

Supporting Information for

Green *in-situ* synthesis of ZIF-8 membranes on the inner-surface of PESf hollow fibers and application in hydrogen separation

Yifan Yang,^{‡a} Tengfei Yang,^{‡a} Lu Liu^a, Hanhan Chen^b, Wenxiu Zhang^a, Shaomin Liu^{b,c} and Xiaobin Wang^{*a}

^aSchool of Chemistry and Chemical Engineering, Shandong University of Technology, Zibo, 255000, Shandong, P.R. China.

^bState Key Laboratory of Organic-Inorganic Composites, College of Chemical Engineering, Beijing University of Chemical Technology, Beijing, 100029, P.R. China.

^cDongguan Key Laboratory of Intelligent Equipment and Smart Industry, School of Engineering, Great Bay University, Dongguan 523000, P.R. China

E-mail: wangxiaobin@sdut.edu.cn

Captions of Figures

Fig. S1 The overall synthesis scheme of ZIF-8 membrane.

Fig. S2 Schematic diagram of HF membrane prepared by phase-inversion-spinning technology.

Fig. S3 Schematic drawing of preparation process for ZIF-8 membrane.

Fig. S4 Schematic diagram of gas permeation device.

Fig. S5 Gas permeances for PESf hollow fibers.

Fig. S6 SEM images of ZIF-8 membranes prepared under static condition at 70 °C and 8 h (a, a1: inner surface; b: cross-section; c: overall-view).

Fig. S7 Photographs of synthesis solutions containing zinc nitrate and Hmim at different reaction time (a), SEM images of ZIF-8 membranes prepared at different growth time (b, b1: 2 h; c, c1: 4 h).

Fig. S8 Photographs of the synthetic solution at the beginning (a) and end (b) of the flow process.

Fig. S9 Gas permeation data of ZIF-8 membranes supported on HFf (a) and ZIF-8 membrane supported on HF4 (b).

Fig. S10 Gas permeation data of ZIF-8 membranes prepared at different flow rates.

Fig. S11 Gas permeation data of ZIF-8 membranes prepared at different Hmim concentrations.

Fig. S12 The thickness and average growth rate of ZIF-8 membranes prepared at different synthesis temperature.

Fig. S13 Gas permeation data of ZIF-8 membranes prepared at different synthesis temperature.

Fig. S14 SEM images of ZIF-8 membranes prepared under different synthesis time at 70 °C (a, a1: 1 min; b, b1: 3 min; c, c1: 5 min; d, d1: 10 min; e, e1: 15 min; f, f1: 30 min; g, g1: 40 min; h, h1: 60 min; a, b, c, d, e, f, g, h: surface morphology; a1, b1, c1, d1, e1, f1, g1, h1: cross-section view).

Fig. S15 The thickness and average growth rate of ZIF-8 membranes prepared at different synthesis time.

Fig. S16 SEM images of ZIF-8 membrane after treatment under different ultrasonication time (a, a1: 20 min; b, b1: 40 min; a, b: surface morphology; a1, b1: cross-section view).

Fig. S17 SEM images (a: surface morphology; a1: cross-section view) and XRD patterns (b) of ZIF-8 membrane after testing.

Fig. S18 Gas permeances as a function of pressure difference.

Captions of Tables

Table S1 ZIF-8 membranes prepared by different synthesis methods.

Table S2 Hazard codes and boiling points of solvents according to the “ECHA C&L Inventory database”.

Table S3 Residence time corresponding to different flow rates.

Table S4 Detailed comparison of H_2/CH_4 and H_2/N_2 separation performance of membranes listed in Fig. 9.

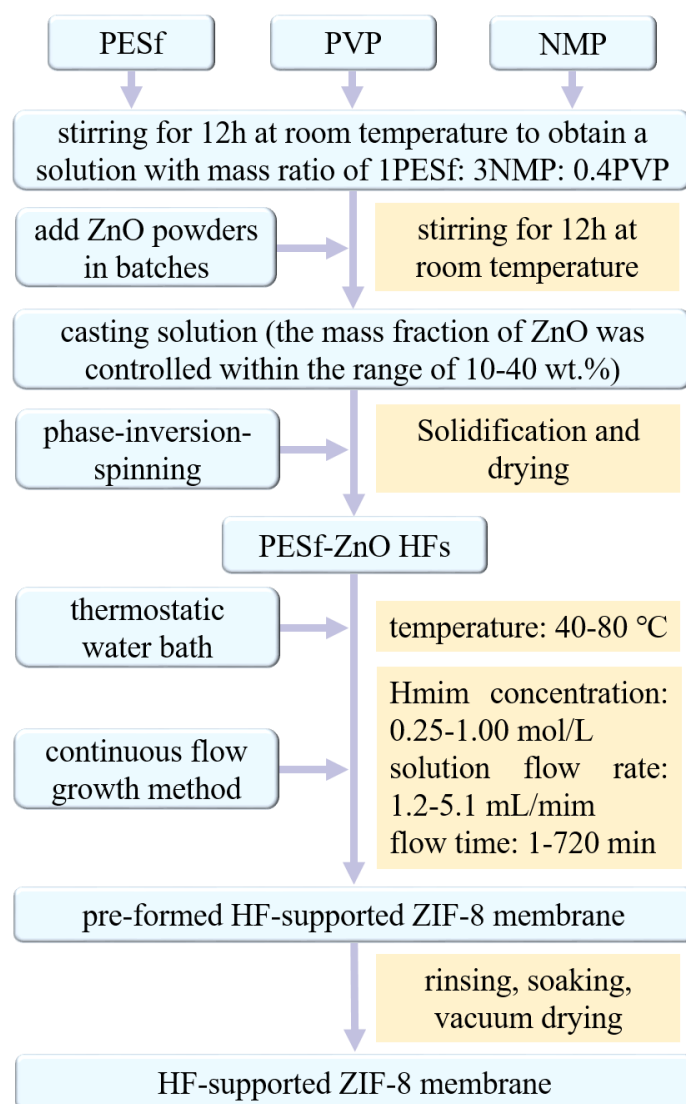


Fig. S1 The overall synthesis scheme of ZIF-8 membrane.

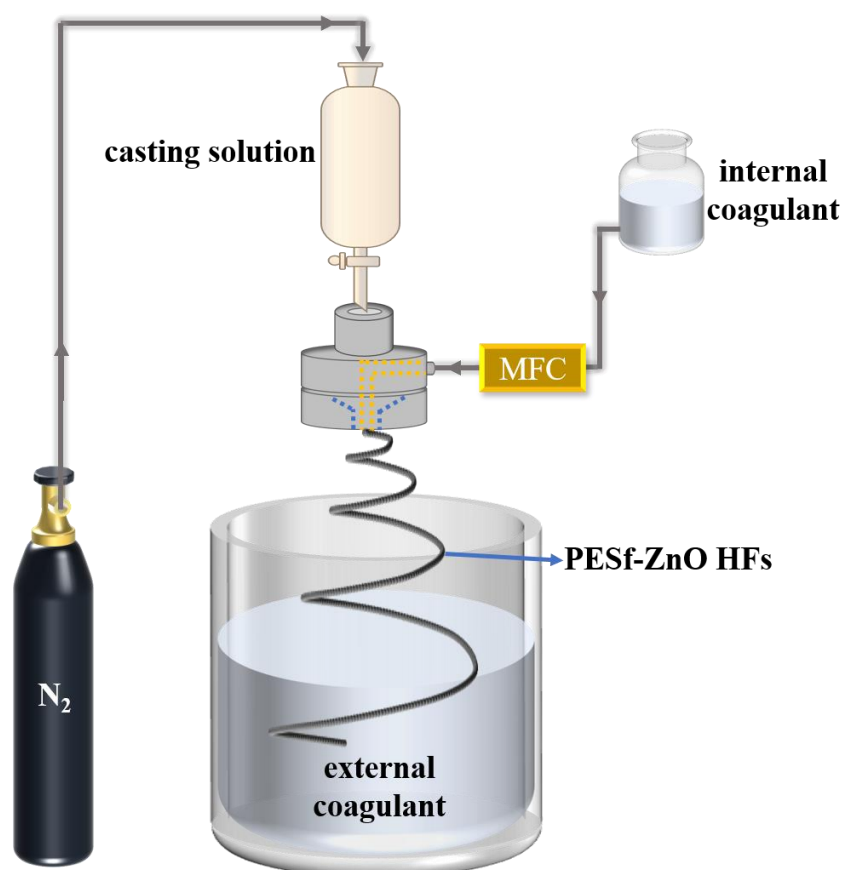


Fig. S2 Schematic diagram of HF membrane prepared by phase-inversion-spinning technology.

