Supporting Information

for

Catalysis Activity and Chemoselectivity Control with the *Trans* Ligand in Ru-H Pincer Complexes

Mita Halder,^a Diana Castillo Cardenas,^a Angela Chartouni,^a and Damien B. Culver^a*

*Email: culver@ameslab.gov

Address where work was performed: ^aAmes National Laboratory, Ames, IA, 50011, United States

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Purification of 1

In a nitrogen filled glovebox, a 25 mL round bottom flask was charged with 1 g of **1** and 20 mL of dichloromethane. After stirring for 30 minutes at room temperature, the suspension was filtered through a layer of celite. The filtrate was concentrated under reduced pressure resulting in a pale yellow colored solid. The solid was further washed with diethyl ether (3×10 mL) and finally dried under vacuum for 1 hour. Percent recovery of **1** was 46 %. Representative ³¹P NMR spectra for **1** before and after purification are shown in Figure S1.



Figure S1. ³¹P{¹H} NMR spectrum of **1** (A) before and (B) after purification; [*]: residual impurities.

Catalysis data and NMR spectra

$$\begin{array}{c} O \\ R_1 \\ R_2 \end{array} \xrightarrow{iPrOH, 80 °C, t h} OH \\ -Me_2CO \end{array} OH$$

| | Table S1. Summary | of the catal | tic transfer h | vdrogenation o | of benzophenone | reactions. ^a |
|--|-------------------|--------------|----------------|----------------|-----------------|-------------------------|
|--|-------------------|--------------|----------------|----------------|-----------------|-------------------------|

| Entry | Catalyst | Catalyst | KO ^t Bu | Time (h) | Yield (%) ^b |
|-------|----------|-------------|--------------------|----------|------------------------|
| | | loading (%) | loading (%) | | |
| 1 | 2a | 2 | 5 | 1 | 96.2 (±0.6) |
| 2 | 2a | 2 | 2.5 | 1 | 91.8 (±0.4) |
| 3 | 2a | 0.5 | 2.5 | 1 | 97.0 (±0.15) |
| 4 | 2a | 0.5 | 0.625 | 4 | 97.4 (±0.9) |
| 5 | 2a | 0.1 | 0.625 | 5.5 | 97.1 (±1) |
| 6 | 1 | 0.1 | 0.625 | 5.5 | 96.6 (±1.46) |
| 7 | 2a | 0.1 | NA | 24 | <1 |
| 8 | none | NA | 0.625 | 24 | 8 |

a. Reaction conditions: benzophenone (0.15 mmol), ⁱPrOH (0.4 mL), catalyst, and KO^tBu were combined in an NMR tube along with a benzene- d_6 capillary and heated at 80 °C. b. Yields were determined by ¹H NMR spectroscopy and performed in duplicate. The values reported are the average and the errors are reported in the parentheses. NA = not applicable.



Figure S2. Representative ¹H NMR spectra to monitor the progress of the reaction between benzophenone (0.15 mmol) and ⁱPrOH in presence of **1** (0.1 mol% of Ru), and KO^tBu (0.625 mol%). [A] after 0 minutes; [B] after 75 minutes; [C] after 2.5 h; [D] after 3.45 h; [E] 5.5 h. * = Benzophenone; # = diphenylmethanol. ~ = solvent signal cutoff.



Figure S3. Representative ¹H NMR spectra to monitor the progress of the reaction between benzophenone (0.15 mmol) and iPrOH in presence of **2a** (0.1 mol% of Ru), and KOtBu (0.625 mol%). [A] after 15 minutes; [B] after 45 minutes; [C] after 2 h; [D] after 2.45 h; [E] 4.15 h. * = Benzophenone; # = diphenylmethanol, § = acetone. ~ = solvent signal cutoff.



