Supplementary Information (SI) for Energy & Environmental Science. This journal is © The Royal Society of Chemistry 2025

# **Supporting Information**

## Enhanced electrical performance of perovskite solar cells via strain engineering

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#### Methods

### Materials

All the materials were used as received without purification. *N*, *N*-dimethylformamide (DMF, 99.8 %), dimethyl sulfoxide (DMSO, 99.7 %) and Nickel (II) acetylacetonate [Ni(acac)<sub>2</sub>, 95 %] was purchased from Sigma Aldrich. Formamidinium iodide (FAI, 99.99 %) and methylammonium iodide (MAI, 99.99 %) were purchased from Great Cell Solar. Caesium iodide (CsI, 99.9 %) was purchased from Alfa-Aesar. Lead iodide (PbI<sub>2</sub>, 98 %) and [4-(3,6-Dimethyl-9H-carbazol-9-yl) butyl] phosphonic acid (Me-4PACz, 99%) were purchased from TCI. 3-fluoro-phenethylammonium iodide (3F-PEAI,  $\geq$ 99.5%) and methylammonium chloride (MACl,  $\geq$ 99.5%) was purchased from Nano-C. Hydrochloric acid (HCl) was purchased from Sinopharm. All other solvents were anhydrous and purchased from Across Organic. The FTO coated glass was purchased from Suzhou Shang Yang Solar Technology Co., Ltd. The synthesis of 2TI is detailed in earlier reports<sup>1,2</sup>.

#### **Device fabrication**

The FTO substrates ( $2.5 \times 2.5$  cm, 7  $\Omega$ /square) were cleaned in ultrasonic bath with detergent, deionized water and ethanol for 10 min. Before film deposition, the FTO substrates were treated with ultraviolet ozone for 15 min. The NiO<sub>x</sub> films were prepared via a sol-gel method as described in the previous work<sup>3</sup>. The Ni(acac)<sub>2</sub> was dissolved in anhydrous ethanol (32.11 mg ml<sup>-1</sup>) with the addition of 17 µl hydrochloric acid, and stirred overnight at room temperature. Prior to use, the solution was filtered through a 0.45 µm PTFE filter. The solution was spin coated on FTO substrates at 4,000 rpm for 40 s and annealed at 150 °C for 10 min. The entire deposition process was carried out in a dry box ( $\leq$ 30% RH). Then, the samples were transferred to a 180 °C preheating hotplate in ambient air ( $\leq$ 50% RH) for 10 min, followed by heating at 400 °C for 45 minutes. After annealing, the NiO<sub>x</sub> samples were transferred to the N<sub>2</sub> glovebox immediately for the subsequent interlayer deposition. Me-4PACz (1mg ml<sup>-1</sup> in anhydrous ethanol; ultrasonic bath for 15 min) was spin coated at 3,000 rpm for 30 s,

and annealed at 100°C for 10 min. We use 1.3 M Cs<sub>0.05</sub>FA<sub>0.9</sub>PbI<sub>3</sub> perovskites as the photoactive layers, and the precursor solutions were prepared by dissolving the 653.25 mg PbI<sub>2</sub>, 12.99 mg CsI, and 212.38 mg FAI in a 1 mL mixed solvent (5:1 v:v) of DMF and DMSO, respectively. 0.97 mg MABr and 3.19 mg PbBr<sub>2</sub> and 9.79 mg MACl were additionally added into the precursor solution for better crystallization and perovskite phase transformation. The precursor solutions were stirred overnight at room temperature in a nitrogen-filled glovebox. The solutions were filtered with a 0.45 µm PTFE filter before use. The perovskite solutions were then spun on the substrates at 1,000 rpm. for 10 s and 4000 rpm. for 40 s. 150 µl of chlorobenzene (with or without 0.6mg ml<sup>-1</sup> 2TI) was added dropwise onto the substrate 5 s before the end. The substrates were then placed on a pre-heated hotplate and annealed at 100 °C for 30 min. After cooling down, a mixture of 3F-PEAI (2 mg ml<sup>-1</sup>) and MAI (1 mg ml<sup>-1</sup>) was dissolved in IPA–DMF (200:1, v/v), spun onto the as-prepared perovskite films. Finally, C<sub>60</sub> (20 nm), BCP (7 nm), and copper electrode (100 nm) were deposited sequentially using thermal evaporation under a high vacuum (≤8×10<sup>-5</sup> Pa).

