

Supporting Information

Synthesis and characterization of In(III) S-thiobezoylthioglycolate complexes and their catalytic applications in CO₂ fixation and multicomponent synthetic reactions. †

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The NMR spectra of following synthesised indium complexes

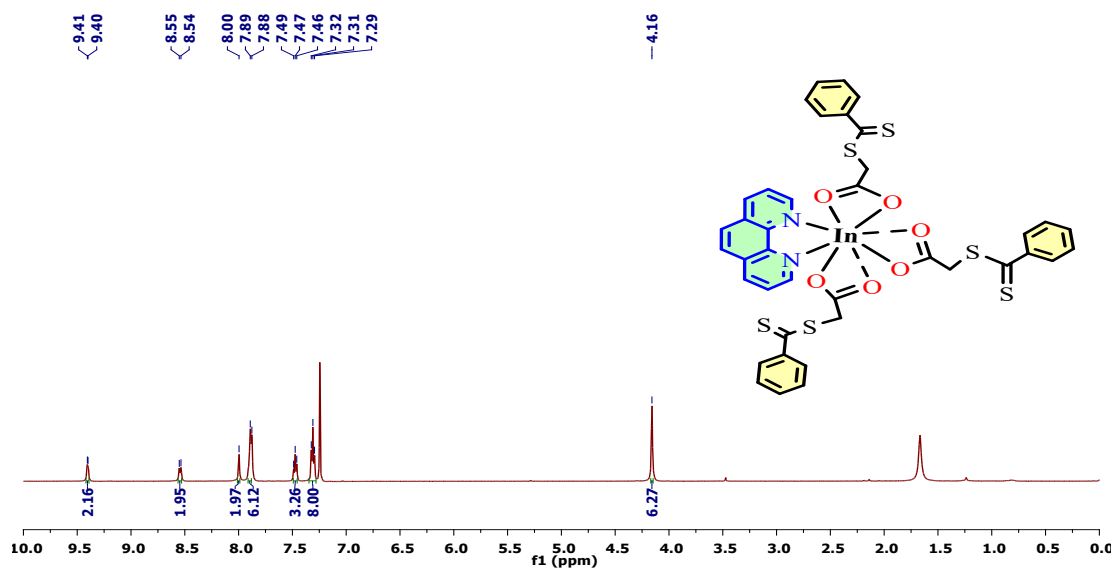


Figure S1 ¹H NMR spectra of complex 1 [In(1,10phen.)(Stbtg)₃]

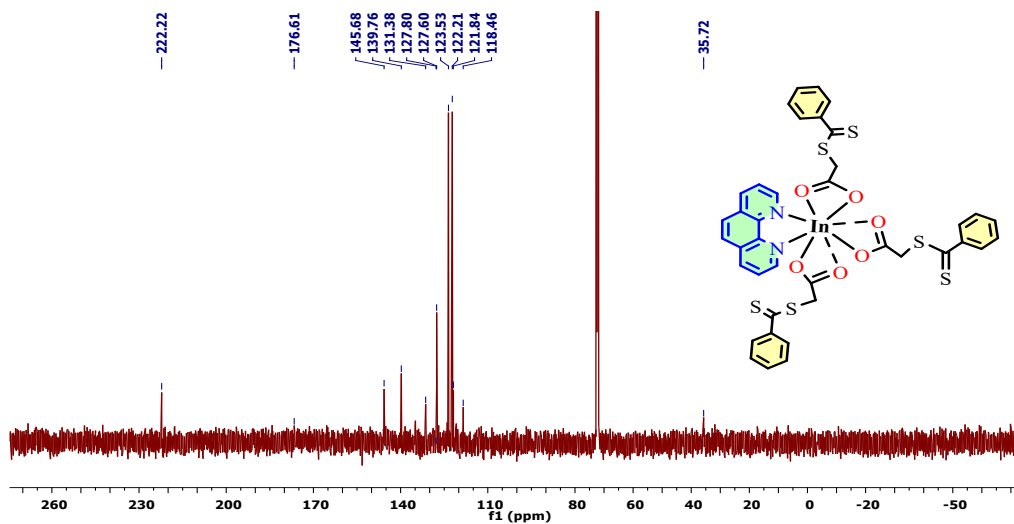


Figure S2 ¹³C NMR spectra of complex 1 [In(1,10phen.)(Stbtg)₃]

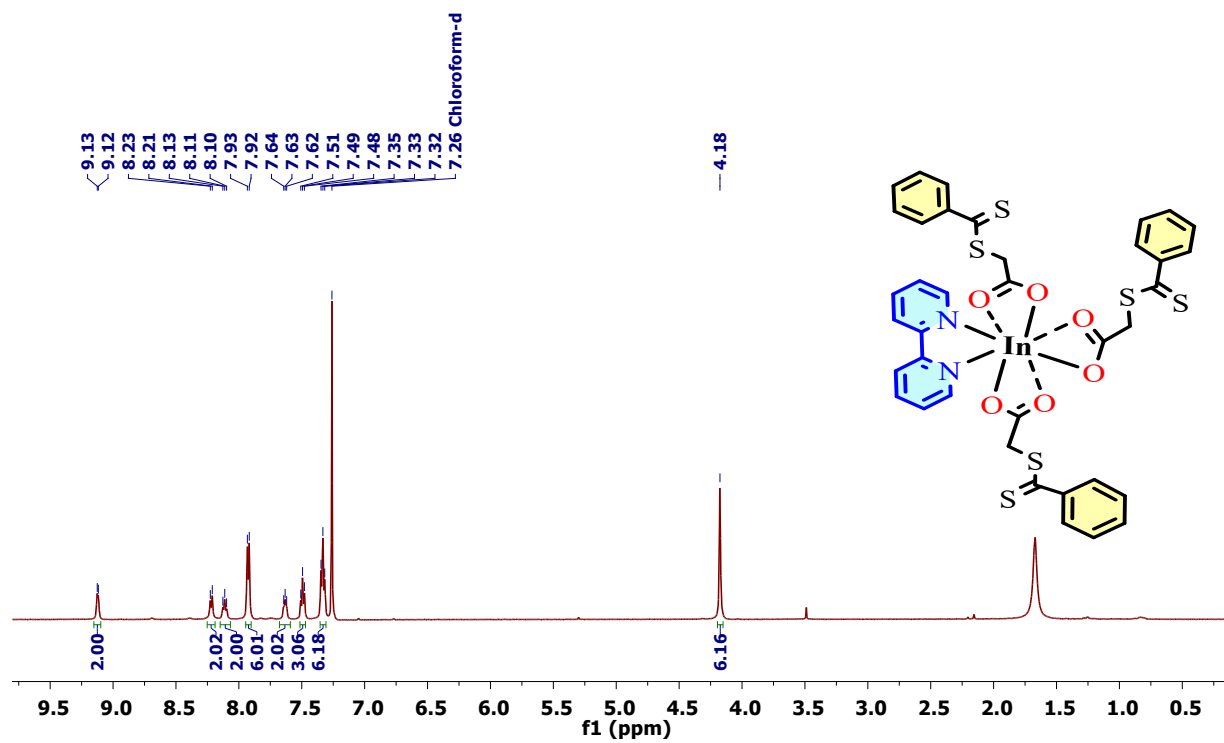


Figure-S3 ¹H NMR spectra of Complex 2, [In(2,2bipyridyl)(stbtg)₃]

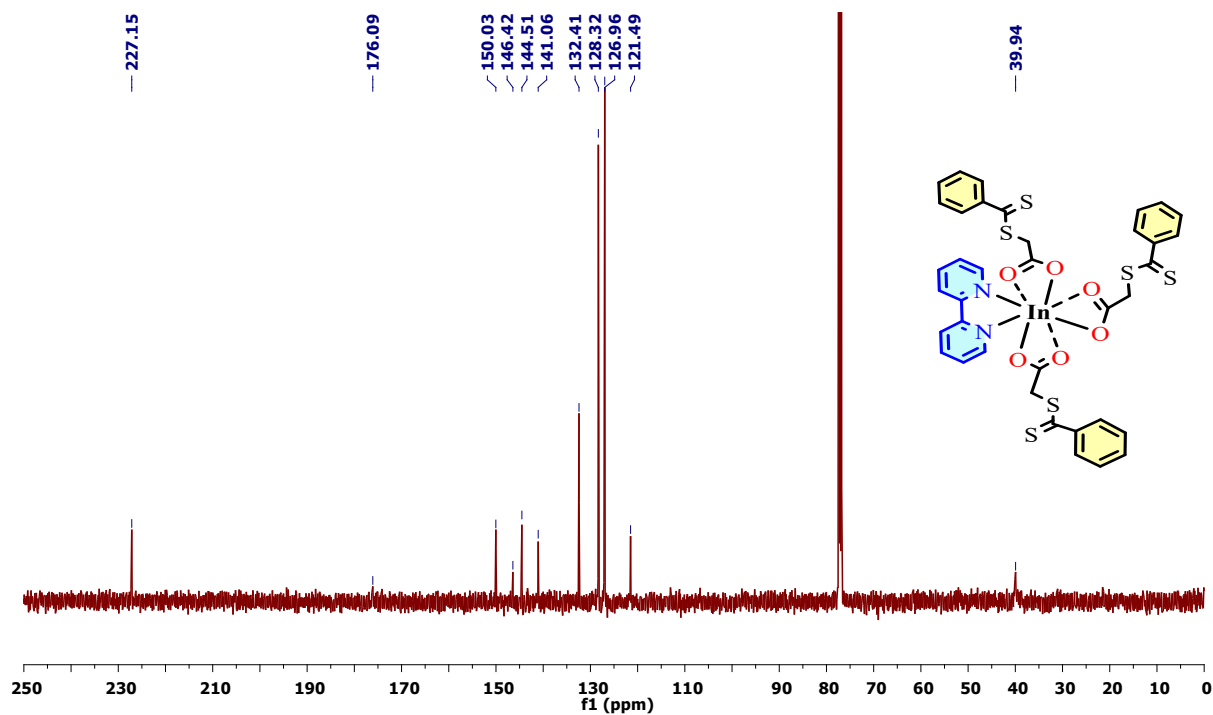


Figure S4. ¹³C NMR spectra of Complex 2, [In(2,2bipyridyl)(Stbtg)₃]

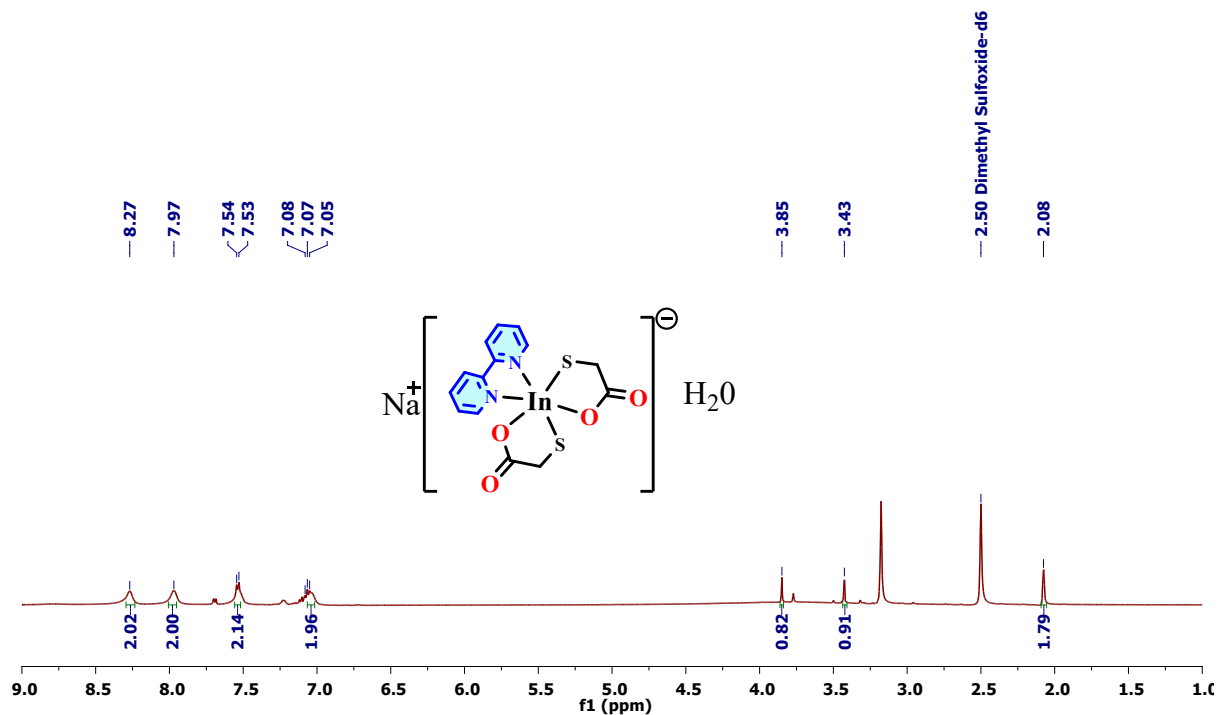


Figure S5. ^1H NMR spectra of Complex 3, $\text{Na} [\text{In} (2,2'\text{-bipyridyl})(\text{SCH}_2\text{COO})_2] \cdot (\text{H}_2\text{O})$

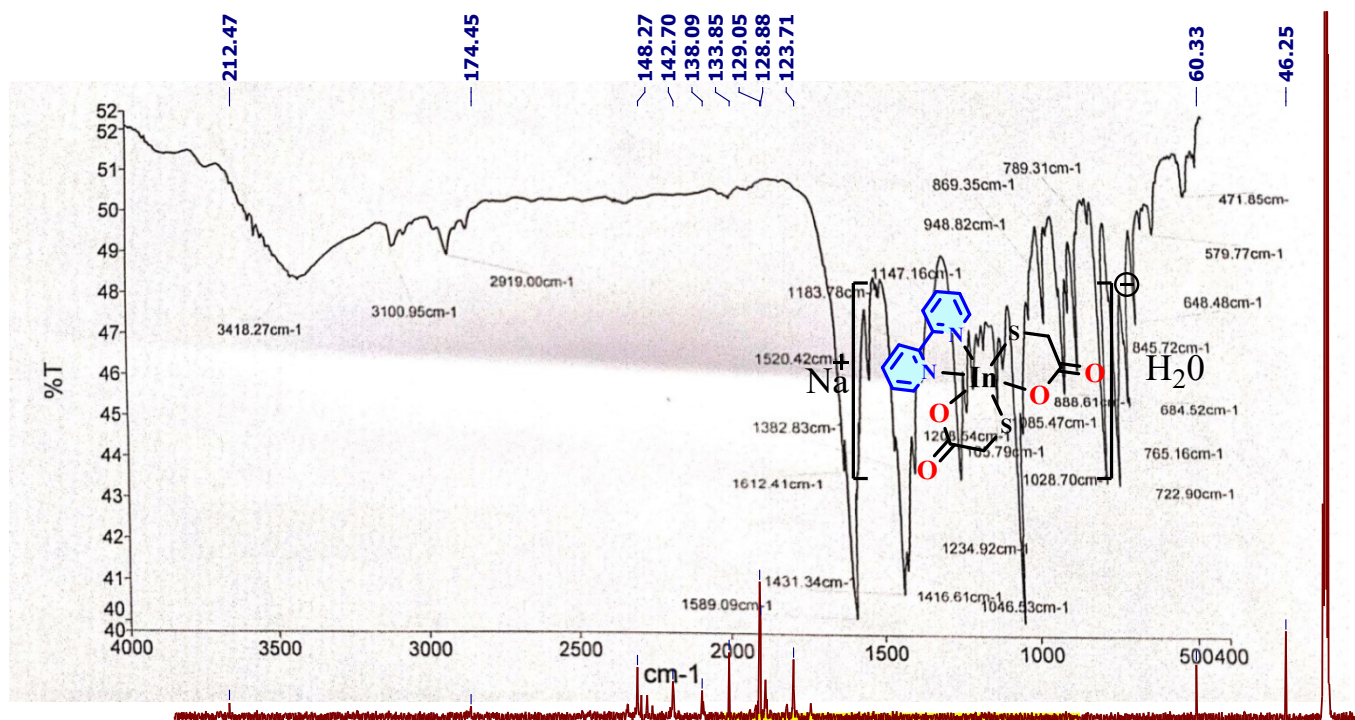


Figure S7. IR spectrum of Complexes $[\text{In}(\text{1,10-phen})(\text{Stbtg})_3] (1)$

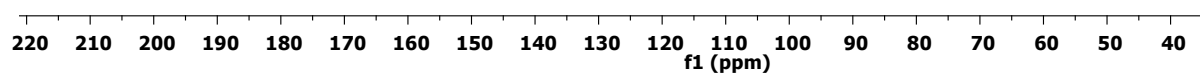


Figure S6 ^{13}C NMR spectrum of Complex 3, $\text{Na} [\text{In} (2,2'\text{-bipyridyl})(\text{SCH}_2\text{COO})_2] \cdot (\text{H}_2\text{O})$

IR Spectra of the complexes

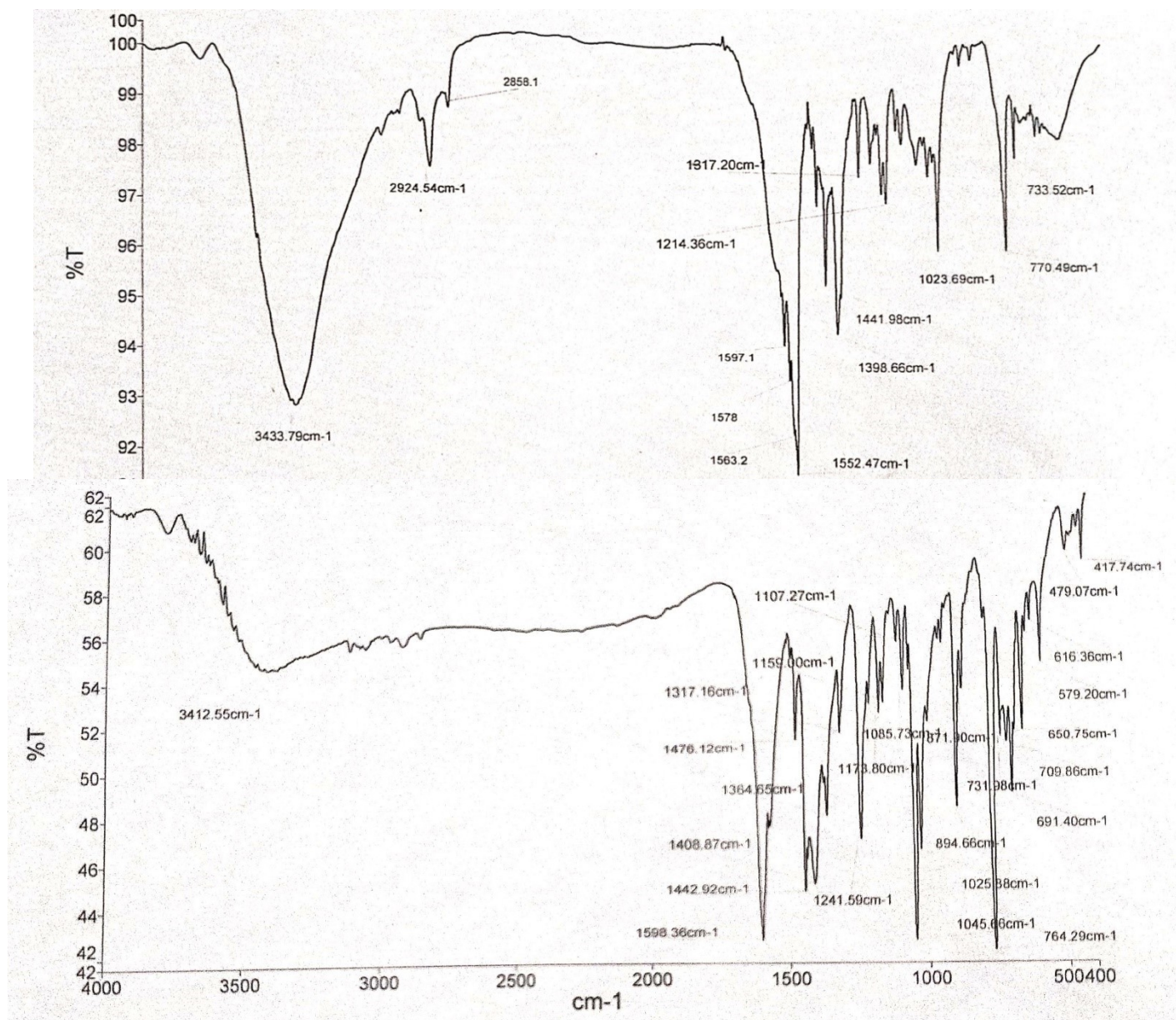


Figure S8. IR spectrum of Complexes $[In(2,2'-bipy)(Stbtg)_3]$ (2)

Single Crystal X-ray diffraction

Crystals with a particular shape were used to gather X-ray intensity data using a Rigaku XtaLAB Synergy Dualflex diffractometer equipped with a HyPix 3000 detector and monochromated MoK α radiation ($\lambda = 0.71073$ Å). Data reduction, scaling, and absorption corrections were performed with the *CrysAlis Pro* software.^{1,2} The structure was solved using the SHELXT program with Intrinsic Phasing and refined by least-squares procedures applying the SHELXL-2018¹ software package through the OLEX2 suite.³ While hydrogen atom positions were geometrically calculated and refined using the riding model. Crystallographic and structure refinement data of the complex is shown in Table S1.

Identification code	Complex 1	Complex 2	Complex 3
CCDC No.	2334707	2405392	2405398
Empirical formula	C ₃₉ H ₂₉ InN ₂ O ₆ S ₆	C ₃₇ H ₂₉ InN ₂ O ₆ S ₆	C ₁₄ H ₁₄ N ₂ O ₅ NaS ₂ In
Formula weight	928.82	904.80	492.20
Temperature/K	293	293	293
Crystal system	monoclinic	orthorhombic	monoclinic
Space group	Pn	Pca2 ₁	P2 ₁ /c
a/Å	10.9278(2)	28.2437(6)	15.7350(3)
b/Å	9.4920(2)	17.5038(3)	7.94580(10)
c/Å	19.0268(4)	14.8073(3)	14.7384(3)
α /°	90	90	90
β /°	97.401(2)	90	110.678(2)

$\gamma/^\circ$	90	90	90
Volume/ $\text{\AA}^3$	1957.14(7)	7320.3(2)	1723.99(6)
Z	2	4	4
$\rho_{\text{calc}}/\text{g cm}^{-3}$	1.576	1.642	1.896
μ/mm^{-1}	0.973	1.038	1.666
Donor NBO (i) F(000)	Acceptor NBO (j) 940.0	E(2) (kcal/mol) 3664.0	E(j)-E(i) (a.u.) 976.0
Radiation	Mo K α ($\lambda = 0.71073$)	Mo K α	Mo K α
LP (2) O 8	LP*(1)In 1	7.29	0.30
2 θ range / $^\circ$	4.804 to 54.136	4.616 to 54.128	5.57 to 54.112
LP (2) O 8	LP*(2)In 1	8.25	0.49
Index ranges	$-13 \leq h \leq 13, -11 \leq k \leq 11, -24 \leq l \leq 23$	$-35 \leq h \leq 35, -20 \leq k \leq 20, -18 \leq l \leq 18$	$-20 \leq h \leq 20, -9 \leq k \leq 9, -18 \leq l \leq 18$
LP (2) O 8	LP*(3)In 1	13.32	0.52
Reflections collected	23670	54672	36996
LP (2) O 10	LP*(1)In 1	19.87	0.35
Independent reflections	7385 [$R_{\text{int}} = 0.0460$, $R_{\text{sigma}} = 0.0569$]	14672 [$R_{\text{int}} = 0.0623$, $R_{\text{sigma}} = 0.0763$]	3676 [$R_{\text{int}} = 0.0529$, $R_{\text{sigma}} = 0.0288$]
LP (2) O 10	LP*(2)In 1	29.67	0.54
Data/restraints/parameters	7385/38/488	14672/1/937	3676/0/230
Goodness-of-fit on F^2	1.037	1.042	1.065
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0420$, $wR_2 = 0.0604$	$R_1 = 0.0570$, $wR_2 = 0.1231$	$R_1 = 0.0273$, $wR_2 = 0.0621$
Final R indexes [all data]	$R_1 = 0.0533$, $wR_2 = 0.0663$	$R_1 = 0.0978$, $wR_2 = 0.1436$	$R_1 = 0.0333$, $wR_2 = 0.0666$
Largest diff.peak/hole / $\text{e}\text{\AA}^{-3}$	0.48/-0.31	1.90/-0.61	0.36/-0.56
Flack parameter	0.03(2)	0.32(2)	

Table S1. Crystal structure information and refinement parameters:

Second Order Perturbation Theory Analysis:

Table S2. For Complexes [In(1,10-phen)(Stbtg)₃] (1)

LP (2) O 11	LP*(1)In 1	20.94	0.31	0.079
LP (2) O 11	LP*(4)In 1	21.19	0.54	0.096
LP (2) O 12	LP*(1)In 1	18.93	0.33	0.077
LP (2) O 12	LP*(3)In 1	21.64	0.56	0.098
LP (2) O 14	LP*(1)In 1	20.32	0.34	0.081
LP (2) O14	LP*(2)In 1	24.42	0.53	0.102
LP (2) O14	LP*(3)In 1	3.09	0.57	0.037
LP (2) O16	LP*(1)In 1	18.32	0.31	0.074
LP (2) O16	LP*(2)In 1	3.48	0.50	0.037
LP (2) O16	LP*(3)In 1	16.01	0.54	0.083
LP (2) O16	LP*(4) In 1	4.08	0.54	0.042
LP (1) N9	LP*(1) In 1	22.50	0.35	0.087
LP (1) N9	LP*(3) In 1	10.65	0.58	0.070
LP (1) N9	LP*(4) In 1	20.56	0.58	0.098
LP (1) N15	LP*(1)In 1	21.95	0.35	0.086
LP (1) N15	LP*(3)In 1	10.00	0.57	0.068
LP (1) N15	LP*(4)In 1	20.60	0.58	0.098

Table S3. For Complex [In(2,2'-bipy)(Stbtg)₃] (2)

