

Synthesis, Characterization and Chemical Bonding Analysis of the Quaternary Cyanamides $\text{Li}_2\text{MnHf}_2(\text{NCN})_6$ and $\text{Li}_2\text{MnZr}_2(\text{NCN})_6$

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Supporting Information

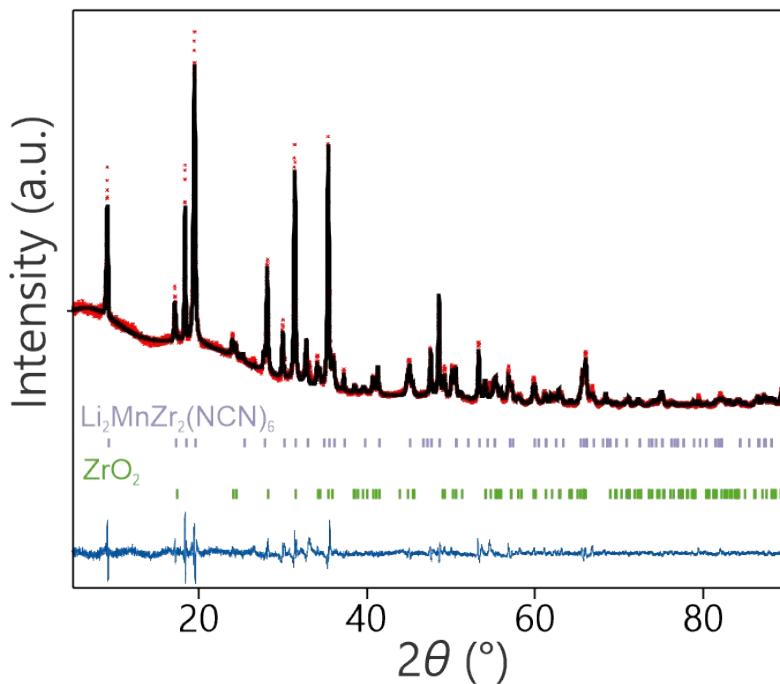


Figure S1. Rietveld fit of $\text{Li}_2\text{MnZr}_2(\text{NCN})_6$ to PXRD data ($\text{Cu K}_{\alpha 1}$), showing observed (red), calculated (black) and difference (blue) intensities. Bragg positions of $\text{Li}_2\text{MnZr}_2(\text{NCN})_6$ (violet) and 20.1(1) wt% ZrO_2 baddeleyite $P2_1/c$ (blue) are denoted by vertical markers.

Table S1. CHN results of $\text{Li}_2\text{MnHf}_2(\text{NCN})_6$ using ZnNCN and $\text{Li}_2\text{MnZr}_2(\text{NCN})_6$ compared to theoretical values corrected for the presence of HfO_2 and ZrO_2 .

		C	H	N
$\text{Li}_2\text{MnHf}_2(\text{NCN})_6$	Theoretical (wt. %)	9.9	0	23.2
	Experimental (wt. %)	8.8	0	18.9
$\text{Li}_2\text{MnZr}_2(\text{NCN})_6$	Theoretical (wt. %)	12.8	0	29.9
	Experimental (wt. %)	12.9	0	26.3

Table S2. Crystallographic data and fractional coordinates for $\text{Li}_2\text{MnZr}_2(\text{NCN})_6$.^a

Atom		x	y	z	occ.	U_{iso} (10^2 \AA^2)
Li1	$2d$	$\frac{1}{3}$	$\frac{2}{3}$	$\frac{1}{2}$	0.944(1)	0.92(4)
Zr2	$2d$	$\frac{1}{3}$	$\frac{2}{3}$	$\frac{1}{2}$	0.056(1)	0.92(4)
Mn	$1b$	0	0	$\frac{1}{2}$	1	1.38(14)
Zr1	$2c$	$\frac{1}{3}$	$\frac{2}{3}$	0	0.944(1)	0.92(4)
Li2	$2c$	$\frac{1}{3}$	$\frac{2}{3}$	0	0.056(1)	0.92(4)
N1	$6k$	0.5945(15)	0	0.8687(7)	1	2.9(2)
N2	$6k$	0.3217(14)	0	0.3849(8)	1	2.9(2)
C	$6k$	0.628(3)	0	0.736(1)	1	2.9(2)

Trigonal, $P\bar{3}1m$ (No. 162), $Z=1$, $a=5.94791(12) \text{ \AA}$, $c=9.6403(3) \text{ \AA}$; $R_{\text{wp}}=5.84\%$, $R_p=4.14\%$, $\chi^2=4.647$, $R_{\text{wpb}}=8.45\%$

^a Standard deviations are given in parentheses.