**Device characterizations.** The Fourier-transform infrared (FTIR) spectroscopy was obtained by using Thermo Fisher NICILET iS10. Nuclear magnetic resonance (NMR) characterization was characterized with a Bruker Avance NEO 600. The solvent in the NMR measurements is deuterated DMSO and the concentration of each solute in the solutions is 10 mg/ml. X-ray photoelectron spectroscopy (XPS) measurements were performed on a Thermo Fisher Scientific ESCALAB Xi+ instrument with a monochromatic Al K Alpha (1486.68 eV) X-ray source, and the samples were measured under an ultrahigh vacuum. The binding energies were calibrated using the C (1s) carbon peak (284.8 eV). All the high-resolution spectra were obtained under constant analyzer energy (CAE) mode with a pass energy of 30 eV and step size of 0.05 eV. X-ray diffraction (XRD) patterns were carried out by a Rigaku Smart lab 3 kW XRD (Rigaku Ltd., Japan) using Cu K<sub>a</sub> radiation ( $\lambda = 0.15405$  nm). The surface morphology of the samples was obtained using a scanning electron microscope (Gemini SEM 500, Zeiss). EQE spectra were measured using a Zolix-SCS600 system, calibrated

by a calibrated crystalline Si reference cell. The custom-made in situ PL testing setup is positioned next to a hot plate, with an excitation laser (640 nm) being directed onto the sample through a fiber optic. During the annealing process of the perovskite, the PL spectral signals are collected every 20 ms by a detector connected to the spectrometer (TE-Cooled Micro Spectrometer, ATP5020R, Optosky Optical Ltd.). The J-V curves were measured (2400 Series Source Meter, Keithley Instruments) in the air under simulated AM 1.5G sunlight at 100 mW/cm<sup>2</sup> irradiation by a WACOM Class AAA (WXS-100S-5) simulator, with the intensity calibrated with a Fraunhofer ISEcalibrated silicon reference cell with KG1 filter. The mismatch factor was calculated to be less than 1% and applied to calibrate the AM 1.5 G 100 mW cm<sup>-2</sup> equivalent irradiance. The active area of the solar cell measured under AM 1.5 G is 0.07 cm<sup>2</sup>, defined by an opaque aperture mask. The J-V curves were measured at a scan rate of 200 mVs<sup>-1</sup> (with no preconditioning) in ambient air (~25 °C and relative humidity ~ 50 %). For stabilized output measurements at the MPP, the device testing chamber was left under ambient conditions. Solar cells were measured at the MPP voltage, (determined from J-V sweeps in reverse scans) and the current output was tracked over time. For TRPL mapping measurement, a laser scanning confocal microscope system equipped 405nm laser (1 MHz, 3.1 µJ cm<sup>-2</sup> per pulse) and galvo-based scanner was used in air. The objective was 100X with numerical aperture of 0.95. The lifetimes were collected based on time-correlated single photon counting (TCSPC), the resolution was set to 7ps with 4096 channels, The diameter of signal collection area for each sample was  $20 \times 20$  um<sup>2</sup>.

### Density-functional theory (DFT) calculation

The Vienna Ab initio Simulation Package (VASP) was used for theoretical calculations based on the density functional theory (DFT). The electron-ion potential and exchangecorrelation functional were described by projected augmented wave (PAW) and generalized gradient approximation (GGA)<sup>4,5</sup>, respectively. The kinetic energy cutoff was set to 500 eV for the relaxation of geometric structures. A  $5 \times 5 \times 5$  Monkhorst-Pack *k*-point mesh was used in the Brillouin zone<sup>6</sup>. To avoid the periodic interaction, a vacuum region of 20 Å was used to avoid the periodic interaction. The stress force and energy convergence criterions are chosen as 0.01 eV/Å and 10<sup>-5</sup> eV, respectively. To treat the van der Waals interaction between perovskites and organic molecules, a semiempirical DFT-D2 method was employed<sup>7</sup>. The uniaxial strain applied to the perovskites is  $\varepsilon = (a - a_0)/a_0 \times 100\%$ , where *a* and  $a_0$  are the strained and unstrained lattice constants.