Donor NBO (i)	Acceptor NBO (j)	E(2) (kcal/mol)	E(j)-E(i) (a.u.)	F(i,j) (a.u.)
LP (2) O 8	LP*(1)In 1	16.97	0.31	0.070
LP (2) O 8	LP*(3)In 1	19.54	0.53	0.091
LP (2) O 10	LP*(1)In 1	16.25	0.32	0.071
LP (2) O 10	LP*(3)In 1	17.28	0.55	0.087
LP (2) O 11	LP*(1)In 1	23.06	0.37	0.089
LP (2) O 11	LP*(2)In 1	23.26	0.55	0.101
LP (2) O 11	LP*(3)In 1	7.99	0.59	0.062
LP (2) O 12	LP*(1)In 1	20.46	0.35	0.082

LP (2) O 12	LP*(2)In 1	27.94	0.53	0.109
LP (2) O 13	LP*(1)In 1	12.73	0.29	0.059
LP (2) O 13	LP*(2)In 1	10.47	0.48	0.063
LP (2) O 13	LP*(3)In 1	9.02	0.51	0.061
LP (2) O 16	LP*(1)In 1	26.21	0.34	0.091
LP (2) O 16	LP*(4)In 1	24.69	0.56	0.106
LP (3) O 16	LP*(2)In 1	1.95	0.38	0.025
LP (1) N 9	LP*(1)In 1	22.78	0.35	0.087
LP (1) N 9	LP*(3)In 1	5.88	0.57	0.052
LP (1) N 9	LP*(4)In 1	24.21	0.57	0.105
LP (1) N 80	LP*(1)In 1	20.79	0.35	0.083
LP (1) N 80	LP*(3)In 1	11.43	0.57	0.073
LP (1) N 80	LP*(4)In 1	16.18	0.57	0.086

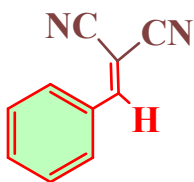
Table S4. For Complex Na[In(2,2'-bipy)(SCH₂COO)₂].(H₂O) (3)

Donor NBO (i)	Acceptor NBO (j)	E(2) (kcal/mol)	E(j)-E(i) (a.u.)	F(i,j) (a.u.)
LP (3) S 2	LP (1)In 1	115.54	0.22	0.151
LP (3) S 2	LP*(2)In 1	42.16	0.40	0.119
LP (3) S 2	LP*(3)In 1	15.89	0.37	0.072
LP (3) S 2	LP*(4)In 1	2.72	0.36	0.029
LP (3) S 3	LP (1)In 1	111.93	0.23	0.155
LP (3) S 3	LP*(2)In 1	49.67	0.42	0.131
LP (3) S 3	LP*(3)In 1	3.81	0.39	0.036
LP (3) S 3	LP*(4)In 1	2.00	0.38	0.026
LP (2) O 4	LP (1)In 1	11.75	0.49	0.078
LP (2) O 4	LP*(2)In 1	10.93	0.68	0.080

LP (2) O 4	LP*(3)In 1	24.52	0.65	0.115
LP (2) O 4	LP*(4)In 1	8.21	0.64	0.066
LP (2) O 5	LP (1)In 1	19.19	0.53	0.104
LP (2) O 5	LP*(4)In 1	39.74	0.67	0.149
LP (1) N 7	LP (1)In 1	12.16	0.39	0.070
LP (1) N 7	LP*(2)In 1	13.11	0.57	0.080
LP (1) N 7	LP*(3)In 1	24.58	0.55	0.106
LP (1) N 21	LP (1)In 1	19.85	0.36	0.087
LP (1) N 21	LP*(3)In 1	10.82	0.52	0.068
LP (1) N 21	LP*(4)In 1	18.48	0.51	0.087
LP (2) O 37	LP*(1)Na 36	4.17	1.01	0.059
LP (1) O 12	LP*(1)Na 36	5.57	1.10	0.071
. LP (1) O 12	LP*(2)Na 36	6.39	0.98	0.071
LP (1) O 12	LP*(3)Na 36	11.48	0.92	0.092
LP (1) O 12	LP*(4)Na 36	2.49	0.92	0.043
LP (2) O 37	LP*(2)Na 36	6.09	0.89	0.066
LP (2) O 37	LP*(4)Na 36	15.17	0.82	0.101
LP (2) O 40	LP*(1)Na 36	9.30	0.96	0.086
LP (2) O 40	LP*(2)Na 36	5.77	0.84	0.063
LP (2) O 40	LP*(3)Na 36	12.21	0.78	0.088
LP (2) O 43	LP*(1)Na 36	11.92	0.95	0.096
LP (2) O 43	LP*(2)Na 36	4.89	0.83	0.058
LP (2) O 41	LP*(1)Na 36	11.29	0.96	0.094
LP (2) O 41	LP*(2)Na 36	11.52	0.84	0.088
LP (2) O 42	LP*(1)Na 36	10.32	1.03	0.093
LP (2) O 42	LP*(2)Na 36	11.52	0.91	0.092
LP*(1)Na 44	LP*(1)Na 45	17.81	0.04	0.076

Data S4-NMR (¹H and ¹³C) spectra of 2 component Knoevenagel Condensation Products:

4a. 2-benzylidenemalononitrile⁴



M. F. $C_{11}H_8N_2$ (168.20), Yield: (0.156 g, 93%). White powder. 1H NMR (500 MHz, $CDCl_3$, ppm) δ 7.54 (t, $J=7.5$ Hz, 2H), 7.64 (t, $J=8$ Hz, 2H), 7.78 (s, 1H), 2.46 (d, 2H); ^{13}C NMR (125 MHz, $CDCl_3$, ppm) δ 160.29, 134.99, 131.34, 131.10, 130.02, 114.07, 112.91, 83.33;

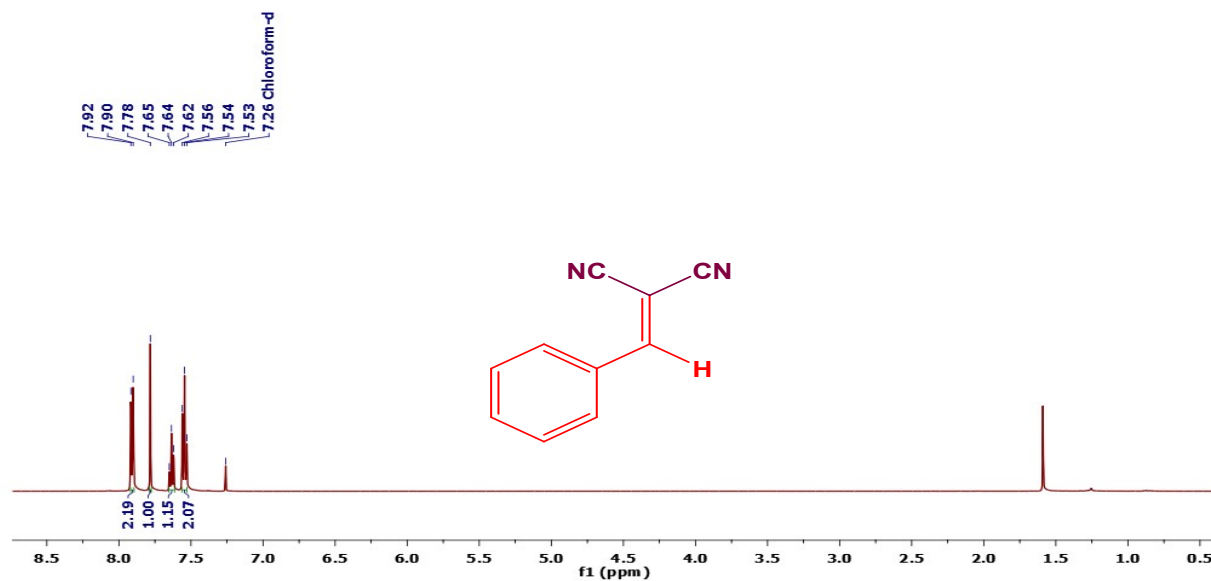


Figure S10. 1H NMR spectra of 4a

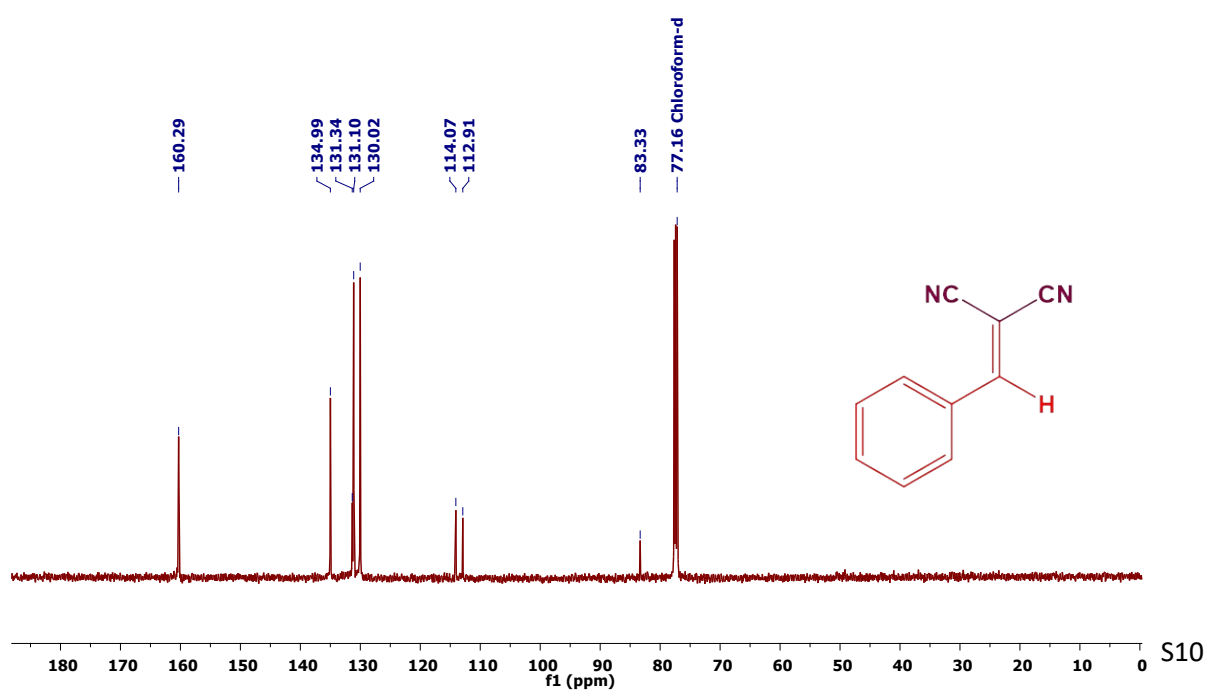
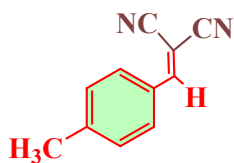


Figure S11. ^{13}C NMR spectra of 4a

4b. 2-(4-methylbenzylidene)malononitrile⁴



M. F. C₁₁H₈N₂ (168.20). Yield: (0.159 g, 95%). White powder. ¹H NMR (500 MHz, CDCl₃, ppm) δ 7.81 (d, *J* = 8 Hz, 2H), 7.72 (s, 1H), 7.34 (d, *J* = 8 Hz, 2H), 2.46 (s, 3H); ¹³C NMR (125 MHz, CDCl₃, ppm) δ 159.81, 146.44, 130.98, 130.45, 128.57, 114.07, 112.92, 81.34, 22.05;

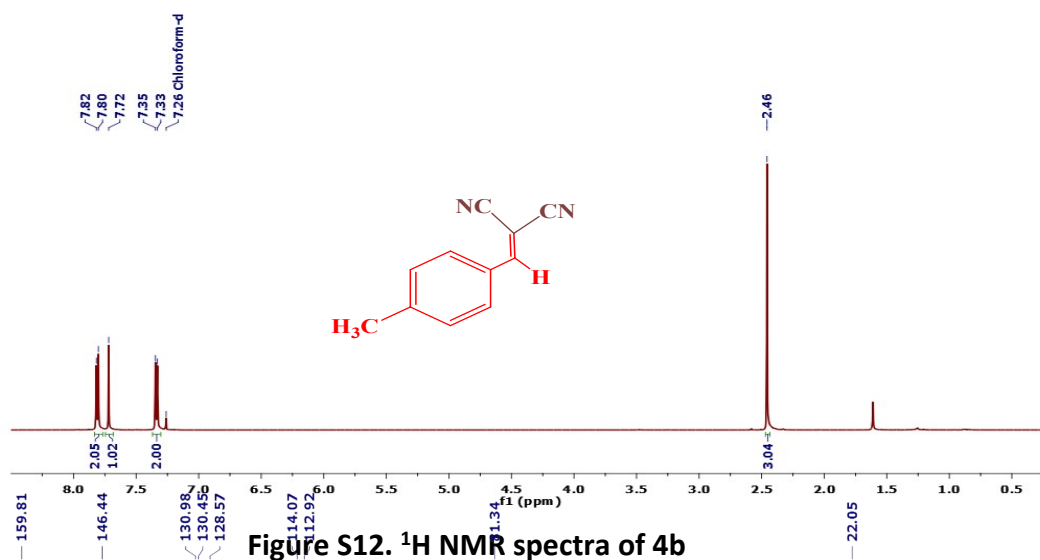


Figure S12. ¹H NMR spectra of 4b

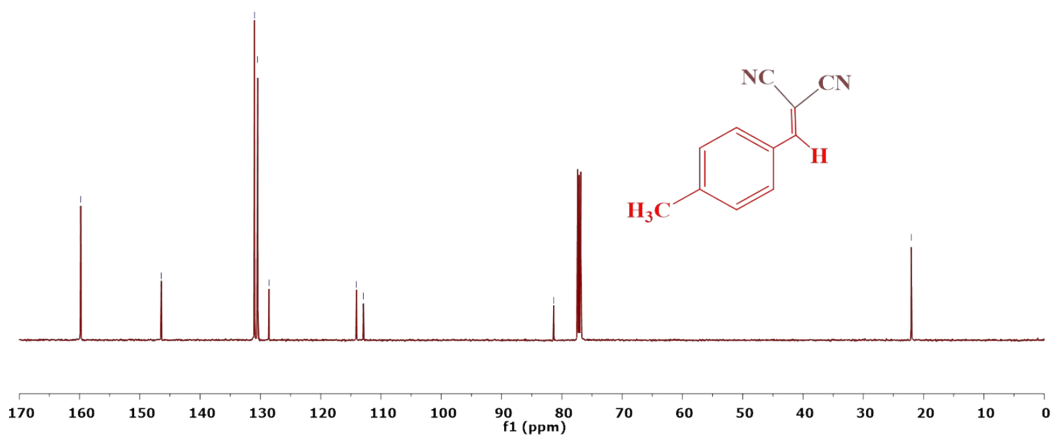
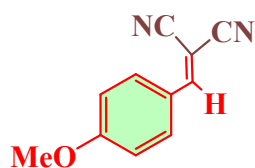


Figure S13. ¹³C NMR spectra of 4b

4c. 2-(4-methoxybenzylidene)malononitrile⁴



M. F. C₁₁H₈N₂O (184.20). Yield: (0.174 g, 95%). yellow powder. ¹H NMR (500 MHz, CDCl₃, ppm) δ 7.91 (d, *J* = 9.5 Hz, 2H), 7.65 (s, 1H), 7.34 (d, *J* = 9.5 Hz, 2H), 3.91 (s, 3H); ¹³C NMR (125 MHz, CDCl₃, ppm) δ 164.82, 158.90, 133.51, 124.12, 115.23, 114.48, 112.49, 78.71, 55.87

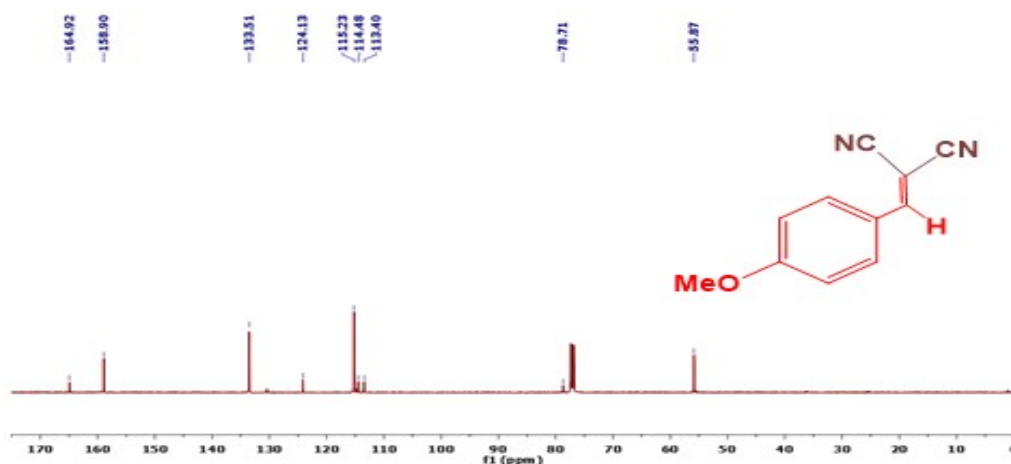
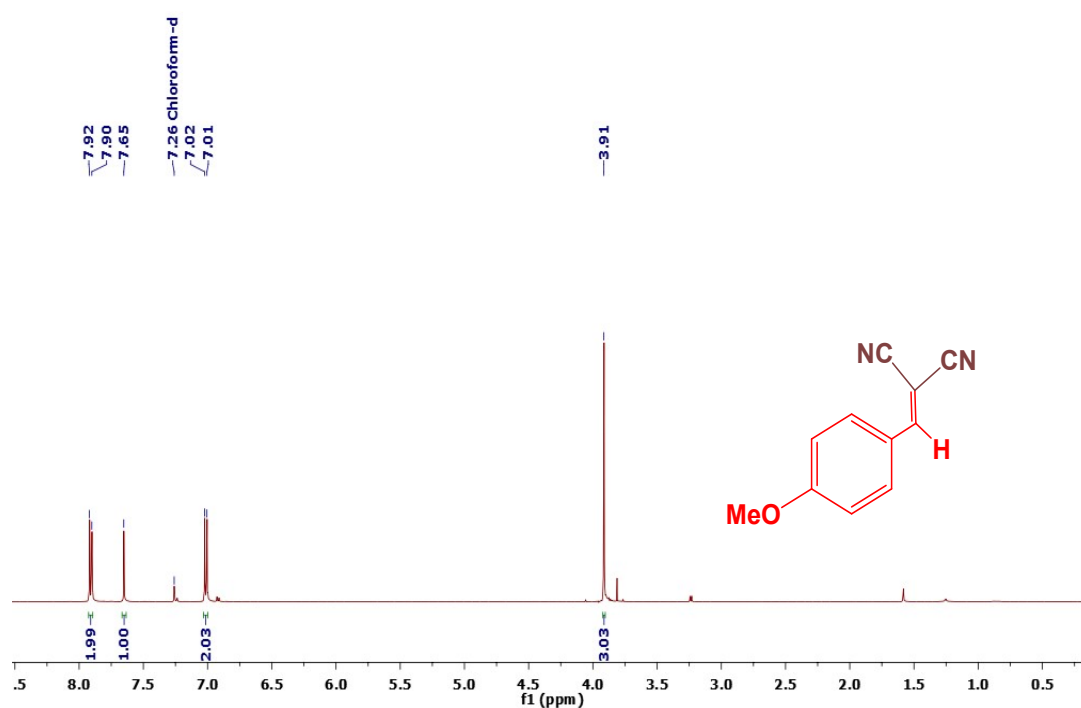
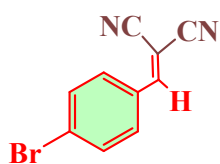


Figure S15. ¹³C NMR spectra of 4c

4d. 2-(4-bromobenzylidene)malononitrile⁵



M.F. $C_{10}H_5N_2Br$ (233.07) Yield: (0.214g, 92%). White powder. 1H NMR (500 MHz, $CDCl_3$, ppm) δ 8.38 (d, J = 9 Hz, 2H), 8.07(d, J = 8.5 Hz, 2H), 7.88 (s, 1H), ^{13}C NMR (125 MHz, $CDCl_3$, ppm) δ 158.42, 133.16, 131.85, 129.98, 129.75, 113.49, 112.38, 83.67

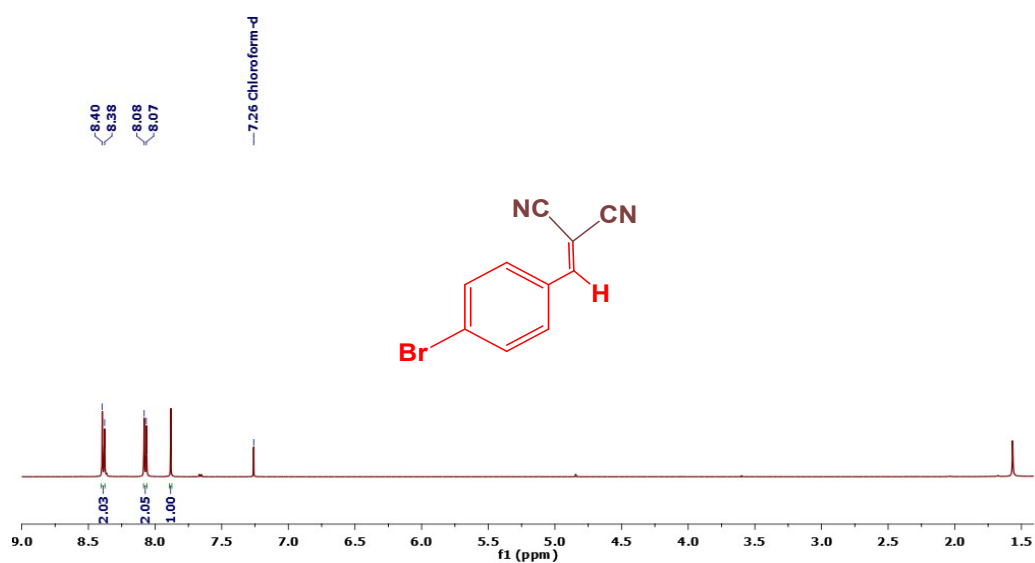


Figure S16. 1H NMR spectra of 4d

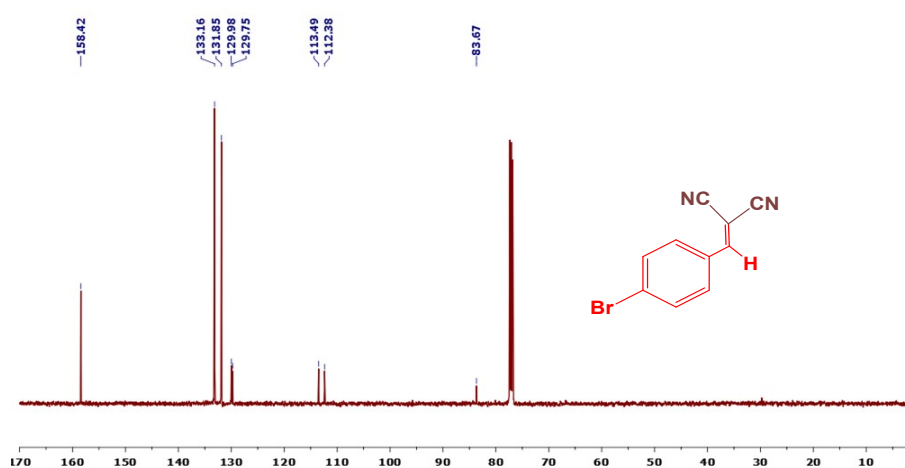
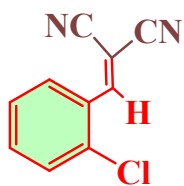


Figure S17. ^{13}C NMR spectra of 4d

4e. 2-(2-chlorobenzylidene)malononitrile⁴



M.F. $C_{10}H_5N_2Cl$ (188.61), Yield: (0.169 g, 92%). White powder. 1H NMR (500 MHz, $CDCl_3$, ppm) δ 8.27 (s, 1H), 8.18 (d, $J = 7.5$ Hz, H), 7.55 (d, $J = 3.5$ Hz, 2H); 7.45(m, 1H), ^{13}C NMR (125 MHz, $CDCl_3$, ppm) δ 156.06, 136.41, 135.07, 130.78, 129.16, 129.16, 127.85, 113.26, 111.95, 85.94;

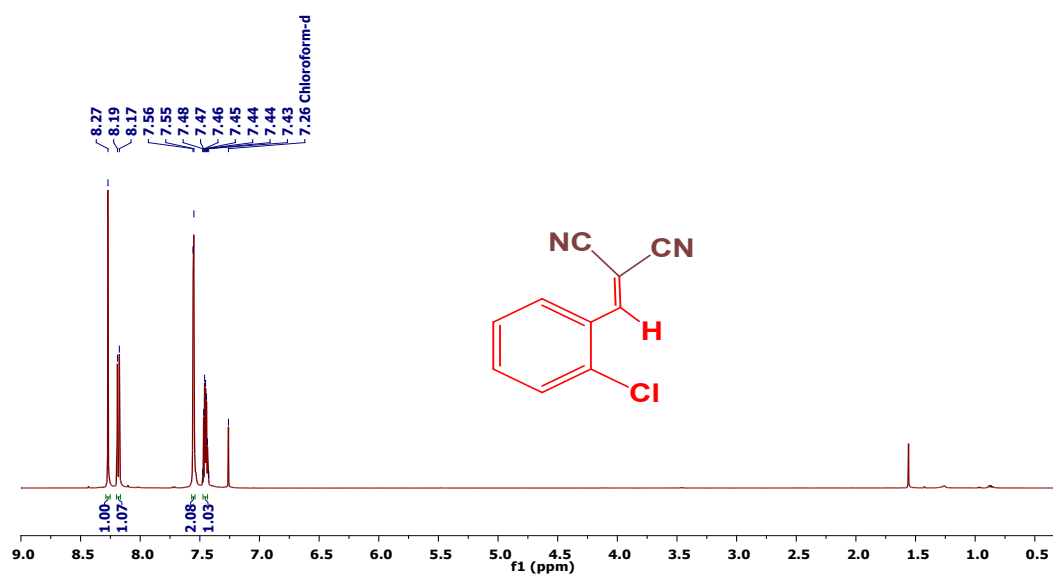


Figure S18. ¹H NMR spectra of 4e

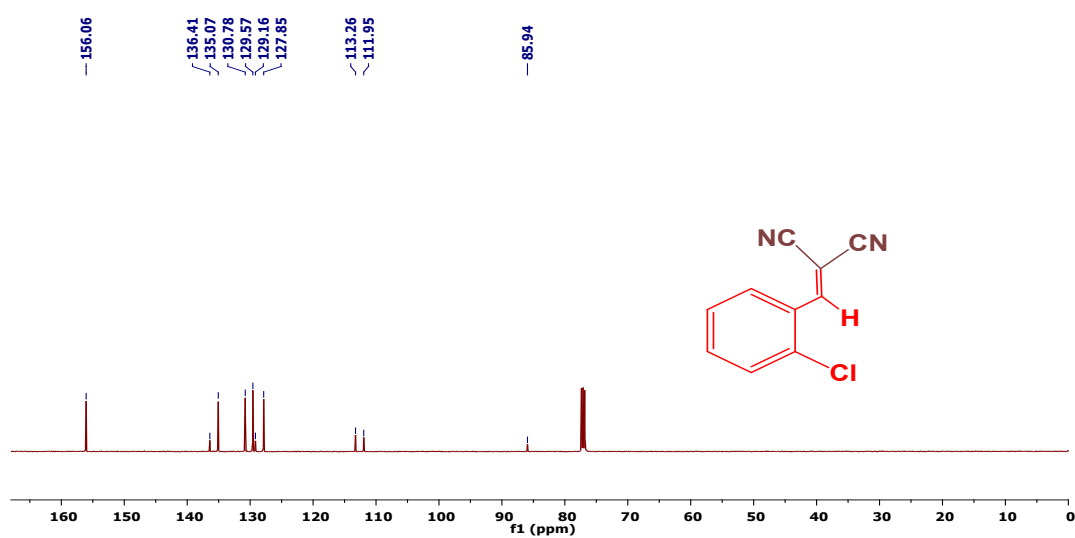
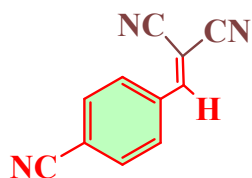