Figure S4. Representative ¹H NMR spectra to monitor the progress of the reaction between benzophenone (0.15 mmol) and iPrOH in presence of **2d** (0.1 mol% of Ru), and KOtBu (0.625 mol%). [A] after 0 h; [B] after 4 h; [C] after 8 h; [D] after 12 h; [E] 16 h; [F] after 20 h; [G] after 24 h. * = Benzophenone; # = diphenylmethanol. ~ = solvent signal cutoff.



Figure S5. Representative ¹H NMR spectra to monitor the progress of the reaction between benzophenone (0.15 mmol) and iPrOH in presence of **3** (0.1 mol% of Ru), and KOtBu (0.625 mol%). [A] after 0 h; [B] after 0.5 h; [C] after 1 h; [D] after 1.5 h. * = Benzophenone; # = diphenylmethanol. ~ = solvent signal cutoff.



Figure S6. Representative ¹H NMR spectra to monitor the progress of the reaction between benzophenone (0.15 mmol) and iPrOH in presence of **4** (0.1 mol% of Ru), and KOtBu (0.625 mol%). [A] after 0 h; [B] after 0.5 h; [C] after 1 h; [D] after 1.5 h. * = Benzophenone; # = diphenylmethanol. ~ = solvent signal cutoff.



8.4 8.2 8.0 7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 3.6 3.4 ¹H NMR chemical shift (ppm)

Figure S7. Representative ¹H NMR spectra for the transfer hydrogenation of 4bromoacetophenone (0.15 mmol) with [A] **1** (0.1 mol%) after 2 h heating, [B] Catalyst **2a** (0.1 mol%) after 2 h heating, [C] Catalyst **3** (0.1 mol%) after 1 h heating, and [D] Catalyst **4** (0.1 mol%) after 1 h heating at 80 °C in presence of KO^tBu (0.625 mol%) and ⁱPrOH (0.4 mL). Δ = starting material; @ = product peaks; * = Internal standard; # = unknown byproducts. Conditions B and C are before the addition of internal standard. ~ = solvent signal cutoff.



Figure S8. Representative ¹H NMR spectra for the transfer hydrogenation of acetophenone (0.15 mmol) with [A] **1** (0.1 mol%) after 2 h heating, [B] Catalyst **2a** (0.1 mol%) after 2 h heating, [C] Catalyst **3** (0.1 mol%) after 1 h heating, and [D] Catalyst **4** (0.1 mol%) after 1 h heating at 80 °C in presence of KO^tBu (0.625 mol%) and ⁱPrOH (0.4 mL). @ = starting material; # = product peaks; * = Internal standard; + = unknown byproducts. ~ = solvent signal cutoff.



8.4 8.2 8.0 7.8 7.6 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0 5.8 5.6 5.4 5.2 5.0 4.8 4.6 4.4 4.2 4.0 3.8 ¹H NMR chemical shift (ppm)

Figure S9. Representative ¹H NMR spectra for transfer hydrogenation of 4methoxyacetophenone (0.15 mmol) with [A] **1** (0.1 mol%) after 2 h heating, [B] Catalyst **2a** (0.1 mol%) after 2 h heating, [C] Catalyst **3** (0.1 mol%) after 3 h heating, and [D] Catalyst **4** (0.1 mol%) after 3 h heating at 80 °C in presence of KO^tBu (0.625 mol%) and PrOH (0.4 mL). @ = starting material; # = product peaks; * = Internal standard; Conditions B is before the addition of internal standard. ~ = solvent signal cutoff.

Catalysis in the presence of excess ligand



Figure S10. ¹H NMR spectra of **3** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL), PMe₃ (3 µmol) and THF-*d*₈ (0.15 mL) [A] after 5 minutes sonication; [B] after 10 minutes heating at 80 °C; [C] after addition of benzophenone (0.15 mmol) followed by 0.5 h of heating at 80 °C. * = benzophenone; # = diphenylmethanol, ~ = solvent signal cutoff.



Figure S11. Expansion of the hydride region of the ¹H NMR spectra of **3** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL), PMe₃ (3 µmol) and THF- d_8 (0.15 mL) [A] after 5 minutes sonication; [B] after 10 minutes heating at 80 °C; [C] after addition of benzophenone (0.15 mmol) followed by 0.5 h of heating at 80 °C.



