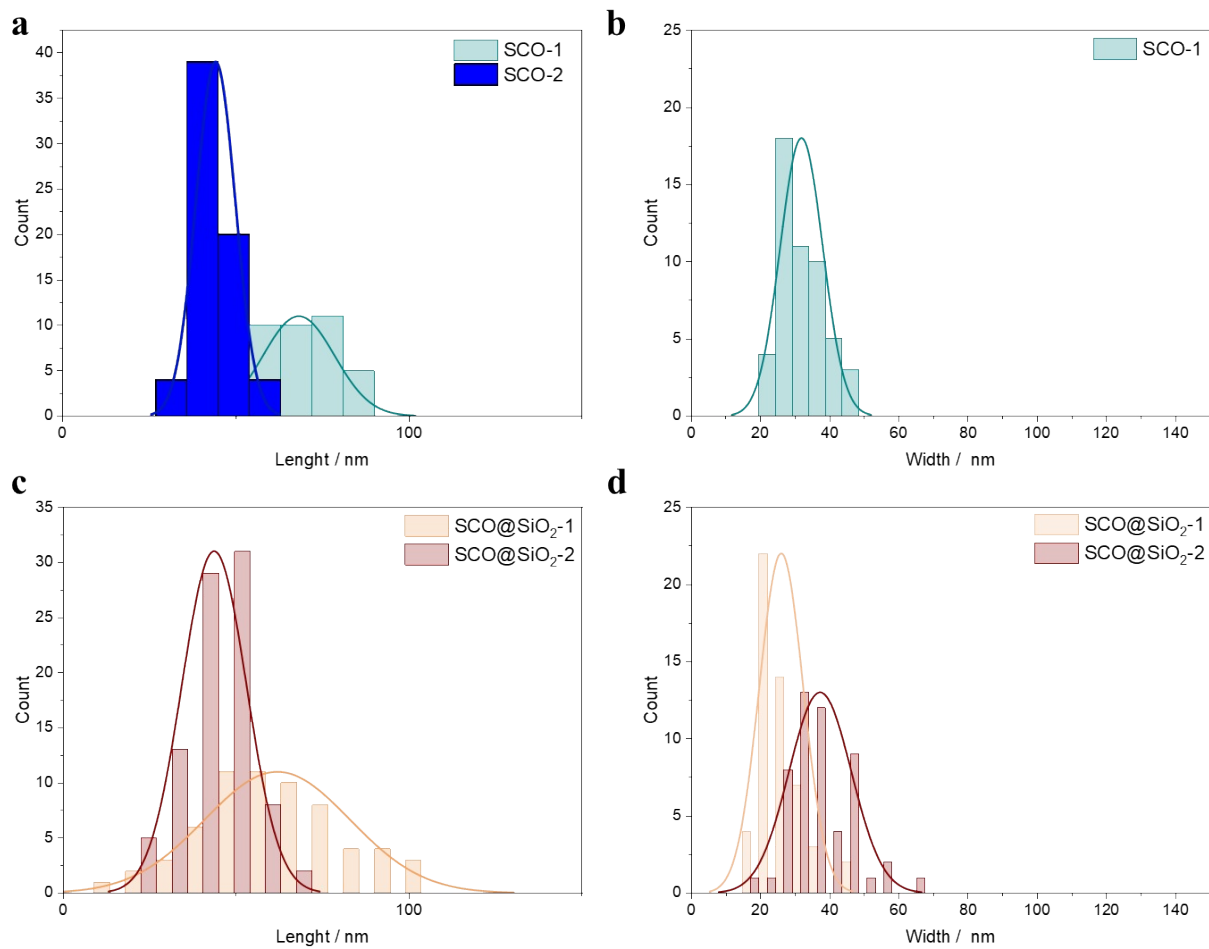


Figure S3. The plot of the length and width size extracted from manual counting from SEM or TEM images. Length of SCO-1 to 2 (a) and width (b), of SCO-SiO<sub>2</sub>-1 to 2 lengths (c) and width (d).