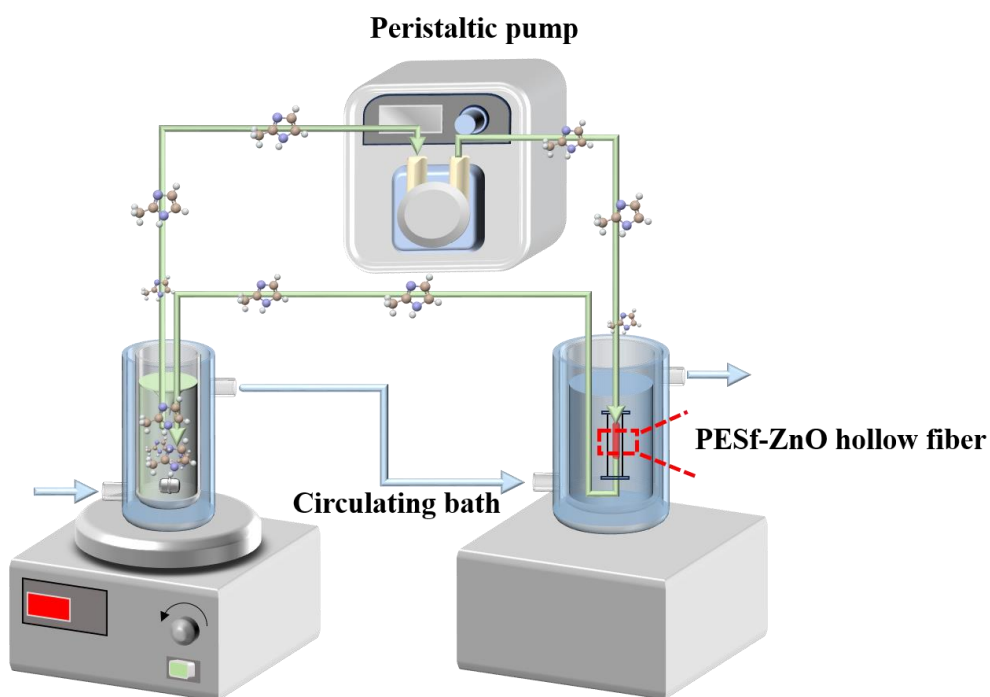


Fig. S3 Schematic drawing of preparation process for ZIF-8 membrane.

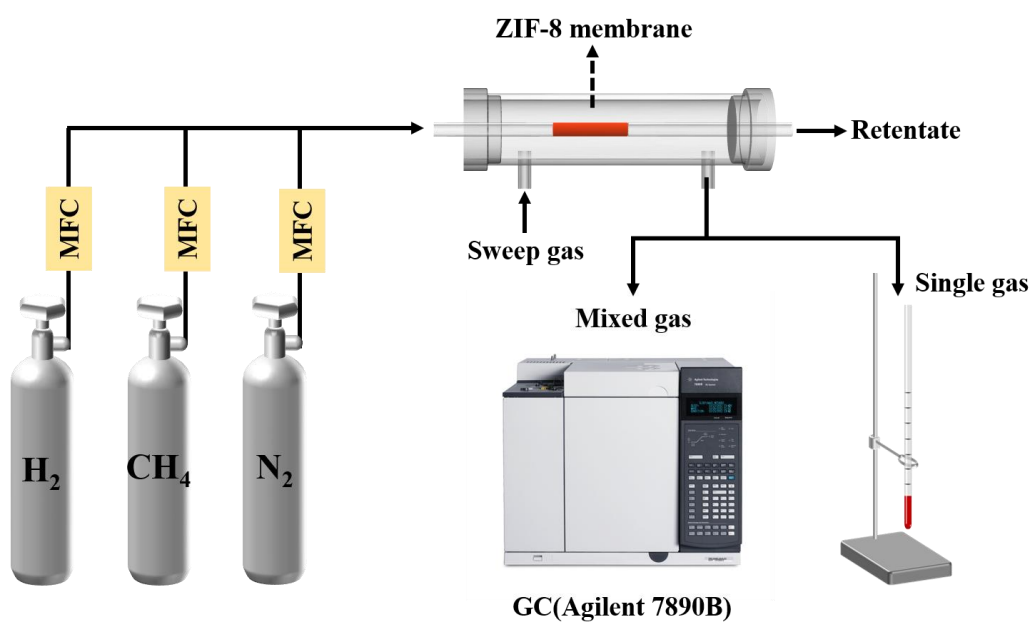


Fig. S4 Schematic diagram of gas permeation device.

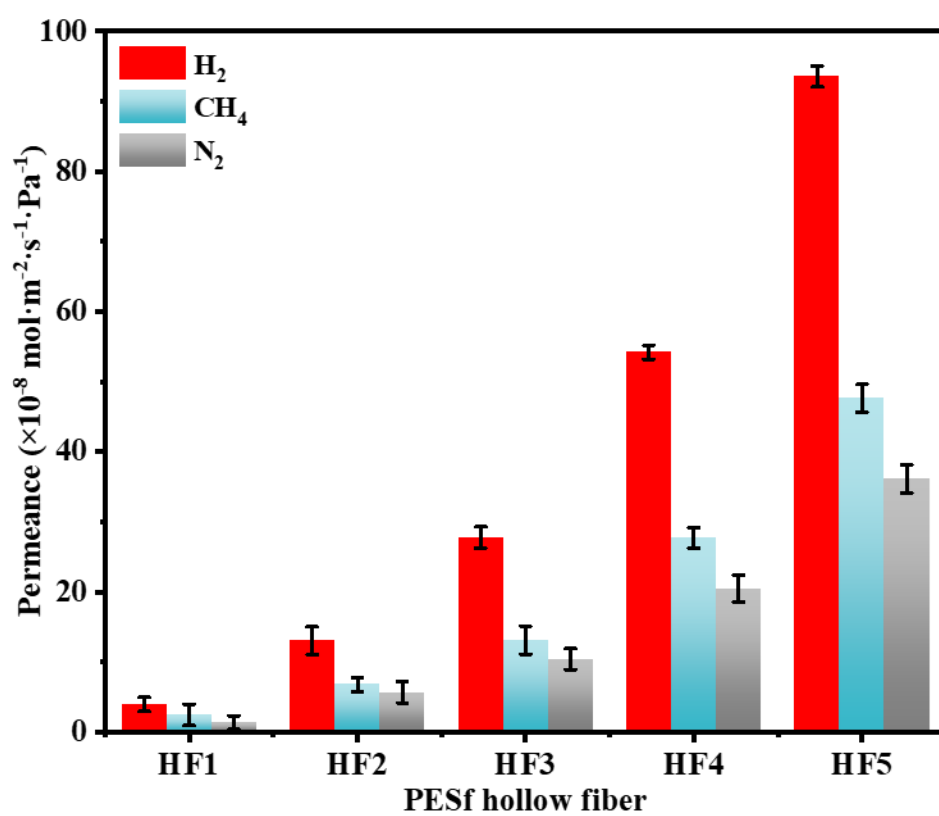


Fig. S5 Gas permeances for PESf hollow fibers.

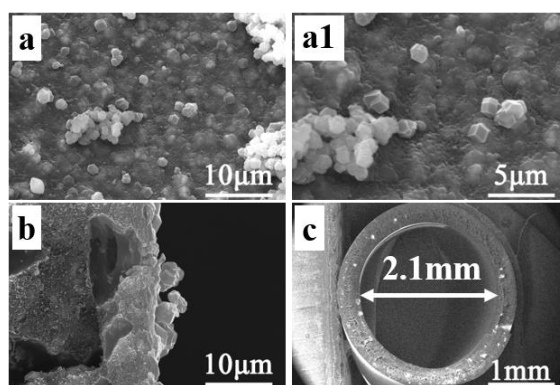


Fig. S6 SEM images of ZIF-8 membranes prepared under static condition at 70 °C and 8 h (a, a1: inner surface; b: cross-section; c: overall-view).

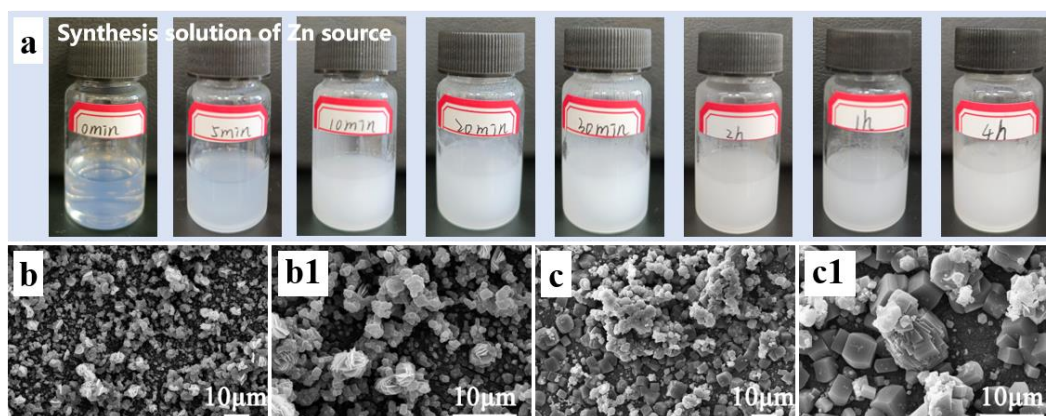


Fig. S7 Photographs of synthesis solutions containing zinc nitrate and Hmim at different reaction time (a), SEM images of ZIF-8 membranes prepared at different growth time (b, b1: 2 h; c, c1: 4 h).