**Device simulation.** Simulations of PSCs were conducted using the heterojunction solar cell simulator SCAPS-1D, version 3.3.11.

**Long-term stability tests.** The perovskite devices were encapsulated with ultraviolet adhesive (LT-U001, Lumtec) and a glass coverslip as a barrier layer in a nitrogen-filled glovebox. Before encapsulation, we blew the devices with a nitrogen gun to remove contamination and stored them in a nitrogen-filled glovebox for overnight to remove any residual moisture. We use a UV-containing large-area LED array as the light source. We do not have control over the humidity but monitored the laboratory humidity, which is around a relative humidity of ~ 50%, during the aging. For the MPP tracking, as the  $V_{\text{MPP}}$  may change over time, we have applied perturb-and-observe algorithms that carry out a periodic perturbation of the operating voltage maximizing the total power output of the device.

#### Note S1. Discussion on crystallization kinetics

Based on the evolution of PL peak positions and intensities, we assume the crystallization process of the perovskite film to occur in four stages:

*Stage I: Rapid Nucleation.* In this stage, the PL intensity increases along with a blue shift in the peak position, transitioning from 788 nm to 772 nm. Previous study revealed that during thermal annealing process, there exists a correlation between the peak position and grain size, where smaller grains lead to shorter emission wavelengths<sup>8</sup>. Therefore, we posit that the blue shift is attributed to the formation of numerous crystalline nuclei at the liquid-gas interface. These nuclei are smaller than those formed during the anti-solvent quenching process, a consequence of surface solvent evaporation driven by the temperature increase in the perovskite wet film<sup>9</sup>. Moreover, as illustrated by the dashed lines in Fig. 1e, g in the main text, the target film reaches the first maximum intensity faster than the control film by 20 ms, implying a faster nucleation rate that is advantageous for the subsequent crystallization. However, given that our in situ PL test involves data collection at 20 ms intervals, we consider this slight difference to be negligible. In summary, there are no significant differences between the two films in this stage.

*Stage II: Crystallization-Dissolution Dynamic Equilibrium.* In this stage, both PL intensity and peak position exhibit relatively stable behaviors. This stage involves a competition between the continued growth of crystalline nuclei in the upper wet perovskite film and the dissolution of the upper crystalline part due to the solvent evaporation of the lower part solution<sup>10</sup>. The former process enhances PL intensity, whereas the latter one leads to a decrease. As the solvent evaporation from the lower part slows down over time, the growing upper crystalline nuclei may hinder the evaporation of the lower solution, resulting in even slower evaporation of the lower solvent. Therefore, initially, there is a slight decrease in PL intensity, suggesting a faster dissolution process due to rapid evaporation of solution at lower part, followed by the opposite scenario. In the control film, PL intensity changes smoothly, indicating a linear progression of the process, with the crystallization process continuously favoring the growth of the crystalline nuclei. Rapid growth of the upper perovskite crystalline nuclei

forming a shell-like structure can lead to many residual solvents in the lower part of the perovskite film, deteriorating the crystallization process and causing lattice distortion or even void formation<sup>11</sup>. In contrast, in the target film, the change in PL intensity is not as smooth, with fluctuations, similar to previous reports, leading to a significant extension of the second stage by 200 ms compared to the control film. As the competitive mechanism previously mentioned, we can reasonably speculate that with the addition of 2TI, the crystallization process in the second stage is modulated, restricting the growth of the upper wet film crystalline nuclei, widening the window for the evaporation of the lower part.

Notably, the PL peak position remains essentially unchanged during this stage, suggesting that the grain size does not change, and the PL intensity at the end of the second stage is close to that at the end of the first stage, implying the existence of this dynamic equilibrium process.

*Stage III: Crystalline Nuclei Grow and Enlarge.* In this stage, there is a rapid rise in PL peak intensity accompanied by a continual red shift in the peak position, signifying the ongoing enlargement of crystalline nuclei and the free growth of perovskite crystals in three dimensions, forming larger crystal structures<sup>12,13</sup>. The control group reaches its peak PL intensity at 600 ms, whereas the target group peaks at 800 ms. The 200 ms difference is attributed to the delay induced by the addition of 2TI in the second stage. Overall, there are no substantial differences between control and target films in this stage.