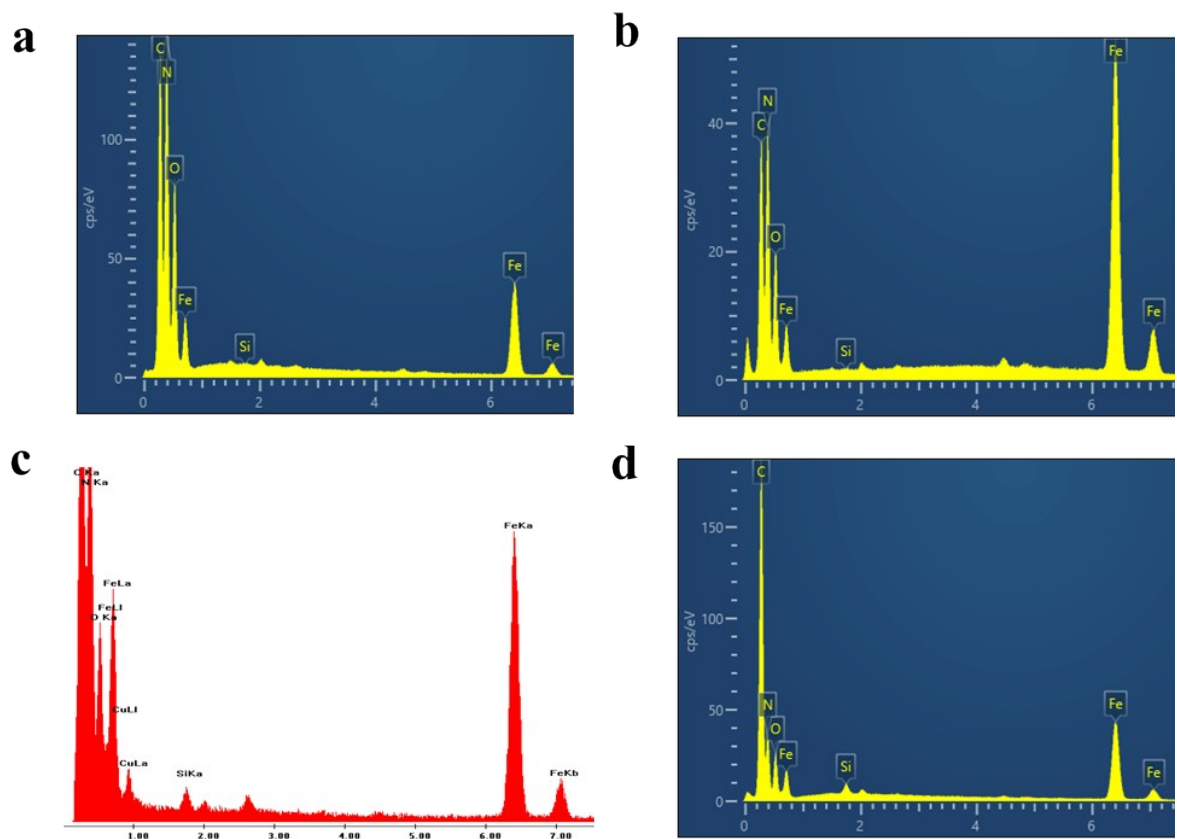


Figure S4. EDX signal of SCO-1 (a), SCO-2 (b), and SCO@SiO<sub>2</sub>-1 (c), and SCO@SiO<sub>2</sub>-2 (d)

