

Engineering the Multifunctionality of $\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}$ Garnet with Sm^{3+} and Tb^{3+} Activators for Solid-State Lighting and Luminescence Thermometry

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Supplementary data

Table S1. The percentage difference of ionic radii (D_r) between host cations and Sm^{3+} ions.

Host cation (CN)	Activator ion (CN)	R_m (Å)	R_d (Å)	D_r (%)
Li^+ (4)	Sm^{3+} (8)	0.59(4)	1.027	-82.8
Y^{3+} (8)	Sm^{3+} (8)	1.01(9)	1.027	-5.9
Te^{6+} (6)	Sm^{3+} (8)	0.56(6)	1.027	-92.7

Table S2. Comparison of $T_{1/2}$ of reported Sm^{3+} based phosphors.

Compositions	$T_{1/2}$ (K)	References
$\text{Li}_3\text{Ba}_2\text{La}_3(\text{MoO}_4)_8: \text{Sm}^{3+}$	512	[1]
$\text{Li}_6\text{CaLa}_{1.94}\text{Sm}_{0.06}\text{Ta}_2\text{O}_{12}$	423	[2]
$\text{Ca}_2\text{MgTeO}_6: \text{Sm}^{3+}$	>480	[3]
$\text{Li}_3\text{Gd}_3\text{Te}_2\text{O}_{12}: \text{Sm}^{3+}$	>480	[4]
$\text{Ca}_2\text{Al}_2\text{SiO}_7: \text{Sm}^{3+}$	>500	[5]
$\text{Sr}_9\text{In}(\text{PO}_4)_7: \text{Sm}^{3+}$	>523	[6]
$\text{NaSrLa}(\text{MoO}_4)\text{O}_3: \text{Sm}^{3+}$	423	[7]
$\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}: \text{Sm}^{3+}$	500	This work

Table S3. Rietveld refinement and crystallographic data of $\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}$: 0.05 Tb^{3+} , 0.07 Sm^{3+} phosphor.

Formula	$\text{Li}_3\text{Y}_{2.88}\text{Tb}_{0.05}\text{Sm}_{0.07}\text{Te}_2\text{O}_{12}$					
Crystal system	Cubic					
Space group	$Ia\bar{3}d$ (230, O_h^{10})					
Cell Parameters	$a = 12.2596(1) \text{ \AA}$					
Reliability factors	$R_{\text{wp}} = 3.40 \%$, $R_p = 2.63 \%$ and $\text{GOF} = 1.56$					
Atom	Site	X	y	Z	Occupancy	$B_{\text{iso}} (\text{\AA}^2)$
$\text{Y}^{3+}/\text{Tb}^{3+}/\text{Sm}^{3+}$	24c	0.125	0.00	0.25	1	0.007(9)
Te^{6+}	16a	0.00	0.00	0.00	1	0.006(9)
Li^{3+}	24d	0.25	0.875	0.00	1	0.018(3)
O^{2-}	96h	0.268(1)	0.110(2)	0.196(8)	1	0.01(1)

Table S4. Comparison of temperature sensing properties of different phosphors.

Phosphors	Temperature range (K)	S_r (% K ⁻¹)	References
$\text{Ca}_2\text{TbSn}_2\text{Al}_3\text{O}_{12}:\text{Sm}^{3+}$	300-500	0.50	[8]
$\text{SrY}_2(\text{MoO}_4)_4:\text{Tb}^{3+}/\text{Sm}^{3+}$	290-440	0.9	[9]
$\text{Ca}_2\text{LaTaO}_6:\text{Mn}^{2+}/\text{Tb}^{3+}$	300-450	3.6	[10]
$\text{CaGdAlO}_4:\text{Mn}^{4+}, \text{Tb}^{3+}$	200-600	2.23	[11]
$\text{Y}_3\text{Al}_5\text{O}_{12}:\text{Dy}^{3+}/\text{Cr}^{3+}$	293- 573	2.32	[12]
$\text{Na}_3\text{Sc}_2\text{P}_3\text{O}_{12}:\text{Eu}^{2+}/\text{Mn}^{2+}$	293-473	1.556	[13]
$\text{Sr}_2\text{LuTaO}_6:\text{Tb}^{3+}/\text{Mn}^{4+}$	313–573	1.98	[14]
$\text{Ba}_3(\text{VO}_4)_2:\text{Sm}^{3+}$	303-463	2.24	[15]
$\text{BaGd}_2\text{O}_4:\text{Bi}^{3+}/\text{Sm}^{3+}$	293-473	1.11	[16]
$\text{LaNbO}_4:\text{Bi}^{3+}/\text{Eu}^{3+}$	303-483	1.89	[17]
$\text{Ca}_2\text{NaMg}_2\text{V}_3\text{O}_{12}:\text{Sm}^{3+}$	303-503	1.889	[18]
$\text{Sr}_2\text{NaMg}_2\text{V}_3\text{O}_{12}:\text{Sm}^{3+}$	300-500	2.01	[19]
$\text{LaNbO}_4:\text{Bi}^{3+}/\text{Tb}^{3+}$	303-483	2.36	[17]
$\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}:\text{Dy}^{3+}$	80-300	1.2	[20]
$\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}:\text{Bi}^{3+}/\text{Pr}^{3+}$	300-500	1.8	[21]
$\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}:\text{Sm}^{3+}$	300-500	1.0	This work
$\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}:\text{Sm}^{3+}/\text{Tb}^{3+}$	300-500	1.8	This work