*Stage IV: Crystal Aggregation.* In this stage, PL intensity decreases and the peak position continues to redshift for a short duration before stabilizing. PL decrease is attributed to the upper perovskite crystals growing large enough to aggregate and form grain boundaries, a key source of non-radiative recombination in the film<sup>8</sup>. Consequently, the perovskite only continues to grow vertically, leading to the observed continued redshift of the PL peak.

#### Note S2. Details of simulation

We use device simulation techniques to analyze the impact of different  $\Delta E$  on device performance. All our simulations of PSCs are conducted in Solar Cell Capacitance Simulator One dimension (SCAPS-1D), version 3.3.11<sup>14</sup>. AM 1.5 solar spectrum (100 mW/cm<sup>2</sup>) is adopted for sunlight incidence. This software is mainly based on three differential equations (equation 1~equation 3) to give the simulation results.

$$\frac{\partial}{\partial x} \left( \varepsilon_0 \varepsilon \frac{\partial \Psi}{\partial x} \right) = -q \left( p - n + N_D^+ - N_A^- + N_{def} \right) \quad (1)$$
$$-\frac{\partial J_n}{\partial x} - U_n + G = \frac{\partial n}{\partial x} \qquad (2)$$
$$-\frac{\partial J_p}{\partial x} - U_p + G = \frac{\partial p}{\partial x} \qquad (3)$$

Here,  $\varepsilon_0$  is the Vacuum Permittivity,  $\varepsilon$  is the relative permittivity,  $\Psi$  is the electrostatic potential, q is the elementary charge, n(p) is the electron (hole) density,  $N_D^+(N_A^-)$  is the density of ionized donors (acceptors),  $N_{def}$  is the possible defect density,  $J_n(J_p)$  is the electron (hole) current densities,  $U_n(U_p)$  is the net electron (hole) recombination rate, G is the electron-hole generation rate. The electron and hole density are given by:

$$J_{n} = qn\mu_{n}E + qD_{n}\frac{\partial n}{\partial x}$$
(4)  
$$J_{p} = qp\mu_{p}E - qD_{p}\frac{\partial p}{\partial x}$$
(5)

Where  $\mu_n(\mu_p)$  is the electron (hole) mobility, and  $D_n(D_p)$  is the diffusion coefficient of the electron (hole).

Typically, perovskite is considered as intrinsic semiconductor, and its Fermi level ( $E_{\rm f}$ ) can be calculated using the following formula:  $E_f = \frac{E_C + E_V}{2} + \frac{1}{2}kTln\left(\frac{N_V}{N_C}\right) = \frac{E_C + E_V}{2} + \frac{3}{4}kTln\left(\frac{m_p^*}{m_n^*}\right)$ (6) . where  $E_c$  is energy level of conduction band,  $E_v$  is energy level of valence band, k is Boltzmann's constant, T is

temperature,  $N_{\rm c}$  is the effective density of states for the conduction bands,  $N_{\rm v}$  is the

effective density of states for valence bands,  $m_n^*(m_p^*)$  is electron(hole) effective mass. Therefore, we can alter the  $E_f$  of perovskite by adjusting the effective density of states ratio between the valence band and the conduction band in SCAPS-1D. In order to simulate the different Fermi level of the top and bottom surfaces of perovskite film, we set the band structure of the entire perovskite film to have a gradient variation in this simulation (near the HTL side, it is pure A, near the ETL side, it is pure B, with a linear distribution in between). It's worth noting that according to Equation 6, the ratio of  $N_c$ to  $N_v$  needs to follow a logarithmic distribution to fulfill the gradient distribution of the overall band structure.