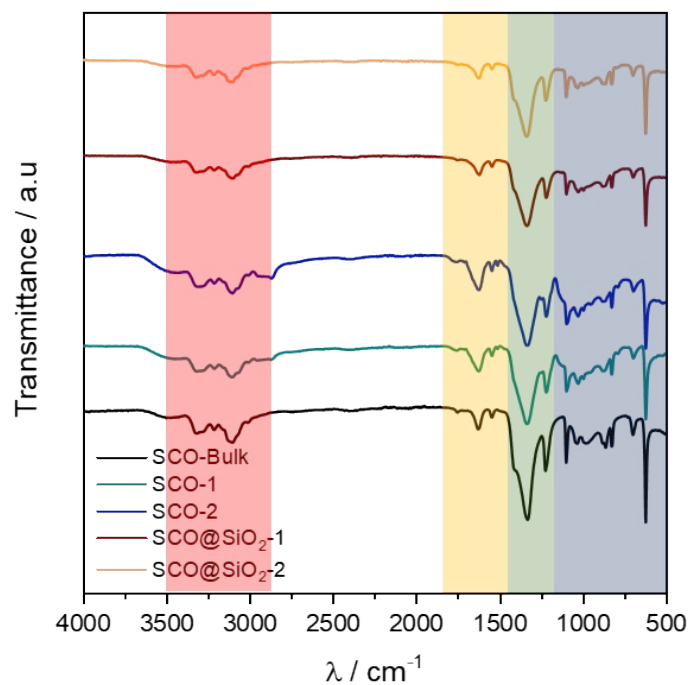


Figure S5. ATR-IR spectra of SCO-Bulk (black), SCO-1 (dark blue), SCO-2 (olive green), and SCO@SiO<sub>2</sub>-1 (burgundy red), and SCO@SiO<sub>2</sub>-2 (orange).

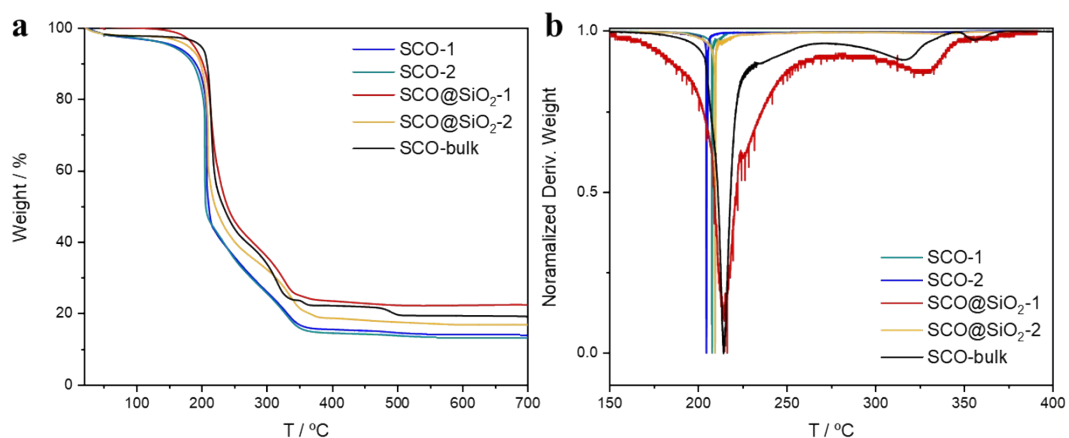


Figure S6. (a) Thermo gravimetric analysis of SCO-bulk (Black), SCO-1 (dark blue), SCO-2 (olive green), and SCO@SiO<sub>2</sub>-1 (burgundy red), and SCO@SiO<sub>2</sub>-2 (orange).(b) Normalized Derivative of the different TGA plots

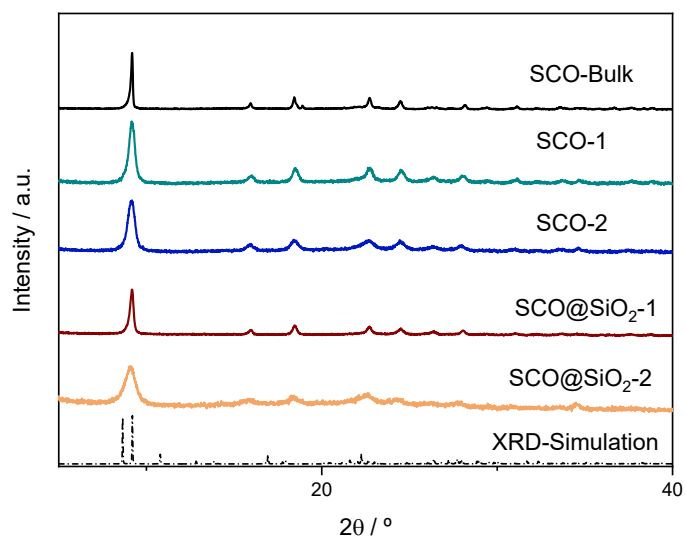


Figure S7. XRD spectra of SCO-bulk (black), SCO-1 (dark blue), SCO-2 (cyan), and SCO@SiO<sub>2</sub>-2 (burgundy red), SCO@SiO<sub>2</sub>-2 (orange), and simulated pattern dashed line.