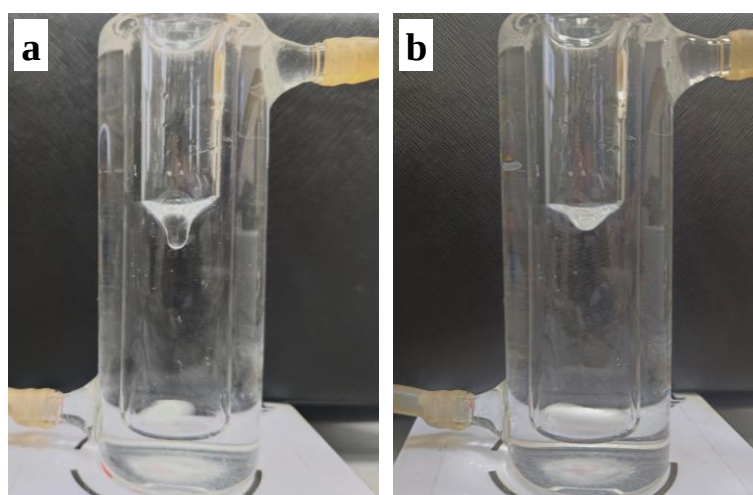


Fig. S8 Photographs of the synthetic solution at the beginning (a) and end (b) of the flow process.

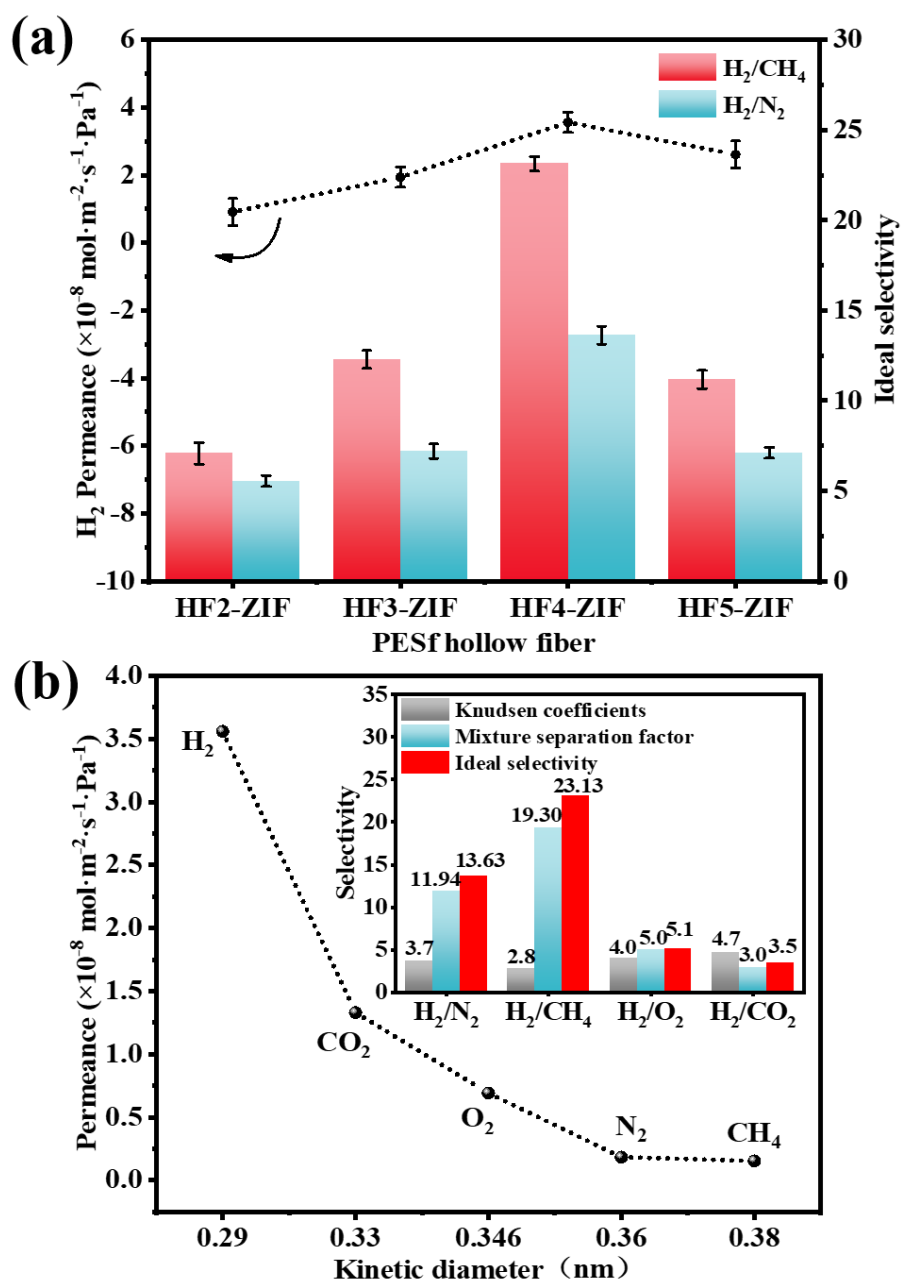


Fig. S9 Gas permeation data of ZIF-8 membranes supported on HFs (a) and ZIF-8 membrane supported on HF4 (b).

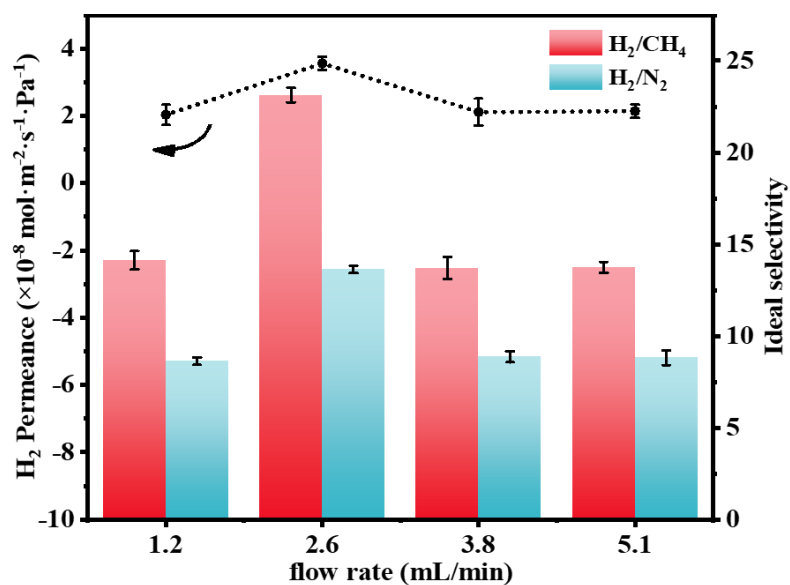


Fig. S10 Gas permeation data of ZIF-8 membranes prepared at different flow rates.

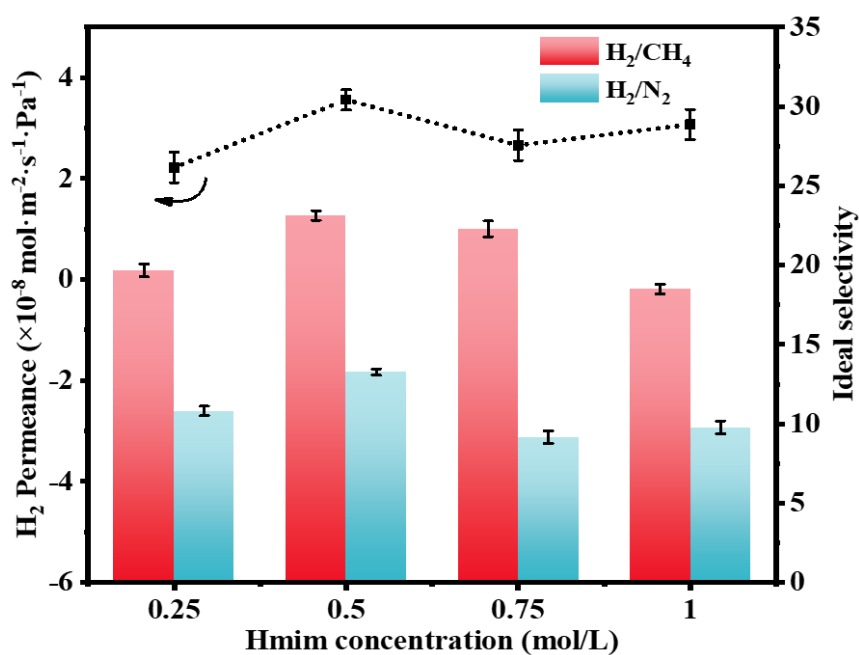


Fig. S11 Gas permeation data of ZIF-8 membranes prepared at different Hmim concentrations.