To simplify the simulation process, we fix the value of  $E_c$ ,  $E_v$  and  $N_v$  at the top and bottom parts of the perovskite to the same value, respectively. Subsequently,  $E_{f, \text{ bottom}}$ can be adjusted by tuning the  $N_c$  value at bottom part of perovskite (pure A). The calculation formula for  $\Delta E$  using the parameters included in SCAPS-1D is as follows:

$$\Delta E = E_{f, \text{ bottom}} - E_{f, \text{ top}}$$

$$= \left(\frac{E_C + E_V}{2} + \frac{1}{2}kTln\left(\frac{N_V}{N_C}\right)_{\text{bottom}} - \left(\frac{E_C + E_V}{2} + \frac{1}{2}kTln\left(\frac{N_V}{N_C}\right)_{\text{top}}\right)_{\text{top}}$$

$$= \frac{1}{2}kTln\left(\frac{N_{v,bottom}}{N_{c,bottom}} * \frac{N_{c,top}}{N_{v,top}}\right) = \frac{3}{4}kTln\left(\frac{m_{p,bottom}}{m_{n,bottom}} / \frac{m_{p,top}}{m_{n,top}}\right)$$

$$= \frac{1}{2}kTln\left(\frac{N_{c,top}}{N_{c,bottom}}\right) \qquad (7)$$

Integrating Equations 6 and 7, it becomes evident that the value of  $\Delta E$  is correlated with the ratio of the effective masses of electrons and holes at the top and bottom surfaces, prompting the discussion in Section 2 of our text.

The relevant simulation parameters are listed in Table S2 and 3.



2TI

Fig. S1 Molecular structure of 2TI.



**Fig. S2** The extracted values of **(a)** emission peak position and **(b)** normalized PL peak intensity of control and target films in a long-time scale. In a long-time scale, the shifts in peak positions are nearly identical for both. However, it can be observed that the peak intensity in the target film decreases more rapidly compared to the control film.



Fig. S3 GIXRD spectrum of the top surface of control (a) and target (b) films, and the bottom surface of control (c) and target (d) films. The dashed lines represent the peak shifts.



Fig. S4 GIXRD spectrum of wide-bandgap perovskite at different tilt angles at the depth of 50 nm for control (a) and target (b) films and 500 nm for control (c) and target (d) films.



Fig. S5 I 3d XPS spectra of (a) the top and (b) bottom surface of control and target films.



**Fig. S6** XPS depth profile analysis of S 2p (**a**) and I 3d (**b**) for perovskite films with the addition of 2TI.



Fig. S7 The 1H NMR spectra of control and target perovskite films dispersed in  $D_2O$ . We observe reduction in peak densities of both the DMF and DMSO signals<sup>15,16</sup>, indicating less solvent residue upon 2TI treatment.



Fig. S8 Band structures under different strain conditions based on first-principles density functional theory. We present three typical cases herein: strain = (a) 0%, (b) - 1.5%, (c) -3%.



Fig. S9 (a) Simulated J-V curves of devices with different representative  $\Delta E$ . (b) Dependence of the  $V_{oc}$ ,  $J_{sc}$  and FF on  $\Delta E$ .

#### Note S3. Further discussion about the impact of $\Delta E$ on charge transport.

In Fig. S10, we schematically depict the ideal scenario for charge transport within a perovskite photo absorber, where the holes and electrons preferentially diffuse towards HTL and ETL, respectively. When  $\Delta E < 0$  eV, after photoexcitation, the electrons tend to diffuse towards the side with low Fermi energy to occupy empty energy states, as illustrated in the schematic diagram shown in Fig. S10b. This results in electron accumulation near the HTL interface, aligning the electron distribution trends with those of hole, leading to a reduction in the hole-to-electron ratio as it approaches the ETL (Fig. S10b). This scenario would be unfavorable for the device operation in inverted cell architecture. When  $\Delta E = 0$  eV, holes and electrons are more uniformly distributed within the perovskite film due to the absence of energy landscape differences (Fig. S10c). When  $\Delta E > 0$  eV, electrons tend to flow towards the ETL side driven by the low Fermi energy near this region, resulting in preferential electron accumulation at the ETL side (Fig. S10d). Meanwhile, the holes tend to diffuse towards the side with high Fermi level (*i.e.* the HTL side), resulting in a high hole-to-electron ratio at this interface. This scenario would be highly favourable for device operational devices. Furthermore, we also investigated the band structure under  $\Delta E$ . It is evident that with the improvement (i.e., increase) of  $\Delta E$  within the perovskite thin films, the quasi-Fermi level correspondingly rises.



