Supporting Information

## Achieving Unprecedented Power-Output in 4-Terminal Mirror-Symmetrical

## Printable Carbon CsPbBr<sub>3</sub> Solar Cells through Dual-Solvent Engineering

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Figure S1. a) CsBr/H<sub>2</sub>O solutions containing H<sub>2</sub>O, CH<sub>3</sub>CN, CH<sub>3</sub>OH, and DMF. b) Tyndall effect of CsBr solution.



Figure S2. Raman spectrum of the interaction between Cs<sup>+</sup> and DMF obtained by DFT calculations.



Figure S3. Contact angle test a) Control, b) Target.



Figure S4. PbBr<sub>2</sub> film a) Surface, b) Cross-section.



Figure S5. FWHM and intensity of the (200) diffraction peak of CsPbBr<sub>3</sub> film.



Figure S6. XPS spectra of Control and Target perovskite films: a) Full spectrum, b) Pb 4f, c) Br 3d, d) Br/Pb ratio.



Figure S7. a) Time-resolved photoluminescence (TRPL) curve of Control and b) Target perovskite films.



Figure S8. a) Temperature-dependent conductivity of Control and b) Target perovskite films.



Figure S9. The relationship between the short-circuit current density  $(J_{SC})$  of perovskite devices and light intensity.



Figure S10. Transient photovoltage curve of perovskite devices.



Figure S11. Equivalent circuit diagram of the EIS curve.



Figure S12. R<sub>S</sub> values of Control and Target under different biases.



**Figure S13.** Statistical distribution of a)  $V_{OC}$ , b)  $J_{SC}$ , c) FF, and d) PCE of 15 cell devices each for Control and Target groups.



Figure S14. J-V curve of large-area devices.



Figure S15. Stability of p-MPSC devices at 25°C.

Solvent	Dielectric constant	Donor number	
H <sub>2</sub> O	80.1	18.0	
CH <sub>3</sub> CN	37.5	14.1	
CH <sub>3</sub> OH	32.7	19.9	
DMF	36.7	26.6	

Table S1. Values of DN and  $\varepsilon$  for different solvents.

Table S2. Detailed parameters for fitting the TRPL curve of the perovskite film.

	$A_1$	$\tau_1$ (ns)	$A_2$	$ au_2$	$ au_{ m ave}$	R <sup>2</sup>
ZrO <sub>2</sub> -Target	0.2471	23.76	0.6325	144.86	137.57	0.996
TiO <sub>2</sub> -Target	0.576	3.24	0.7936	19.38	17.63	0.997
ZrO <sub>2</sub> -Control	0.4528	21.07	0.4527	68.45	57.30	0.997
TiO <sub>2</sub> -Control	0.4791	9.53	0.5595	39.30	34.18	0.997

	Applied voltage (V)	$R_{\mathrm{S}}\left(\Omega\right)$	$R_{\rm ct}(\Omega)$	high f C (F)	$R_{ m rec}\left(\Omega\right)$	low f C (F)
Target	1.35	58.96	23.40	9.14×10 <sup>-9</sup>	1097.00	2.85×10 <sup>-8</sup>
	1.40	58.84	23.00	1.11×10 <sup>-8</sup>	357.50	4.88×10 <sup>-8</sup>
	1.45	57.91	21.64	1.31×10 <sup>-8</sup>	131.30	1.04×10 <sup>-7</sup>
	1.50	58.99	16.50	1.80×10 <sup>-8</sup>	59.50	2.13×10 <sup>-7</sup>
Control	1.35	66.51	51.28	3.87×10 <sup>-9</sup>	740.80	3.09×10 <sup>-8</sup>
	1.40	59.98	55.40	2.83×10 <sup>-9</sup>	251.50	6.43×10 <sup>-8</sup>
	1.45	58.62	49.89	2.61×10-9	103.60	1.43×10 <sup>-7</sup>
	1.50	61.4	41.02	3.20×10-9	53.14	2.77×10 <sup>-7</sup>

**Table S3.** Fitting parameters of the device's EIS.

**Table S4.** Photovoltaic performance parameters of Control and Target devices.

1		1				
	$V_{\rm OC}$ (V)	$J_{\rm SC}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)	HI	
Control Reverse Scan	1.50	7.41	74.46	8.26	2.30%	
Control Forward Scan	1.49	7.43	73.10	8.07		
Target Reverse Scan	1.58	7.69	83.70	10.18	1.87%	
Target Forward Scan	1.58	7.65	82.74	9.99		