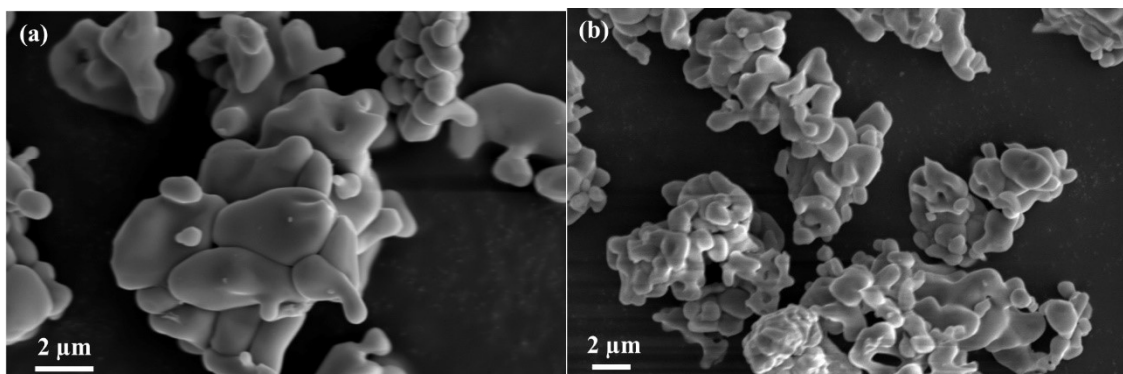


Fig. S1 SEM images of (a) LYTO, and (b) LYTO: 0.05 Sm³⁺ phosphors.

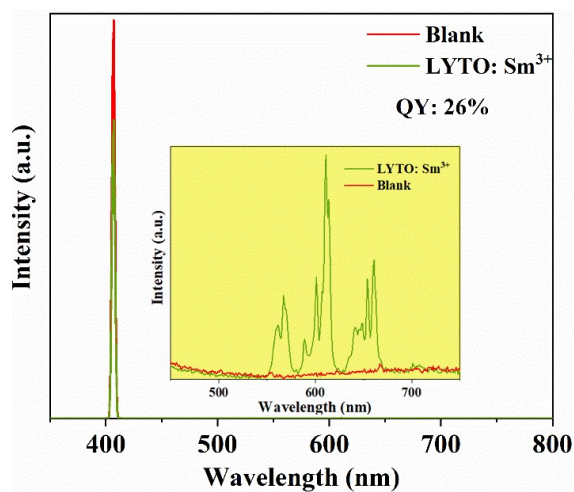


Fig. S2 Quantum yield of LYTO: Sm³⁺ under 407 nm excitation and 611 nm emission wavelength.

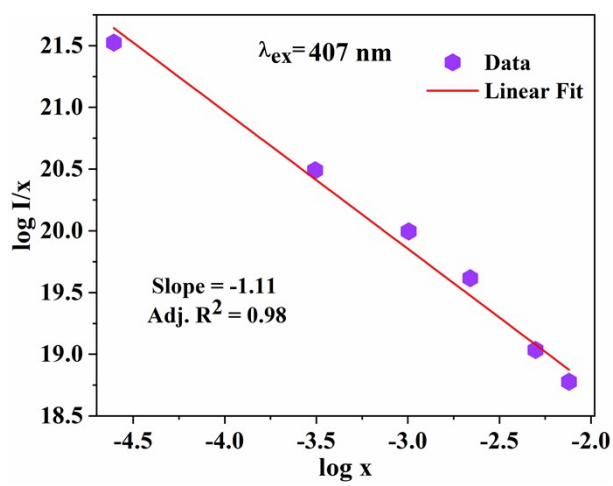


Fig. S3 The $\log(I/x)$ - $\log(x)$ plot for the transition of Sm^{3+} ions in LYTO: Sm^{3+} phosphor.