Fig. S10 Simulation analysis of different energy band landscapes on device performance. (a) Schematic diagrams illustrating charge transport within perovskite photo absorber in an inverted cell. The ETL and HTL stand for electron-transporting layer and hole-transporting layer, respectively. The  $E_{f, \text{ top}}$  and  $E_{f, \text{ bottom}}$  represent the Fermi level of the top (close to ETL) and bottom (close to HTL) region of the perovskite film. The  $\Delta E$  is defined as the  $E_{f, \text{ bottom}}$  subtracted by the  $E_{f, \text{ top}}$ . The blue and red circles depicit photogenerated holes (h<sup>+</sup>) and electrons (e<sup>-</sup>), respectively. (b) – (d) Charge transport and distribution of hole and electron concentrations and the ratio of hole to electron of the three different  $\Delta E$  scenarios: (b) > 0, (c) = 0, (d) < 0.



Fig. S11 The simulated band diagram of three different  $\Delta E$  scenarios: (a) > 0, (b) = 0, (c) < 0.



Fig. S12 Tauc plots of absorption spectra of control and target films. The extracted  $E_g$  values are 1.54 eV.



**Fig. S13** (a) EQE spectrum and integrated  $J_{sc}$  of control and target device. The measured  $J_{sc}$  values from *J-V* analysis is consistent with the integrated  $J_{sc}$  values from the EQE analysis with only exhibit less than 1.9% mismatch. (b) Inflection points of EQE. The inflection points of the control and target perovskite are both at 805nm, and the obtained bandgap value is 1.54 eV, which is consistent with the results obtained by Tauc plot.



Fig. S14 Histograms showing the device (a)  $V_{oc}$ , (b) FF and (c)  $J_{sc}$  of 30 cells per type, fitted with Gaussian distributions (solid lines). The values in parentheses represent the average values. (d) Stable power output (SPO) of the best-performing control and target devices.



福建省计量科学研究院 FUJIAN METROLOGY INSTITUTE (国家光伏产业计量测试中心) National PV Industry Measurement and Testing Center

报告编号: 24Q3-00500

检测结果/说明:

1. 标准测试条件 Standard Te	st Condition (STC):
	总辐照度 Total Irradiance: 1000 W/m <sup>2</sup> 被测电池温度 Temperature: 25.0 °C
	光谱分布 Spectral Distribution: AM1.5G

2.STC条件下测量数据 Measurement Data and I-V/P-V Curves under STC 正扫 Forward Scan

	$I_{sc}$ (mA)	V <sub>oc</sub> (V)	I <sub>MPP</sub> (mA)	V <sub>MPP</sub> (V)	P <sub>MPP</sub> (mW)	FF(%)	$A (\mathrm{cm}^2)$
	1.786	1.169	1.688	0.9878	1.667	79.84	0.0698
反扫 Reverse \$	Scan						

I <sub>sc</sub> (mA)	$V_{oe}(V)$	I <sub>MPP</sub> (mA)	V <sub>MPP</sub> (V)	P <sub>MPP</sub> (mW)	FF(%)	$A (\mathrm{cm}^2)$
1.790	1.172	1.732	1.014	1.756	83.70	0.0698

失配因子 Mismatch factor: 1.0003



检测报告续页专用

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**Fig. S15** Certification report of the 0.0698-cm<sup>2</sup> cells tested by the Chinese National PV Industry Measurement and Testing Center (NPVM), showing the I-V and P-V characteristic curves measured under standard test condition (STC).



**Fig. S16 (a)** *J-V* curves of the best-performing control (blue) and target (red) devices of wide-bandgap perovskite solar cell. The devices are scanned in both reverse (solid lines) and forward (dashed lines) directions. **(b)** SPO of control and target devices.



**Fig. S17** Histograms showing the device (a)  $V_{oc}$ , (b)  $J_{sc}$ , (c)FF, (d) $V_{oc}$ ×FF and (e)PCE of 24 cells per type, fitted with Gaussian distributions (solid lines). (f) Tauc plots of absorption spectra of wide-bandgap perovskite films. The extracted  $E_g$  values are 1.68 eV.



