

Support information

A Simple but Efficient Lithium-doping Approach for Enhancing Supercapacitor Performance of the BiFeO₃ Perovskite Nanostructures

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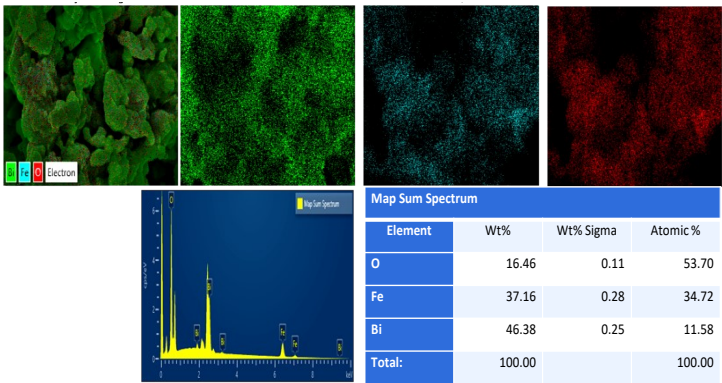
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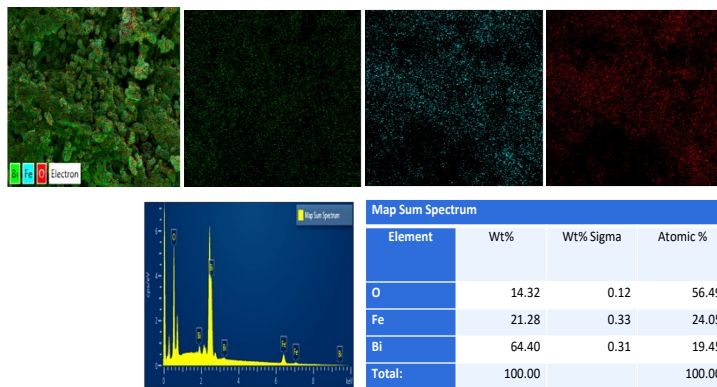
Table S1: Elemental wt.% details obtained for Bi₁Li₀FeO₃ (BLF-A), Bi_{0.95}Li_{0.05}FeO₃ (BLF-B), Bi_{0.90}Li_{0.10}FeO₃ (BLF-C), Bi_{0.85}Li_{0.15}FeO₃ (BLF-D), and Bi_{0.8}Li_{0.2}FeO₃ (BLF-E).

Electrode	Fe wt. %	Bi wt. %	O wt. %	Bi/Fe	Electrode
Bi ₁ Li ₀ FeO ₃ (BLF-A)	37.16	46.38	16.46	1.25	a
Bi _{0.95} Li _{0.05} FeO ₃ (BLF-B)	21.28	64.40	14.32	3.03	b
Bi_{0.90}Li_{0.10}FeO₃ (BLF-C)	20.03	65.48	14.50	3.27	c
Bi _{0.85} Li _{0.15} FeO ₃ (BLF-D)	33.15	55.22	11.63	1.66	d
Bi _{0.8} Li _{0.2} FeO ₃ (BLF-E)	33.67	46.80	19.81	1.33	e

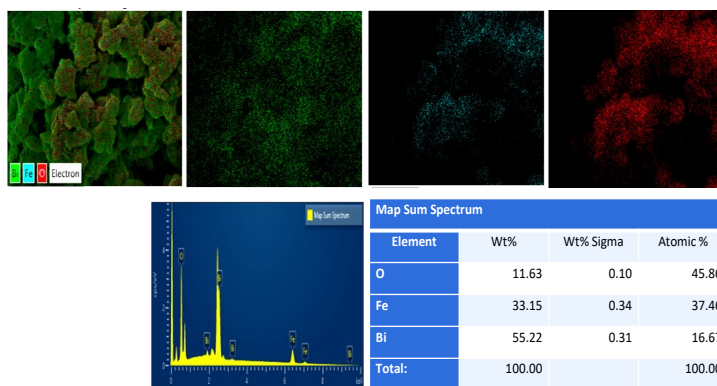
A)



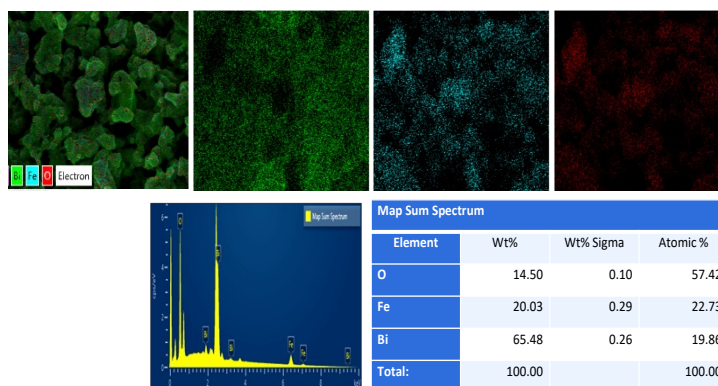
B)



c)



D)



E)