Figure S12. ³¹P{¹H} NMR spectra of **3** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL), PMe₃ (3 µmol) and THF- d_8 (0.15 mL) [A] after 5 minutes sonication; [B] after 10 minutes heating at 80 °C; [C] after addition of benzophenone (0.15 mmol) followed by 0.5 h of heating at 80 °C. * = Complex **3**'; # = PMe₃ ligand; + = unknown.

Complex speciation under catalysis conditions



Figure S13. ¹H NMR spectra of **1** (3 μ mol, 2 mol%), KO^tBu (7.5 μ mol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 m L) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).



Figure S14. Expansion of the hydride region of the ¹H NMR spectra of **1** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).



Figure S15. ³¹P{¹H} NMR spectra of **1** (3 μ mol, 2 mol%), KO^tBu (7.5 μ mol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).



Figure S16. ¹H NMR spectra of **2b** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C). ~ = signal cutoffs for isopropanol and THF-*d*₈.



Figure S17. Expansion of the hydride region of the ¹H NMR spectra of **2b** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).



Figure S18. ³¹P{¹H} NMR spectra of **2b** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF- d_8 (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C). # denotes **2b**' and * is an unknown species.



Figure S19. ¹H NMR spectra of **3** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF- d_8 (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C). ~ = signal cutoffs for isopropanol and THF- d_8 .



Figure S20. Expansion of hydride region in the ¹H NMR spectra of **3** (3 μ mol, 2 mol%), KO^tBu (7.5 μ mol, 5 mol%), ⁱPrOH (0.4 mL) and THF-*d*₈ (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).



³¹P NMR chemical shift (ppm)

Figure S21. ³¹P {¹H} NMR spectra of **3** (3 µmol, 2 mol%), KO^tBu (7.5 µmol, 5 mol%), ⁱPrOH (0.4 mL) and THF- d_8 (0.15 mL) before (A) and after heating at 80 °C for 30 minutes (B); and after addition of benzophenone (0.15 mmol) followed by 30 minutes of heating at 80 °C (C).

SCXRD structures and data



Figure S22. SCXRD for **2b**, thermal ellipsoids are drawn at 50% probability, and most hydrogens are omitted for clarity.



Figure S23. SCXRD for **2d**, thermal ellipsoids are drawn at 50% probability, and most hydrogens are omitted for clarity.



Figure S24. SCXRD for **3**, thermal ellipsoids are drawn at 50 % probability, and most hydrogens are omitted for clarity.



Figure S25. SCXRD for **4**, thermal ellipsoids are drawn at 50 % probability, and most hydrogens are omitted for clarity.

Table S2. Summary of relevant bond lengths and angles for the SCXRD structures of **2b**, **2d**, **3** and **4**.^a

| | Bond length (Å) | | | Bond angles | | | | |
|---------|--------------------------------------------------|--------------|----------------------------------------|---------------------------------------|------------|------------|-------------|------------------------|
| Complex | Ru-C/P (ligand <i>trans</i> to hydride) | Ru-C (CO) | Ru-P (Ph P N [⊬] P) | Ru-N (PhP N ^H P) | P-Ru- P | X-Ru- H | N-Ru- CO | Ru-C-N (isonitrile) |
| 2b | 2.048(6) | 1.841(6) | 2.3258(14)/ 2.3199(14) | 2.180(4) | 164.11 | 175.09 | 169.93 | 173.60 |
| 2d | 2.043(3) | 1.856(3) | 2.3145(7)/ 2.3237(7) | 2.189(2) | 165.7 | 173.5 | 170.45 | 172.23 |
| 3 | 2.4358(6) | 1.847(2) | 2.3160(5)/ 2.3248(5) | 2.2027(17) | 159.61 | 176.94 | 174.01 | NA |
| 4 | 2.168(4) | 1.819(4) | 2.2885(9)/ 2.3450(9) | 2.224(3) | 161.74 | 179.68 | 166.26 | NA |

a. Errors are indicated in the parantheses. NA = not applicable.