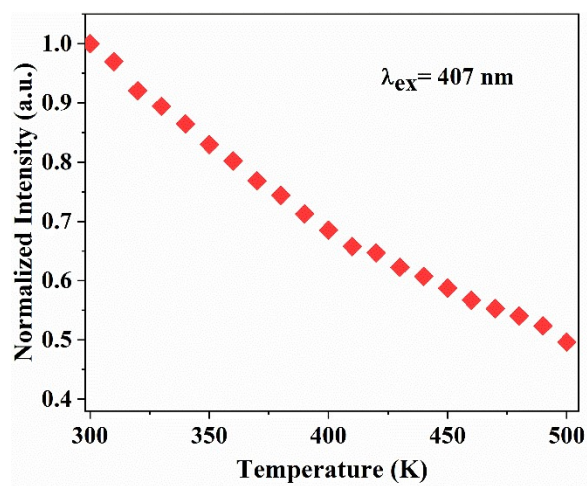


Fig. S4 The variation of emission intensity of LYTO: Sm^{3+} with temperature.

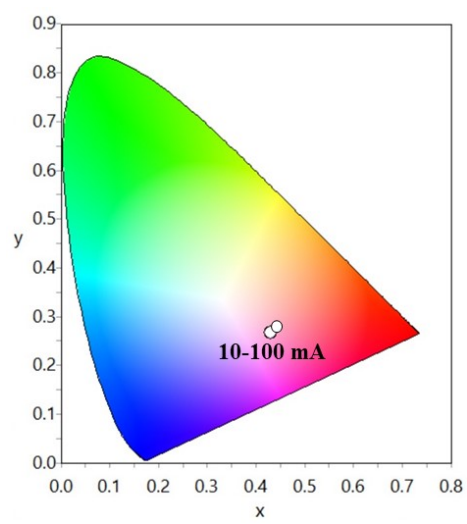


Fig. S5 The shift of CIE coordinates of the fabricated LED at higher input bias currents.