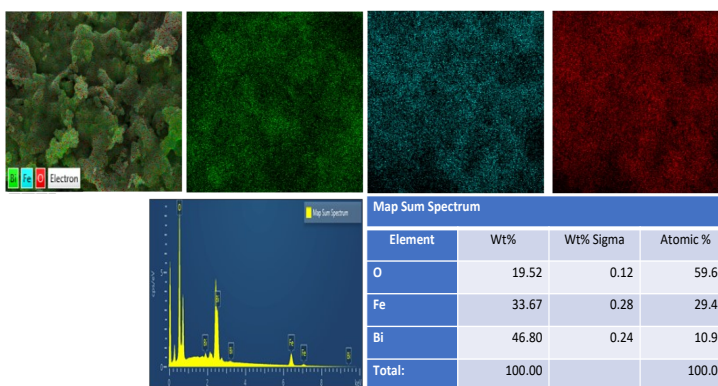


Fig. S1: a-e) Obtained O, Fe and Bi wt.% proportions confirmed by the surfaces of different electrodes.

Table S2. Comparison of Energy Storage Performance of perovskite as Positive Electrodes of a Supercapacitor-Hybrid Device with Other Electrode Active Materials and Devices Reported in the Literature.

	Electrode	electrolyte	Ed (Wh kg ⁻¹)	Pd (W kg ⁻¹)	SC (F/g)	RE
1	TiO ₂ /BiFeO ₃	0.5 M Na ₂ SO ₄	58.5	1200	440	1
2	MnFe ₂ O ₄ /rGO	6 M KOH	15.9	324.5	271	2
3	rGO-NiFe ₂ O ₄	1 M Na ₂ SO ₄	23.7	225	210.9	3
4	BiFeO ₃ -RGO	3 M KOH	6.65	190	92.43	4
5	LCM050/ LCM050	1M KOH	7.6	160	21.7	5
7	BFMO// BFMO	3 M NaOH	19	325	81.57	6
8	Co–Ni/Bi// Co–Ni/Bi	1 M Na ₂ SO ₄	26.21	100	125.79	7
10	BLF-C//Bi₂S₃	6 M KOH	48.65	750	155.6	Our work

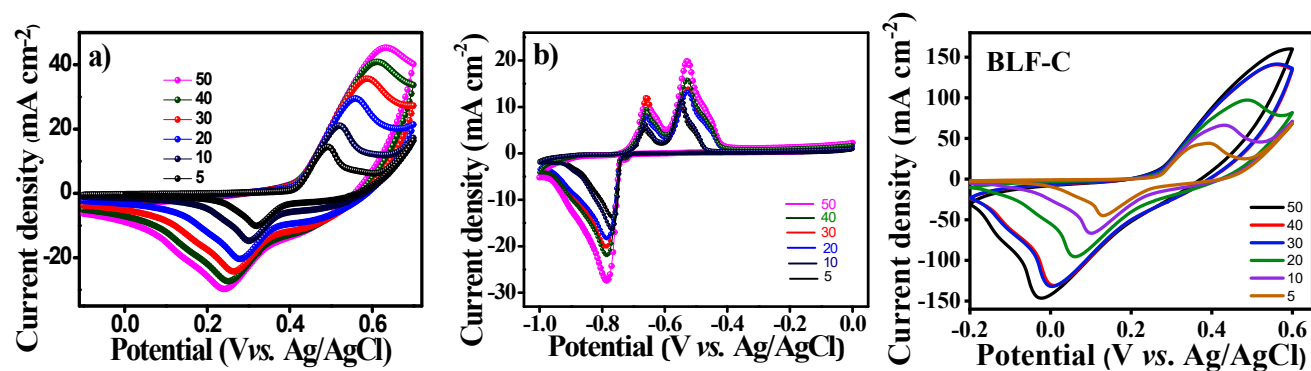


Fig. S2: CV spectrums of the; (a) Bi₂O₃, (b) Fe₂O₃, and (c) BLF-C electrodes.

Reference

- 1 A. Sarkar, A. K. Singh, D. Sarkar, G. G. Khan and K. Mandal, *ACS Sustain. Chem. & Eng.*, 2015, **3**, 2254–2263.
- 2 A. G. Tabrizi, N. Arsalani, A. Mohammadi, H. Namazi, L. S. Ghadimi and I. Ahadzadeh, *New J. Chem.*, 2017, **41**, 4974–4984.
- 3 Y.-Z. Cai, W.-Q. Cao, Y.-L. Zhang, P. He, J.-C. Shu and M.-S. Cao, *J. Alloys Compd.*, 2019, **811**, 152011.
- 4 D. Moitra, C. Anand, B. K. Ghosh, M. Chandel and N. N. Ghosh, *ACS Appl. Energy Mater.*, 2018, **1**, 464–474.
- 5 H. Mo, H. Nan, X. Lang, S. Liu, L. Qiao, X. Hu and H. Tian, *Ceram. Int.*, 2018, **44**, 9733–9741.
- 6 H. Kuzhandaivel, Y. Selvaraj, M. C. Franklin, S. Manickam and K. S. Nallathambi, *New J. Chem.*, 2021, **45**, 15223–15233.
- 7 A. N. Alqarni, E. Cevik, M. A. Almessiere, A. Baykal, M. A. Gondal, M. Hassan, A. Bozkurt, A. Iqbal, S. M. Asiri and Y. Slimani, *J. Phys. Chem. Solids*, 2023, **177**, 111288.