| Parameter | 2b | 2d |
|-------------------------------------|--------------------------------------------------------------------|--------------------------------------------------------------------|
| Empirical formula | C ₅₈ H ₅₉ BN ₂ OP ₂ Ru | C ₆₁ H ₅₇ BN ₂ OP ₂ Ru |
| Formula weight | 973.95 | 1006.89 |
| Temperature/K | 173(2) K | 173(2) K |
| Crystal system | triclinic | monoclinic |
| Space group | P-1 | P 1 21/c 1 |
| a/Å | 12.7846(8) | 13.1411(5) |
| b/Å | 13.8907(11) | 12.9946(5) |
| c/Å | 17.4106(13) | 29.4189(11) |
| α/° | 101.625(3) | 90 |
| β/° | 105.684(2) | 90.2580(10) |
| γ/° | 94.257(3) | 90 |
| Volume/Å ³ | 2888.4(4) | 5023.6(3) |
| Z | 2 | 4 |
| ρ _{calc} g/cm ³ | 1.286 | 1.331 |
| µ/mm ⁻¹ | 0.375 | 0.420 |
| F(000) | 1176.0 | 2092.0 |
| Crystal size/mm ³ | 0.25 × 0.24 × 0.1 | 0.42 × 0.38 × 0.12 |
| Radiation | ΜοΚα (λ = 0.71073) | ΜοΚα (λ = 0.71073) |
| 2O range for data collection/° | 2.29 to 23.17 | 2.09 to 27.50 |
| Index ranges | -14 ≤ h ≤ 14, -15 ≤ k ≤ 15, - 19 ≤ l ≤ 19 | -16 ≤ h ≤ 16, -15 ≤ k ≤ 15, - 36 ≤ l ≤ 36 |

Table S3. Crystallographic data and refinement parameters for **2b** and **2d**.

| Reflections collected | 93094 | 226272 |
|-------------------------------------------|----------------------------------------|---------------------------|
| Independent reflections | 8193 (Rint = 0.1290, Rsig = 0.0566) | 11499 |
| Data/restraints/parameters | 8193 / 0 / 679 | 11499 / 0 / 617 |
| Goodness-of-fit on F ² | 1.023 | 1.038 |
| Final R indexes [I>=2σ (I)] | R1 = 0.0562, wR2 = 0.1387 | R1 = 0.0395, wR2 = 0.0896 |
| Final R indexes [all data] | R1 = 0.0768, wR2 = 0.1531 | R1 = 0.0515, wR2 = 0.0979 |
| Largest diff. peak/hole /e Å ⁻ | 2.404/-0.771 | 0.753/ -0.585 |

Table S4. Crystallographic data and refinement parameters for **3** and **4**.

| Parameter | 3 | 4 |
|-------------------|---------------|--------------------------------------------------------------------|
| Empirical formula | C56H59BNOP3Ru | C ₅₈ H ₅₈ BN ₃ OP ₂ Ru |
| Formula weight | 966.90 | 986.95 |
| Temperature/K | 273 K | 173 K |
| Crystal system | monoclinic | monoclinic |
| Space group | P 1 21/n 1 | P 1 21/n 1 |
| a/Å | 9.9730(3) | 11.4158(5) |
| b/Å | 22.1902(7) | 34.6443(15) |
| c/Å | 22.4709(7) | 12.7068(5) |
| α/° | 90 | 90 |
| β/° | 97.363(10) | 99.007(10) |
| γ/° | 90 | 90 |