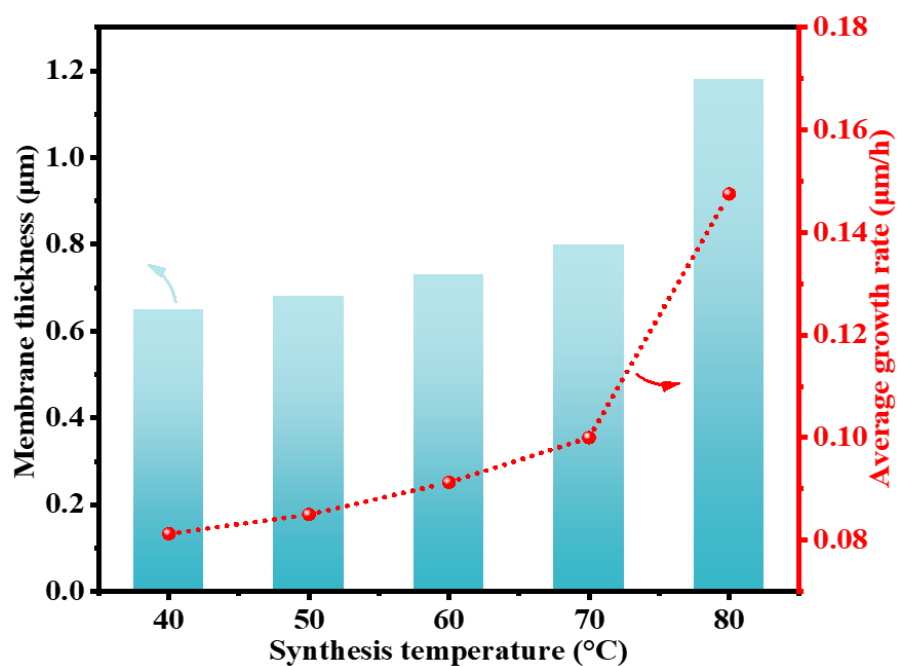


Fig. S12 The thickness and average growth rate of ZIF-8 membranes prepared at different synthesis temperature.

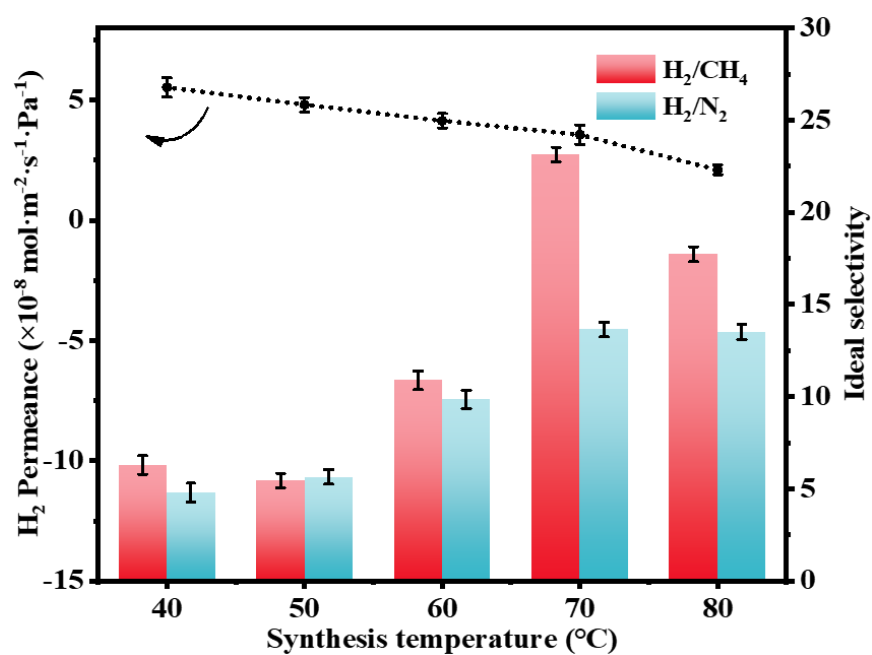


Fig. S13 Gas permeation data of ZIF-8 membranes prepared at different synthesis temperature.

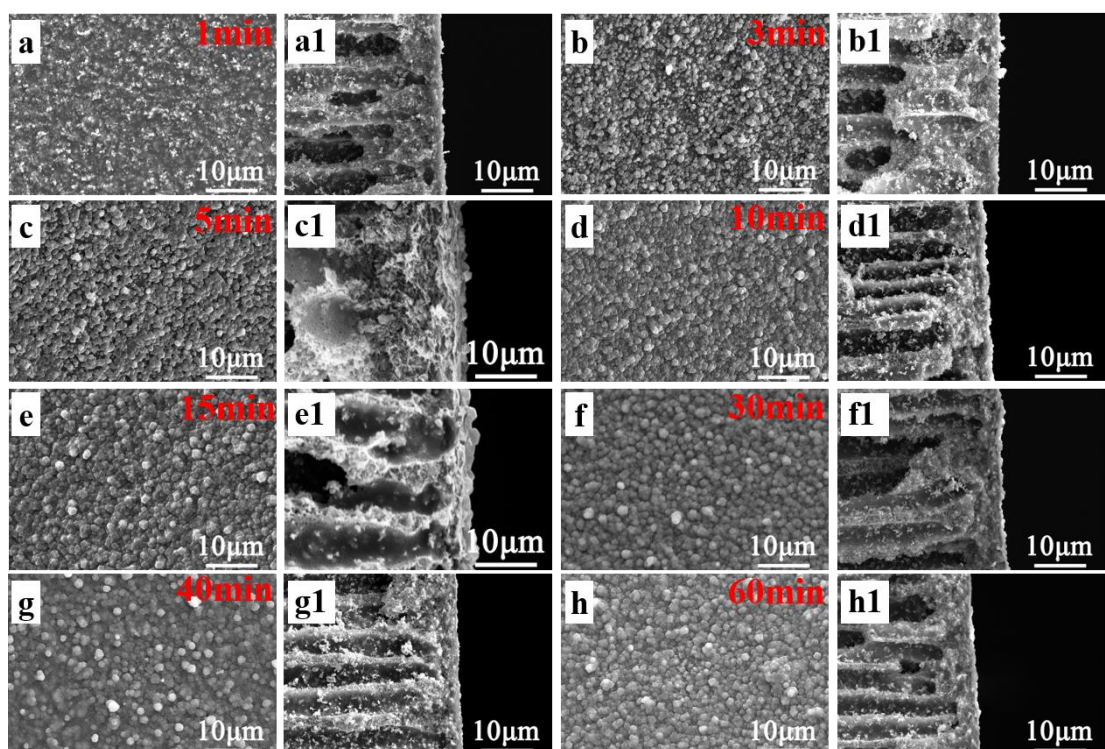


Fig. S14 SEM images of ZIF-8 membranes prepared under different synthesis time at 70 °C (a, a1: 1 min; b, b1: 3 min; c, c1: 5 min; d, d1: 10 min; e, e1: 15 min; f, f1: 30 min; g, g1: 40 min; h, h1: 60 min; a, b, c, d, e, f, g, h: surface morphology; a1, b1, c1, d1, e1, f1, g1, h1: cross-section view).

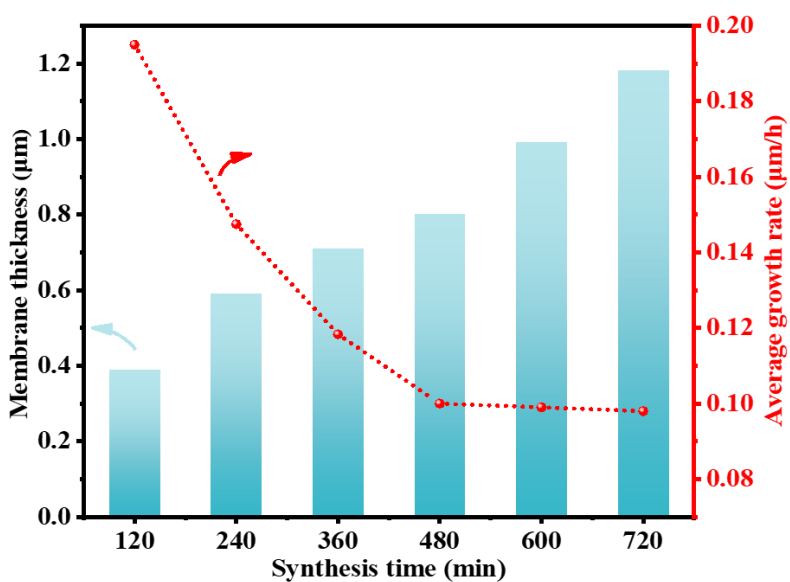


Fig. S15 The thickness and average growth rate of ZIF-8 membranes prepared at different synthesis time.

