## Two 2D spin crossover coordination polymers constructed by [Pd(SCN)<sub>4</sub>]<sup>2-</sup> building blocks

## Kai-ping Xie<sup>1\*</sup>, Zhi-zhen Peng<sup>1</sup>, Ze-yu Ruan<sup>2</sup>, Wei-ding Fan<sup>1</sup>, Yan-ru Chen<sup>2</sup>, Xiao-dan Zheng<sup>1</sup>, Yu-bo Zou<sup>1</sup>, Si-guo Wu<sup>2\*</sup>, Zi-cheng Xiao<sup>1</sup>

<sup>1</sup>School of Chemistry and Materials Engineering, Huizhou University, Huizhou, 516007, P. R. China; <sup>2</sup>Key Laboratory of Bioinorganic and Synthetic Chemistry of Ministry of Education, School of Chemistry, Sun, Yat-Sen University, Guangzhou, 510275, P. R. China



Figure S1. Powder X-ray diffraction data of 1 (a) and 2 (b).



Figure S2. Thermogravimetric analyses of 1 (a) and 2 (b).



Figure S3. IR spectra of 1 (a) and 2 (b).



Figure S4. Differential scanning calorimetry (DSC) measurements of 1 (a) and 2 (b).



Figure S5. Asymmetric units of 1 (a) and 2 (b). Hydrogen atoms are omitted for clarity.



Figure S6. The dihedral angles of 1 (a) and 2 (b).

Table S1. Selected structural parameters for 1 and 2 at different temperatures.

	1		2	
Parameter	100 K	245 K	110 K	200 K
<fe-nscn></fe-nscn>	<fe1-n3>1.931(1)</fe1-n3>	<fe1-n3>2.126(1)</fe1-n3>	<fe1-n2>1.947(4)</fe1-n2>	<fe1-n2>2.134(4)</fe1-n2>
	<fe1-n4>1.939(2)</fe1-n4>	<fe1-n4>2.134(2)</fe1-n4>	<fe1-n3>1.950(4)</fe1-n3>	<fe1-n3>2.103(5)</fe1-n3>
			<fe2-n5>1.937(4)</fe2-n5>	<fe2-n5>2.119(4)</fe2-n5>
			<fe2-n6>1.920(3)</fe2-n6>	<fe2-n6>2.114(4)</fe2-n6>
<fe-n<sub>py&gt;</fe-n<sub>	<fe1-n1>2.006(2)</fe1-n1>	<fe1-n1>2.200(2)</fe1-n1>	<fe1-n1>2.003(5)</fe1-n1>	<fe1-n1>2.180(6)</fe1-n1>
			<fe2-n4>2.004(5)</fe2-n4>	<fe2-n4>2.177(5)</fe2-n4>

Table S2. The shortest Fe…Fe distance and dihedral angles of 1 and 2.

	1		2	
Parameter	100 K	245 K	110 K	200 K
Fe…Fe <sup>[a]</sup>	8.398(8)	8.577(8)	8.284 (2)	8.448(2)
Fe…Fe <sup>[b]</sup>	9.407(7)	9.592(7)	13.302(3)	13.455(3)
Dihedral angles <sup>[c]</sup>	21.651	23.524	37.351	41.870
Dihedral angles <sup>[d]</sup>	14.986	15.410	26.325	26.891
Dihedral angles <sup>[e]</sup>	14.450	14.740	27.194	27.940

[a] The shortest Fe···Fe distance within the layer (Å); [b] The shortest Fe···Fe distance between layers (Å); [c] The dihedral angles between the two [FeN<sub>4</sub>] faces (°); [d] The dihedral angles between the two Fe···Pd···Fe faces in the quadrilateral (°); [e] The dihedral angles between the two Fe···S–Pd–S···Fe faces in the quadrilateral (°).

 Table S3. Ligand-ligand interaction between adjacent layers of 1.

	1	
Parameter	100 K	245 K
Centroid <sup>a</sup> ····Centroid <sup>b</sup>	4.548(4) Å	4.555(4) Å
O1 <sup>a</sup> ····N2 <sup>b</sup>	3.752(2) Å	3.712(2) Å
O1 <sup>a</sup> O1 <sup>b</sup>	4.199(4) Å	4.290(4) Å

N2 <sup>a</sup> ····N2 <sup>b</sup>	4.322(4) Å	4.446(4) Å
N1 <sup>a</sup> ····N2 <sup>b</sup>	4.719(4) Å	4.731(4) Å
N2 <sup>a</sup> ···N1 <sup>b</sup>	4.416(3) Å	4.449(3) Å

Centroid: N1-N2, C1-C4; a: x, y, z; b: x, 1-y, -0.5+z.

	2	
Parameter	110 K	200 K
N7 <sup>a</sup> ····N9 <sup>b</sup>	3.598(2) Å	3.830(7) Å
N8 <sup>a</sup> ····N9 <sup>b</sup>	3.654(8) Å	3.794(9) Å
N7 <sup>a</sup> ···N10 <sup>b</sup>	3.778(8) Å	3.913(9) Å
N8 <sup>a</sup> ···N10 <sup>b</sup>	4.189(1) Å	4.232(1) Å
N1 <sup>a</sup> ····C18 <sup>b</sup>	3.899(7) Å	3.890(7) Å
N1 <sup>a</sup> ····C17 <sup>b</sup>	4.224(6) Å	4.188(6) Å
N1 <sup>a</sup> ····C19 <sup>b</sup>	4.266(9)	4.263(9)
C11 <sup>a</sup> ···N4 <sup>b</sup>	3.892(8) Å	3.925(8) Å
C10 <sup>a</sup> ···N4 <sup>b</sup>	4.194(1) Å	4.247(1) Å
C6 <sup>a</sup> ···N4 <sup>b</sup>	4.371(8) Å	4.283(9) Å

 Table S4. In-layer ligand-ligand interaction of 2.

a: x, y, z; b: 1-x, 1-y, 1-z.