Figure S8. SCO@SiO<sub>2</sub>-1 freshly redispersed in EtOH (pink solution), and after heated to 60°C, and cool down to room temperature (white solution).

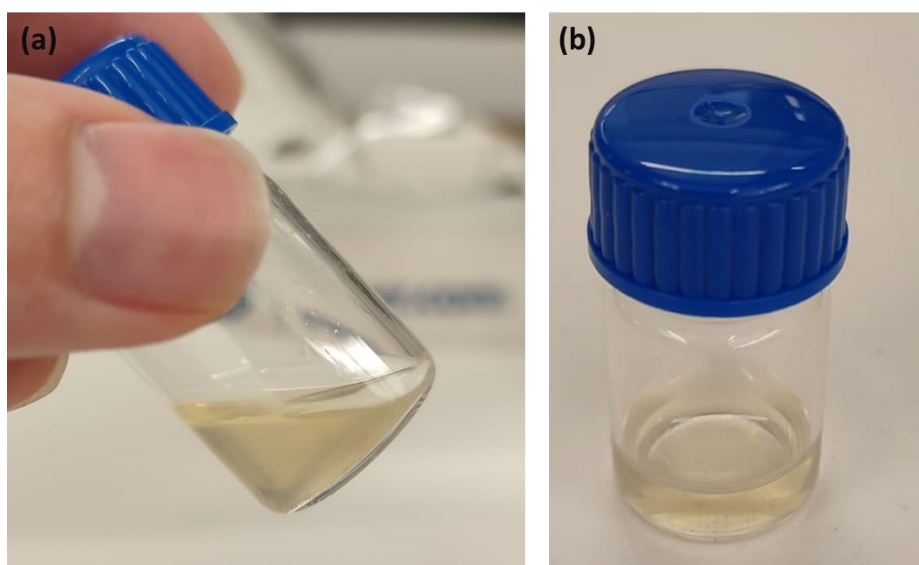


Figure S9. SCO-1(a) and 2 (b) redispersed in H<sub>2</sub>O.

