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Supplementary Information

Synergistic Effect of a Berlin Green Framework for Highly Efficient

Moisture–Electric Energy Transformation

Minjae Song,^{‡a} Daewoong Kim,^{‡a} Hyewon Lee,^a Hyunsoo Han^a and Sangmin Jeon^{*a}

^aDepartment of Chemical Engineering, Pohang University of Science and Technology

(POSTECH), 77 Cheongam-Ro, Pohang, Gyeongbuk, Republic of Korea

[‡]These authors contributed equally to this work.

* Author to whom correspondence should be addressed. E-mail: jeons@postech.ac.kr



Fig. S1. SEM images of (a) BG particles in the BGC layer and (b) NaCl particles in the NC layer. NaCl and BG were incorporated in each layer in the form of crystalline particles, with sizes of ~1 μ m (BG) and ~0.5 μ m (NaCl).



Fig. S2. Time-dependent changes in (a) open-circuit voltage and (b) short-circuit current density of BGC–NC at 90% RH for 10 h. Although the current density decreased from its peak value of 2,910 μ A/cm², it sustained a level of 140 μ A/cm² even after 10 h.



Fig. S3. Zeta potential of BGC and GC in DI water. Since BGC contains carboxyl and cyano functional groups, its surface becomes negatively charged at -57.8 mV.



Fig. S4. Time-dependent changes in (a) voltage and (b) current density of GC–NC (black), PGC–NC (blue), and BGC–NC (green) at 90% RH for 3 h. The voltage (0.65 V) and current outputs (1,070 μ A/cm²) of PGC–NC were higher than those of GC–NC due to the increased surface area for the physical adsorption of Na⁺ ions, but lower than those of BGC–NC.



Fig. S5. Time-dependent changes in (a) voltage and (b) current density of BGC–NC at various RH values for 3 h. Inset provides a magnified view for below 75% RH. The voltage output increased with increasing RH, reaching its peak at 90% RH (1.28 V). Similarly, the current output increased with increasing RH, reaching 2,910 μ A/cm² at 90% RH.



Fig. S6. Variations of the voltage (red) and current density (blue) output as a function of BGC layer thickness of BGC–NC. While keeping the thickness of the NC layer fixed at 0.2 mm, the thickness of the BGC layer was varied to 0.3, 0.6, 1.1, and 2.2 mm. The MEET performance increased as the BGC thickness increased, whereas it remained constant when the BGC layer thickness reached 2.2 mm.



Fig. S7. Variations of the voltage (red) and current density (blue) output as a function of NC layer thickness of BGC–NC. While keeping the thickness of the BGC layer fixed at 1.1 mm, the thickness of the NC layer was varied to 0.1, 0.2, and 0.4 mm. The changes in the voltage and current outputs were negligible, indicating that the NC layer contains excess sodium ions that can reduce BG to PB.



Fig. S8. Time-dependent changes in (a) voltage and (b) current density of BGC–NC (green) and BC–NC (black) at 90% RH for 3 h. (c) Electrical conductivity of BGC and BC. BC–NC reaches similar voltage (1.28 V), but has lower current and electrical conductivity (59.3 μ A/cm², 0.347 μ S/cm) compared to BGC–NC (2,910 μ A/cm², 2,230 μ S/cm), highlighting that GO acts as a conductive material within BGC.



Fig. S9. Time-dependent changes in voltage when a 2000 μ F capacitor is charged by three GC–NCs (black) or three BGC–NCs (green) at 90 % RH. When a 2000 μ F capacitor was connected to three BGC-NCs, the BGC-NCs exhibited a much faster charging rate and reached a higher voltage than the GC-NCs.