Figure S19. ¹³C NMR spectra of 4e

4f. 2-(4-cyanobenzylidene)malononitrile⁶



M.F.C₁₁H₅N₃ (199.17), Yield: (0.181g, 91%). Light yellow powder. ¹H NMR (500 MHz, CDCl₃, ppm) δ 7.98 (d, *J* = 8 Hz, 2H), 7.82 (d, *J* = 8.5 Hz, 2H); 7.81 (s, 1H), ¹³C NMR (125 MHz, CDCl₃, ppm) 157.36, 134.33, 133.21, 130.75, 117.37, 112.79, 111.74, 87.03;

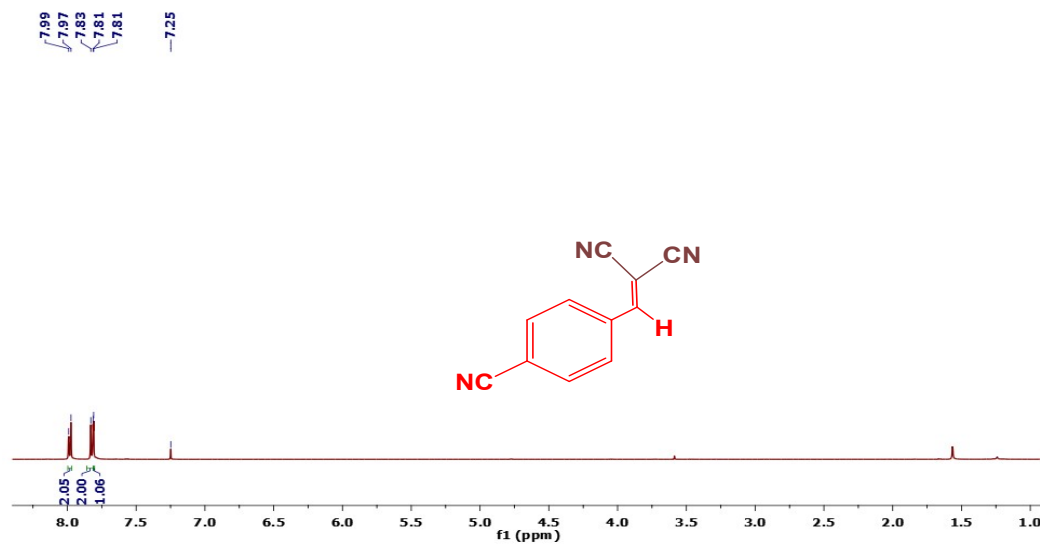


Figure S20. ^1H NMR spectra of 4f

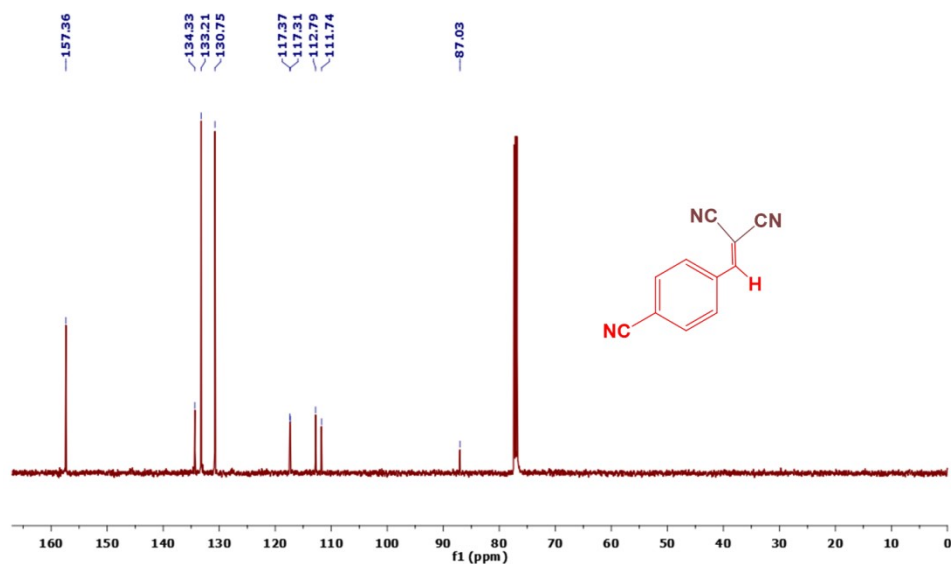
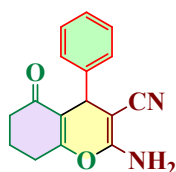


Figure S21. ^{13}C NMR spectra of 4f

The Products of Multicomponent reaction

Data S5 -NMR (^1H and ^{13}C) spectra of 2- Amino-4H-chromeneDerivatives of cyclohexane 1,3-dione

5a. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-phenyl-4H-chromene-3-carbonitrile⁴



M. F. C₁₆H₁₄N₂O₂ (266.29), Yield: (0.239g, 90%), white powder. ¹H NMR (500 MHz, CDCl₃, ppm) δ 7.31 – 7.19 (m, 5H), 4.51 (s, 2H), 4.44 (s, 1H), 2.66-2.53 (m, 2H), 2.43 – 2.31 (m, 2H), 2.09 – 1.98 (m, 2H) ; ¹³C NMR (125 MHz, CDCl₃, ppm) δ :195.95, 163.32, 157.48, 143.23, 128.66, 127.62, 127.24, 118.66, 115.38, 62.07, 36.87, 35.44, 27.08, 20.19

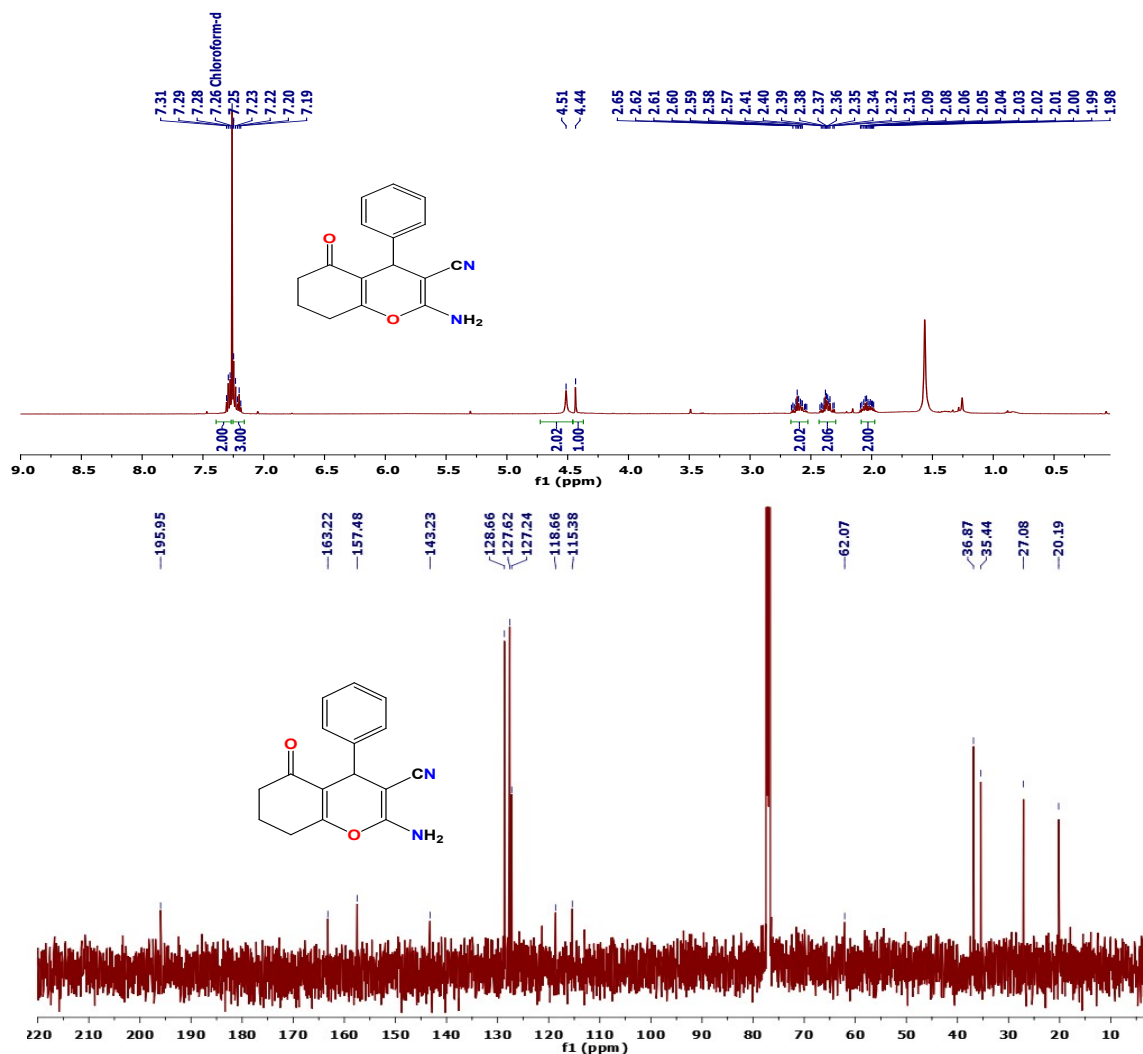
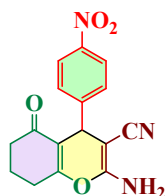


Figure S23. ¹³C NMR Spectrum of 5a

5b. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-(4-nitrophenyl)-4H-chromene-3-carbonitrile⁶



M. F. C₁₆H₁₃N₃O₄ (311.29), Yield: (0.255 g, 82%), off white powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) :δ 8.11 (d, *J* = 9 Hz, 2H), 7.45 (d, *J* = 9 Hz, 2H), 7.17 (s, 2H), 4.36 (s, 1H), 2.67–2.58 (m, 2H), 2.35 – 2.21 (m, 2H), 2.00 – 1.88 (m, 2H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ

196.43, 165.66, 159.06, 152.84, 146.77, 129.10, 124.17, 119.88, 113.24, 57.42, 36.73, 36.09, 27.04, 20.25.

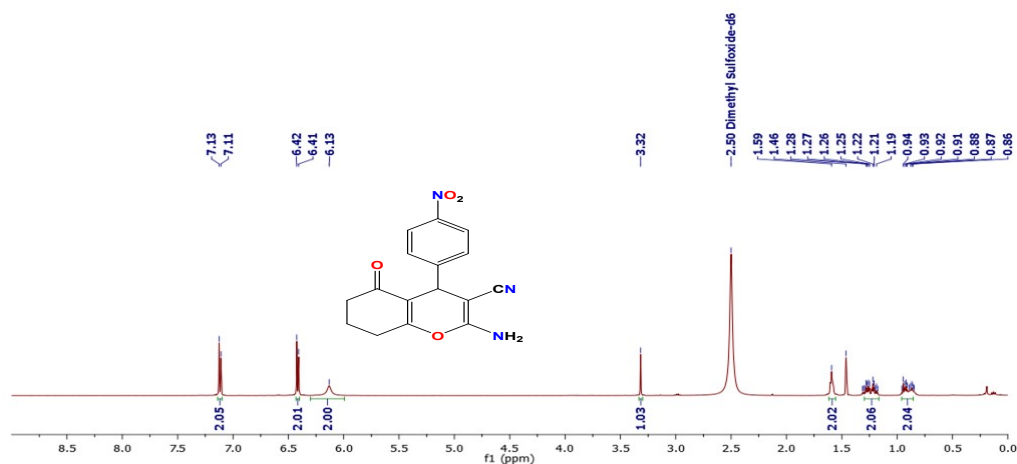


Figure S24. ¹H NMR Spectrum of 5b

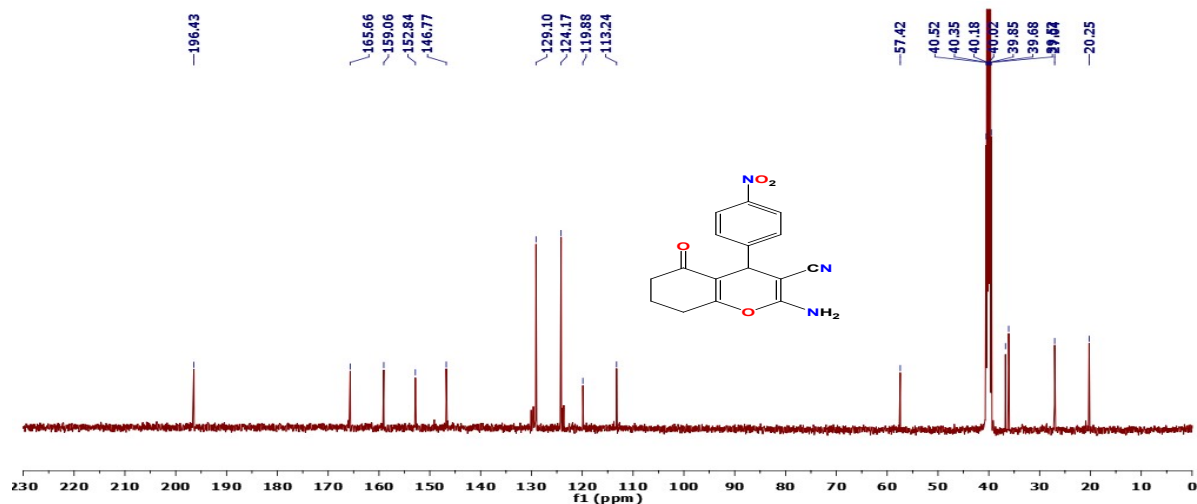
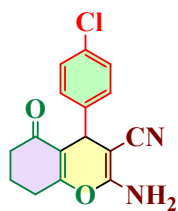


Figure S25. ¹³C NMR Spectrum of 5b

5c. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-(4-chlorophenyl)-4H-chromene-3-carbonitrile⁴



M. F. C₁₆H₁₃N₂O₃Cl (300), Yield: (0.264g, 88%), Pale yellow powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) δ 7.33 (d, *J* = 9 Hz, 2H), 7.18 (d, *J* = 8 Hz, 2H), 7.05 (s, 2H), 4.19 (s, 1H), 2.64 – 2.58 (m, 2H), 2.33 – 2.21 (m, 2H), 1.98 – 1.83 (m, 2H); ¹³C NMR (125 MHz,

DMSO-d₆, ppm) δ 196.41, 165.17, 158.98, 144.31, 131.60, 129.61, 128.79, 120.11, 113.87, 58.21, 36.80, 35.51, 29.99, 26.99, 20.28

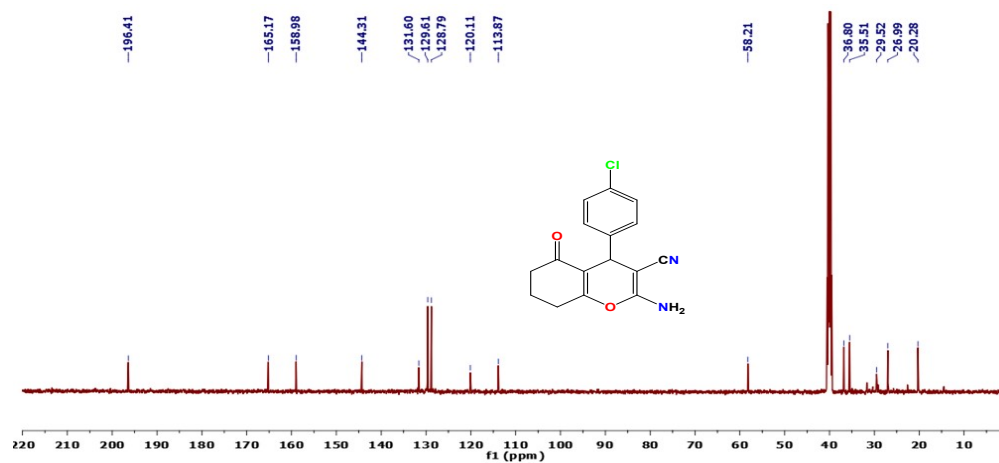
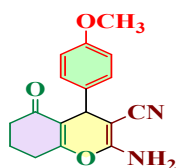


Figure S27. ¹H NMR Spectrum of 5c

5d. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-(4-methoxyphenyl)-4H-chromene-3-carbonitrile ⁷



M. F. C₁₇H₁₆N₂O₃ (296.32), Yield: (0.281 g, 95%), pale yellow powder. ¹H NMR (500 MHz, CDCl₃, ppm) δ 7.16 (d, *J* = 9.5 Hz, 2H), 6.82 (d, *J* = 8 Hz, 2H), 4.55 (s, 2H), 4.38 (s, 1H), 3.716(s, 3H), 2.64 – 2.51 (m, 2H), 2.40 – 2.33 (m, 2H), 2.07 – 1.94 (m, 2H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ 196.15, 163.04, 158.69, 157.44, 135.59, 128.72, 118.72, 115.54, 114.03, 63.74, 55.30, 36.90, 34.70, 27.06, 20.20

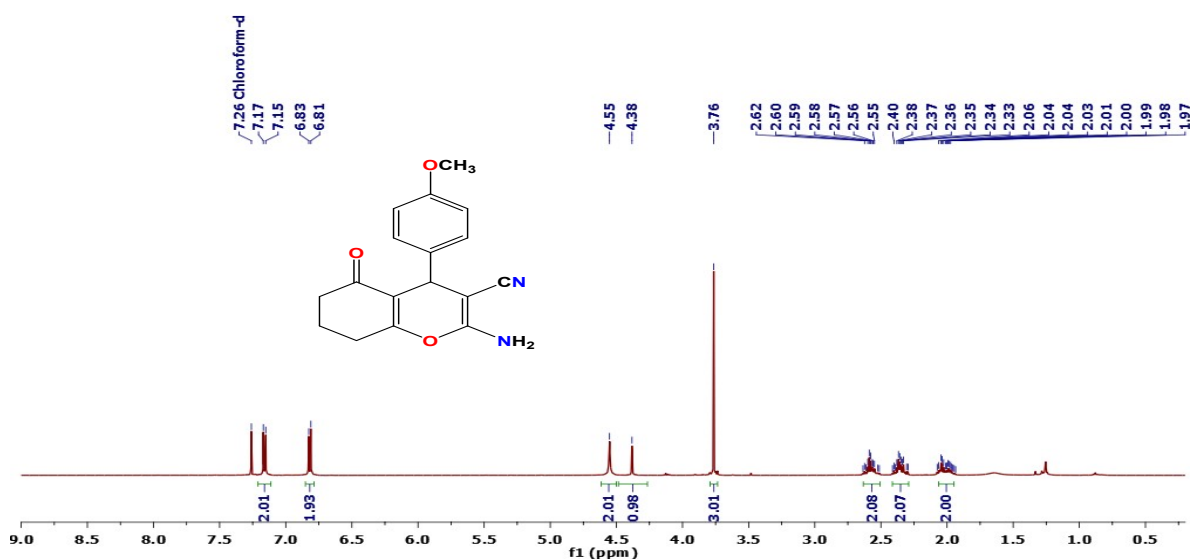


Figure S28. ¹H NMR Spectrum of 5d

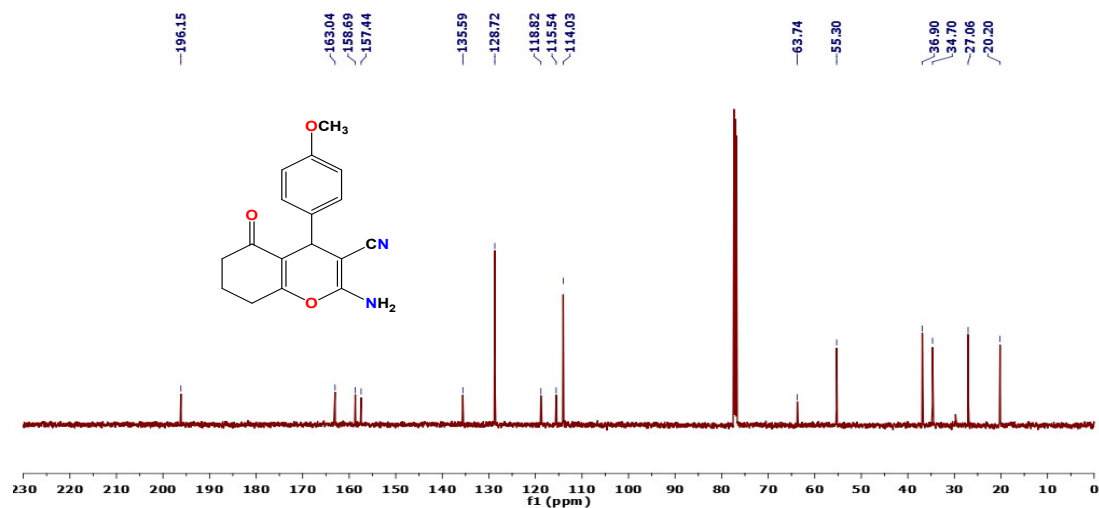
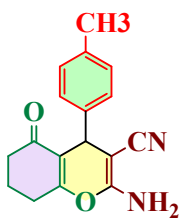


Figure S29. ¹³C NMR Spectrum of 5d

5e. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-(4-methylphenyl)-4H-chromene-3-carbonitrile^{4,6}



M. F. $C_{16}H_{16}N_2O_2$ (280.32), Yield: (0.266 g, 95%), white powder. 1H NMR (500 MHz, $CDCl_3$, ppm) δ 7.13 (d, J = 8 Hz, 2H), 7.09 (d, J = 8 Hz, 2H), 4.55 (s, 2H), 4.39 (s, 1H), 2.64 – 2.52 (m, 2H), 2.39 – 2.32 (m, 2H), 2.29 (s, 3H), 2.08 – 1.95 (m, 2H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ 196.10, 163.17, 157.47, 140.38, 136.79, 127.49, 118.80, 115.45, 63.69, 36.89, 35.06, 29.76, 27.08, 21.15, 20.19

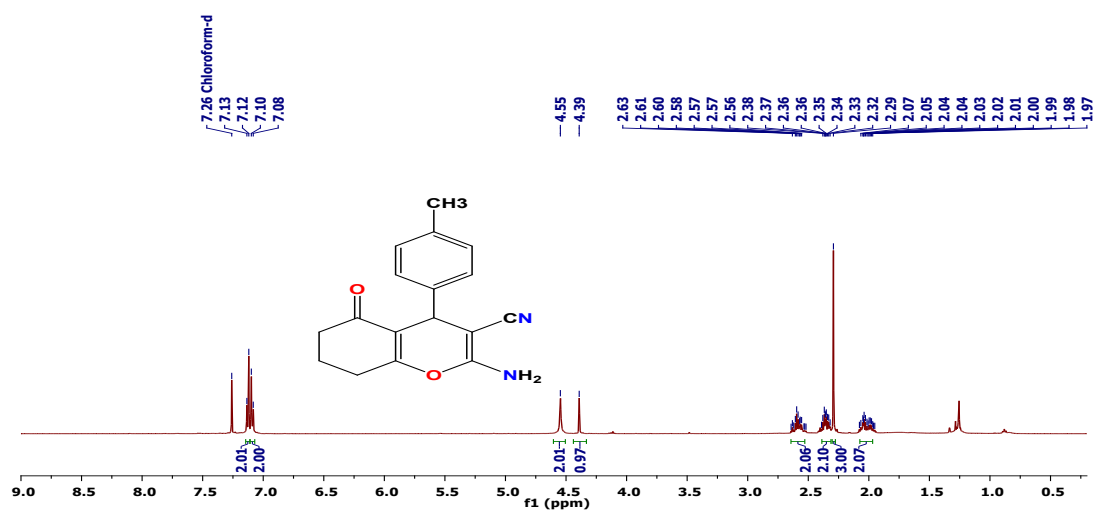


Figure S30. ¹H NMR Spectrum of 5e

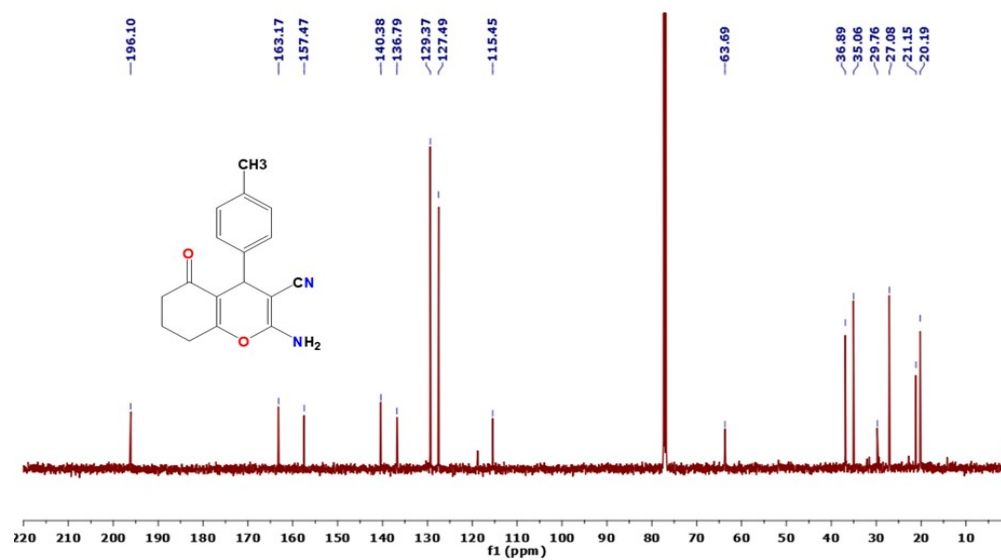
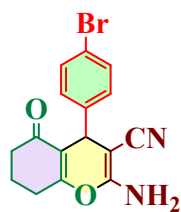