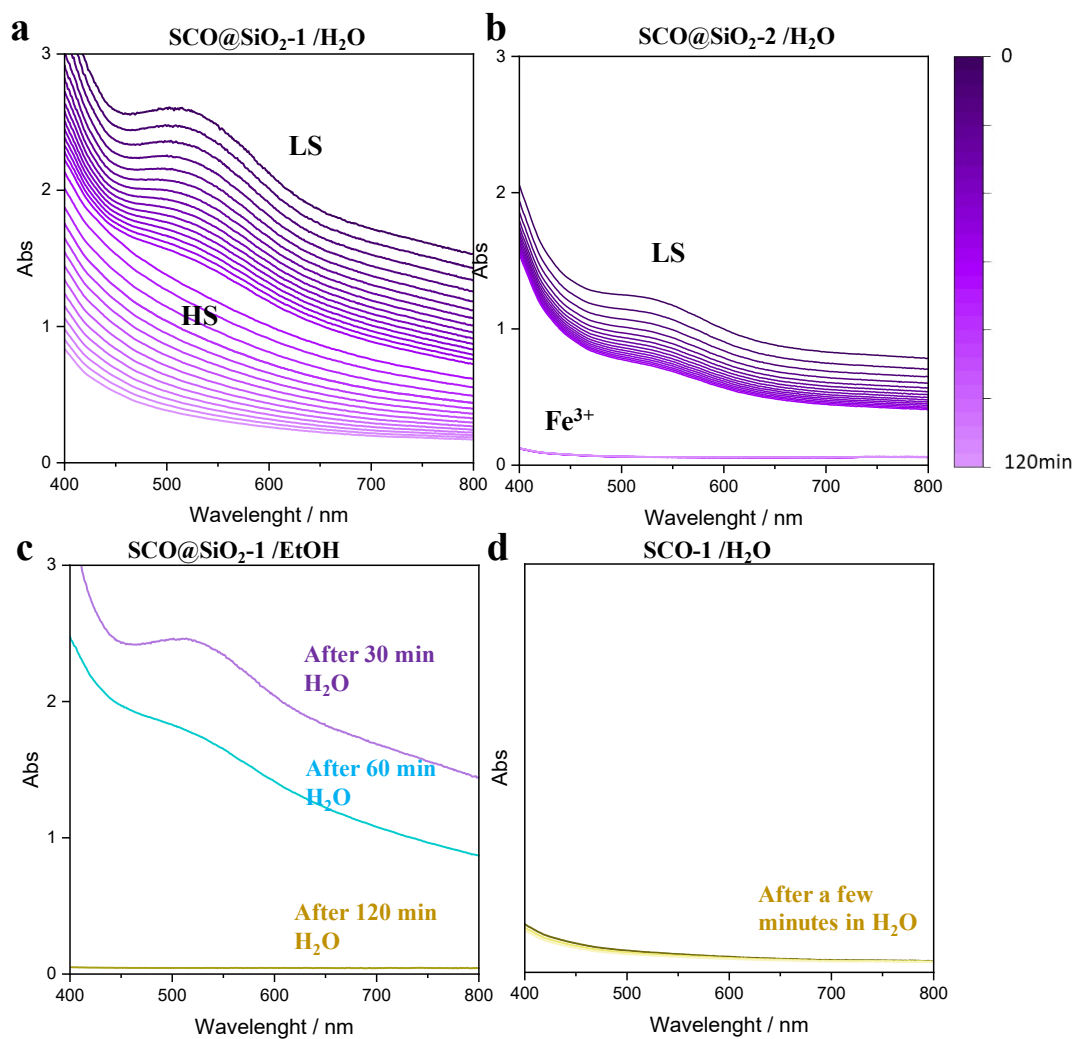


Figure S10. a) Dependency of the UV-VIS absorbance of SCO@SiO<sub>2</sub>-1 redispersed in water as a function of the time. (b) Dependency of the UV-vis absorbance as a function of the time of SCO@SiO<sub>2</sub>-2. The different measurements in the plot were carried out with a 2.5 min delay. (c) UV-vis spectra of SCO@SiO<sub>2</sub>-1 redispersed in EtOH, after 30 min in water (purple), 60 min (cyan), 120 min (dark yellow). (d) UV vis spectra of SCO-1 freshly redispersed in water.