Fig. S18 GIXRD spectrum of wide-bandgap perovskite at different tilt angles at the depth of 50 nm for control (a) and target (b) films and 500 nm for control (c) and target (d) films. Bottom: strain distribution for the films (measured (points) and Gauss fitted (line) diffraction strain data as a function of  $\sin^2 \varphi$ ).

### Note S4. Experimental procedures of vertical charge transport

Inspired by previous work, we utilize transient PL spectroscopy with emission wavelength resolution to investigate charge transport behavior in the vertical direction. The transient PL spectroscopy is recorded with a time-correlated single photon counter (PicoQuant GmbH, TimeHarp 260) equipped with a monochromator (Omni  $\lambda$ 3008i, the wavelength resolution is 0.1 nm), excited with a picosecond diode laser (640 nm, LDH-P-C-640B) at a repetition rate of 1 MHz. The laser excitation fluence is 3.2 nJ cm<sup>-2</sup>. We use the monochromator to perform single time-resolved photoluminescence (TRPL) tests on emission signals every 1 nm within the 765-845 nm range, with each test having an integration time of 100 seconds. The PL1 signal corresponds to the aggregated emission from 805 to 845 nm, whereas the PL2 signal corresponds to the aggregated emission from 765 to 805 nm.



Fig. S19 Transient decay curve of the control and target films, collected at different wavelengths (PL1 > 805 nm and PL2 < 805 nm).



Fig. S20 The (a)  $V_{\text{MPP}}$  and (b)  $J_{\text{MPP}}$  of control and target devices in Fig. 5a.



Fig. S21 GIXRD spectrum at different tilt angles at the depth of 50 nm for control (a) and target (b) films and 500 nm for control (c) and target (d) films after aging in ambient conditions. Bottom: strain distribution for the films (measured (points) and Gauss fitted (line) diffraction strain data as a function of  $\sin^2 \varphi$ ).



**Fig. S22 (a)** Long-term stability of the devices aged at 85 °C and 50% relative humidity under 1-sun equivalent illumination and the corresponding  $V_{\text{MPP}}$  (b) and  $J_{\text{MPP}}$  (c) evolution. (d-f) Corresponding normalized data for clearer observation of the performance evolution of the control and target device.



**Fig. S23** Long-term stability of the **(a)** target and **(b)** control devices aged at 85 °C in the nitrogen glove box without illumination during 300 h.

	50	nm	500 nm		
Ψ()	ω (°)	$arPhi\left(^{\circ} ight)$	ω (°)	$arPhi\left(^{ m o} ight)$	
5	0.3114	-17.5492	3.4807	-21.7563	
15	0.3213	-43.0644	3.6065	-50.1414	
25	0.3427	-56.8029	3.8835	-63.4648	
35	0.3796	-64.3174	4.3772	-70.5945	

**Table S1.** Instrumental angles  $(\Psi, \omega, \Phi)$  set for (012) crystal plane in different depths of GIXRD residual stress measurements.

Parameter HTL Perovskite ETL Thickness(nm) 20 680 25 Bandgap(eV) 3.6 1.54 2.2 Composition А В 2 **Electron affinity (eV)** 3.88 3.88 3.87 **Dielectric permittivity** 11.75 30 30 4.2 (relative) **CB** effective density of  $2.5 \times 10^{20}$ Various  $1 \times 10^{18}$ 8×10<sup>19</sup> states (cm<sup>-3</sup>) VB effective density of 2.5×10<sup>20</sup> 8×10<sup>19</sup> 2.2×10<sup>19</sup>  $2.2 \times 10^{19}$ states (cm<sup>-3</sup>) **Electron/hole thermal**  $1 \times 10^{7}$  $1 \times 10^{7}$  $1 \times 10^{7}$  $1 \times 10^{7}$ velocity (cm/s) **Electron mobility** 1×10-3 10 10 20 (cm<sup>2</sup>/Vs) Hole mobility (cm<sup>2</sup>/Vs) 1×10-3 10 10 1×10<sup>-2</sup> Shallow uniform donor 0  $1 \times 10^{18}$ 0 0 density N<sub>D</sub> (cm<sup>-3</sup>) Shallow uniform 1×10<sup>19</sup> acceptor density N<sub>A</sub> 0 0 0 (cm<sup>-3</sup>)

 Table S2. Relevant parameters used for simulation in SCAPS-1D. The relevant

 parameters are derived from reference literature<sup>17–21</sup>.