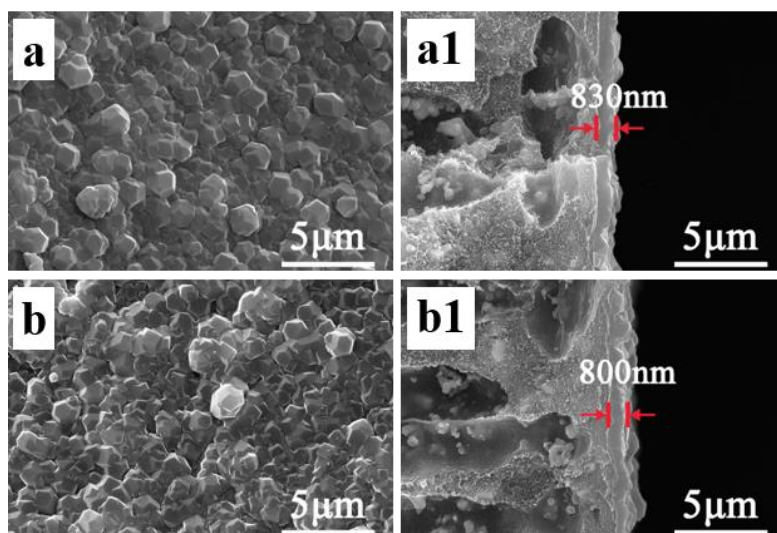


Fig. S16 SEM images of ZIF-8 membrane after treatment under different ultrasonication time (a, a1: 20 min; b, b1: 40 min; a, b: surface morphology; a1, b1: cross-section view).

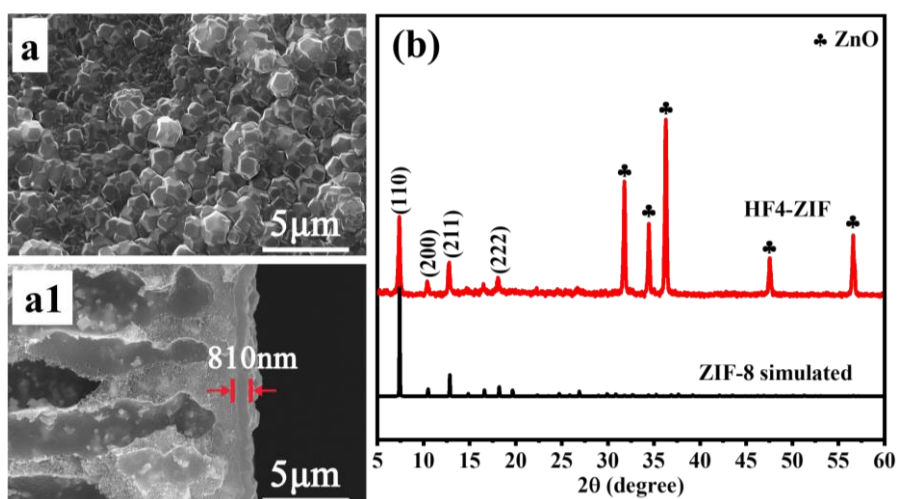


Fig. S17 SEM images (a: surface morphology; a1: cross-section view) and XRD patterns (b) of ZIF-8 membrane after thermal and long-term stability tests.

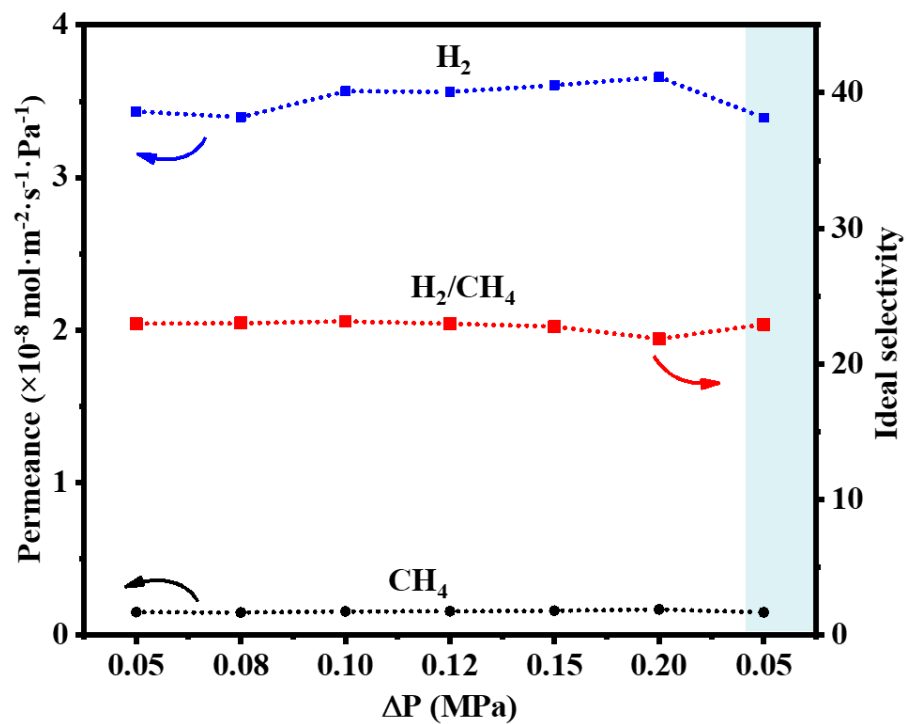


Fig. S18 Gas permeances as a function of pressure difference.

Table S1 ZIF-8 membranes prepared by different synthesis methods.

Preparation method	Support	The composition of synthetic solution (molar ratio)	Metal source in the solution	Alkaline additives in the solution	Solvents	Coupling agents or other reagents	Synthesis temperature	Synthesis time	Reaction vessel	Ref.
<i>In-situ</i> solvothermal synthesis	α -Al ₂ O ₃ disc	1ZnCl ₂ : 8.3Hmim: 5.6HCOONa: 250MeOH	ZnCl ₂ (0.55 g)	HCOONa (1.43 g)	MeOH (40 mL)	-	120 °C	4 h	Teflon-lined stainless steel autoclave	[1]
	PDA-modified α -Al ₂ O ₃ disc	1ZnCl ₂ : 2Hmim: 1HCOONa: 313MeOH	ZnCl ₂ (0.538 g)	HCOONa (0.268 g)	MeOH (50 mL)	Dopamin (2 mg/mL); Tris-HCl (10 mM)	85 °C	24 h	Teflon-lined stainless steel autoclave	[2]
	PDA-functionalized stainless-steel-net (SSN)	1ZnCl ₂ : 2Hmim: 1HCOONa: 313MeOH	ZnCl ₂ (0.538 g)	HCOONa (0.268 g)	MeOH (50 mL)	Dopamine (2 mg/mL); Tris-HCl (10 mM)	85 °C	24 h	Teflon-lined stainless steel autoclave	[3]
	PDA-functionalized stainless-steel-net (SSN)	1ZnCl ₂ : 2Hmim: 1HCOONa: 313MeOH	ZnCl ₂ (0.538 g)	HCOONa (0.268 g)	MeOH (50 mL)	Dopamine (2 mg/mL); Tris-HCl (10 mM)	85 °C	48 h	Teflon-lined stainless steel autoclave	[4]
Secondary seeding growth method	Hollow ceramic fiber	1ZnCl ₂ : 2.1Hmim: 1HCOONa: 306MeOH	ZnCl ₂ (0.55 g)	HCOONa (0.272 g)	MeOH (50 mL)	C ₃ H ₆ O	120 °C	8 h	Teflon-lined stainless steel autoclave	[5]

	α -Al ₂ O ₃ disc	1Zn(NO ₃) ₂ ·6H ₂ O: 74.5Hmim: 6000H ₂ O	Zn(NO ₃) ₂ ·6H ₂ O (0.11 g)	-	H ₂ O (40 mL)	MeOH	90 °C	24 h	Autoclave	[6]
	Al ₂ O ₃ hollow fiber	1ZnCl ₂ : 1.5Hmim: 1.5HCOONa: 250MeOH	ZnCl ₂ (0.674 g)	HCOONa (0.506 g)	MeOH (50 mL)	-	100 °C	2 h	Teflon-lined stainless steel autoclave	[7]
	ZIF-8- PVDF/PVDF dual-layer polymeric hollow fiber	1ZnCl ₂ : 1.5Hmim: 1.5HCOONa: 250MeOH	ZnCl ₂	HCOONa	MeOH	DMAc (N, N dimethylacet amide); PEG 400 (polyethylen e glycol 400)	100 °C	2.5 h	Teflon-lined stainless steel autoclave	[8]
	Zn foam substrates	1Zn(NO ₃) ₂ ·6H ₂ O: 4Hmim: 2HCOONa: 451MeOH	Zn(NO ₃) ₂ ·6H ₂ O (0.9766 g)	HCOONa (0.4477 g)	MeOH (60 mL)	CH ₃ CH ₂ OH	25 °C	24 h	Beaker	[9]
Plasma- assisted synthesis	α -Al ₂ O ₃ disc	1ZnCl ₂ : 2Hmim: 1.4HCOONa: 313MeOH	ZnCl ₂ (0.538 g)	HCOONa (0.3694 g)	MeOH (50 mL)	-	85 °C	24 h	Teflon-lined stainless steel autoclave	[10]
Electrochemic al deposition	Polyamide	1Zn(OAc) ₂ : 2Hmim: 500MeOH	Zn(OAc) ₂ (0.827 g)	-	MeOH (90 mL)	-	25 °C	15-75 min	Electrolytic cell	[11]
	Polypropylene support	1Zn(OAc) ₂ ·2H ₂ O: 60Hmim: 53H ₂ O	Zn(OAc) ₂ ·2H ₂ O (0.18 g)	-	H ₂ O (80 mL)	-	25 °C	1 h	Electrolytic cell	[12]
Contra- diffusion growth	α -Al ₂ O ₃ disc	A: 1ZnCl ₂ : 137MeOH B: 1Hmim: 0.12HCOONa: 15.6MeOH	ZnCl ₂ (0.98 g)	HCOONa (0.5 g)	MeOH (80 mL)	-	B: 120 °C	A: 1 h B: 4 h	Teflon-lined stainless steel autoclave	[13]