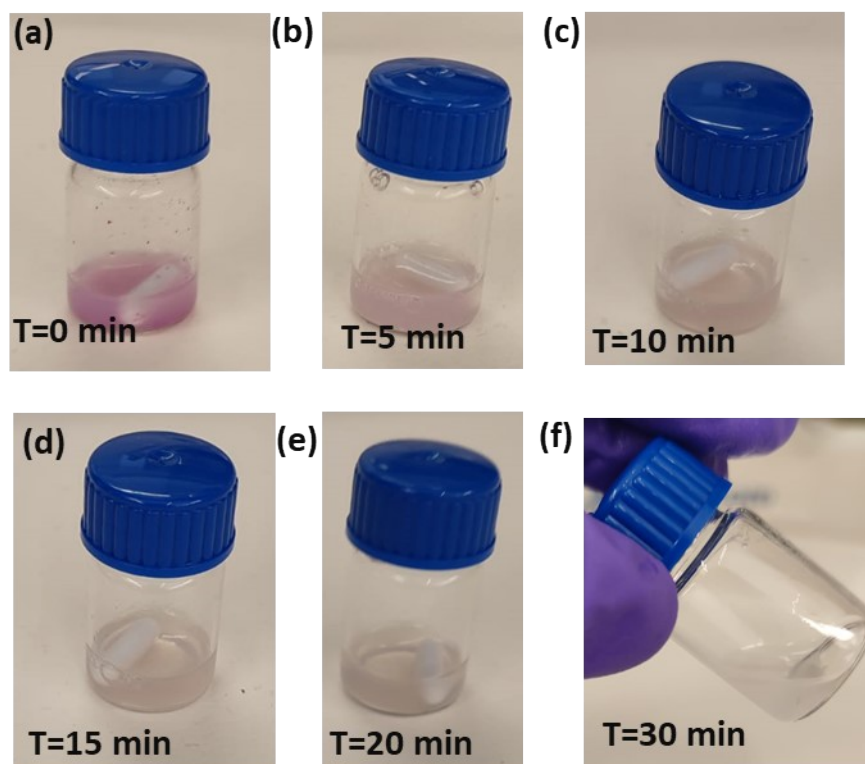


Figure S11. SCO@SiO<sub>2</sub>-1 redispersed in H<sub>2</sub>O at different times after the redispersion.

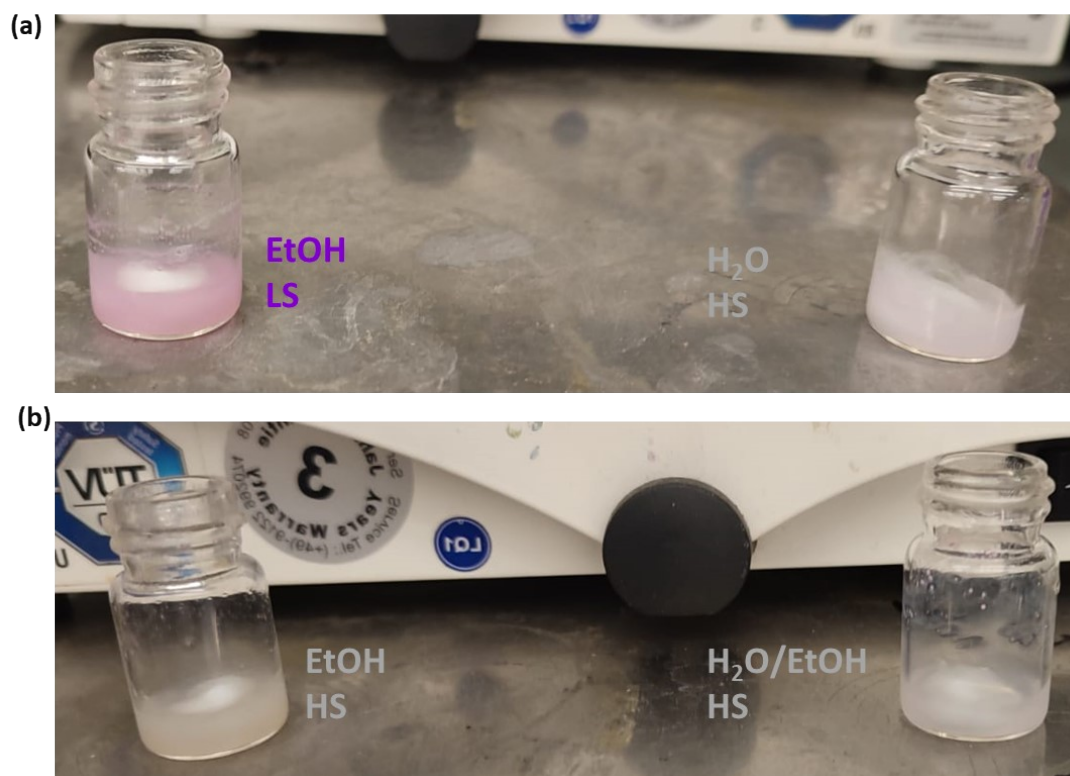


Figure S12. (a) Comparison of SCO@SiO<sub>2</sub>-1 redispersed in EtOH (left) and H<sub>2</sub>O (right) after 15 min. (b) Comparison of the same SCO@SiO<sub>2</sub>-1 presented in (a) after being heated at 60°C, left. At right, we show the SCO@SiO<sub>2</sub>-1 redispersed in water (b) after being centrifuged, and the solid resuspended in EtOH.