| Volume/Å ³ | 4931.9(3) | 4963.5(4) |
|-------------------------------------------|----------------------------------------------|--------------------------------------------------|
| Z | 4 | 4 |
| $\rho_{calc}g/cm^3$ | 1.302 | 1.321 |
| µ/mm ⁻¹ | 0.455 | 0.424 |
| F(000) | 2016 | 2056.0 |
| Crystal size/mm ³ | 0.29 × 0.14 × 0.11 | 0.34 × 0.30 × 0.08 |
| Radiation | ΜοΚα (λ = 0.71073) | ΜοΚα (λ = 0.71073) |
| 2O range for data collection/° | 2.33 to 28.71 | 2.15 to 25.68 |
| Index ranges | -12 ≤ h ≤ 13, -29 ≤ k ≤ 29, - 30 ≤ l ≤ 30 | -13 ≤ h ≤ 13, -42 ≤ k ≤ 42, - 15 ≤ l ≤ 15 |
| Reflections collected | 126831 | 167482 |
| Independent reflections | 12745 (Rint = 0.0532, Rsig = 0.0263) | 9416 ($R_{int} = 0.1034$, $R_{sig} = 0.0380$) |
| Data/restraints/parameters | 12745 / 0 / 578 | 9416 / 0 / 604 |
| Goodness-of-fit on F ² | 1.040 | 1.044 |
| Final R indexes [I>=2σ (I)] | R1 = 0.0367, wR2 = 0.0946 | R1 = 0.0450, wR2 = 0.0951 |
| Final R indexes [all data] | R1 = 0.0459, wR2 = 0.1023 | R1 = 0.0698, wR2 = 0.1082 |
| Largest diff. peak/hole /e Å ⁻ | 1.002/ -0.717 | 0.569/ -0.449 |



Figure S26. FTIR spectrum of 2a.

Figure S27. ¹H NMR spectrum of **2a** in THF-*d*₈.

Figure S28. ³¹P{¹H} NMR spectrum of **2a** in THF-*d*₈; [*]: residual impurities.

Figure S30. ¹H-¹H COSY NMR spectrum of **2a** in THF-*d*₈.

Figure S31. Expansions of the aromatic and aliphatic regions of the ¹H-¹H COSY NMR spectrum of **2a** in THF- d_8 .

Figure S32.¹H-¹³C HSQC NMR spectrum of **2a** in THF-*d*₈.

Figure S33.¹H-¹³C HMBC NMR spectrum of **2a** in THF-*d*₈.

Figure S34. Expansions of the aliphatic regions of the ${}^{1}H{}^{-13}C$ HMBC NMR spectrum of **2a** in THF-*d*₈.

Figure S35. ¹H-¹H NOSY NMR spectrum of **2a** in THF-*d*₈.

Figure S36. FTIR spectrum of **2b**.

Figure S37. ¹H NMR spectrum of **2b** in THF-*d*₈.

105 100 95 90 85 80 75 70 65 60 55 50 45 40 35 30 25 20 15 10 ³¹P NMR chemical shift (ppm)

Figure S38. ³¹P{¹H} NMR spectrum of **2b** in THF-*d*₈; [*]: residual impurities.

Figure S41. Expansions of the aromatic and aliphatic regions of the ¹H-¹H COSY NMR spectrum of **2b** in THF- d_8 .

Figure S42. ¹H-¹³C HSQC NMR spectrum of **2b** in THF-*d*₈.

Figure S43. ¹H-¹³C HMBC NMR spectrum of **2b** in THF-*d*₈.

Figure S44. Expansions of the aromatic regions of the ${}^{1}H{}^{-13}C$ HMBC NMR spectrum of **2b** in THF-*d*₈.

Figure S45. ¹H-¹H NOSY NMR spectrum of **2b** in THF-*d*₈.

Figure S46. FTIR spectrum of **2b**.

Figure S48. ³¹P{¹H} NMR spectrum of **2c** in THF-*d*₈; [*]: residual impurities.

Figure S49. ¹³C{¹H} NMR spectrum of **2c** in THF- d_8 . ~ = solvent signal cutoff.

Figure S50. ¹H-¹H COSY NMR spectrum of **2c** in THF-*d*₈.

Figure S51. ¹H-¹³C HSQC NMR spectrum of **2c** in THF-*d*₈.