Figure S31. ¹³C NMR Spectrum of 5e

5f. 2-amino-6,6,7,7,8,8-hexahydro-5-oxo-4-(4-Bromophenyl)-4H-chromene-3-carbonitrile⁷



M. F. $C_{16}H_{13}N_2O_3Br$ (345.19), Yield: (0.290g, 84%), Pale yellow powder. 1H NMR (500 MHz, DMSO- d_6 , ppm) δ 7.47 (d, J = 8 Hz, 2H), 7.12 (d, J = 8Hz, 2H), 7.03 (s, 2H), 4.18 (s, 1H), 2.64 – 2.58 (m, 2H), 2.33

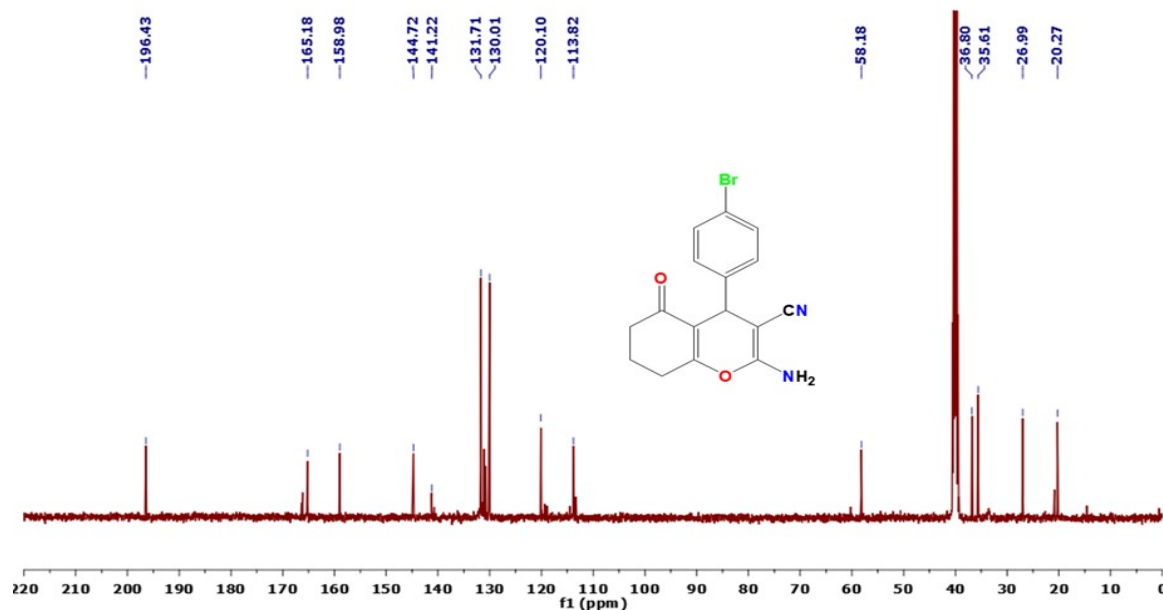


Figure S33. 1H NMR Spectrum of 5f

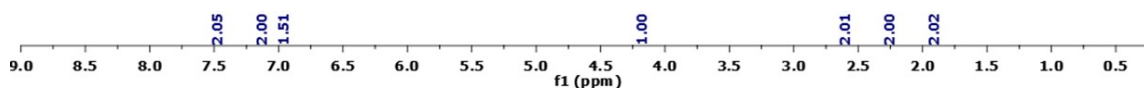
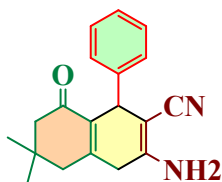


Figure S32. 1H NMR Spectrum of 5f

– 2.21 (m, 2H), 1.98 – 1.84 (m, 2H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ 196.43, 165.18, 158.98, 144.72, 141.22, 131.71, 130.01, 120.10, 113.82, 58.18, 36.80, 35.61, 26.99, 20.27

Datas6- NMR (1H and ^{13}C) spectra of 2- Amino-4H-chromeneDerivatives of 5,5' dimethyl cyclohexane 1,3-dione

6a. 2-amino-6,6,8,8-tetrahydro-7,7-dimethyl-5-oxo-4-phenyl-4H-chromene-3-carbonitrile⁴



M. F. $C_{18}H_{18}N_2O_2$ (294.34). Yield: (0.264g, 90%). Pale yellow powder. 1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 7.26 (t, J = 7 Hz, 2H), 7.16 (t, J = 7.5 Hz, 1H), 7.13 (d, J = 7 Hz, 2H), 6.98(s,2H),4.15 (s, 1H), 2.49(d, J = 4.5 Hz, 2H), 2.09 (d, J = 16 Hz, 1H), 2.08 (d, J =16Hz 1H), 102 (s, 3H), 0.93 (s, 3H); ^{13}C NMR (125 MHz, DMSO d_6 , ppm) δ 196.22, 163.04, 159.02, 145.26, 128.85, 127.66, 127.10, 120.24, 113.26, 58.86, 50.50, 32.32, 28.91, 27.32.



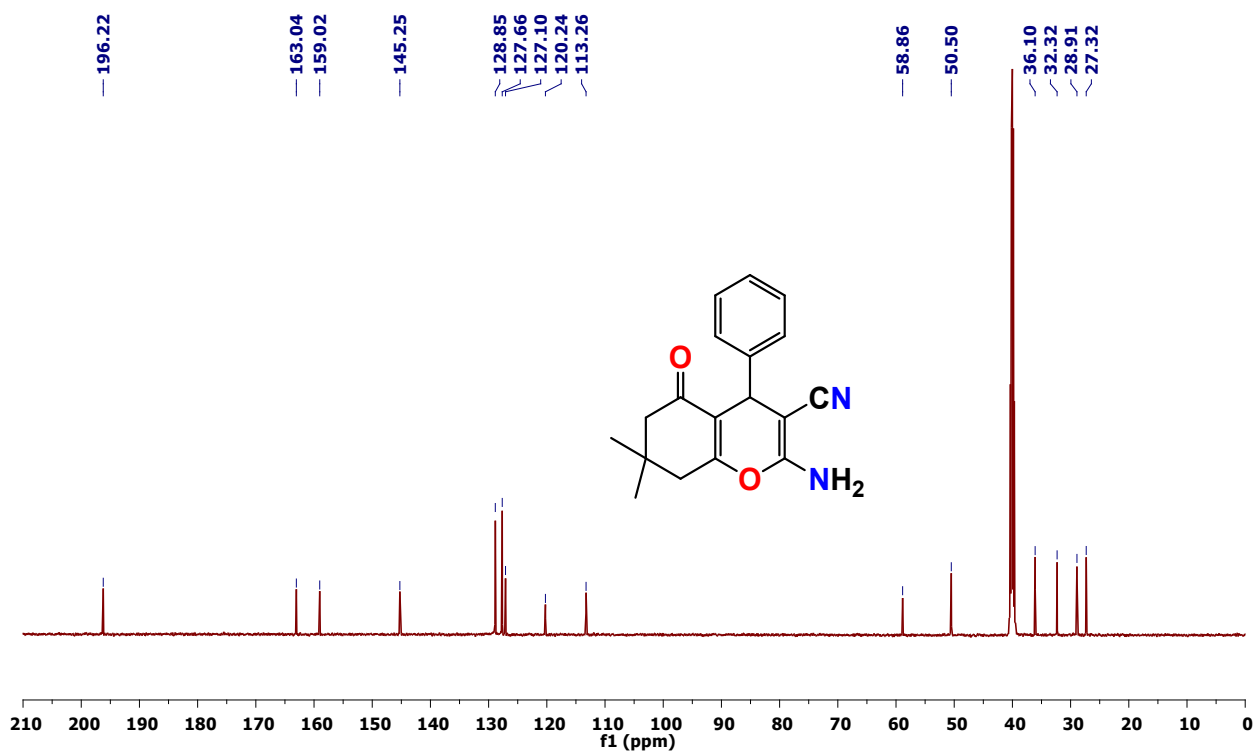
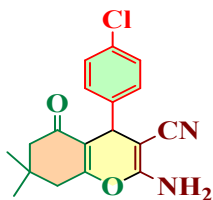


Figure S35. ¹³C NMR Spectrum of 6a

6b.2-amino-6,6,8,8-tetrahydro-7,7-dimethyl-5-oxo-4-(4-Chlorophenyl)-4H-chromene-3-carbonitrile⁴



M.F. C₁₆H₁₃ClN₂O₂ (328.79), Yield: (0.270g, 82%). White powder, ¹H NMR (500 MHz, DMSO d₆, ppm) δ 7.34 (d, *J* = 8.5 Hz, 2H), 7.17 (d, *J* = 8.0 Hz, 2H), 7.05 (s, 2H), 4.19 (s, 1H), 2.24 (d, *J* = 16 Hz, 2H), 2.09 (d, *J* = 16.5 Hz, 1H), 1.24 (d, *J* = 10 Hz, 1H), 1.17 (t, *J* = 6.5 Hz, 1H), 103 (s, 3H), 0.94 (s, 3H). ¹³C NMR (125 MHz, DMSO d₆, ppm) δ 196.30, 163.20, 159.01, 144.26, 131.64, 129.64, 128.82, 120.10, 112.82, 58.29, 50.45, 35.62, 32.32, 28.82, 27.36.

7.35
7.33
7.17
7.16
7.05

4.19

2.50 Dimethyl Sulfoxide-d₆

2.26
2.22
2.11
2.08

1.23
1.18
1.17
1.15
1.03
0.94

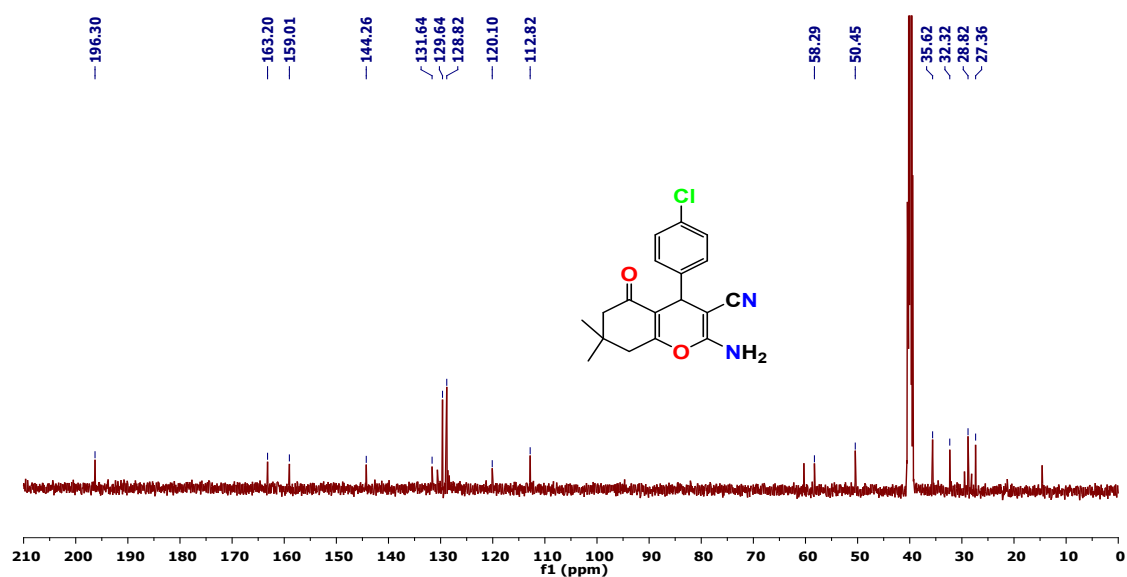
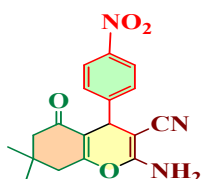
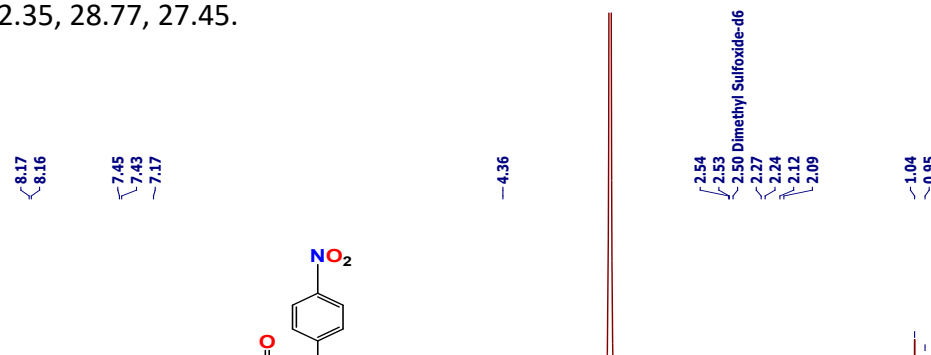


Figure S37. ¹³C NMR Spectrum of 6b

6c. 2-amino-6,6,8,8-tetrahydro-7,7-dimethyl-5-oxo-4-(4-Nitrophenyl)-4H-chromene-3-carbonitrile⁷



M. F. C₁₈H₁₇N₃O₄(339.35), Yield: (0.292g, 86%). Off white powder, ¹H NMR (500 MHz, DMSO-d₆, ppm) δ 8.17 (d, *J* = 9 Hz, 2H), 7.44 (d, *J* = 9 Hz, 2H), 7.17(s, 2H), 4.36 (s, 1H), 2.54 – 2.53 (m, 2H), 2.27 – 2.09 (m, 2H), 1.04 (s, 3H), 0.95 (s, 3H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ 196.32, 163.69, 159.10, 152.81, 146.78, 129.14, 124.21, 119.86, 112.22, 57.50, 50.37, 36.16, 32.35, 28.77, 27.45.



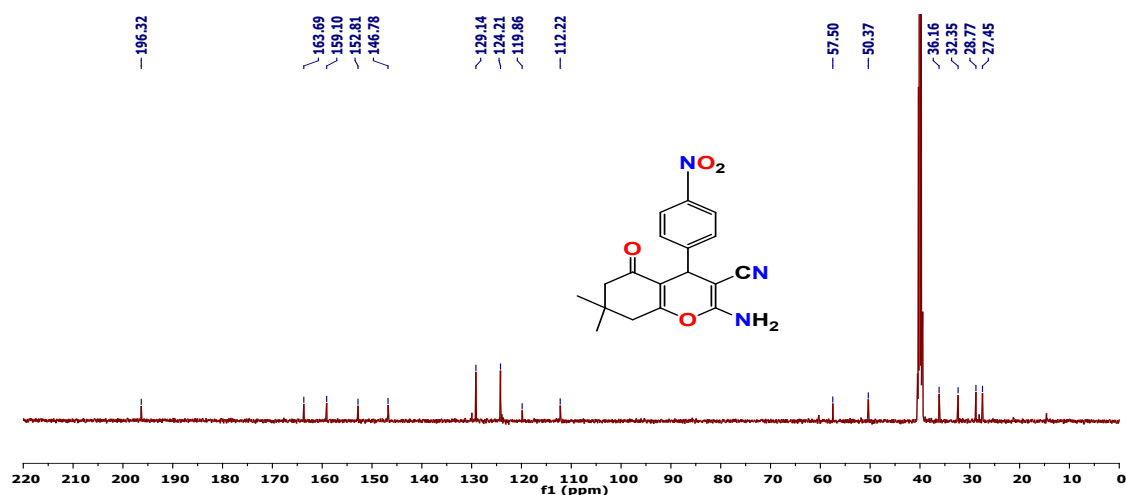
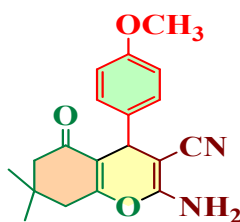
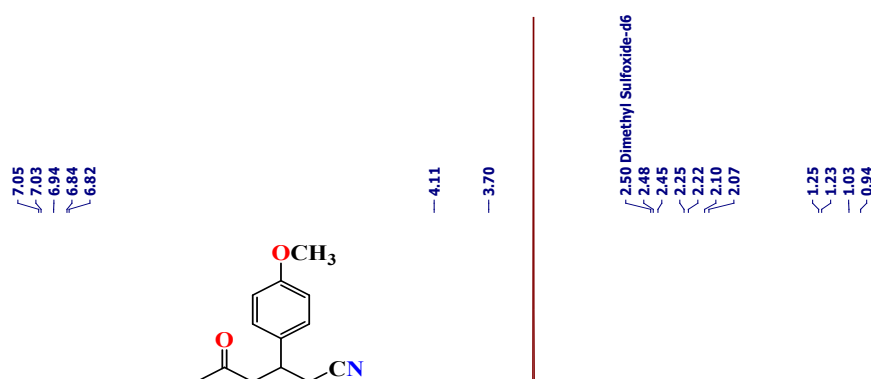


Figure S39. ¹³C NMR Spectrum of 6c

6d. 2-amino-6,6,8,8-tetrahydro-7,7-dimethyl-5-oxo-4-(4-methoxyphenyl)-4H-chromene-3-carbonitrile⁶



M. F. C₁₉H₂₀N₂O₃ (324.38), Yield: (0.308 g, 95%). Off white powder, ¹H NMR (500 MHz, DMSO d₆ppm) δ: 7.04 (d, *J* = 9 Hz, 2H), 6.94 (s, 2H), 6.31 (d, *J* = 8 Hz, 2H), 4.11 (s, 1H), 3.70 (s, 3H), 2.46 (d, *J* = 17.5 Hz, 2H), 2.24 (d, *J* = 16 Hz, 1H), 2.08 (d, *J* = 16 Hz, 1H), 1.24 (t, *J* = 11 Hz, 1H), 103 (s, 3H), 0.94 (s, 3H) ¹³C NMR (125 MHz, DMSO d₆, ppm) δ 196.22, 162.68, 158.94, 158.44, 137.36, 128.73, 120.32, 114.20, 113.51, 59.11, 55.52, 50.53 35.27, 32.30, 28.92, 27.29.



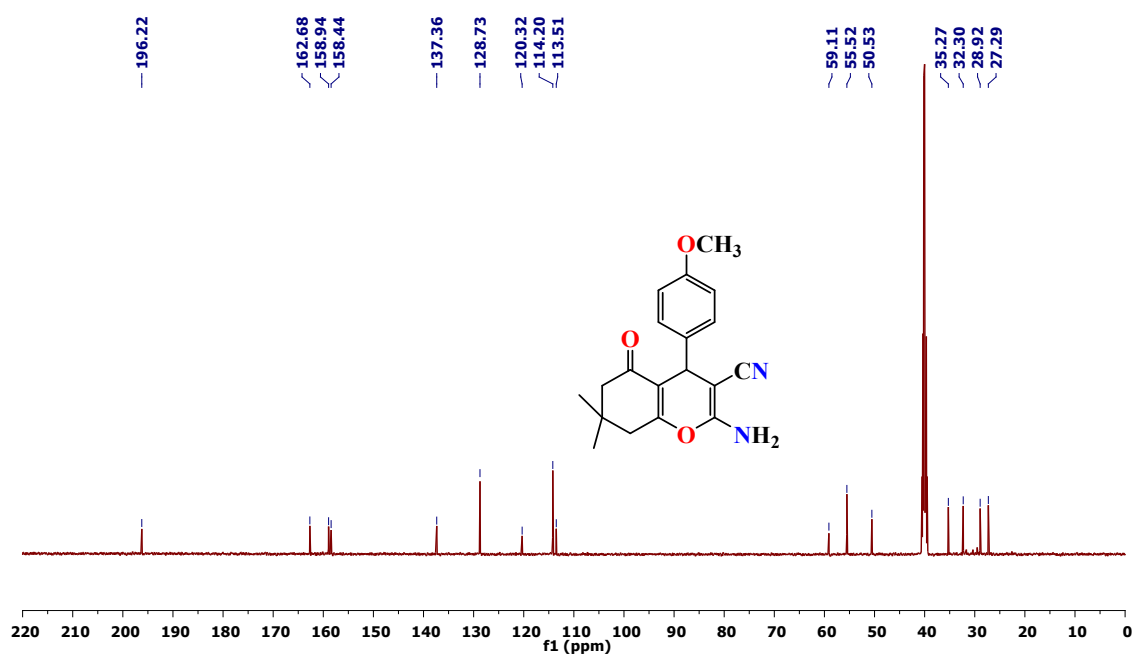
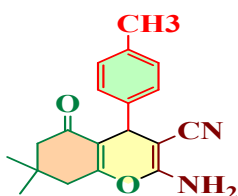


Figure S41. ¹³C NMR Spectrum of 6d

6e. 2-amino-6,6,8,8-tetrahydro-7,7-dimethyl-5-oxo-4-(4-methylphenyl)-4H-chromene-3-carbonitrile ⁷



M. F. C₁₉H₂₀N₂O₂ (308.38) Yield: (0.293 g, 95%). Off white powder, ¹H NMR (500 MHz, DMSO d₆, ppm) δ: 7.07 (d, *J* = 7.5 Hz, 2H), 7.00 (d, *J* = 8 Hz, 2H), 6.94 (s, 2H), 4.11 (s, 1H), 2.49 (d, *J* = 4 Hz, 2H), 2.23 (s, 3H), 2.25-2.22 (d, *J* = 16 Hz, 1H), 2.08 (d, *J* = 16 Hz, 1H), 1.02 (s, 3H), 0.94 (s, 3H); ¹³C NMR (125 MHz, DMSO d₆, ppm) δ: 196.19, 162.83, 158.96, 142.33, 136.15, 129.40, 127.59, 120.27, 113.39, 58.99, 50.52, 35.70, 32.31, 28.93, 27.28, 21.10.



Chemical structure of 2-amino-6-bromo-4-cyano-3-methyl-2H-chromene. The structure features a chromene core with a bromine atom (Br) at position 6, a cyano group (CN) at position 4, an amino group (NH₂) at position 2, and a methyl group at position 3.

Chemical structure of 2-bromo-1-(4-cyanophenyl)propan-1-one (SMILES: BrC(=O)C(c1ccc(C#N)cc1)C) and its ¹³C NMR spectrum (CDCl₃).

The chemical structure shows a benzene ring substituted with a bromine atom (green), a cyano group (blue), and a 1-oxo-2-bromopropyl group. The carbonyl carbon is red, and the bromine atom is green.