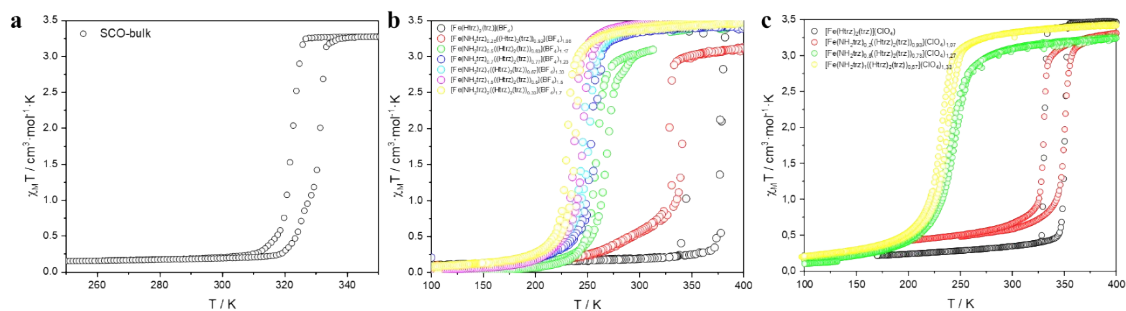


Figure S13. thermal variation of the  $\chi_M T$  product for the SCO-Bulk (a) and different SCO based on  $[\text{Fe}(\text{NH}_2\text{trz})_{3x}((\text{Htrz})_2(\text{trz}))_{1-x}](\text{BF}_4)_{1+x}$  (b), and  $[\text{Fe}(\text{NH}_2\text{trz})_{3x}((\text{Htrz})_2(\text{trz}))_{1-x}](\text{ClO}_4)_{1+x}$  (c).

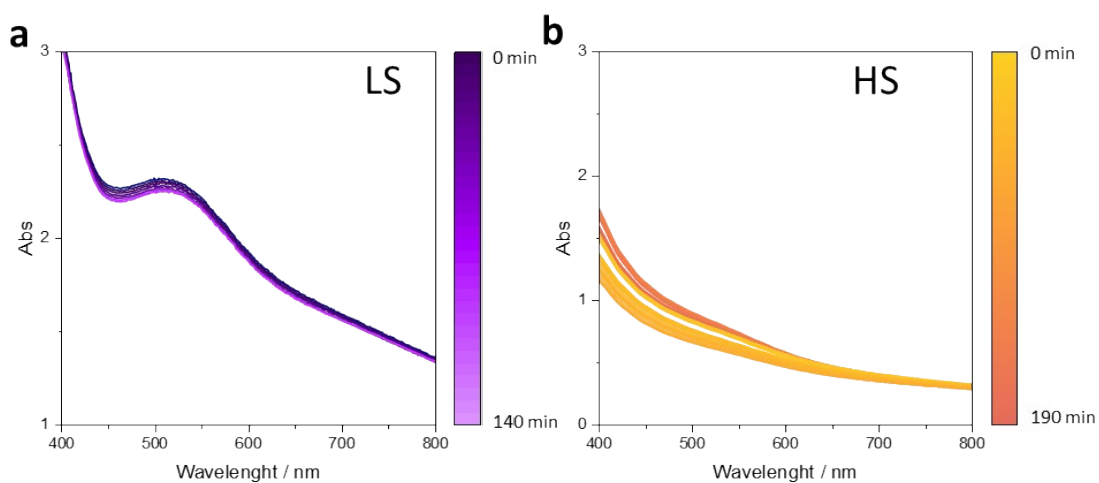


Figure S14. UV vis spectra of SCO@SiO<sub>2</sub>-1 redispersed in EtOH(a), and after heated to 60°C, and cooled down to room temperature (b). The different measurements in the plot were carried out with a 20 minutes delay for (a) and 5-minute for (b).