Constructing a novel super-crosslinked triazine COFs through

molecular expansion for enhanced photocatalytic performance under

visible light

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	ç		Element		We	ight?	%		Wt	% Sig	gma			Ato	omi	ic%	2	
-			Atomic		9	9.12				0.19				9	9.2	4		
- 60			Nitrogen		C	0.88				0.19				(0.76	5		
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Fig.S1 EDX mapping of HCTF-2.



Fig.S2 The degradation curves of RhB by CTF-2 and HCTF-2;





Fig.S4 (a-d) the effects of different dosages of trapping agents on the photodegradation of LEF by HCTF-2



Fig.S5 Variations in the relative intensity of intermediate products of LEF with different reaction time, as obtained in the LC-MS spectra.

Tubles Teomparison of field 2 for photocalarytic Te acguation.										
Catalyst / mg	V (mL) / C ₋₃₀ (mg·L ⁻¹)	Light source	Time (min)	Result (%)	TOF	Ref				
HCTF-2/10	50/20	Visible light	21	99.0	4.714	This work				
TpMa/5	20/20	Visible light	70	19.7	0.225	[1]				
COF-Ph/10	50/20	Visible light	90	29.0	0.322	[2]				
TpPa-1/10	20/80	Sun light	40	70.0	2.800	[3]				

4-1-4: TO 1-CI ſ

 $TOF = \frac{C_{-30} \times V_{TC} \times Degradation \, rate}{m_{photocatalyst} \times t}$

Table.52 Comparison of I	IC 11-2 101 pilo	tocatalytic Π_2	evolution.		
Photocatalysts/mg	Light source	H ₂ generation rate µmol·g ⁻¹ h ⁻¹	Ref		
HCTF-2/30	Visible light	2728.61	This work		
TpPa-1/40	Visible light	137.00	[4]		
TpBD/25	Visible light	141.00	[5]		
TpPa-2/10	Visible light	2200.00	[6]		
TFPT-COF/10	Ultraviolet light	230.00	[7]		

Table.S2 Comparison of HCTF-2 for photocatalytic H₂ evolution.

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