The ¹³C NMR spectrum (CDCl₃) shows the following chemical shifts (ppm):

- 200.0 (C=O)
- 144.4 (C-Br)
- 143.3 (C-Br)
- 127.07 (C=C)
- 120.6 (C=C)
- 120.0 (C=C)
- 114.3 (C=C)
- 113.3 (C=C)
- 111.1 (C=C)
- 99.9 (C≡N)
- 90.0 (C≡N)

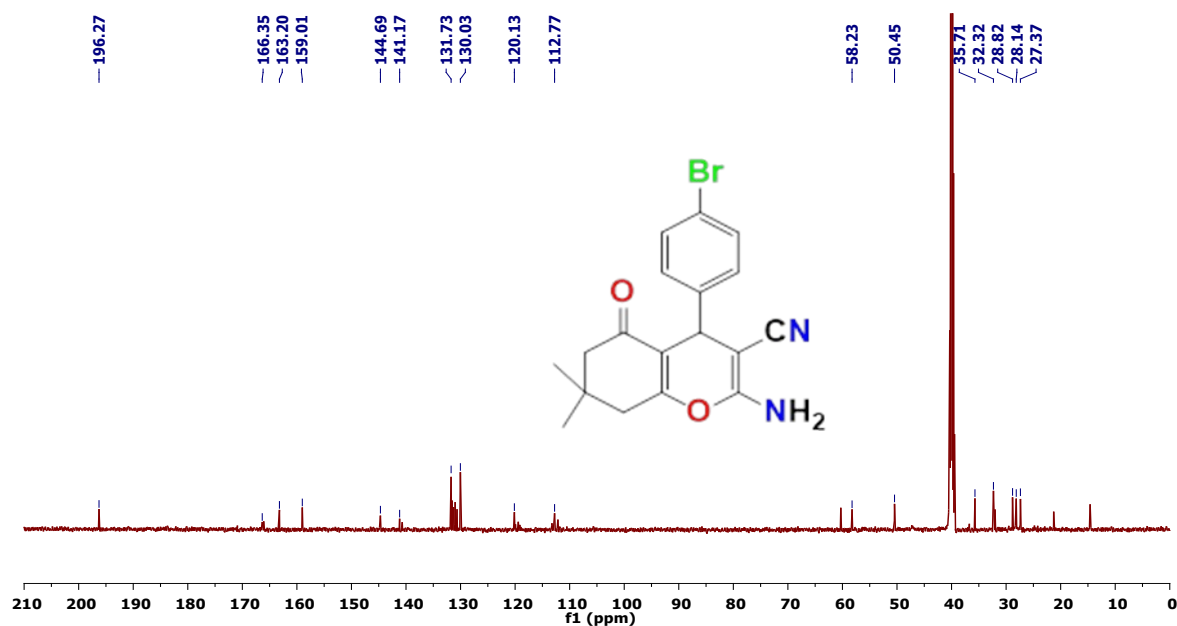
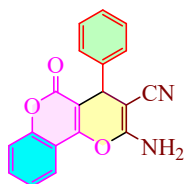


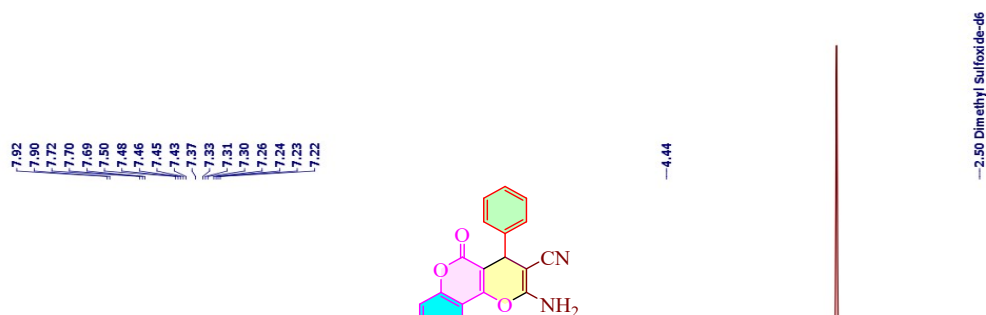
Figure S45. ¹³C NMR Spectrum of 6f

Data S7- NMR (¹H and ¹³C) spectra of 2- Amino-4chromeneDerivatives of 4-Hydroxycoumarin

7a. 2-amino-5-hydroxy-4-phenyl-4H,5H-pyrano[3,2-c]chromene-3-carbonitrile⁷



M. F. C₁₉H₁₂N₂O₃ (316.32), Yield: (0.284g, 90%), white powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) δ: 7.91 (d, *J* = 8 Hz, H), 7.70 (t, *J* = 8.5Hz, H), 7.48(t, *J* = 7.5Hz, H), 7.44(d, *J* = 8.5Hz, H), 7.37 (s, 2H), 7.31(t, *J* = 8Hz, 2H), 7.25(t, *J* = 7.25Hz, 2H) 4.44 (s, 1H), ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ: 160.05, 158.90, 158.48, 153.65, 152.64, 135.89, 133.37, 129.25, 125.16, 123.01, 119.74, 117.05, 114.40, 113.54, 104.84, 58.84, 55.59, 36.72, 31.45



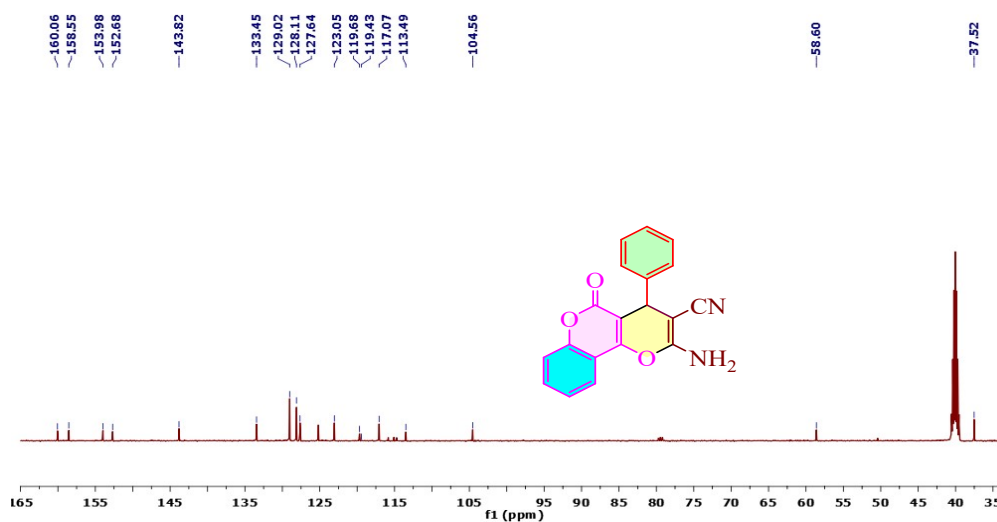
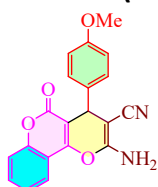


Figure S47. ¹³C NMR Spectrum of 7a

7b. 2-amino-4-(4-methoxyphenyl)-5-oxo-4H,5H-pyrano[3,2-c]chromene-3-carbonitrile⁶:



M. F. C₂₀H₁₄N₂O₄ (346.34), Yield: (0.325 g, 94%), white powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) δ: 7.86 (d, *J* = 7.5 Hz, H), 7.66 (t, *J* = 8 Hz, H), 7.44 (t, *J* = 8 Hz, H), 7.40 (d, *J* = 8.5 Hz, H), 7.28 (s, 2H), 7.13 (d, *J* = 8 Hz, H), 6.82 (d, *J* = 8 Hz, 2H), 4.39 (s, 1H), 3.72 (s, 3H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ: 160.05, 158.90, 158.48, 153.65, 152.64, 135.89, 133.37, 129.25, 125.16, 123.01, 119.74, 117.05, 114.40, 113.54, 104.84, 58.84, 55.59, 36.72, 31.45

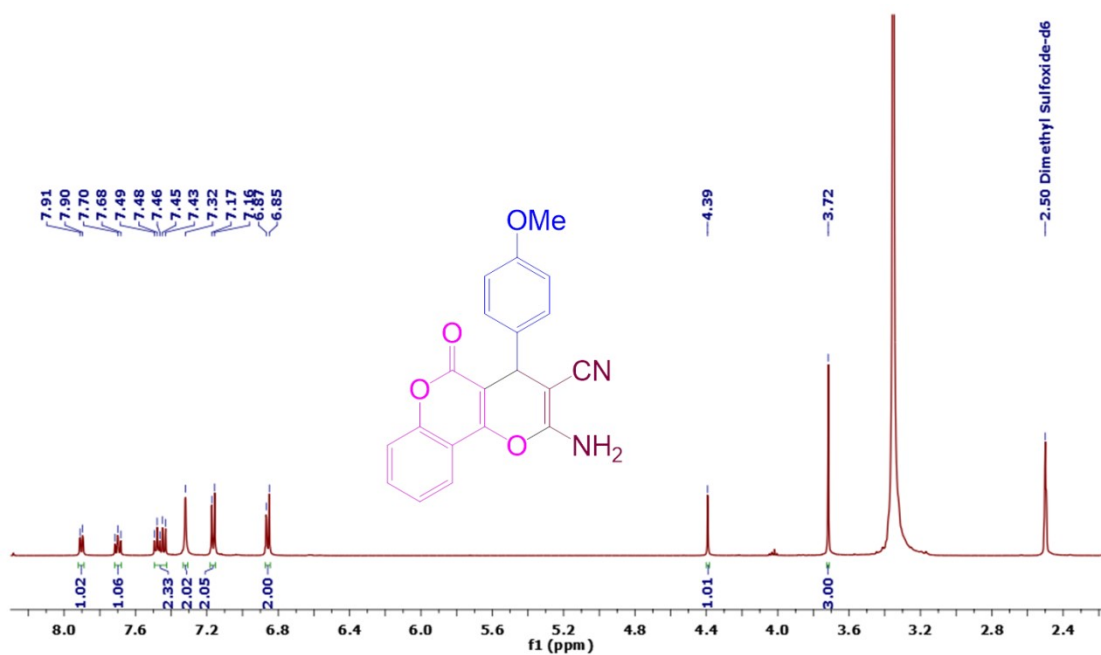


Figure S48. ¹H NMR Spectrum of 7b

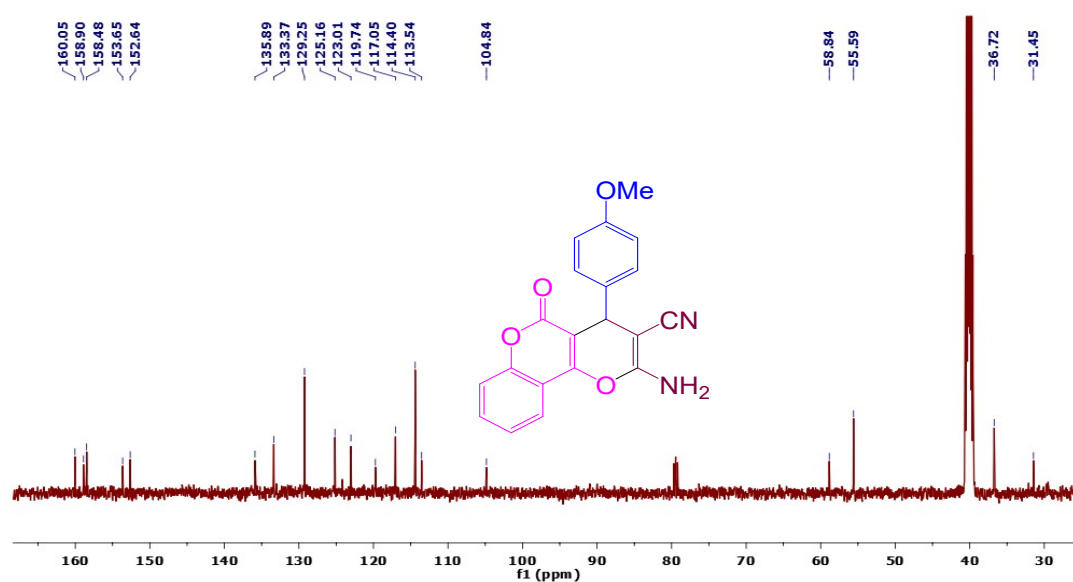
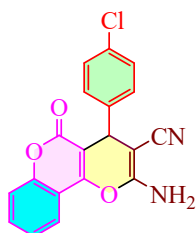


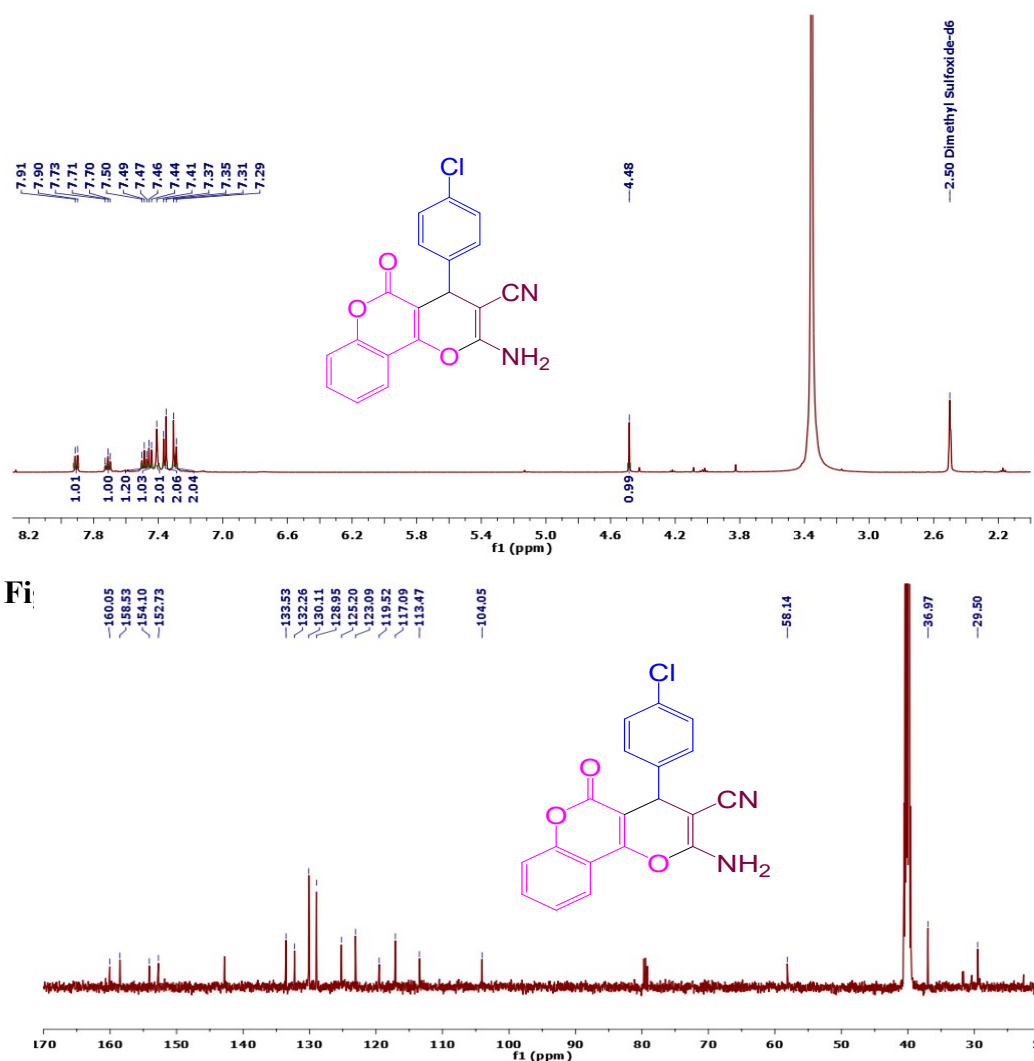
Figure S49. ¹³C NMR Spectrum of 7b

7c. 2-amino-4-(4-chlorophenyl)-5-oxo-4H,5H-pyrano[3,2-c]chromene-3-carbonitrile⁶

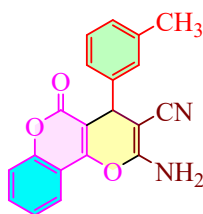


M. F. C₁₉H₁₁N₂O₃Cl (350.76), Yield: (0.308 g, 88%), reddish brown powder. . ¹H NMR (500 MHz, DMSO-d₆, ppm) δ: 7.91 (d, *J* = 8 Hz, H), 7.71 (t, *J* = 8.5Hz, H), 7.48(t, *J* = 7.5Hz, H),

7.45(d, $J = 8.5\text{Hz}$, H), 7.41 (s, 2H), 7.36(d, $J = 8\text{Hz}$, 2H), 7.29(d, $J = 9\text{Hz}$, 2H) 4.48 (s, 1H), ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 160.05, 158.53, 154.10, 152.73, 133.53, 132.26, 130.11, 128.59, 125.20, 123.09, 119.52, 117.09, 113.47, 104.05, 58.14, 36.97, 29.50



7d. 2-amino-5-oxo-4-(m-tolyl)-4H,5H-pyrano[3,2-c]chromene-3-carbonitrile⁷



M. F. $\text{C}_{20}\text{H}_{14}\text{N}_2\text{O}_3$ (330.34), Yield: (0.304 g, 92%), reddish brown powder. ^1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 7.91 (d, $J = 10\text{ Hz}$, H), 7.71 (t, $J = 7.5\text{Hz}$, H), 7.49(t, $J = 7.5\text{Hz}$, H), 7.45(d, $J = 8\text{Hz}$, H), 7.36 (s, 2H), 7.19(d, $J = 8\text{Hz}$, H), 7.04(d, $J = 10\text{Hz}$, 3H) 4.40 (s, 1H), 2.27(s, 3H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 160.06, 158.50, 153.94, 152.69,

143.82, 138.18, 133.91, 128.53, 128.37, 125.35, 125.19, 123.02, 119.72, 117.10, 113.55, 104.61, 58.70, 37.47, 21.54

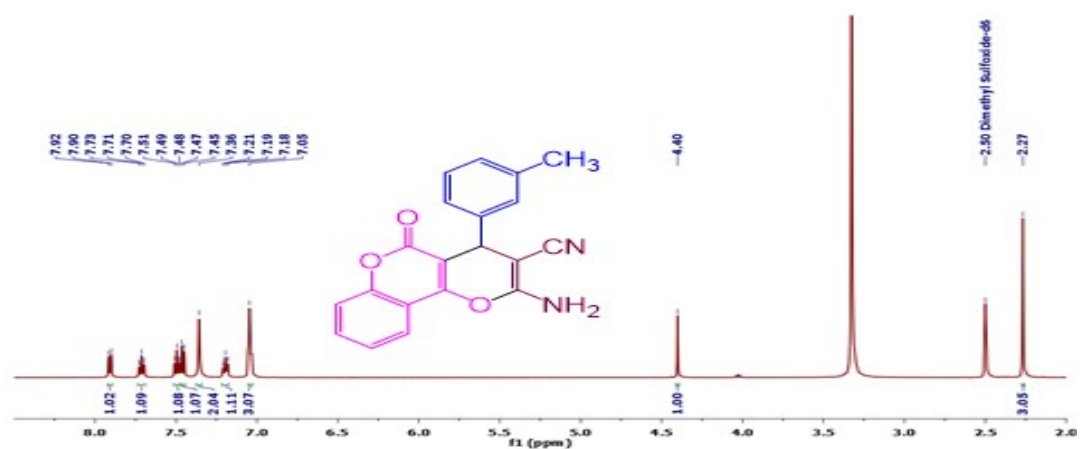


Figure S52. ¹H NMR Spectrum of 7d

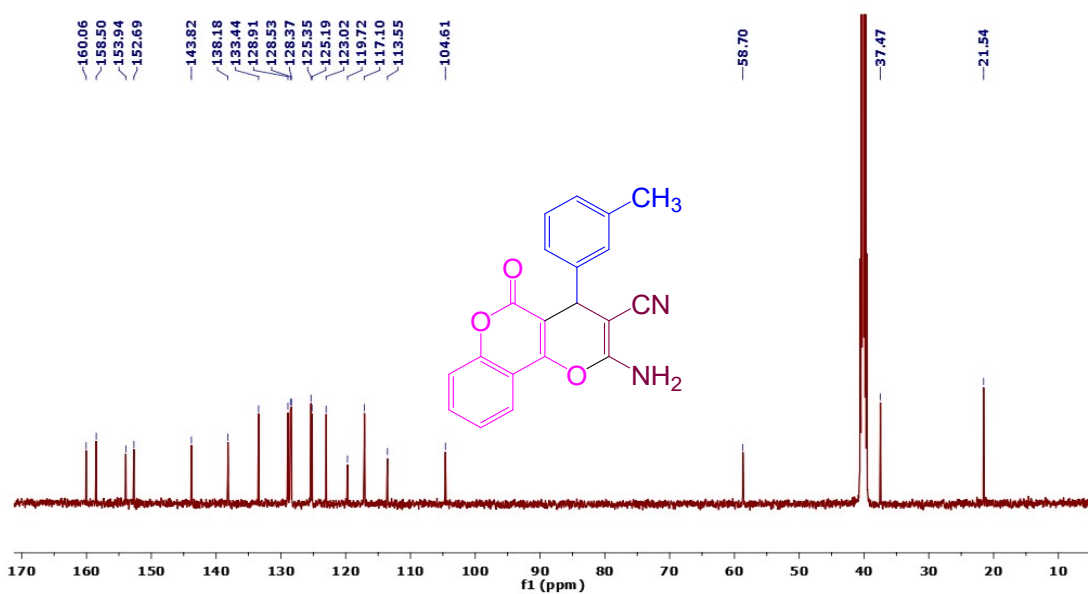
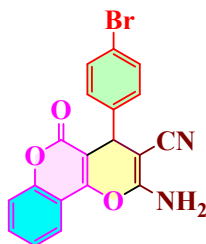


Figure S53. ¹³C NMR Spectrum of 7d

7e. 2-amino-4-(4-bromophenyl)-5-oxo-4H,5H-pyrano[3,2-c]chromene-3-carbonitrile⁶



M. F. C₁₉H₁₁N₂O₃Br (395.21), Yield: (0.340g, 86%), reddish brown powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) δ: 7.90 (d, *J* = 8 Hz, H), 7.72 (t, *J* = 7.5Hz, H), 7.49(t, *J* = 7.5Hz, 2H), 7.46(d, *J* = 8.5Hz, 2H), 7.43 (s, 2H), 7.24(d, *J* = 8Hz, 2H), 7.29(d, *J* = 8Hz, 2H) 4.47 (s, 1H), ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ: 160.05, 158.52, 154.11, 152.74, 143.27, 133.54, 131.86, 130.51, 125.21, 123.06, 120.75, 119.52, 117.11, 113.49, 104.00, 58.09, 37.06, 29.50

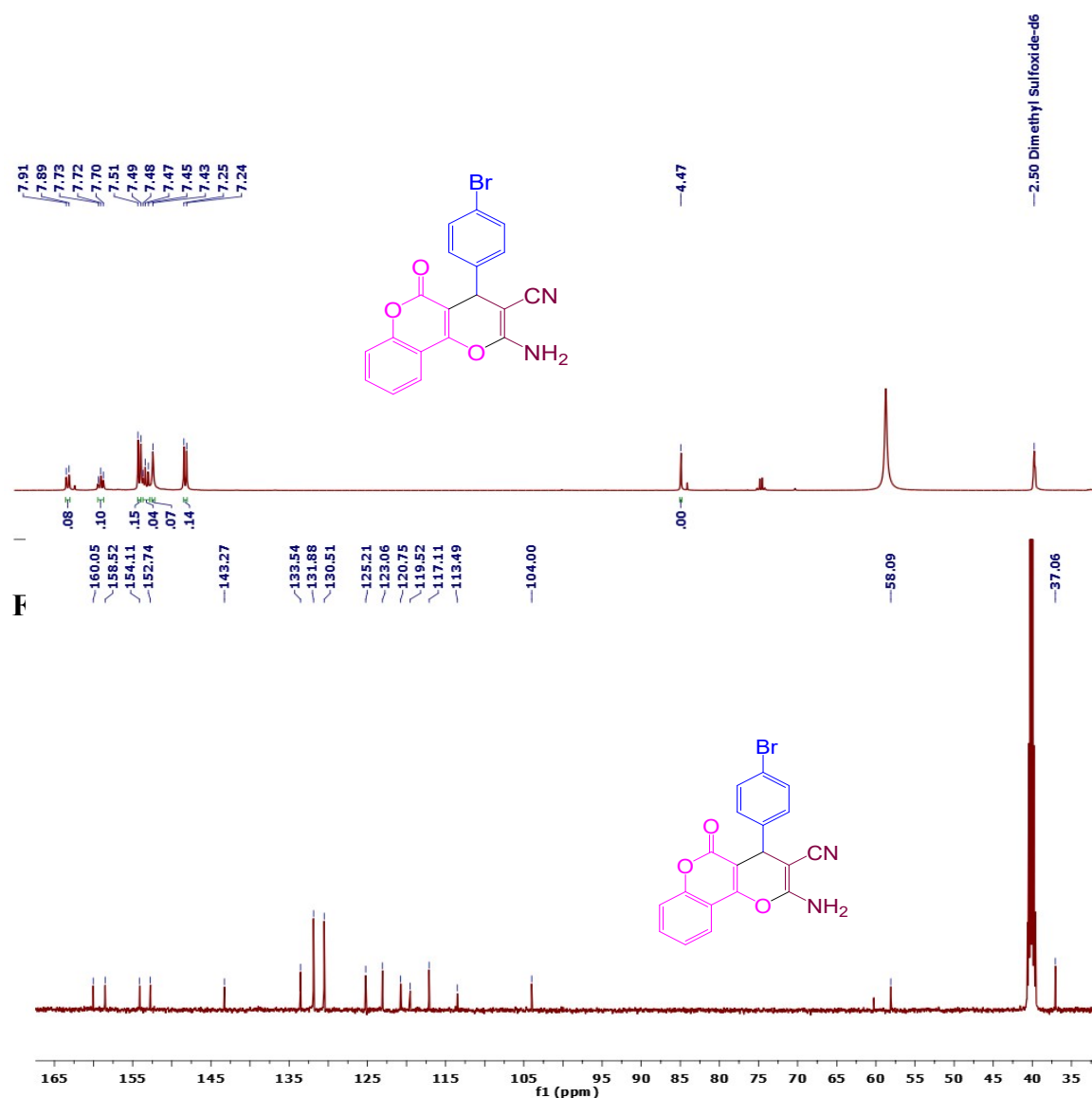
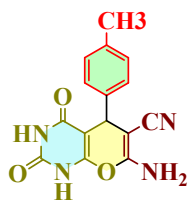


Figure S55. ¹³C NMR Spectrum of 7e

Data S8- NMR (¹H and ¹³C) spectra of 2- Amino4-chromeneDerivatives of of Burbuteric acidderivative

8a. 7-amino-2,4-dioxo-5-(p-tolyl)-1,3,4,5-tetrahydro-2H-pyrano[2,3-d]pyrimidine-6-carbonitrile ⁶



M. F. $C_{15}H_{12}N_4O_3$ (296.29), Yield: (0.284g, 96%), white powder, 1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 12.03 (s, 1H), 11.11(s, 1H), 7.08(d, J = 2Hz, 4H), 7.06(s, 2H), 4.17(s, 1H), 2.25(s, 3H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 168.25, 162.95, 158.10, 152.17, 141.73, 136.31, 129.35, 127.70, 89.17, 59.63, 35.81; 21.13;

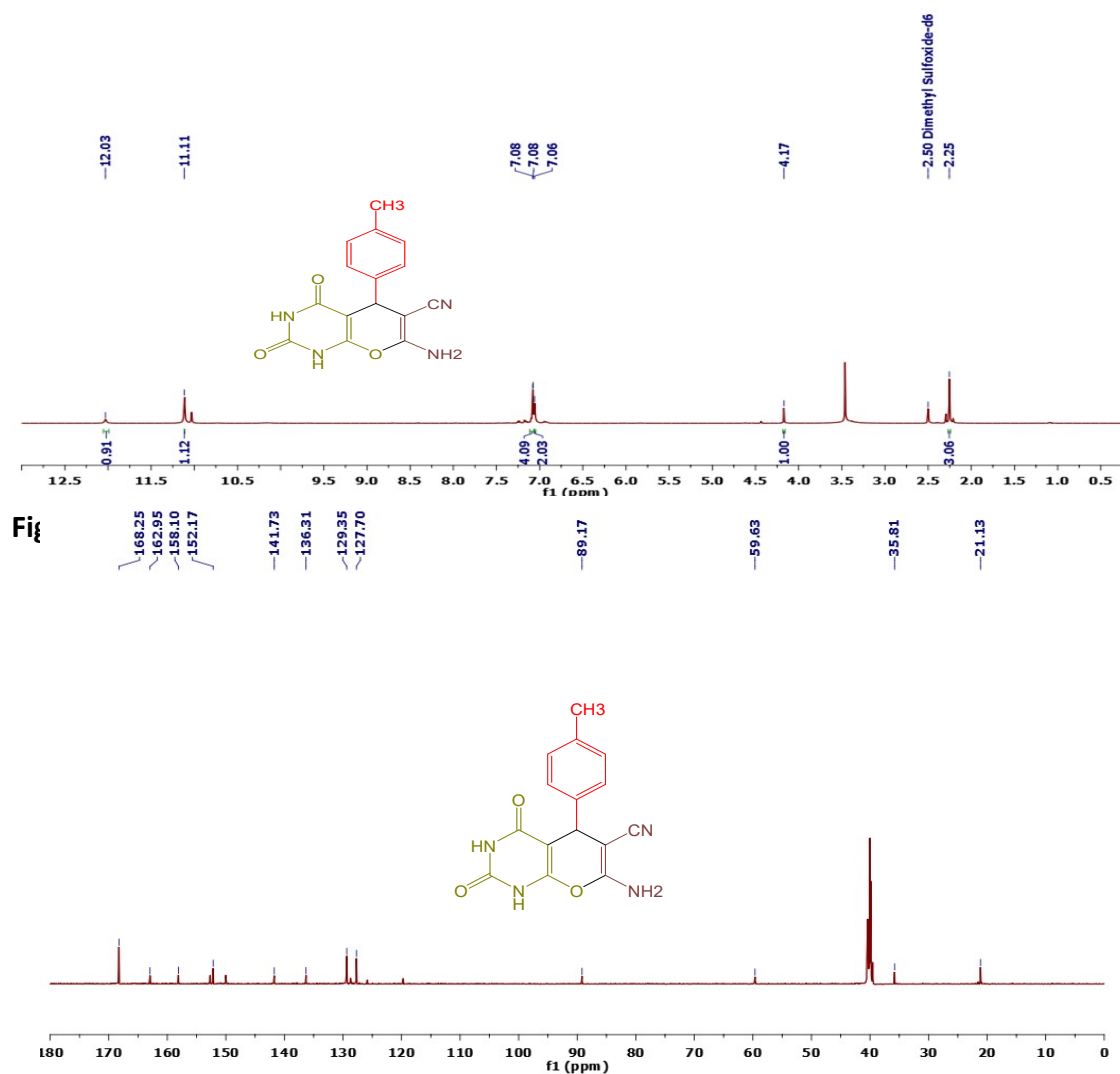
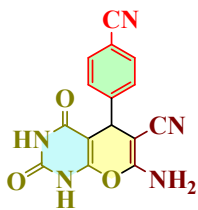