Defect parameter	HTL	HTL /Perovskite interface	Perovskite	ETL /Perovskite interface	ETL	
Defect type	Neutral	Neutral	Neutral	Neutral	Neutral	
Capture cross section (electrons) (cm <sup>-3</sup> )	1×10 <sup>-15</sup>	1×10 <sup>-19</sup>	1×10 <sup>-15</sup>	1×10 <sup>-19</sup>	1×10 <sup>-15</sup>	
Capture cross section (hole) (cm <sup>-3</sup> )	1×10 <sup>-15</sup>	1×10 <sup>-19</sup>	1×10 <sup>-15</sup>	1×10 <sup>-19</sup>	1×10 <sup>-15</sup>	
Energetic distribution	Single	Single	Single	Single	Single	
Energy level (eV)	0.6	0.6	0.6	0.6	0.6	
Total defect density (cm <sup>-2</sup> )	1×10 <sup>14</sup>	1×10 <sup>12</sup>	1×10 <sup>14</sup>	1×10 <sup>13</sup>	1×10 <sup>14</sup>	

 Table S3. Defect parameters used for simulation in SCAPS-1D.

Device		V <sub>oc</sub> (V)	$J_{\rm sc}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)	
Control	Forward	1.125	24.81	81.40	22.72	
	Reverse	1.147	25.23	82.78	23.96	
Target	Forward	1.177	25.59	84.29	25.39	
	Reverse	1.184	25.73	85.21	25.96	

Table S4. Performance parameters of the best device in in Fig. 4a in the main text.

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Eg	V <sub>oc</sub>	$J_{ m sc}$	FF	PCE	V <sub>oc</sub> *FF	R	Ref.
(eV)	(V)	(mA/cm <sup>2</sup> )	(%)	(%)	(V)	(%)	
1.535	1.161	26.30	80.00	24.40	0.929	81.53	22
1.550	1.195	25.77	80.19	24.69	0.958	83.00	23
1.528	1.174	25.90	83.26	25.32	0.977	86.21	24
1.550	1.190	24.70	83.60	24.56	0.995	86.20	25
1.530	1.150	26.19	83.90	25.27	0.973	85.72	26
1.560	1.187	25.69	84.73	25.86	1.006	86.48	27
1.560	1.201	24.80	84.50	25.16	1.015	87.26	28
1.560	1.160	24.40	82.50	23.30	0.957	82.27	29
1.550	1.184	25.68	82.32	25.00	0.975	84.47	30
1.550	1.159	25.02	83.05	24.09	0.963	83.43	31
1.550	1.177	24.80	84.30	24.60	0.992	85.94	32
1.550	1.190	24.55	84.78	24.80	1.009	87.41	13
1.550	1.208	25.08	84.37	25.60	1.019	88.28	33
1.550	1.200	24.70	82.00	24.30	0.984	85.25	34
1.540	1.160	25.73	82.50	24.50	0.957	83.67	35
1.540	1.170	25.50	82.50	24.60	0.965	84.37	36
1.530	1.160	26.30	83.00	25.40	0.963	84.84	37
1.520	1.164	26.14	85.74	26.09	0.998	88.59	38
1.520	1.176	25.88	82.50	25.10	0.970	86.11	39
1.534	1.165	26.22	82.20	25.12	0.958	84.14	40
1.540	1.170	25.72	83.67	25.17	0.979	85.60	41
1.530	1.190	25.40	84.60	25.60	1.007	88.72	42
1.520	1.193	26.66	81.31	25.90	0.988	87.71	43
1.540	1.191	25.49	85.14	25.85	1.014	88.69	This work

**Table S5.** Summary of reported state-of-the-art perovskite solar cell. R represents the  $V_{oc}$ ×FF relative to the S-Q limits of its corresponding bandgap.

Device		V <sub>oc</sub> (V)	$J_{\rm sc}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
Control	Forward	1.180	21.20	81.73	20.45
	Reverse	1.183	21.56	81.72	20.84
Target	Forward	1.184	21.60	84.27	21.54
	Reverse	1.196	21.60	84.66	21.87

**Table S6.** Performance parameters of the best device in in Fig. S16a.

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