	Porous nylon	1Zn(NO ₃) ₂ ·6H ₂ O: 2Hmim: (8-64)NH ₃ ·H ₂ O: 2200H ₂ O	Zn(NO ₃) ₂ ·6H ₂ O (0.147 g)	-	H ₂ O (40 mL)	NH ₃ ·H ₂ O	22 ± 2 °C	48 h	Home-made setup	[14]
Current-driven synthesis process	AAO	1Zn(OAc) ₂ ·2H ₂ O: 2Hmim: 291MeOH: (0.1-0.4)dclm: 129DMF	Zn(OAc) ₂ ·2H ₂ O (0.66 g)	-	MeOH (60 mL); DMF (N, N- dimethylfor mamide, 30 mL)	CH ₃ COOH	25 °C	20 min	Electrolytic cell	[15]
Gel–vapour deposition	PVDF hollow fiber	Hmim	-	-	-	CH ₃ CH ₂ OH; C ₂ H ₇ NO; PEG (poly(ethylen e glycol)); DMAc (N,N- dimethylacet amide)	150 °C	6 h	Teflon-lined stainless steel autoclave	[16]
Interfacial synthesis	PES supports	A: Zn(NO ₃) ₂ ·6H ₂ O in H ₂ O B: Hmin in 1-Octanol	Zn(NO ₃) ₂ ·6H ₂ O	-	H ₂ O; 1-Octanol	PEG; NMP; C ₆ H ₁₄	25 °C	0.5-4 h	-	[17]
Interfacial microfluidic processing	PSf hollow fiber	Zn(NO ₃) ₂ ·6H ₂ O (0.1 mol/L) solution with a Hmim (0.3 mol/L) and HCOONa solution in MeOH	Zn(NO ₃) ₂ ·6H ₂ O (0.357 g)	HCOONa	MeOH (12 mL)	-	25 °C	2.5 h	Polypropylene syringes	[18]

Microwave assisted heating	α -Al ₂ O ₃ hollow fiber	1ZnCl ₂ : 2Hmim: 1.3HCOONa: 323MeOH	ZnCl ₂ (2.39 g)	HCOONa (1.49 g)	MeOH (180 mL)	-	140 °C	5 h	Teflon vessel	[19]
	TiO ₂ disc	1 ZnCl ₂ : 1.5 Hmim: 1 HCOONa: 249.5 MeOH	ZnCl ₂ (1.078 g)	HCOONa (0.54 g)	MeOH (80 mL)	-	100 °C	4 h	Teflon-lined stainless steel autoclave	[20]
Semisolid processing method	Al ₂ O ₃ disc	1 Zn(OAc) ₂ ·2H ₂ O: 2: Hmim: 18 DMAc: 46 H ₂ O	Zn(OAc) ₂ ·2H ₂ O (1.32 g)	-	DMAc (10 mL); H ₂ O (5 mL)	MeOH; CHCl ₃	200 °C	15 min	-	[21]
Homologous metal induction	PESf hollow fiber	1Zn(NO ₃) ₂ ·6H ₂ O: 1.5Hmim: 0.65HCOONa: 450MeOH	Zn(NO ₃) ₂ ·6H ₂ O	HCOONa	MeOH	NMP	60 °C	6 h	Teflon-lined stainless steel autoclave	[22]
	Al ₂ O ₃ -ZnO hollow fiber	1Zn(NO ₃) ₂ ·6H ₂ O: 1.4Hmim: 1HCOONa: 448MeOH	Zn(NO ₃) ₂ ·6H ₂ O (2 g)	HCOONa (0.46 g)	MeOH (122 mL)	CH ₃ CH ₂ O; NMP	100 °C	8 h	Teflon-lined stainless steel autoclave	[23]
	α -Al ₂ O ₃ disc	Hmim solution in DI water (MeOH; DMF; DI water–MeOH mixtures) (0.691 mol/L)	Zn(NO ₃) ₂ ·6H ₂ O	-	DI water (MeOH; DMF; DI water–MeOH mixtures)	HNO ₃	30-90 °C	12 h (48 h)	-	[24]
	ZnO hollow fiber	1Zn(NO ₃) ₂ ·6H ₂ O: 1.5Hmim: 0.65HCOONa: 450MeOH	Zn(NO ₃) ₂ ·6H ₂ O (2 g)	HCOONa (0.46 g)	MeOH (122.5 mL)	NMP	80 °C	8 h	Teflon-lined stainless steel autoclave	[25]

In-situ synthesis	PESf hollow fiber	1Hmim: 111H ₂ O	-	-	H ₂ O (50 mL)	NMP	70 °C	8 h	Glass container	This work
-------------------	-------------------	----------------------------	---	---	--------------------------	-----	-------	-----	-----------------	-----------

Table S2 Hazard codes and boiling points of solvents according to the “ECHA C&L Inventory database”.

Solvent	Hazard codes	Boiling point (°C)
N, N-Dimethylformamide (DMF)	H302, H312, H315, H332, H319	153
N, N-Dimethylacetamide (DMAc)	H360D, H312, H332	165
N-Methyl Pyrrolidone (NMP)	H315, H319, H335, H360D	203
Methanol (MeOH)	H225, H301, H311, H331, H370	64.5
Ethanol (EtOH)	H225	78.3
1-Octanol (C ₈ H ₁₈ O)	H302, H315, H319, H332, H335	195
Acetone (C ₃ H ₆ O)	H225, H315, H319, H336	56
Zinc nitrate (Zn(NO ₃) ₂ ·6H ₂ O)	H272, H302, H315, H319	105-131
Zinc chloride (ZnCl ₂)	H314, H302, H400, H410	732
Zinc acetate (Zn(OAc) ₂)	H302, H318, H319, H335, H411	330
Zinc acetate dihydrate (Zn(OAc) ₂ ·H ₂ O)	H302, H315, H319, H335, H4110	242-244

2-Methylimidazole (Hmim)	H302, H314, H360, H371, H373, H341, H351	267-268
4, 5-dichloroimidazole (dclm)	H315, H319, H335	334.1±22
Sodium formate (HCOONa)	H319, H336	360
Dopamine (DPA)	-	337.7
Tris Hydrochloride (Tris HCl)	H315, H319, H335	225
polyethylene glycol (PEG)	H302, H3115, H319, H335	200-250
Ethanolamine (C ₂ H ₇ NO)	H302, H312, H314, H332, H335	170
Ammonium hydroxide (NH ₃ ·H ₂ O)	H314, H318, H335, H400	36
Hexane (C ₆ H ₁₄)	H225, H315, H336, H304, H373, H361e, H411	69
Chloroform (CHCl ₃)	H331, H302, H315, H319, H351, H372, H361d	61.3
Acetic acid (CH ₃ COOH)	H226, H314, H315, H319	117-118
Nitric acid (HNO ₃)	H272, H314, H330	83

Table S3 Residence time corresponding to different flow rates.

The speed of pump (RPM)	Flow rate of solution (mL/min)	Residence time (min)
5	1.2	2.6
10	2.6	1.2
15	3.8	0.8
20	5.1	0.6

Table S4 Detailed comparison of H₂/CH₄ and H₂/N₂ separation performance of membranes listed in Fig. 9.