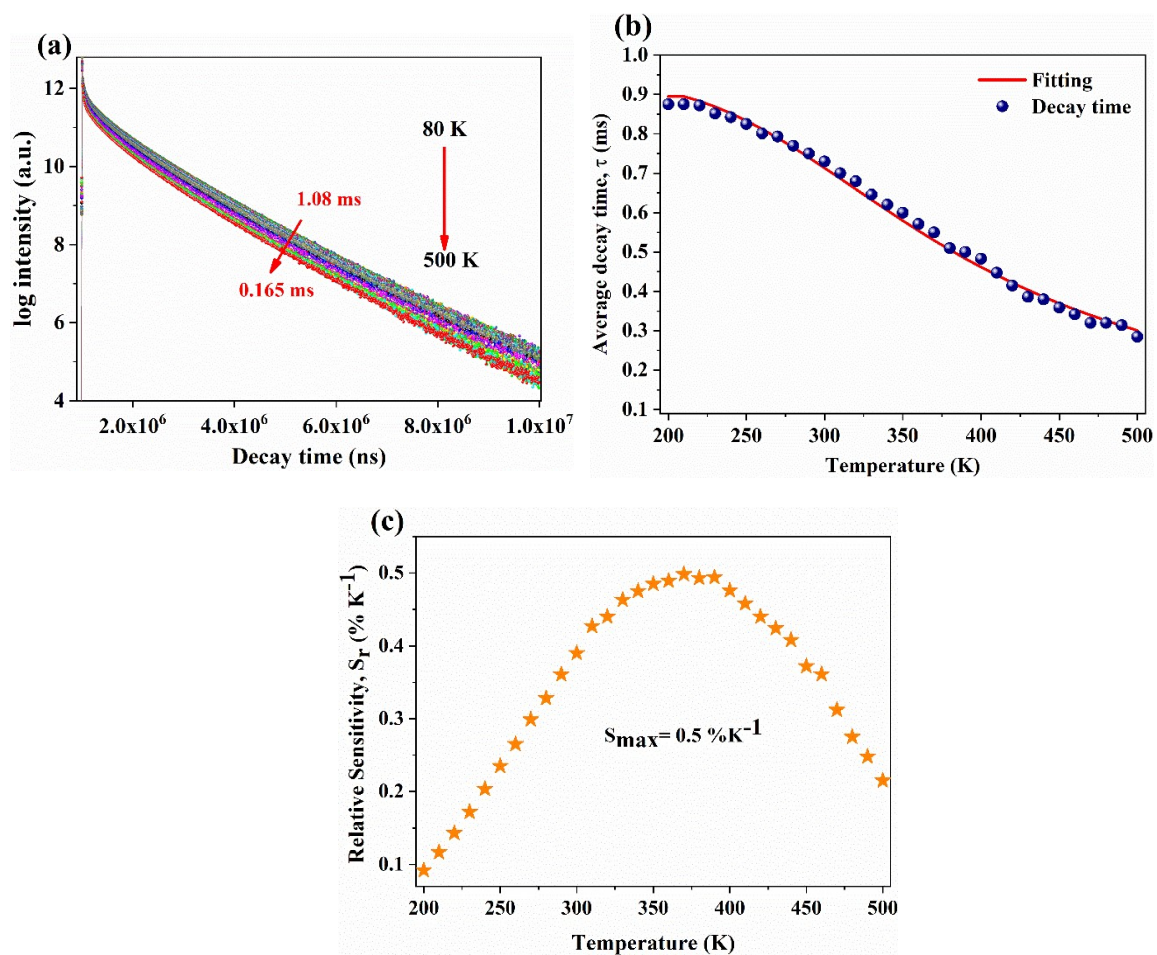


Fig. S6 (a) The temperature-dependent decay curves of LYTO: Sm phosphor, (b) the average decay time with temperature, and (c) the dependence of relative sensitivity, Sr Vs. T in LYTO: Sm phosphor determined by decay time method.

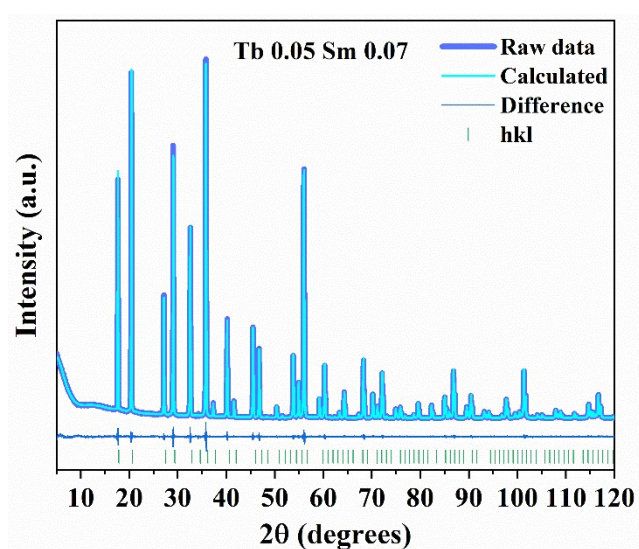


Fig. S7 The Rietveld refinement pattern of $\text{Li}_3\text{Y}_3\text{Te}_2\text{O}_{12}: 0.05 \text{ Tb}^{3+}, 0.07 \text{ Sm}^{3+}$ phosphor.

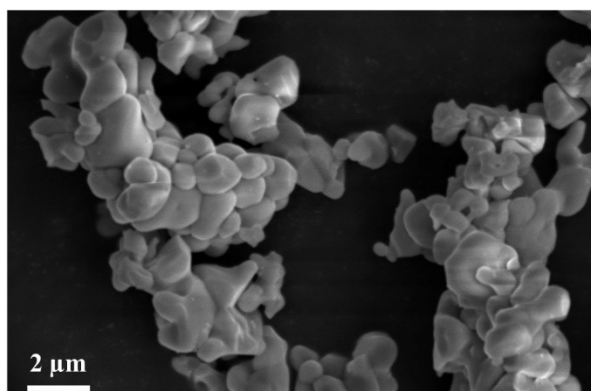


Fig. S8 SEM images of LYTO: Sm³⁺, Tb³⁺ phosphors.

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