

## Supplementary Materials

### Emerging applications of deep eutectic solvents in the preparation and functionalization of biomass-derived carbonaceous materials: Challenges and prospects

Yiyi Shen,<sup>a, b, #</sup> Haiqin Zhou,<sup>a, b, #</sup> Xiaotong He,<sup>a, b</sup> Feng Shen,<sup>c</sup> Zhixiang Xu,<sup>d</sup> Bo Yang,<sup>c</sup> Lingzhao Kong,<sup>f</sup> Lichun Dai<sup>a, b, \*</sup>

<sup>a</sup> Key Laboratory of Development and Application of Rural Renewable Energy, Biogas Institute of Ministry of Agriculture and Rural Affairs, Chengdu 610041, Sichuan, China.

<sup>b</sup> Research Center for Rural Energy and Ecology, Chinese Academy of Agricultural Sciences, Chengdu 610041, Sichuan, China.

<sup>c</sup> Agro-Environmental Protection Institute, Chinese Academy of Agricultural Sciences, No. 31, Fukang Road, Nankai District, Tianjin 300191, China.

<sup>d</sup> School of Energy and Power Engineering, Jiangsu University, Zhenjiang 212013, China.

<sup>e</sup> Zhai Mingguo Academician Work Station, Sanya University, Sanya 572022, China.

<sup>f</sup> School of Environmental Science and Engineering, Suzhou University of Science and Technology, Suzhou 215009, China.

# These two authors contributed equally to this work.

\* Corresponding author. Lichun Dai: Biogas Institute of Ministry of Agriculture and Rural Affairs, Section 4-13, Renmin South Road, Chengdu 610041, China. E-mail: dailichun@caas.cn (Lichun Dai)

Table S1 Selected examples for the applications of DESs in the preparation and functionalization of BCMs.

Roles of DESs	Preparation/functionalization routes	Biomass source	As-obtained BCMs	Applications of BCMs	Performances	Ref.
STC media	STC of ChCl-EG-glucose DES with Al(NO <sub>3</sub> ) <sub>3</sub>	Glucose	AlOOH/Al(OH) <sub>3</sub> functionalized solvothermal carbon (Al-STC)	Cr(VI) removal	99.9% removal by filtration	<sup>1</sup>
				Malachite green removal	99.9% removal by filtration	
				Methylene blue removal	99.9% removal by filtration	
				Ciprofloxacin removal	99.9% removal by filtration	
				Oxytocin removal	88.8% removal by filtration	
				CTAB removal	99.9% removal by filtration	
	STC of biomass in ChCl-CA DES	Corn straw	Carboxyl and phenol groups co-enriched solvothermal carbon	U(VI) adsorption	$Q_m = 353 \text{ mg/g, } 25^\circ\text{C, 1 g/L}$	<sup>2</sup>
Co-pyrolysis	STC of biomass in DES	Wood	Solvothermal carbon	Cr(VI) adsorption	$Q_m = 270 \text{ mg/g}$	<sup>3</sup>
	STC of ChCl-EG-glucose DES with FeSO <sub>4</sub>	Glucose	Fe-doped oxygenated solvothermal carbon	Malachite green adsorption	$Q_m = 690 \text{ mg/g}$	<sup>4</sup>
	STC of biomass in ChCl-H <sub>2</sub> O-FeCl <sub>3</sub> DES	Garden waste	Solvothermal carbon	Methylene blue adsorption	$Q_m = 148 \text{ mg/g}$	<sup>5</sup>
	Solvothermal carbon synthesized in DES was further modified by activation and doping of Mn <sub>x</sub> O <sub>y</sub> in DES- and KMnO <sub>4</sub> -containing solution	Parthenium hysterophorus leaf	Mn <sub>x</sub> O <sub>y</sub> -doped porous carbon	Methylene blue adsorption	99.68% removal by filtration	<sup>6</sup>
	Pyrolysis of biomass with ChCl-urea DES and H <sub>3</sub> PO <sub>4</sub>	Palm shell	Nitrogen-doped porous carbon	Pb adsorption	97 mg/g	<sup>7</sup>
	Pyrolysis of biomass with FeCl <sub>3</sub> -urea DES	Peanut shell	Nitrogen-doped magnetic biocarbon	Cr(VI) adsorption	$Q_m = 133 \text{ mg/g, } 25^\circ\text{C}$	<sup>8</sup>
	Successive microwave heating and pyrolysis of glucose-urea DES	Glucose	Nitrogen-doped porous carbon	Methylene blue adsorption 4-nitrophenol adsorption	$Q_m = 440 \text{ mg/g}$ $Q_m = 487 \text{ mg/g}$	<sup>9</sup>
Co-pyrolysis	Pyrolysis of biomass with ChCl-glycolic acid DES	Pumpkin seed shell	Porous carbon	Doxycycline hydrochloride adsorption	$Q_m = 30.35 \text{ mg/g}$	<sup>10</sup>

	Pyrolysis of the Immersion of biochar with ChCl-urea	<i>Boehmeria nivea</i> (L.) Gaud.	Fibrils separation and nitrogen-doped biochar	Tetracycline adsorption Escherichia coli inactivation	$k_{\text{obs}} = 0.0221 \text{ min}^{-1}$ , PDS (1 mM), 25 °C, TC (20 mg/L) inactivation efficiencies 5.19-log decline	<sup>11</sup>
Post-modification	Impregnation magnetic biocarbon with ChCl-glycerol DES	Pamelo peels	DES-modified magnetic biocarbon	Methcathinone adsorption	$Q_m = 1.4 \text{ mg/g}$	<sup>12</sup>
	Impregnation active carbon with ChOH-urea/glycerol	Palm shell	DES-modified active carbon	$\text{CO}_2$ adsorption	$Q_m = 37.2 \text{ mg/g}$ , 25 °C, 200 mL/min, $\text{CO}_2$ (10%)	<sup>13</sup>
	Impregnation of kaolin-biocarbon composite with hexadecyltrimethylammonium Bromide (CTAB)-glycerol DES	Vitex doniana residue	DES-modified kaolin-biocarbon composite	Acid blue adsorption Ciprofloxacin adsorption Acetaminophen adsorption	$Q_m$ was increased from 35 to 499 mg/g $Q_m$ was increased from 21 to 158 mg/g $Q_m$ was increased from 4 to 133 mg/g	<sup>14</sup>
	Loading of 1-ethyl-3-methylimidazole chloride ([emim]Cl) - imidazole DES into nanofibrous carbon microspheres by physical dispersion	Chitin	DES-loaded Nanofibrous carbon microspheres	$\text{SO}_2$ adsorption	14.66 mol/kg, 25 °C, 1 bar	<sup>15</sup>

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