Membrane	Support	Thickness (μm)	H ₂ permeance (mol·m ⁻² ·s ⁻¹ ·Pa ⁻¹)	H ₂ /CH ₄ selectivity	H ₂ /N ₂ selectivity	Synthetic method; solvent	Stability	Ref.
ZIF-8	PESf hollow fiber	0.8	3.56×10 ⁻⁸	23.1	13.6			This work
	α-Al ₂ O ₃ disc	25	2.4×10 ⁻⁷	14	9.6	Solvothermal synthesis; methanol	-	[1]
	α-Al ₂ O ₃ disc	20	1.8×10 ⁻⁷	31.5	16.2	Solvothermal synthesis; methanol	Pressure	[2]

	Stainless-steel-net (SSN)	-	2.12×10^{-5}	23.2	15	Solvothermal synthesis; methanol	Pressure	[3]
	Stainless-steel-net (SSN)	-	2.02×10^{-5}	-	4.111	Solvothermal synthesis; methanol	-	[4]
	Hollow ceramic fiber	20	7.29×10^{-7}	10.8	9.2	Solvothermal synthesis; methanol	Heating, pressure	[5]
	Al ₂ O ₃ disc	0.9	1.81×10^{-6}	9.56	-	Solvothermal synthesis; DI water	Long-time, heating	[6]
	Al ₂ O ₃ hollow fiber	7	14.6×10^{-8}	-	30.9	Solvothermal synthesis; methanol	Long-time	[7]
	ZIF-8-PVDF/PVDF dual-layer polymeric hollow fiber	7.8	1.5×10^{-7}	-	69.5	Solvothermal synthesis; methanol	Heating, pressure	[8]
	Zn foam substrates	244	8.1×10^{-6}	7.3	6	Room temperature; methanol	Heating, pressure, long-time	[9]
	Porous nylon	2.5	11.3×10^{-7}	-	4.6	Room temperature; water	-	[14]

	PVDF hollow fiber	0.87	1.19×10^{-5}	27.3	22.4	Gel–vapour deposition	-	[16]
	PSf hollow fiber	3.6	4.7×10^{-9}	17.2	18.3	Microfluidic synthesis; methanol	Pressure	[18]
	α -Al ₂ O ₃ hollow fiber	22	6.1×10^{-7}	12.4	1.8	Microwave assisted; methanol	Heating, pressure	[19]
	TiO ₂ disc	42	6.74×10^{-8}	11.2	4.1	Solvothermal synthesis; methanol	Heating	[20]
	PESf hollow fiber	6.2	1.06×10^{-7}	2.95	4.57	Solvothermal synthesis; methanol	Ultrasonic, long- time, reproducibility	[22]
	ZnO hollow fiber	10.5	1.2×10^{-6}	4.29	3.73	Solvothermal synthesis; methanol	Ultrasonic, reproducibility	[25]
	α -Al ₂ O ₃ hollow fiber	4.8	1.19×10^{-6}	9.6	10.2	Solvothermal synthesis; methanol	Ultrasonic, reproducibility	[26]
	α -Al ₂ O ₃ tube	6	1.88×10^{-7}	11.4	10.3	Solvothermal synthesis; methanol	-	[27]
	SSN	10	1.47×10^{-7}	12.4	9	Electrodeposition method; methanol	Long-time	[28]

	AAO disc	2.5	4.71×10^{-6}	9.76	12.53	Room temperature; ethanol-water	-	[29]
	α -Al ₂ O ₃ tube	1	2.3×10^{-7}	14.8	13.7	Room temperature; water	-	[30]
	α -Al ₂ O ₃ tube	8	2.08×10^{-7}	10.4	10.3	Solvothermal synthesis; methanol	-	[31]
	α -Al ₂ O ₃ disc	12	1.0×10^{-7}	16	-	Solvothermal synthesis; methanol	-	[32]
	α -Al ₂ O ₃ hollow fibre	2	4.3×10^{-7}	12.13	11.06	cycling flow; DI water	Long-time	[33]
	PESf disc	35	2.0×10^{-7}	10.5	12.4	Solvothermal synthesis; methanol	Heating	[34]
	Al ₂ O ₃ disc	0.29	1.8×10^{-8}	2.8	3.2	Vapor-solid synthesis	-	[35]
	α -Al ₂ O ₃ disc	0.5	1.9×10^{-8}	9.5	10	stepwise deposition method; methanol	Long-time	[36]
ZIF-8/GO	AAO disc	0.1	5.5×10^{-8}	11.2	11.1	mutual diffusion; methanol	-	[37]

ZIF-67	Al ₂ O ₃ -ZnO hollow fiber	7.5	5.24×10 ⁻⁷	5.17	7.74	Solvothermal synthesis; methanol	Heating, ultrasonic, Reproducibility	[23]
	Al ₂ O ₃ disc	7	2.22×10 ⁻⁷	14.92	-	Solvothermal synthesis; methanol	Reproducibility	[10]
	ZnO hollow fiber	13.8	8.98×10 ⁻⁸	15.89	11.75	Solvothermal synthesis; methanol	Heating, long-time, reproducibility	[38]
	α-Al ₂ O ₃ tube	5	8.21×10 ⁻⁸	33.1	28.7	Solvothermal synthesis; water	Long-time	[39]
ZIF-7	PSf hollow fiber	2.4	2.2×10 ⁻⁹	34.6	35.1	Microfluidic synthesis; ethanol	Pressure	[18]
	Al ₂ O ₃ disc	2	4.5×10 ⁻⁸	14.7	20.7	Solvothermal synthesis; DMF	Long-time	[40]
	α-Al ₂ O ₃ disc	1.5	7.4×10 ⁻⁷	6.29	6.73	Microwave synthesis; DMF	-	[41]
ZIF-9	α-Al ₂ O ₃ tube	10	1.85×10 ⁻⁷	7.6	8.9	Solvothermal synthesis; DMF	Long-time, heating	[42]
	α-Al ₂ O ₃ tube	4	1.13×10 ⁻⁷	14.7	8.2	Solvothermal synthesis; DMF	Long-time	[43]
	α-Al ₂ O ₃ disc	50	7.43×10 ⁻⁶	5.16	3.24	Solvothermal synthesis; DMF	-	[44]
UiO-66	Al ₂ O ₃ disc	1.6	4.1×10 ⁻⁷	4.2	6.4	Solvothermal synthesis; DMF-	Long-time	[45]

							HAc		
		Al ₂ O ₃ hollow fiber	2	7.2×10 ⁻⁷	6.4	22.4	Solvothermal synthesis; DMF	-	[46]
		α-Al ₂ O ₃ tube	6	5.4×10 ⁻⁷	11.4	8.2	Solvothermal synthesis; DMF- HAc	-	[47]
UiO-66-NH ₂		α-Al ₂ O ₃ hollow fiber	5.7	1.39×10 ⁻⁷	21.7	3.13	Solvothermal synthesis; DMF- HAc	Long-time, heating, ultrasonic, hydrothermal, reproducibility	[48]
		α-Al ₂ O ₃ hollow fiber	3	1.57×10 ⁻⁷	18.9	-	Solvothermal synthesis; DMF- HAc	Heating	[49]
Zn ₂ (bim) ₄		ZnO disc	1.8	1.25×10 ⁻⁸	9.2	-	Solvothermal synthesis; toluene	Heating	[50]
Polymer membrane	PLA	-	25	3.35×10 ⁻¹⁰	10	5	-	-	[51]
	PPSU/PBNPI	-	40-50	3.38×10 ⁻¹⁰	5.05	-	-	-	[52]
	PIM-1	-	65	1.74×10 ⁻⁸	5.9	8.5	-	-	[53]
	PIM-NH ₂	-	67	7.25×10 ⁻⁹	6.8	10.8	-	-	[53]
	PIM-deBOC	-	80	7.12×10 ⁻⁹	6.7	10	-	-	[53]
	PIM-T-BOC	-	71	6.13×10 ⁻¹⁰	26	36	-	-	[53]
	PIM-1	-	44	1.46×10 ⁻⁸	7.99	10	-	Long-time	[54]
	PIM-1-550	-	34	9.63×10 ⁻⁹	4.54	5	-	Long-time	[54]
	PIM-EA-TB	-	129	2.02×10 ⁻⁸	11.1	14.8	-	-	[55]
	PI	-	40-60	6.45×10 ⁻⁹	15.5	12.9	-	-	[56]

	CTB1-DMN	-	48	9.04×10^{-9}	13.5	17	-	-	[57]
MMMs membrane	ZnO/ZIF-8	-	-	3.29×10^{-8}	16.6	13.1	-	Heating	[58]