Figure S52. Expansions of the aromatic and aliphatic regions of the ¹H-¹³C HSQC NMR spectrum of **2c** in THF- d_8 .

Figure S53. ¹H-¹³C HMBC NMR spectrum of **2c** in THF-*d*₈.

Figure S54. ¹H-¹H NOSY NMR spectrum of **2c** in THF-*d*₈.

Figure S55. FTIR spectrum of 2d.

Figure S56. ¹H NMR spectrum of **2d** in THF-*d*₈.

Figure S57. ³¹P{¹H} NMR spectrum of **2d** in THF-*d*₈.

Figure S58. ¹³C{¹H} NMR spectrum of **2d** in THF- d_8 . ~ = solvent signal cutoff.

Figure S59. ¹H-¹H COSY NMR spectrum of **2d** in THF-*d*₈.

Figure S60.¹H-¹³C HSQC NMR spectrum of **2d** in THF-*d*₈.

Figure S61. Expansion of the aliphatic regions of the ¹H-¹³C HSQC NMR spectrum of **2d** in THF- d_8 .

Figure S62. ¹H-¹³C HMBC NMR spectrum of **2d** in THF-*d*₈.

Figure S63. ¹H-¹H NOSY NMR spectrum of **2d** in THF-*d*₈.

Figure S64. FTIR spectrum of 3.

Figure S65. ¹H NMR spectrum of **3** in THF-*d*₈.

Figure S66. ³¹P{¹H} NMR spectrum of **3** in THF-*d*₈; [*]: residual impurities.

Figure S68. ¹H-¹H COSY NMR spectrum of **3** in THF-*d*₈.

Figure S69. ¹H-¹³C HSQC NMR spectrum of **3** in THF-*d*₈.

Figure S70. ¹H-¹³C HMBC NMR spectrum of **3** in THF-*d*₈.

Figure S71. ¹H-¹H NOSY NMR spectrum of **3** in THF- d_8 .

Figure S72. FTIR spectrum of 4.

Figure S73. ¹H NMR spectrum of **4** in THF-*d*₈.

Figure S74. ¹H NMR spectrum of **4** in THF- d_8 at 50 °C.

Figure S75. Variable temperature ¹H NMR spectrum of **4** in THF- d_8 . Blue lines highlight region containing the NHC backbone protons.

Figure S76. ³¹P{¹H} NMR spectrum of **4** in THF-*d*₈; [*]: residual impurities.

Figure S78. ¹H-¹H COSY NMR spectrum of **4** in THF-*d*₈.

Figure S79. ¹H-¹³C HSQC NMR spectrum of **4** in THF-*d*₈.

Figure S80. ¹H-¹³C HMBC NMR spectrum of **4** in THF-*d*₈.

Figure S81. Expansions of the aromatic (A) and aliphatic (B) regions of the ${}^{1}H{}^{-13}C$ HMBC NMR spectrum of **4** in THF-*d*₈.

Figure S82. ¹H-¹H NOSY NMR spectrum of **4** in THF-*d*₈.

Figure S83. ¹H NMR spectra to monitor the progress of the reaction between **2a** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C; \sim = signal cutoff.

¹H NMR chemical shift (ppm)

Figure S84. Expansion of the hydride region of the ¹H NMR spectra to monitor the progress of the reaction between **2a** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C.

Figure S85. ³¹P{¹H} NMR spectra to monitor the progress of the reaction between **2a** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C.

Figure S86. ¹H NMR spectra to monitor the progress of the reaction between **4** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C; \sim = signal cutoff.

Figure S87. Expansion of the hydride region of the ¹H NMR spectra to monitor the progress of the reaction between **4** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C.

Figure S88. ³¹P{¹H} NMR spectra to monitor the progress of the reaction between **4** and PMe₃. [A] after 30 minutes heating; [B] after addition of KO^tBu followed by 30 minutes heating at 80 °C. \sim = signal cutoff for PMe₃.