Figure S57. ^{13}C NMR Spectrum of 8a

8b. 7-amino-5-(4-cyanophenyl)-2,4-dioxo-1,3,4,5-tetrahydro-2H-pyrano[2,3-d]pyrimidine-6-carbonitrile⁷



M. F. $C_{15}H_9N_5O_3$ (307.27), Yield: (0.251 g, 82%), white brown powder. 1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 12.11 (s, 1H), 11.09(s, 1H), 7.72 (t, J = 8 Hz, 2H), 7.39(d, J = 8Hz, 2H), 7.22(s, 2H), 4.34(s, 1H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 162.99, 158.28, 153.14, 150.18, 150.00, 132.82, 129.09, 119.35, 110.10, 87.98, 58.23, 36.36;

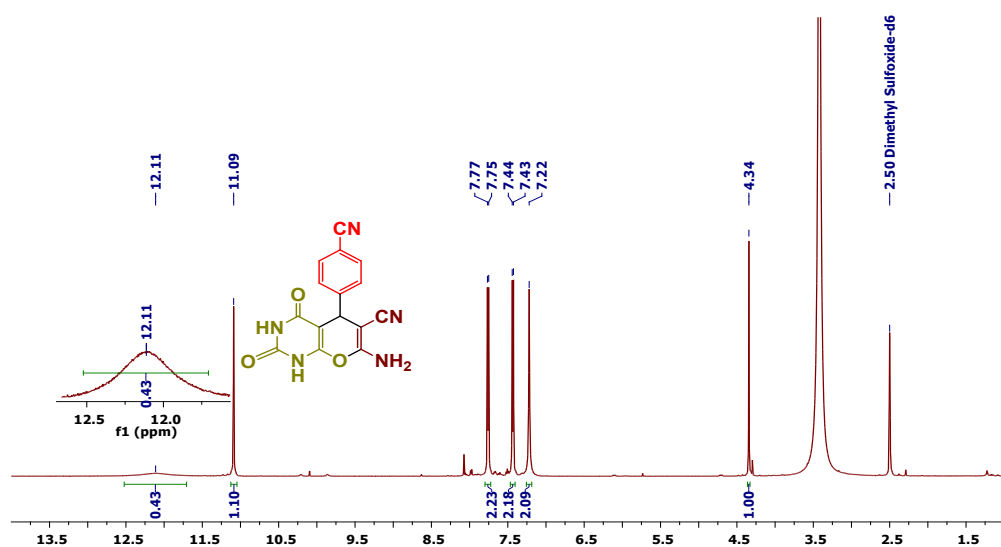


Figure S58. 1H NMR Spectrum of 8b

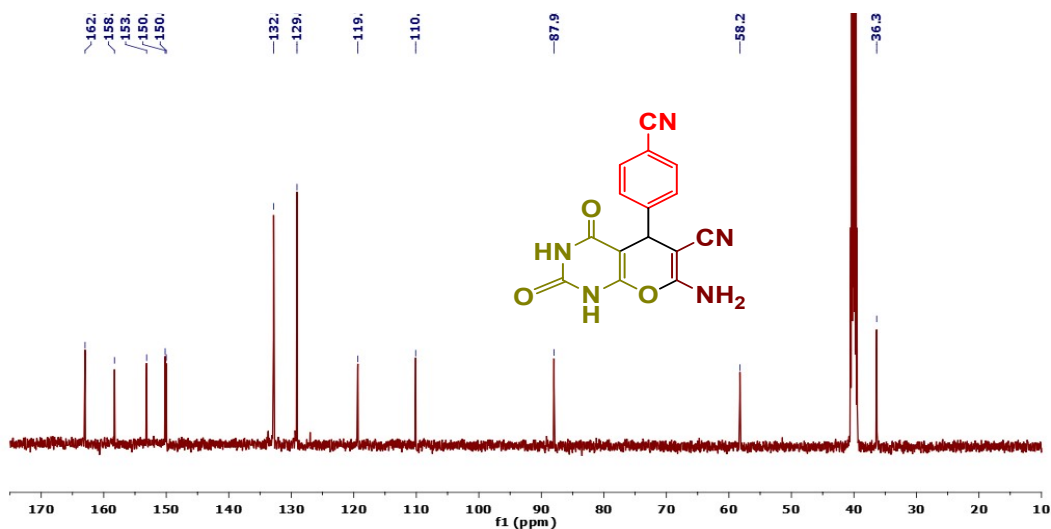
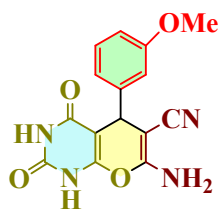


Figure S59. ^{13}C NMR Spectrum of 8b

8c. 7-amino-5-(3-methoxyphenyl)-2,4-dioxo-1,3,4,5-tetrahydro-2H-pyrano[2,3-d]pyrimidine-6-carbonitrile ⁶



M. F. $C_{15}H_{12}N_4O_4$ (312.29), Yield: (0.296 g, 95%), white powder. 1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 12.01 (s, 1H), 11.06 (s, 1H), 7.21 (t, J = 8 Hz, H), 7.06 (s, 2H), 6.75 (d, J = 10 Hz, H), 6.72 (d, J = 7.5 Hz, H), 6.68 (s, H), 4.20 (s, 1H), 3.72 (s, 3H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 162.98, 159.70, 158.22, 152.82, 150.00, 146.22, 129.94, 119.92, 119.66, 114.00, 112.20, 88.92, 59.36, 55.49, 36.10;

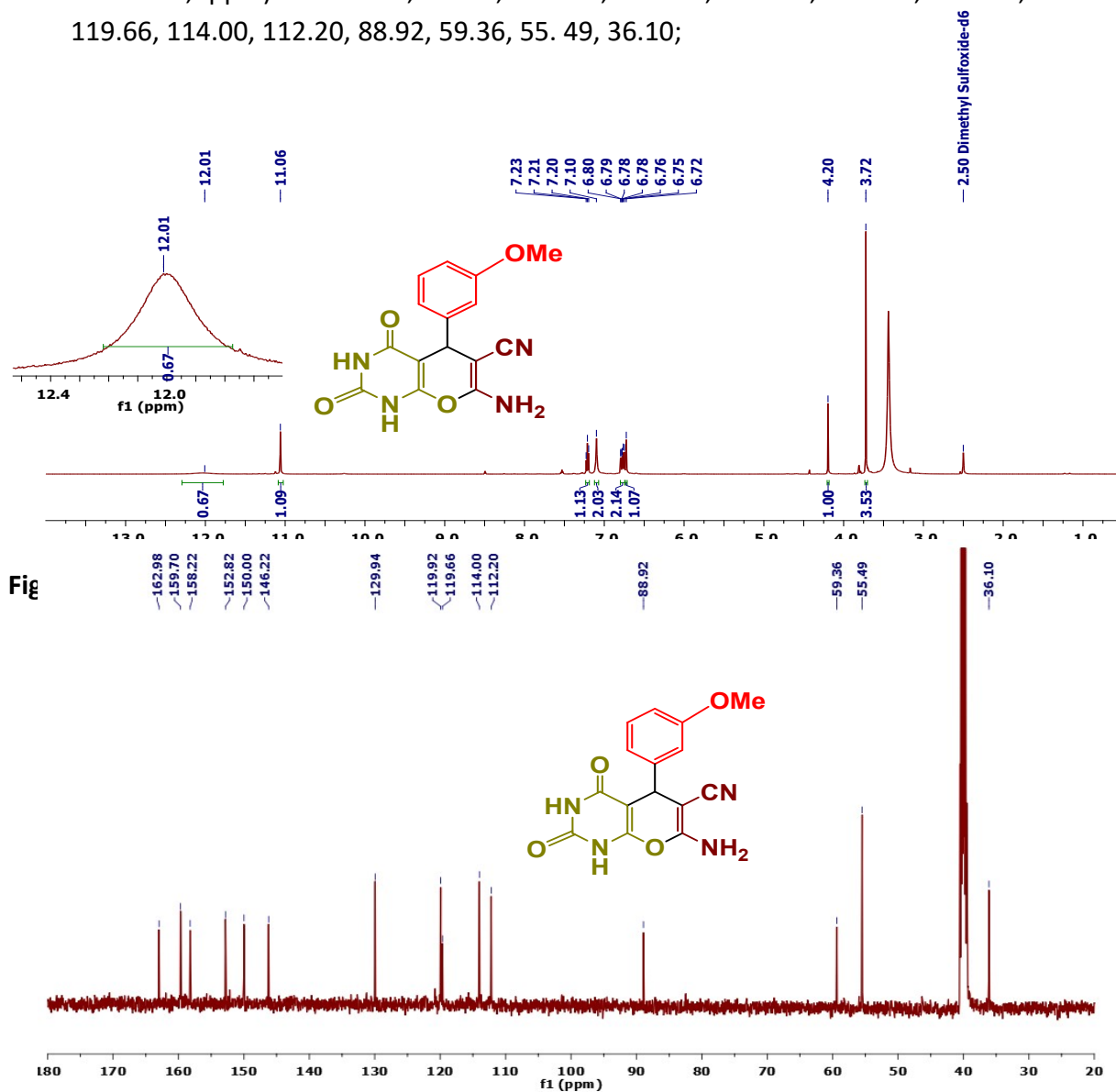
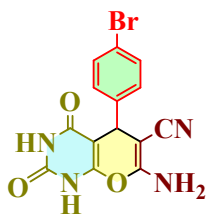


Figure S61. ^{13}C NMR Spectrum of 8c

8d. 7-amino-5-(4-bromophenyl)-2,4-dioxo-1,3,4,5-tetrahydro-2H-pyrano[2,3-d]pyrimidine-6-carbonitrile⁶



M. F. $C_{14}H_9BrN_4O_3$ (361.16), Yield: (0.324 g, 89%), off white powder. 1H NMR (500 MHz, DMSO- d_6 , ppm) δ : 12.01 (s, 1H), 11.07 (s, 1H), 7.43 (t, J = 8 Hz, 2H), 7.14(d, J = 9Hz, 2H), 7.10(s, 2H), 4.23(s, 1H), ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ : 162.97, 158.16, 152.87,

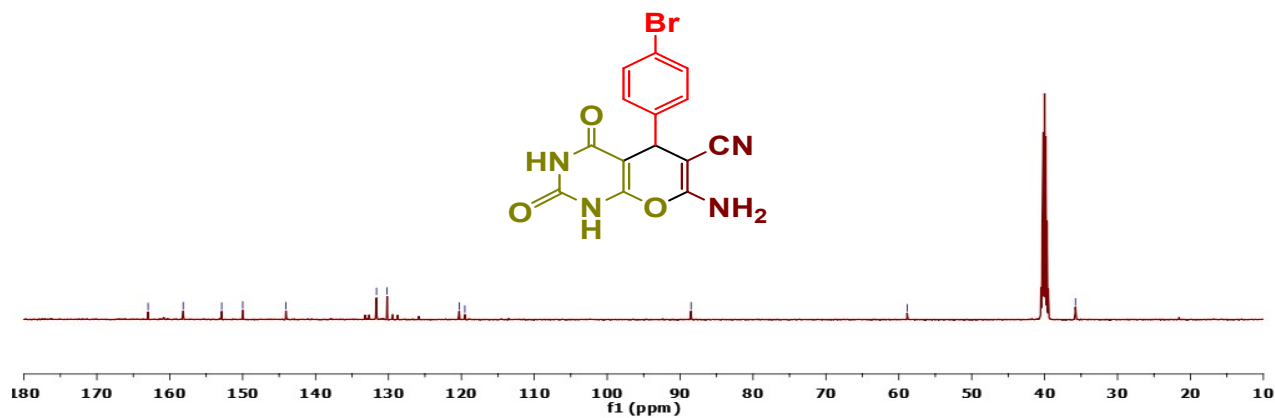


Figure S63. ^{13}C NMR Spectrum of 8d

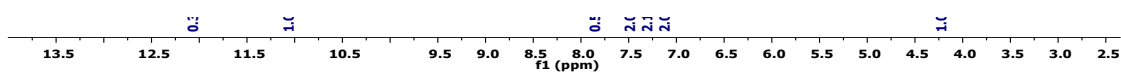
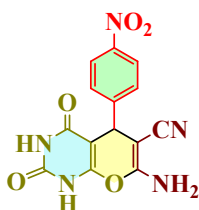


Figure S62. 1H NMR Spectrum of 8d

8e. 7-amino-5-(4-nitrophenyl)-2,4-dioxo-1,3,4,5-tetrahydro-2H-pyrano[2,3-d]pyrimidine-6-carbonitrile⁷



M. F. C₁₄H₉N₅O₅ (327.26), Yield: (0.261g, 80%), light brown powder. ¹H NMR (500 MHz, DMSO-d₆, ppm) δ: 12.15 (s, 1H), 11.10(s, 1H), 8.16 (s, 2H), 7.53(s, 2H), 7.25(s, 2H), 4.42(s, 1H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ: 167.73, 163.99, 157.89, 156.97, 154.76, 153.14, 151.70, 134.09, 128.79, 124.07, 92.72, 62.82, 40.94;

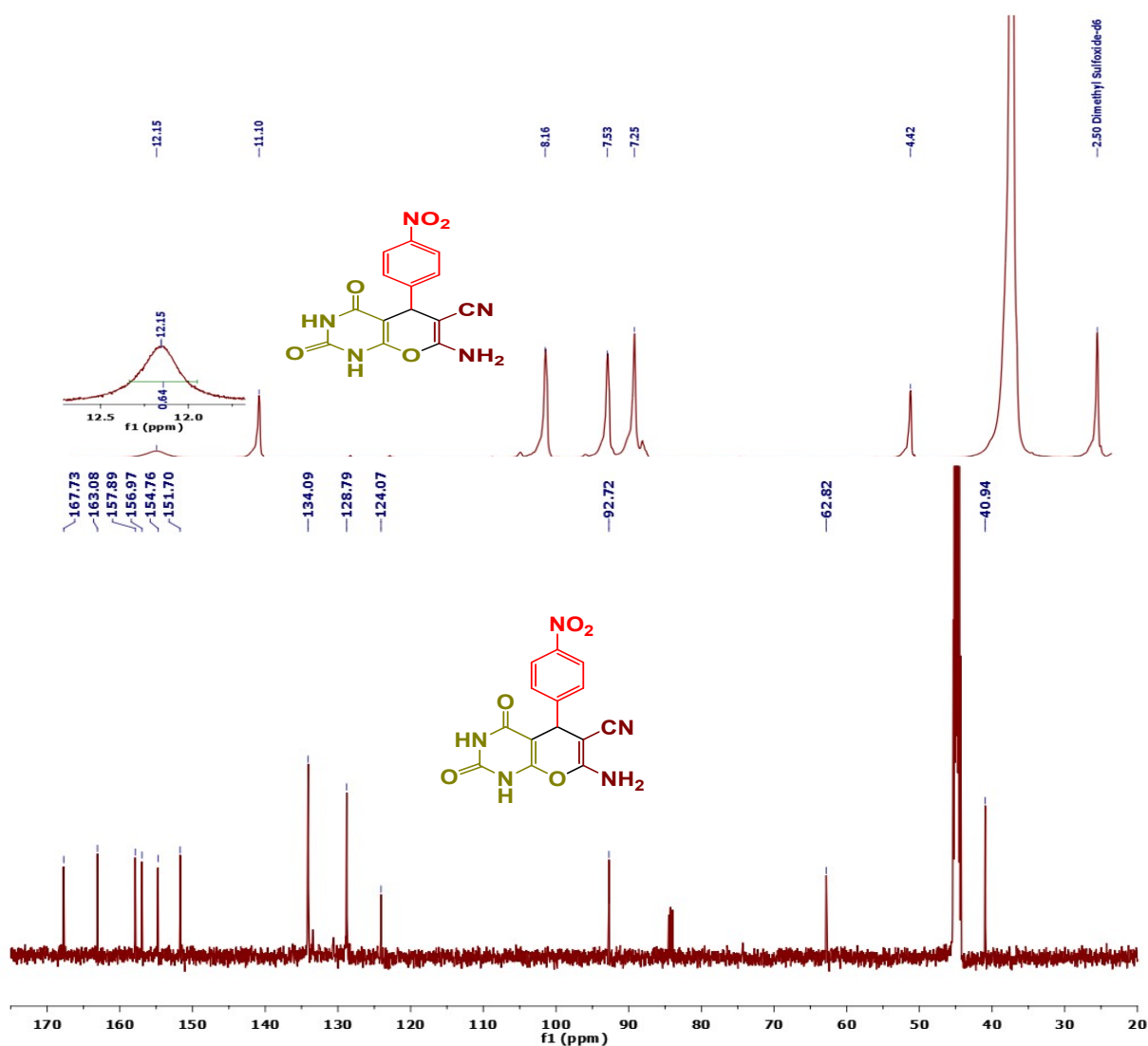
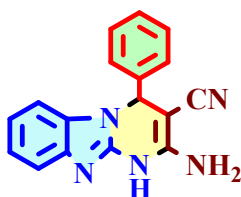


Figure S65. ¹³C NMR Spectrum of 8e

DataS9- NMR (^1H and ^{13}C) spectra of Imidazopyrimidine derivatives:

9a. 2-amino-4-phenyl-1,4-dihydrobenzo[4,5]imidazo[1,2-a]pyrimidine-3-carbonitrile⁴



M. F. $\text{C}_{17}\text{H}_{13}\text{N}_5$ (287.31), Yield: (0.253 g, 88%). Light yellow powder. ^1H NMR (500 MHz, DMSO- d_6 , ppm) δ 9.03 (s, 1H, NH), 7.68 (d, J = 8 Hz, 1H), 7.38 – 7.35 (m, 2H), 7.31 (dd, J = 7, 8 Hz, 4H), 7.16 (t, J = 8 Hz, 1H), 7.06 (t, J = 7 Hz, 1H), 6.87 (s, 2H), 5.28 (s, 1H); ^{13}C NMR (125 MHz, DMSO- d_6 , ppm) δ 156.37, 154.13, 147.86, 134.011, 134.03, 133.29, 131.35, 128.92, 125.73, 124.20, 120.78, 117.98, 67.45, 58.60.

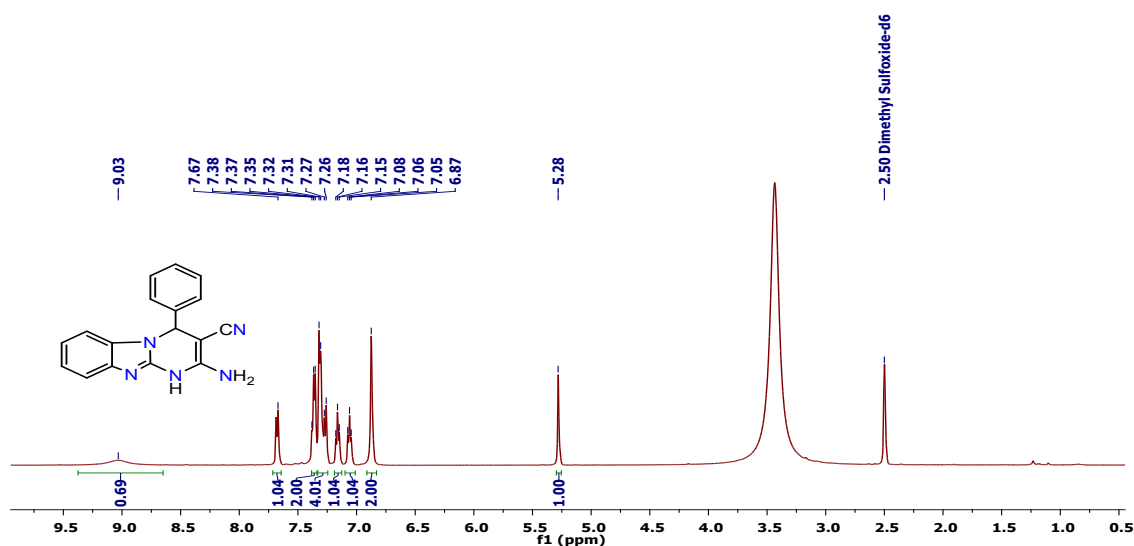


Figure S66. ^1H NMR Spectrum of 9a

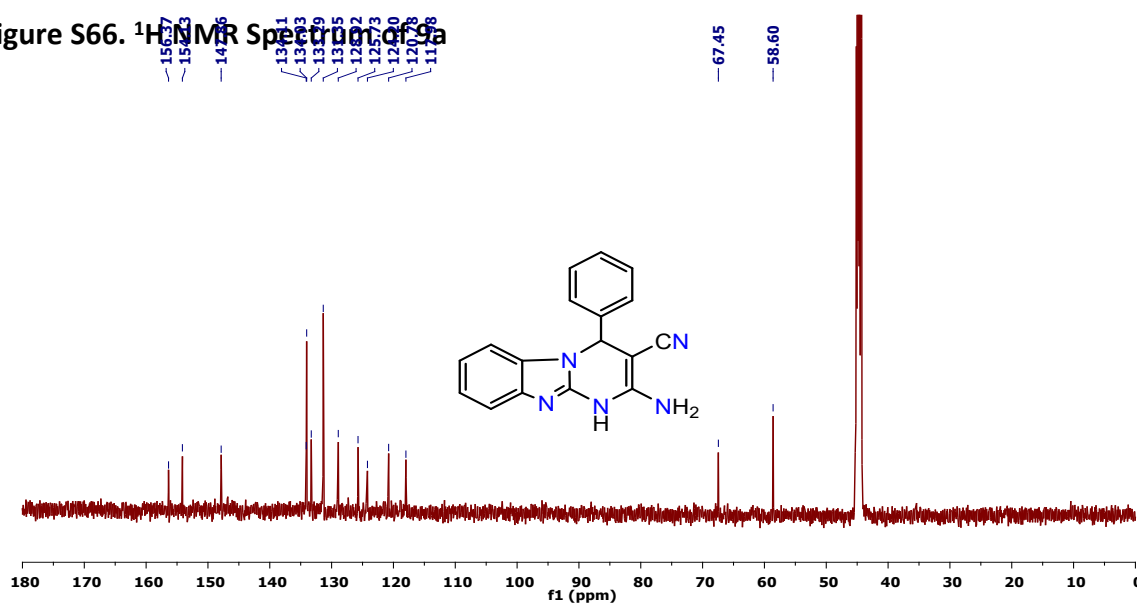
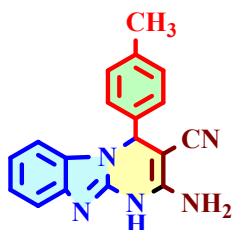


Figure S67. ^{13}C NMR Spectrum of 9a

9b. 2-amino-4-(4-methylphenyl)-1,4-dihydrobenzo[4,5]imidazo[1,2-a]pyrimidine-3 carbonitrile⁵



M. F. C₁₈H₁₅N₅ (301.13), Yield: (0.280g, 93%). Off white powder. ¹H NMR (500 MHz, DMSOd₆, ppm) δ 8.83 (s, 1H, NH), 7.65 (d, *J* = 8 Hz, 1H), 7.24 (d, *J* = 7.5 Hz, 1H), 7.19 (d, *J* = 8 Hz, 4H), 7.15 (d, *J* = 7.5 Hz, 1H),), 7.13 (d, *J* = 7.5 Hz, 1H) 7.03(t, *J* = 7.5 Hz, 1H), 6.83 (s, 2H), 5.20 (s, 1H), 2.26 (s, 3H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ 152.06, 149.50, 140.31, 137.70, 129.73, 126.48, 123.95, 120.58, 119.60, 116.35, 113.01, 62.76, 53.58, 21.16.