CsBr solution	$V_{\rm OC}$ (V)	$J_{\rm SC}$ (mA cm <sup>-2</sup> )	FF (%)	PCE (%)
H <sub>2</sub> O	1.50	7.41	74.46	8.26
CH <sub>3</sub> CN	1.35	4.59	63.73	3.94
CH <sub>3</sub> CN/H <sub>2</sub> O 0.5/9.5 (V/V)	1.50	7.14	81.96	8.77
CH <sub>3</sub> CN/H <sub>2</sub> O 1/9 (V/V)	1.50	7.23	81.51	8.86
CH <sub>3</sub> CN/H <sub>2</sub> O 2/8 (V/V)	1.51	7.08	80.32	8.62
CH <sub>3</sub> OH	1.29	4.78	50.59	3.12
CH <sub>3</sub> OH/H <sub>2</sub> O 0.5/9.5 (V/V)	1.52	7.62	80.80	9.33
CH <sub>3</sub> OH/H <sub>2</sub> O 1/9 (V/V)	1.54	7.72	80.32	9.53
CH <sub>3</sub> OH/H <sub>2</sub> O 2/8 (V/V)	1.50	7.51	79.39	8.96
DMF	1.30	5.32	49.24	3.40
DMF/H <sub>2</sub> O 0.5/9.5 (V/V)	1.59	7.70	80.55	9.83
DMF/H <sub>2</sub> O 1/9 (V/V)	1.58	7.69	83.70	10.18
DMF/H <sub>2</sub> O 2/8 (V/V)	1.53	7.39	80.85	9.13

Table S5. Photovoltaic performance in different CsBr solvents.

Using the double exponential decay equation (Equation 1) to fit the TRPL curve in Figure S7, the carrier lifetimes  $\tau_1$  and  $\tau_2$  of the non-radiative recombination and radiative recombination processes of perovskite are obtained. Then the average carrier lifetime ( $\tau_{ave}$ ) is calculated with the help of Equation (2).<sup>[S1]</sup>

$$Y = A_1 \exp\left(-\frac{t}{\tau_1}\right) + A_2 \exp\left(-\frac{t}{\tau_2}\right) + y_0$$

$$A_1 \tau_1^2 + A_2 \tau_2^2$$
(1)

$$\tau_{ave} = \frac{A_1 \tau_1 + A_2 \tau_2}{A_1 \tau_1 + A_2 \tau_2} \tag{2}$$

Where  $A_1$  and  $A_2$  represent the relative amplitudes of  $\tau_1$  and  $\tau_2$ , and  $y_0$  is the offset constant of the baseline.

SCLC testing trap density  $(N_t)$  calculation Equation (3):

$$N_t = \frac{2\varepsilon\varepsilon_0 V_{TFL}}{qL^2} \tag{3}$$

In this equation, q is the elementary charge, L is the thickness of the perovskite film,  $\varepsilon$  is the relative dielectric constant (according to the literature, the value of  $\varepsilon$  for CsPbBr<sub>3</sub> is 22 <sup>[S2]</sup>), and  $\varepsilon_0$  is the vacuum permittivity.

tDOS testing curve fitting Equation (4).<sup>[S3]</sup>

$$N_t(E_{\omega}) = -\frac{V_{bi} dC \ \omega}{qW d\omega k_B T} \tag{4}$$

Here,  $\omega$  is the angular frequency,  $K_{\rm B}$  is the Boltzmann constant, T is the temperature,  $E_{\omega}$  is the level depth, C is the capacitance, q is the elementary charge,  $V_{\rm bi}$ , and W are the built-in potential and semiconductor depletion layer width obtained from the Mott-Schottky test, and  $E_{\omega}$  can be obtained through Equation (5).

$$E_{\omega} = K_B T \times In \frac{2\beta_{\rho} N_V}{\omega}$$
<sup>(5)</sup>

Here,  $\beta_{\rho}$  is the hole capture coefficient, and  $N_{V}$  is the effective density of states at the valence band maximum.

The Gaussian16 software was used to perform density functional theory (DFT) calculations using the B3LYP method for  $H_2O$ , acetonitrile (CH<sub>3</sub>CN), methanol, and N, N-dimethylformamide (DMF), where the 6-311+G(D) basis set was selected for C, H, O, and N calculations. The pseudopotential basis set LanL2DZ was used for cesium calculations, and the results were visualized using GAUSS VIEW.

## REFERENCES

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