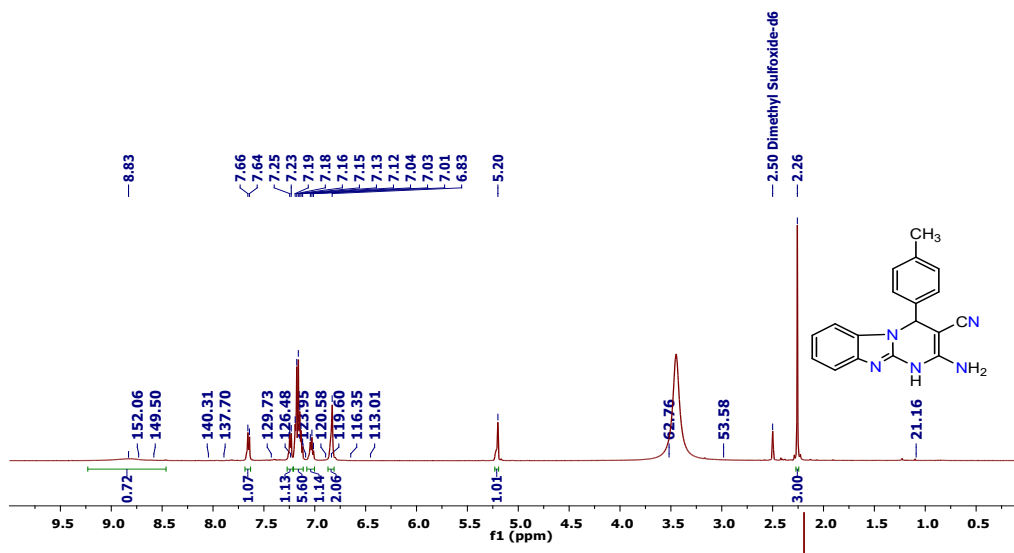


Figure S68. ¹H NMR Spectrum of 9b

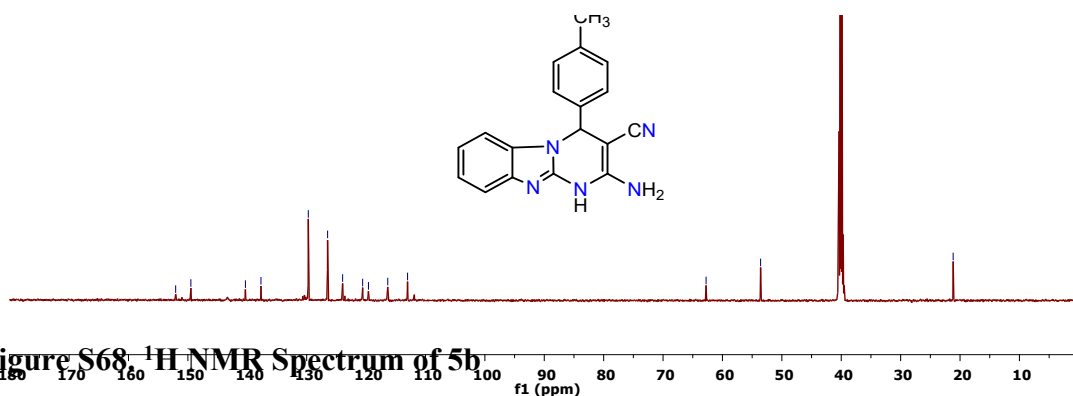
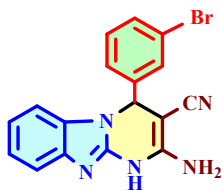


Figure S68. ¹H NMR Spectrum of 5b

Figure S69. ¹³C NMR Spectrum of 9b

9c. 2-Amino-4-(3-bromophenyl)-1,4-dihydrobenzo[4,5]imidazo[1,2-a]pyrimidine-3-carbonitrile⁵



M. F. C₁₇H₁₂BrN₅ (366.21), Yield: (0.311 g, 85%). yellow powder. ¹H NMR (500 MHz, DMSO-*D*₆) δ 8.83 (s, 1H), 7.66 (d, *J* = 8.0 Hz, 1H), 7.52 (s, 2H), 7.50 (d, *J* = 8.5 Hz, 1H), 7.35 – 7.29 (m, 2H), 7.25 (d, *J* = 8.0 Hz, 1H), 7.14 (t, *J* = 8.0 Hz, 1H), 7.04 (t, *J* = 7.5 Hz, 1H), 6.93 (s, 2H), 5.31 (s, 1H); ¹³C NMR (126 MHz, DMSO-*d*₆) δ 151.77, 149.78, 145.96, 143.27, 131.56, 131.31, 129.54, 125.63, 124.07, 122.39, 120.75, 119.46, 116.47, 113.09, , 61.72, 53.10.

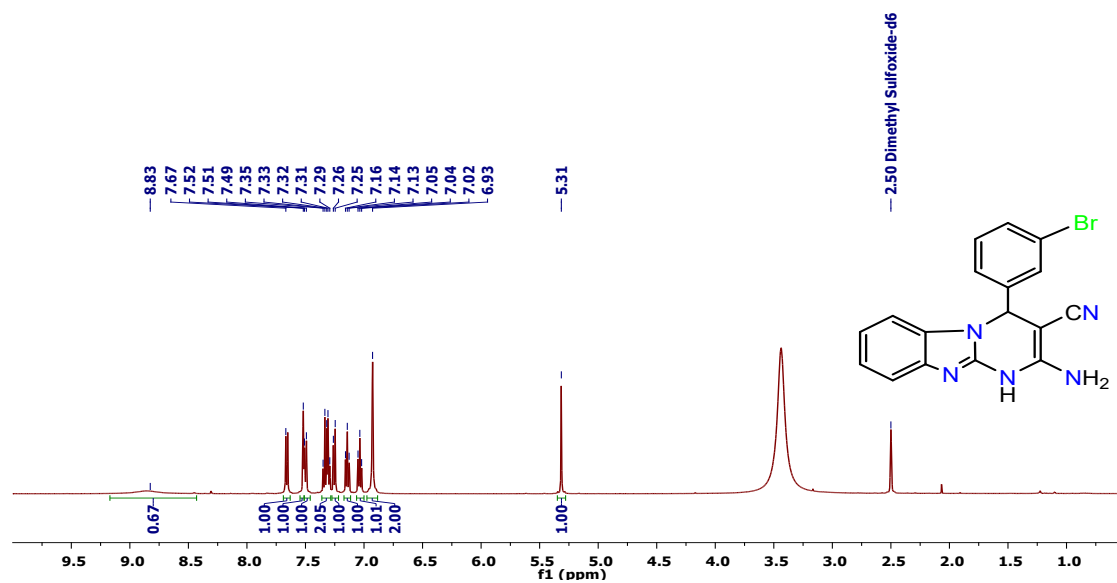


Figure S70. ¹H NMR Spectrum of 9c

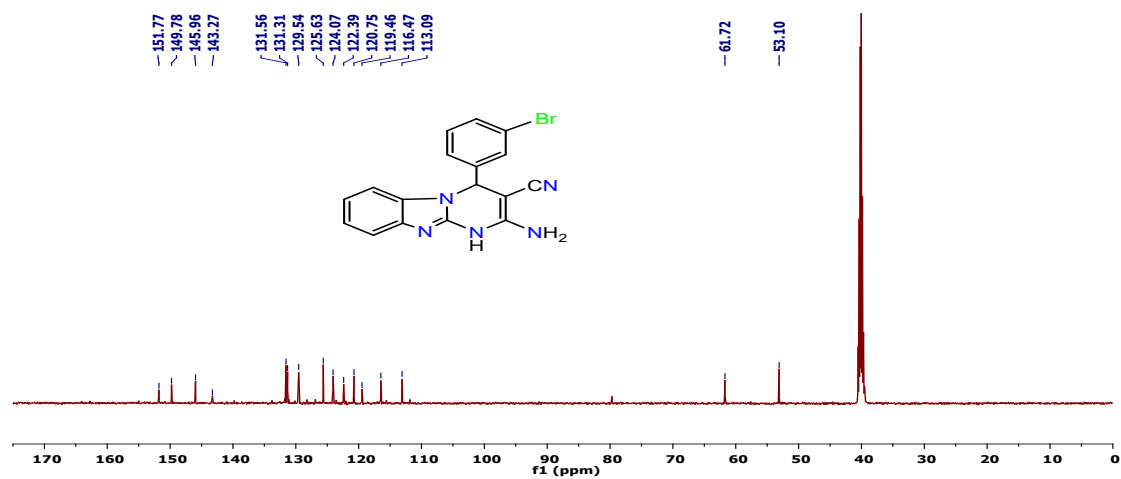
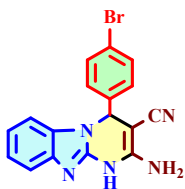


Figure S71. ¹³C NMR Spectrum of 9c

9d. 2-Amino-4-(4-Bromophenyl)-1,4-dihydrobenzo[4,5]imidazo[1,2-a]pyrimidine-3-carbonitrile⁴



M. F. $C_{17}H_{12}N_5Br$ (366.21), Yield: (0.314 g, 86%). yellow powder. 1H NMR (500 MHz, DMSO- d_6 , ppm): δ 8.60 (s, 1H, NH), 7.63 (d, J = 8.5 Hz, 1H), 7.55 (d, J = 7.5 Hz, 2H), 7.24 (dd, J = 8, 6.5 Hz, 3H), 7.12 (t, J = 7.5 Hz, 1H), 7.00 (d, J = 7.5 Hz, 1H), 6.86 (s, 2H), 5.24 (s, 1H); ^{13}C NMR (125 Hz, DMSO- d_6 , ppm): δ 152.07, 149.76, 144.00, 142.73, 142.68, 132.15, 129.76, 128.80, 123.94, 121.51, 120.51, 119.54, 116.64, 112.98, 61.95, 53.17.

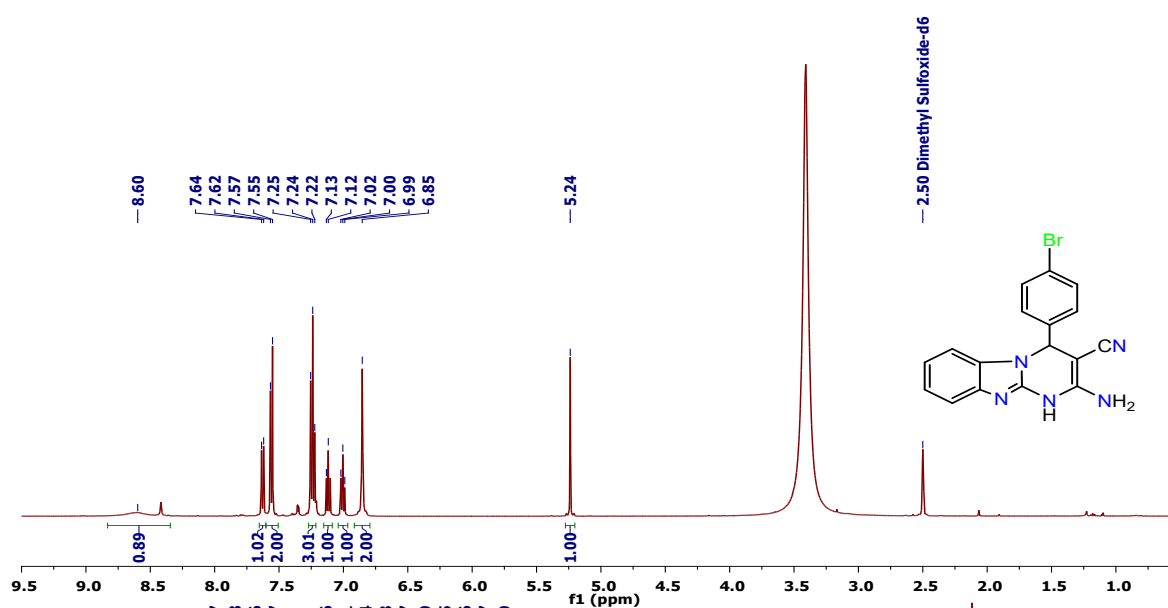
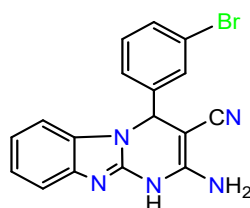
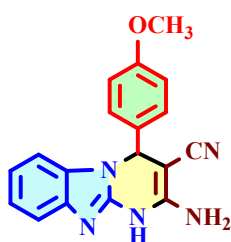


Figure S72. 1H NMR Spectrum of 9d



9e. 2-amino-4-(4-methoxyphenyl)-1,4-dihydrobenzo[4,5]imidazo[1,2-a]pyrimidine-3 carbonitrile⁴



M. F. C₁₈H₁₅N₅O (317.34), Yield: (0.288g,91%). Light yellow powder. ¹H NMR (500 MHz,DMSO-d₆, ppm) δ 8.53 (s, 1H, NH), 7.63 (d, *J* = 7.5 Hz, 1H), 7.20 (t, *J* = 8.0 Hz, 3H), 7.11 (t, *J* = 7.5 Hz, 1H), 6.99 (t, *J* = 8.0 Hz, 1H), 6.90 (d, *J* = 9.0 Hz, 2H), 6.79 (s, 2H), 5.15 (s, 1H), 3.71 (s, 3H); ¹³C NMR (125 MHz, DMSO-d₆, ppm) δ 159.45, 152.24, 149.56, 144.07, 135.32, 129.80, 127.85, 123.82, 112.036, 116.53, 114.55, 112.90, 79.68,62.87, 55.63, 53.31.

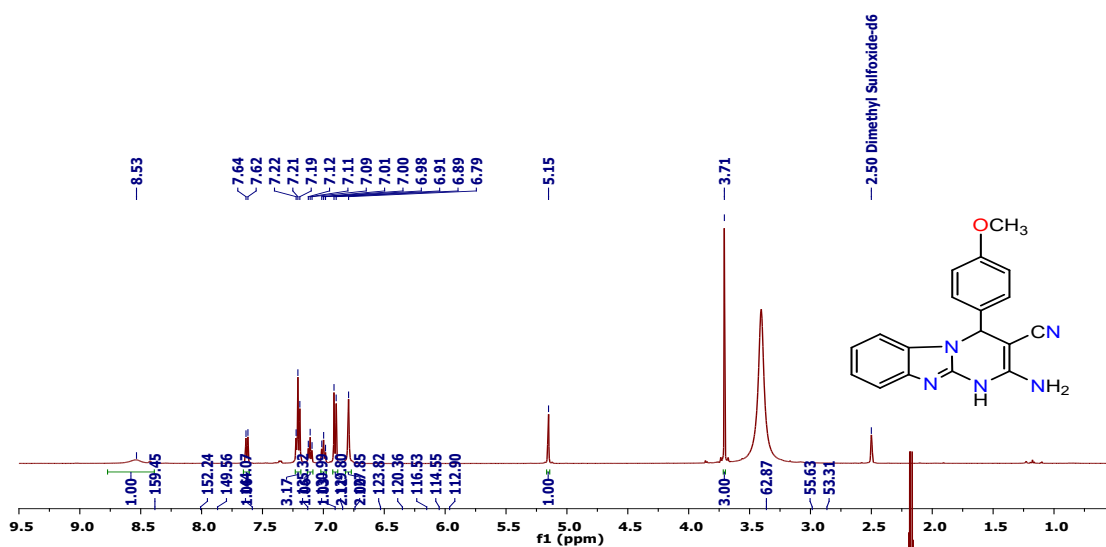
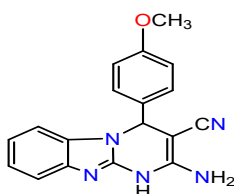
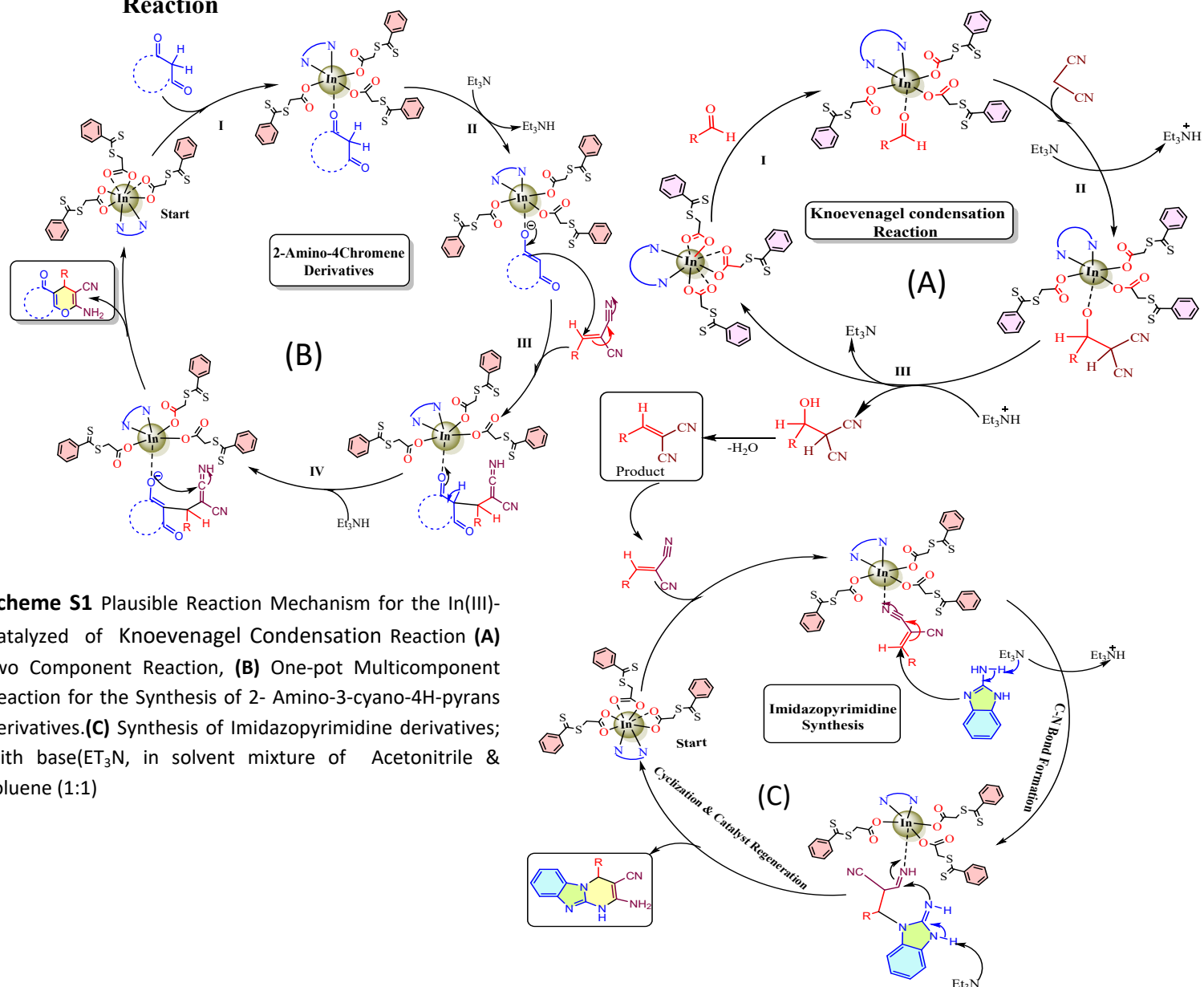


Figure S74. ¹H NMR Spectrum of 9e



Plausible Reaction Mechanism for the In(III)-Catalyzed One-pot Multicomponent Reaction



Data S10- NMR(¹H & ¹³C) spectra For CO₂ conversion Reaction

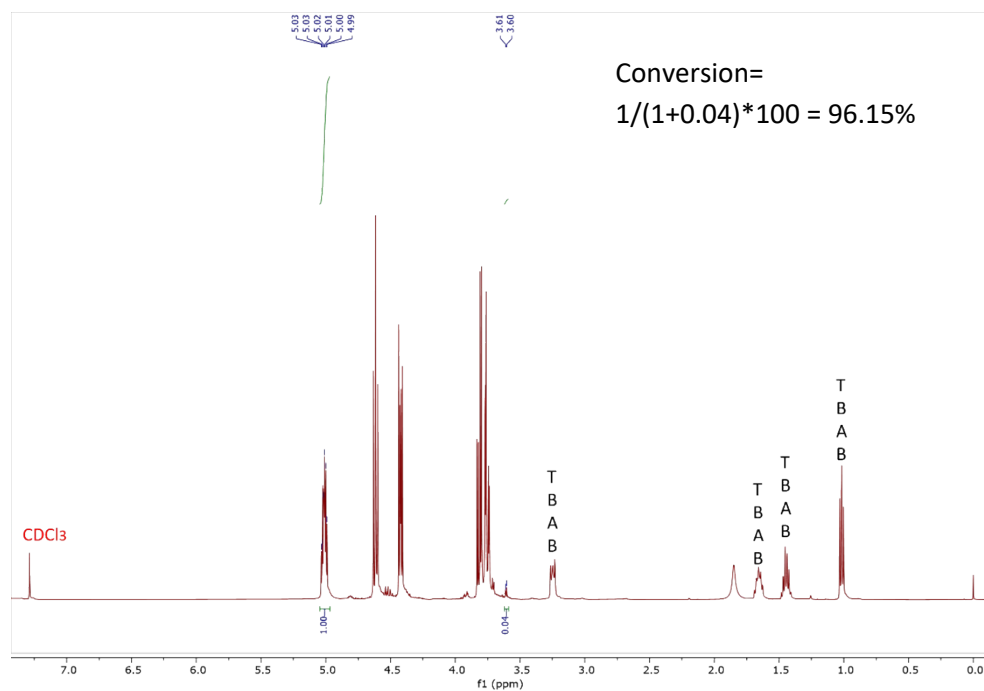


Figure S76: ¹H NMR spectrum of **catalyst 1**(Epichlorohydrin) **EPH** in CDCl₃ at 100°C

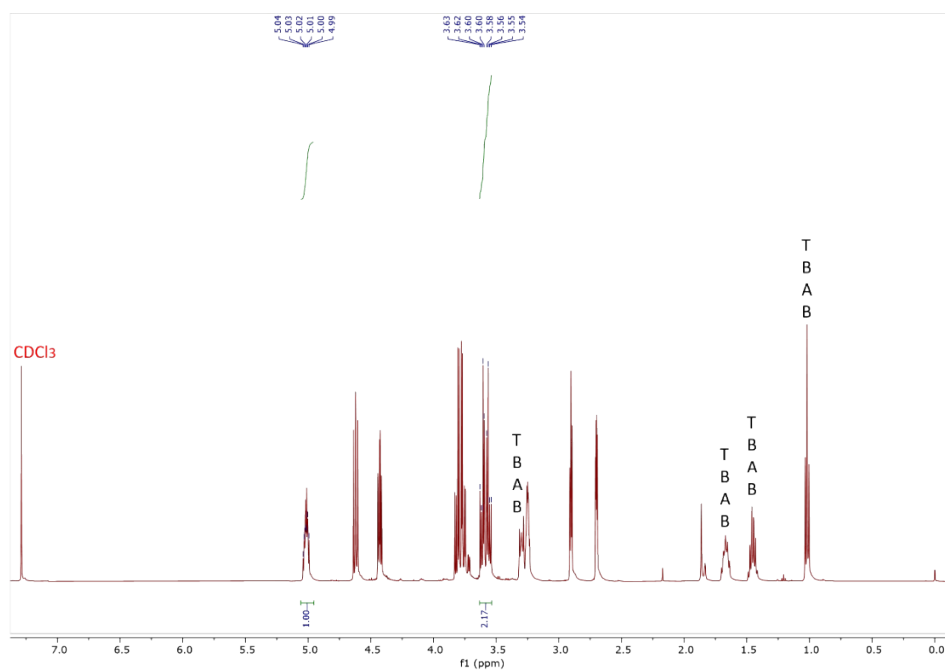


Figure S77: ¹H NMR spectrum of **catalyst 1** (Epichlorohydrin) **EPH** in CDCl₃ at 40 C

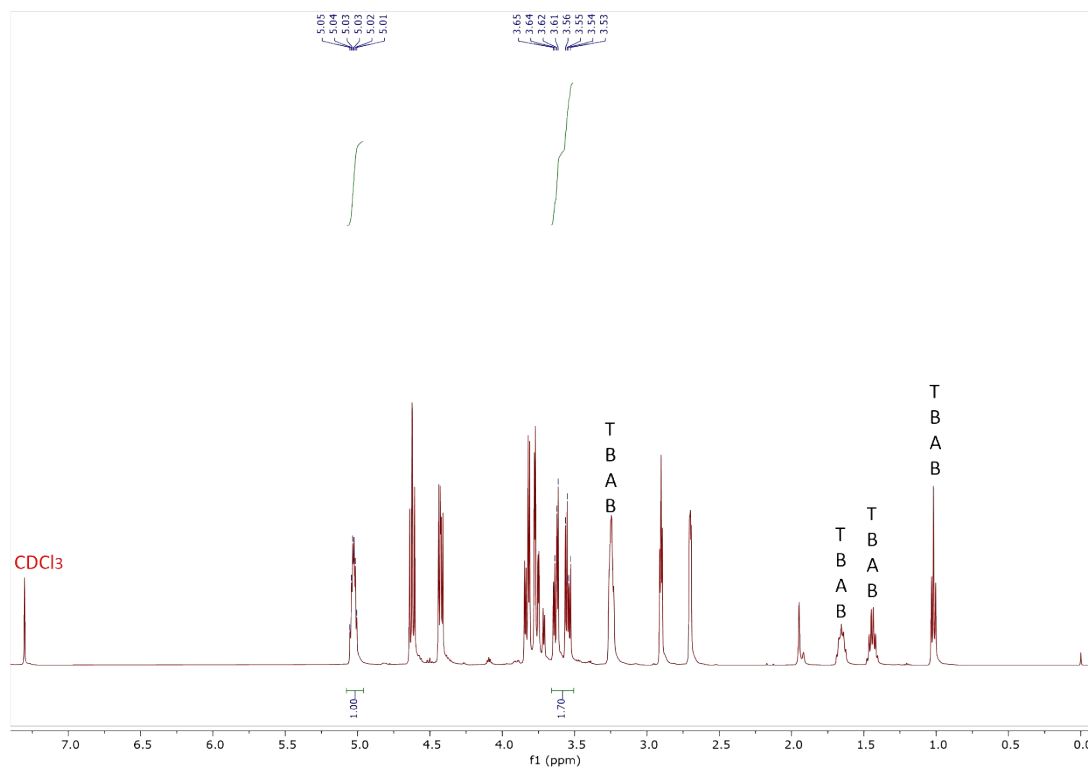
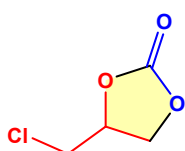


Figure S78: ^1H NMR spectrum of **catalyst 1** Epichlorohydrin **EPH** in CDCl_3 at 60 C

The residue was dissolved in Dichloromethane (5 mL) and washed with water (5 mL) and brine (5 mL $\times$ 2), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane: ethyl acetate = 1 : 1) to afford the desired cyclic carbonate

Epichlorohydrin**EPH**



M.F. $\text{C}_4\text{H}_5\text{O}_3\text{Cl}$ (136.0) (130g, 96%yield); ^1H NMR (500 MHz, CDCl_3) δ : 4.94-4.99 (m, 1H, -CH-), 4.56 (t, J = 8.0 Hz, 1H - CH_2 - O(CO)O-), 4.34.3.37 (m, 1H - CH_2 -O(CO)O-), 3.76-3.80 (m, 1H - CH_2 -Cl), 3.67-3.71 (m, 1H - CH_2 -Cl); ^{13}C NMR (126 MHz, CDCl_3) δ 154.55(C, -O(CO)O-), 74.64(- CH_2), 67.06(-CH-), 44.11(- CH_2 -Cl)

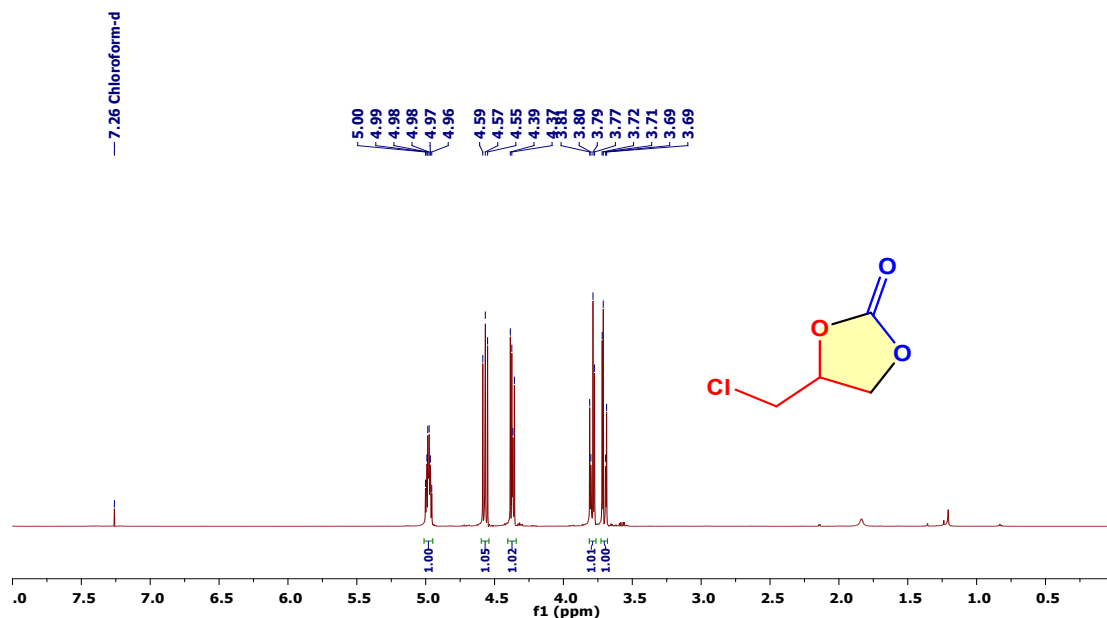
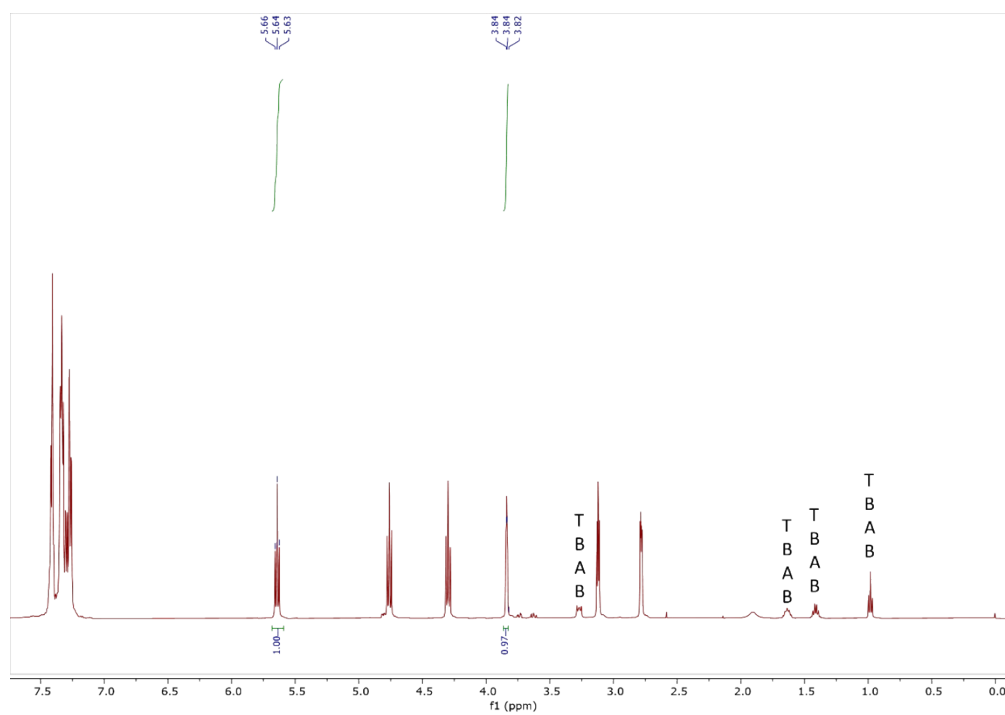
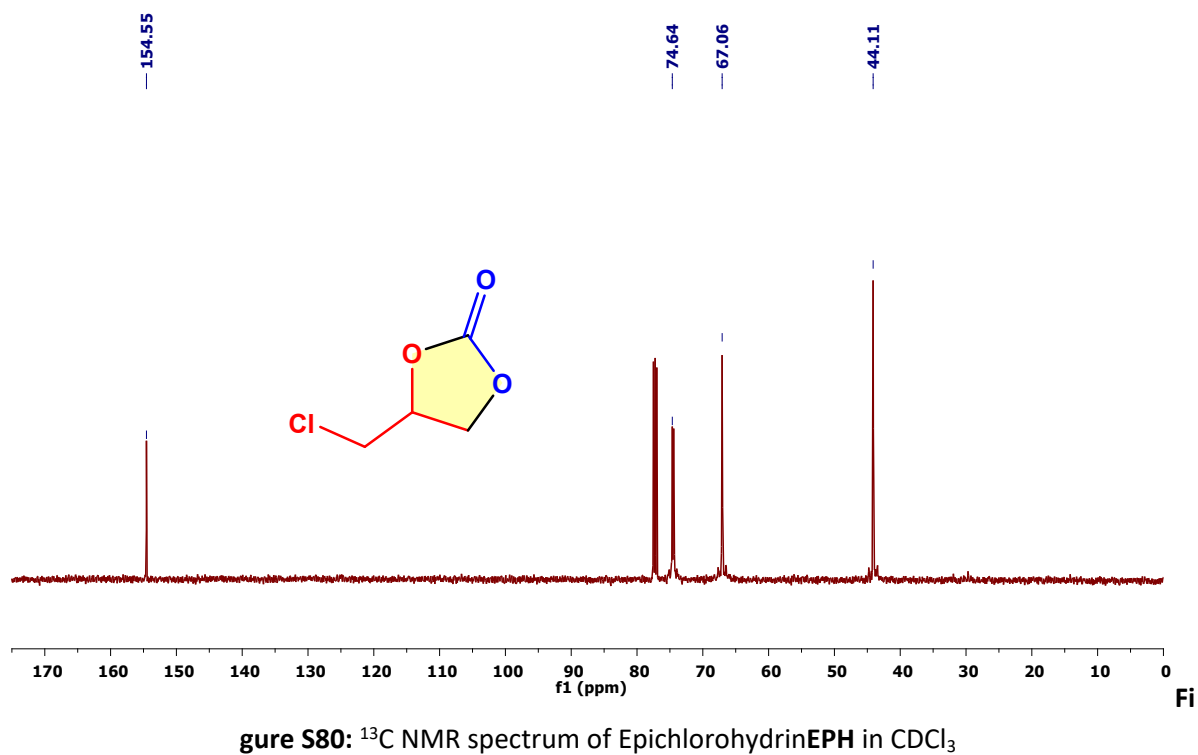


Figure S79: ^1H NMR spectrum of Epichlorohydrin (**EPH**) in CDCl_3



The residue was dissolved in Dichloromethane (5mL) and washed with water (5 ml) and brine (5 mL × 2), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane: ethyl acetate = 3 : 1) to afford the desired cyclic carbonate.

Styrene oxide (STO)

M. F. C₉H₆O₃ (M.W 162.0) (80g, 49%yield) (¹H NMR (500 MHz, CDCl₃) δ : 7.35-7.46 (m, 5H, Ph), 5.67 (t, *J*=8Hz, 1H, -CH-O(CO)O-), 4.48 (t, *J* = 8.0 Hz, 1H -CH₂-O(CO)O-), 4.33 (t, *J* = 8.0 Hz, 1H -CH₂-O(CO)O-); ¹³C NMR (126 MHz, CDCl₃) δ 154.99(C, -O(CO)O-), 135.91(C, Ph), 129.81(C, Ph), 126.00(C, Ph), 71.28(-CH₂), 78.11(-CH-).

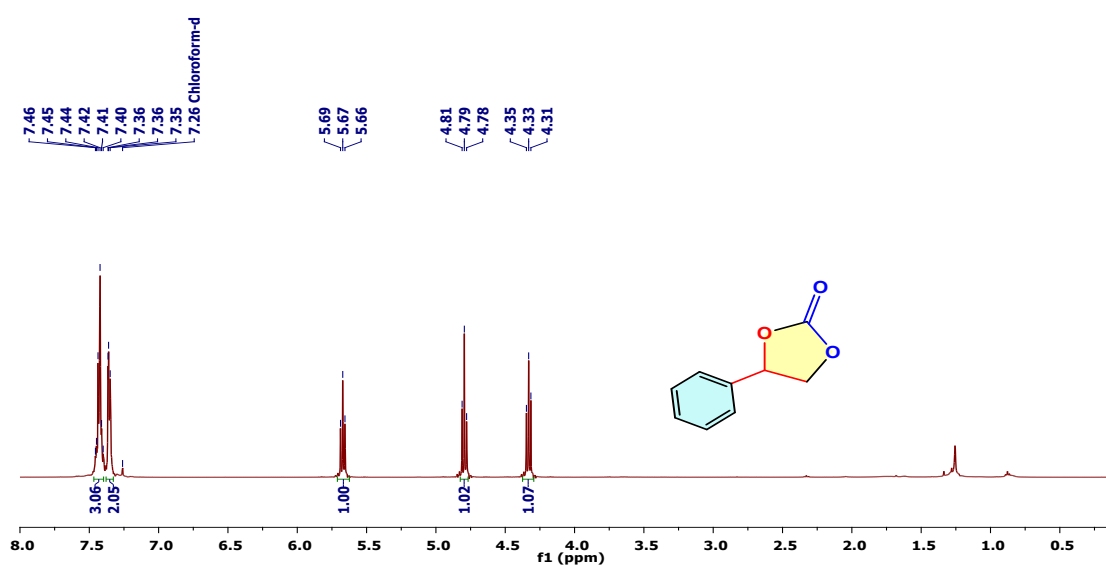


Figure S82: ¹H NMR spectrum of styrene oxide (STO) in CDCl₃

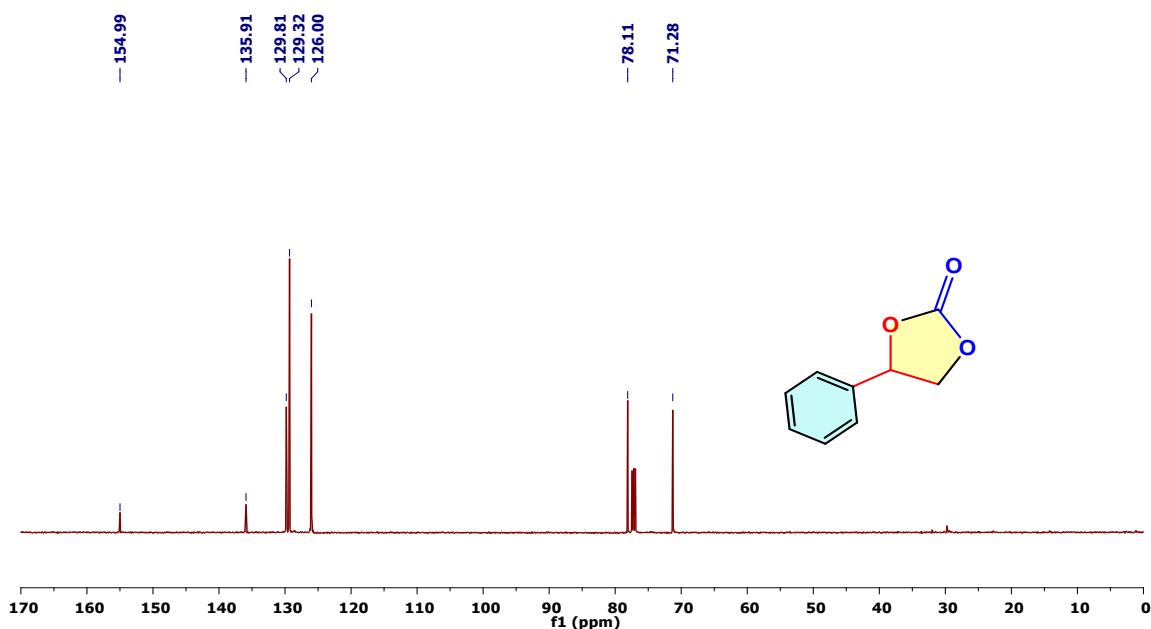


Figure S83: ¹³C NMR spectrum of styrene oxide (STO) in CDCl₃

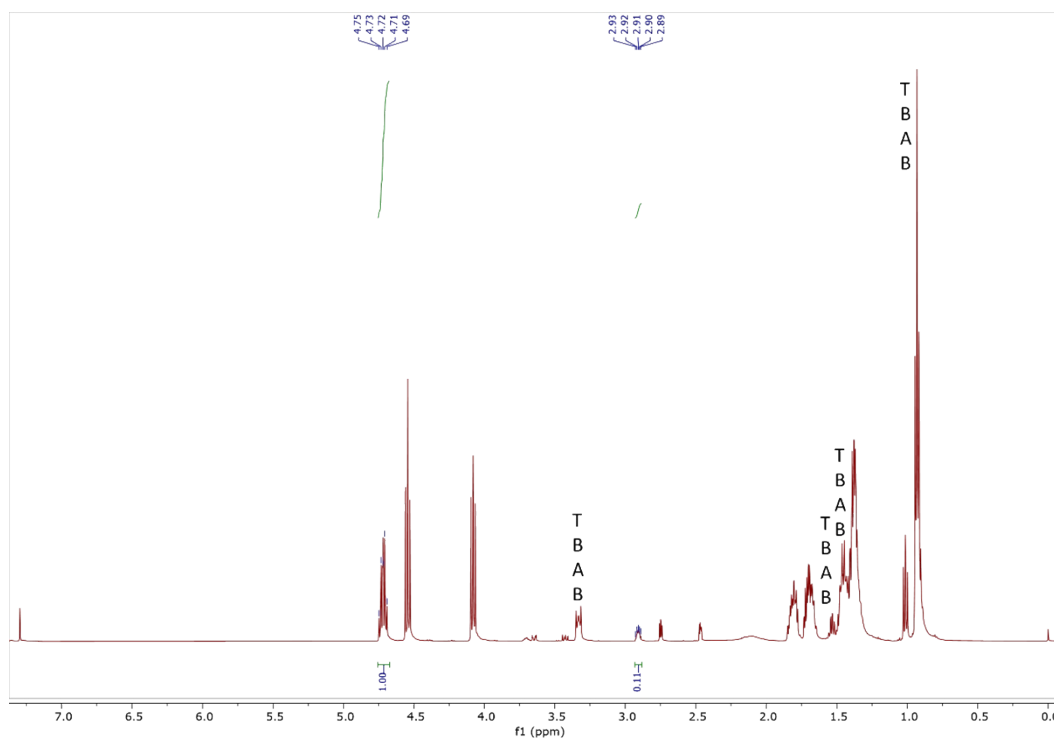
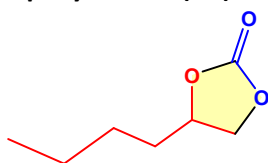


Figure S84: ^1H NMR spectrum of **1 Epoxy hexane (EH)** in CDCl_3

The residue was dissolved in Dichloromethane (5mL) and washed with water (5 ml) and brine (5 mL $\times$ 2), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography (hexane: ethyl acetate = 2 : 1) to afford the desired cyclic carbonate

Epoxy hexane(EH)



F. $\text{C}_7\text{H}_{12}\text{O}_3$ (M.W 144.1); (131g, 91% yield) ^1H NMR (500 MHz, CDCl_3) δ : 4.66 (m, , 1H,), 4.51 (t, J = 8.0 Hz, 1H $-\text{CH}_2-$ O(CO)O-), 4.06 (t, J = 7.5 Hz, 1H $-\text{CH}_2-\text{O}(\text{CO})\text{O}-$), 1.79-1.83(m, 1H, $-\text{CH}_2-$), 1.64-1.71 (m, 1H, $-\text{CH}_2-$), 1.31-1.47 (m, 4H, $-\text{CH}_2-$), 0.91(t, J =6.5Hz, 3H, CH_3); ^{13}C NMR (126 MHz, CDCl_3) δ : 155.16(C, $-\text{O}(\text{CO})\text{O}-$), 69.46(C, $-\text{CH}_2-$), 32.62($-\text{CH}-$), 29.74($-\text{CH}_2-$), 26.49($-\text{CH}_2-$), 22.31($-\text{CH}_2-$), 13.84($-\text{CH}_3$),

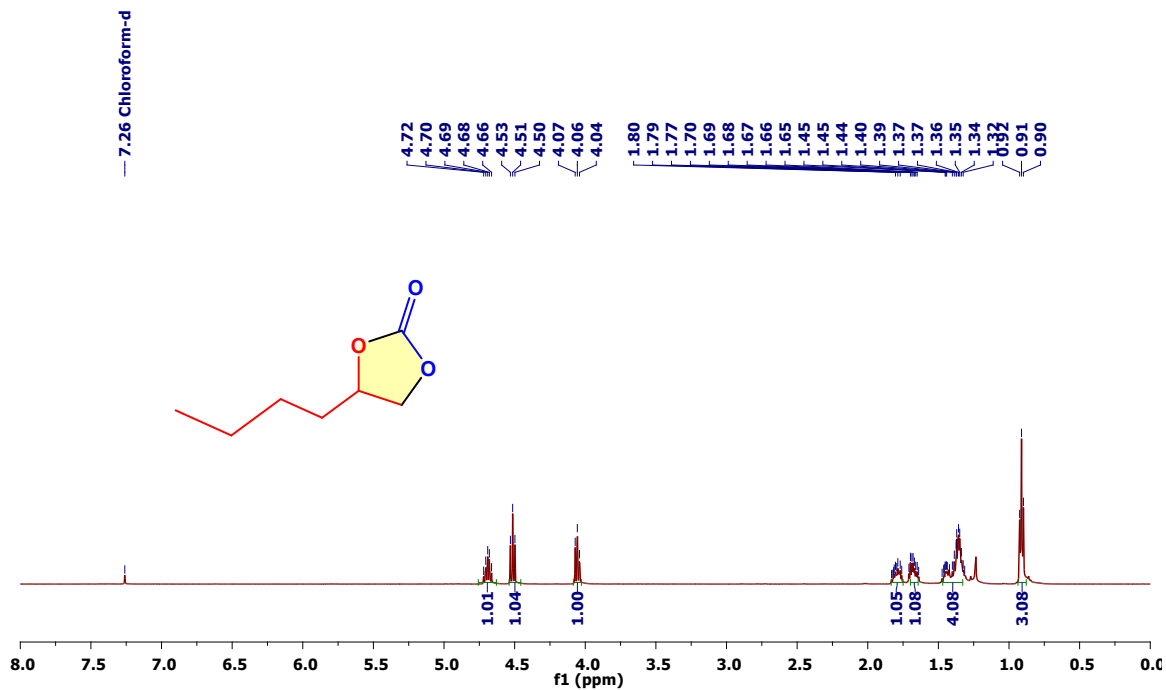


Figure S85: ^1H NMR spectrum of Epoxy hexane (EH) in CDCl_3

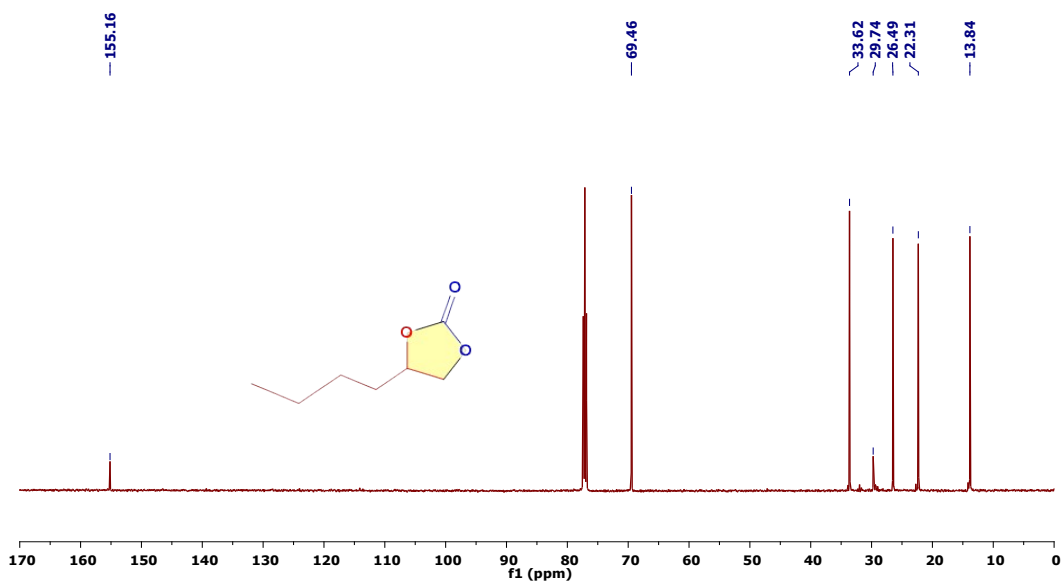
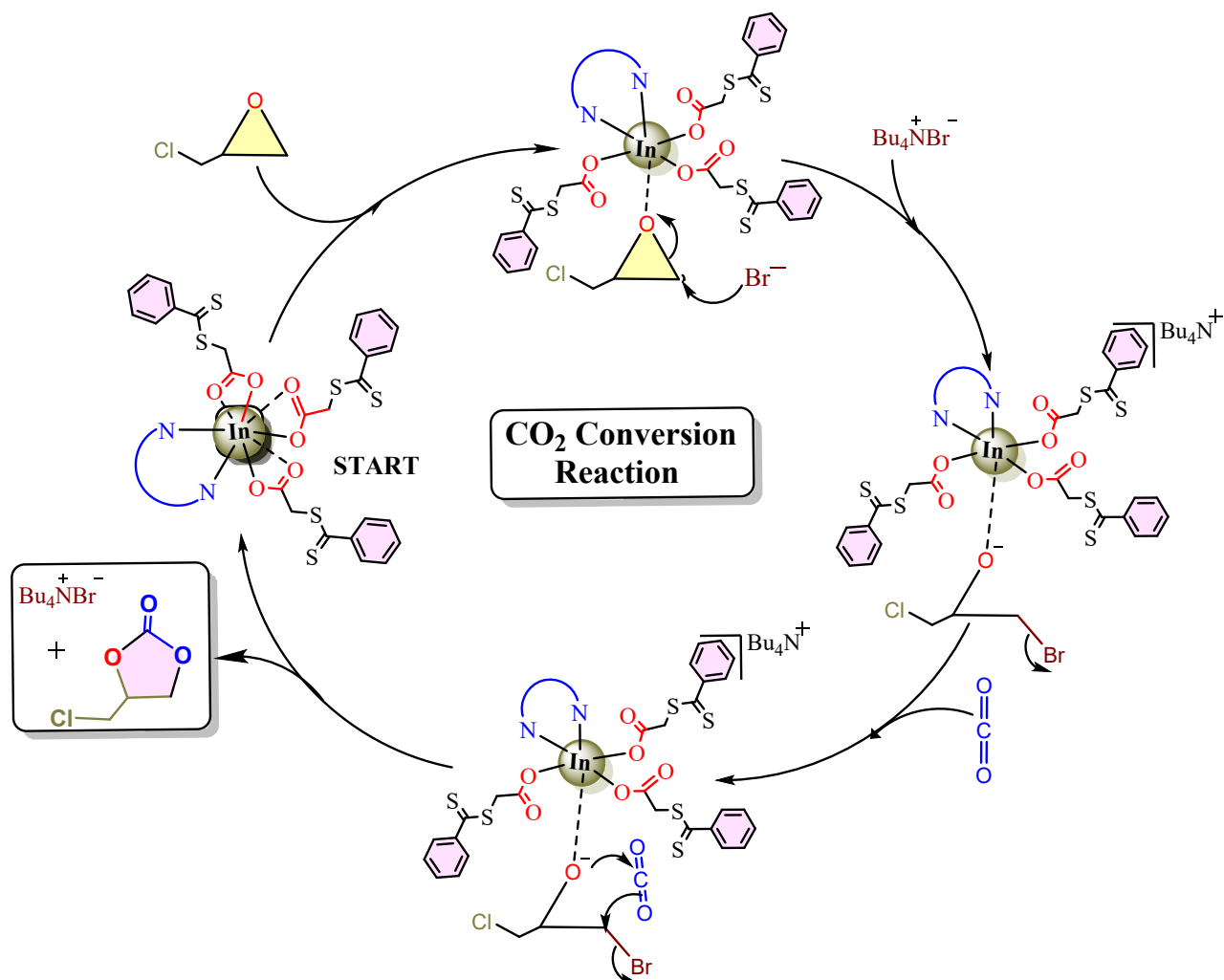


Figure S86: ^{13}C NMR spectrum of Epoxy hexane (EH) in CDCl_3

The catalyst **1** was added to TBAB as a cocatalyst to check the conversion of epoxide to cyclic carbonate using ^1H NMR spectra at different temperatures (100 C, 40 C and 60 C) Figure S1-3. The peak emerged around 5 ppm, depicting the formation of cyclic carbonate and the peak around 3.6

corresponds to the epichlorohydrin (EPH). The % conversion was calculated by using the integral of product and reactant peak.

Plausible Reaction Mechanism for the conversion of CO₂ into cyclic carbonate



Scheme S2. A Proposed mechanism for the coupling of epoxide and CO₂ catalyzed by In/TBAB

Reference

- 1 G. M. Sheldrick, *Acta Crystallogr. Sect. C Struct. Chem.*, 2015, **71**, 3–8.
- 2 G. M. Sheldrick, *Acta Crystallogr. Sect. Found. Adv.*, 2015, **71**, 3–8.
- 3 O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. Howard and H. Puschmann, *J. Appl. Crystallogr.*, 2009, **42**, 339–341.
- 4 K. Kumar, R. K. Sahani, S. Garai and S. Bhattacharya, *Dalton Trans.*, 2023, **52**, 17499–17513.
- 5 N. V. Lakshmi, S. E. Kiruthika and P. T. Perumal, *Can. J. Chem.*, 2013, **91**, 479–485.
- 6 M. R. Maurya, N. Kumar and F. Avecilla, *Inorg. Chem.*, 2024, **63**, 2505–2524.

- 7 S. A. Patra, M. Nandi, M. R. Maurya, G. Sahu, D. Mohapatra, H. Reuter and R. Dinda, *ACS Omega*, 2024, **9**, 31910–31924.