References

- 1 M. Shah, H. T. Kwon, V. Tran, S. Sachdeva and H.K. Jeong, *Microporous Mesoporous Mater.*, 2013, **165**, 63-69.
- 2 Q. Liu, N. Y. Wang, J. Caro and A. S. Huang, *J. Am. Chem. Soc.*, 2013, **135**, 17679-17682.
- 3 A. S. Huang, Q. Liu, N. Y. Wang and J. Caro, *J. Mater. Chem. A*, 2014, **2**, 8246-8251.
- 4 T. M. T. Nguyen, J. W. Chen, M. T. Pham, H. M. Bui, C. C. Hu, S. J. You and Y. F. Wang, *Environ. Technol. Inno.*, 2023, **31**, 103169.
- 5 K. Tao, L. J. Cao, Y. C. Lin, C. L. Kong and L. Chen, *J. Mater. Chem. A*, 2013, **1**, 13046.
- 6 Y. J. Shan, M. L. He, F. Zhang, Y. D. Yang, Y. F. Wang, T. Gui, Y. Q. Li, Z. P. Wang and X. S. Chen, *Ind. Eng. Chem. Res.*, 2024, **63**, 7335-7347.
- 7 X. K. Wu, Y. W. Yang, X. F. Lu and Z. B. Wang, *J. Membr. Sci.*, 2020, **613**, 118518.
- 8 R. M. Zhu, L. Wang, H. Y. Zhang, C. J. Liu and Z. B. Wang, *Sep. Purif. Technol.*, 2024, **335**, 126209.
- 9 F. Yang, J. Q. Sun, Y. X. Sun, Z. H. Qiao and C. L. Zhong, *Sep. Purif. Technol.*, 2023, **325**, 124751.
- 10 Y. J. Shan, M. L. He, F. Zhang, Y. F. Wang, Y. X. Liu, Y. D. Yang, X. Q. Wang, X. K. Zhang, Y. Q. Li, Z. P. Wang and X. S. Chen, *Sep. Purif. Technol.*, 2023, **317**, 123871.
- 11 S. Y. Zhu, Q. J. Lin, X. P. Huang, L. J. Chen, L. Z. Liu, Z. Z. Yao and S. C. Xiang, *ACS Omega*, 2021, **6**, 33018-33023.
- 12 W. Shao, Y. W. Zhou, Z. Chen, Y. L. Chen, Y. Li, Y. J. Ban, W. S. Yang, M. Xue and X. M. Chen, *J. Colloid Interface Sci.*, 2024, **665**, 693-701.
- 13 H. T. Kwon and H. K. Jeong, *J. Am. Chem. Soc.*, 2013, **135**, 10763-10768.
- 14 M. He, J. F. Yao, L. X. Li, Z. X. Zhong, F. Y. Chen and H. T. Wang, *Microporous Mesoporous Mater.*, 2013, **179**, 10-16.
- 15 Y. L. Zhao, X. T. Cai, W. F. Wu, L. B. Li, Y. Y. Wei and H. H. Wang, *J. Membr. Sci.*, 2024, **712**, 123254.
- 16 W. B. Li, P. C. Su, Z. J. Li, Z. H. Xu, F. Wang, H. Ou, J. H. Zhang, G. L. Zhang and E. Zeng, *Nat. Commun.*, 2017, **8**, 406.
- 17 Y. B. Li, L. H. Wee, J. A. Martens and I. F. J. Vankelecom, *J. Membr. Sci.*, 2017, **523**, 561-566.
- 18 F. Cacho-Bailo, S. Catalán-Aguirre, M. Etxeberria-Benavides, O. Karvan, V. Sebastian, C. Téllez and J. Coronas, *J. Membr. Sci.*, 2015, **476**, 277-285.
- 19 N. H. Suhaimi, Y. F. Yeong, H. N. A. Aziz and L. S. Lai, *Chemosphere*, 2022, **308**, 136167.
- 20 H. Bux, F. Y. Liang, Y. S. Li, J. Cravillon, M. Wiebcke and J. Caro, *J. Am. Chem. Soc.*, 2009, **131**, 16000-16001.
- 21 Q. Ma, K. Mo, S. S. Gao, Y. F. Xie, J. Z. Wang, H. Jin, A. Feldhoff, S. T. Xu, J. Y. S. Lin and Y. S. Li, *Angew. Chem. Int. Ed.*, 2020, **59**, 21909-21914.
- 22 Y. Liu, Y. Wang, L. Liu, T. F. Yang, H. Z. Bao and X. B. Wang, *Mater. Lett.*, 2023, **336**, 133842.
- 23 H. H. Chen, X. B. Wang, Y. Liu, T. F. Yang, N. T. Yang, B. Meng, X. Y. Tan and S. M. Liu, *J. Membr. Sci.*, 2021, **640**, 119851.
- 24 J. H. Lee, D. Kim, H. Shin, S. J. Yoo, H. T. Kwon and J. Kim, *J. Ind. Eng. Chem.*, 2019, **72**, 374-379.
- 25 Y. Wang, H. H. Chen, X. B. Wang, B. Meng, N. T. Yang, X. Y. Tan and S. M. Liu, *Ind. Eng. Chem. Res.*, 2020, **59**, 15576-15585.
- 26 Y. Wang, H. L. Zhang, X. B. Wang, C. X. Zou, B. Meng and X. Y. Tan, *Ind. Eng. Chem. Res.*, 2019, **58**, 19511-19518.
- 27 X. F. Zhang, Y. G. Liu, S. H. Li, L. Y. Kong, H. O. Liu, Y. S. Li, W. Han, K. L. Yeung, W. D. Zhu, W. S. Yang and J. S. Qiu, *Chem. Mater.*, 2014, **26**, 1975-1981.
- 28 S. Zhou, Y. Y. Wei, L. B. Zhuang, L. X. Ding and H. H. Wang, *J. Mater. Chem. A*, 2017, **5**, 1948-1951.
- 29 J. W. Li, W. Cao, Y. Y. Mao, Y. L. Ying, L. W. Sun and X. S. Peng, *CrystEngComm*, 2014, **16**, 9788-9791.
- 30 S. Tanaka, K. Okubo, K. Kida, M. Sugita and T. Takewaki, *J. Membr. Sci.*, 2017, **544**, 306-311.
- 31 X. F. Zhang, Y. G. Liu, L. Y. Kong, H. O. Liu, J. S. Qiu, W. Han, L. T. Weng, K. L. Yeung and W. D. Zhu, *J. Mater. Chem. A*, 2013, **1**, 10635-10638.
- 32 H. Bux, A. Feldhoff, J. Cravillon, M. Wiebcke, Y. S. Li and J. Caro, *Chem. Mater.*, 2011, **23**, 2262-2269.
- 33 K. Huang, Z. Y. Dong, Q. Q. Li and W. Q. Jin, *Chem. Commun.*, 2013, **49**, 10326-10328.
- 34 F. Cacho-Bailo, B. Seoane, C. Téllez and J. Coronas, *J. Membr. Sci.*, 2014, **464**, 119-126.
- 35 B. Reif, J. Somboonvong, F. Fabisch, M. Kaspereit, M. Hartmann and W. Schwieger, *Microporous Mesoporous Mater.*, 2019, **276**, 29-40.
- 36 O. Shekhah, R. Swaidan, Y. Belmabkhout, M. du Plessis, T. Jacobs, L. J. Barbour, I. Pinnau and M. Eddaoudi, *Chem. Commun.*, 2014, **50**, 2089-2092.
- 37 Y. X. Hu, J. Wei, Y. Liang, H. C. Zhang, X. W. Zhang, W. Shen and H. T. Wang, *Angew. Chem. Int. Ed.*, 2016, **128**, 2088-2092.
- 38 H. H. Chen, C. Li, L. Liu, B. Meng, N. T. Yang, J. Sunarso, L. M. Liu, S. M. Liu and X. B. Wang, *J. Membr. Sci.*, 2022, **653**, 120550.
- 39 Y. J. Li, C. C. Ma, P. Nian, H. O. Liu and X. F. Zhang, *J. Membr. Sci.*, 2019, **581**, 344-354.
- 40 Y. S. Li, F. Y. Liang, H. Bux, W. S. Yang and J. Caro, *J. Membr. Sci.*, 2010, **354**, 48-54.
- 41 Y. S. Li, F. Y. Liang, H. Bux, A. Feldhoff, W. S. Yang and J. Caro, *Angew. Chem. Int. Ed.*, 2010, **122**, 558-561.
- 42 P. Nian, Y. J. Li, X. Zhang, Y. Cao, H. O. Liu and X. F. Zhang, *ACS Appl. Mater. Interfaces*, 2018, **10**, 4151-4160.
- 43 J. Q. Liu, C. Y. Liu and A. S. Huang, *Int. J. Hydrogen Energy*, 2020, **45**, 703-711.
- 44 Y. Y. Huang, D. H. Liu, Z. P. Liu and C. L. Zhong, *Ind. Eng. Chem. Res.*, 2016, **55**, 7164-7170.
- 45 R. Rong, Y. W. Sun, T. T. Ji and Y. Liu, *J. Membr. Sci.*, 2020, **610**, 118275.
- 46 X. L. Liu, N. K. Demir, Z. T. Wu, K. Li, *J. Am. Chem. Soc.*, 2015, **137**, 6999-7002.
- 47 F. C. Wu, L. Lin, H. O. Liu, H. T. Wang, J. S. Qiu and X. F. Zhang, *J. Membr. Sci.*, 2017, **544**, 342-350.
- 48 Y. Liu, L. Liu, Y. F. Yang, T. F. Yang, C. Li, S. Kawi and X. B. Wang, *J. Membr. Sci.*, 2023, **684**, 121851.

- 49 H. Guo, J. Q. Liu, Y. H. Li, J. Caro and A. S. Huang, *Microporous Mesoporous Mater.*, 2021, **313**, 110823.
- 50 L. Liu, Y. F. Yang, R. Y. Zhang, T. F. Yang, Z. W. Xue and X. B. Wang, *J. Membr. Sci.*, 2024, **689**, 112128.
- 51 A. Iulianelli, C. Algieri, L. Donato, A. Garofalo, F. Galiano, G. Bagnato, A. Basile and A. Figoli, *Int. J. Hydrogen Energy*, 2017, **42**, 22138-22148.
- 52 T. H. Weng, H. H. Tseng and M. Y. Wey, *Int. J. Hydrogen Energy*, 2008, **33**, 4178-4182.
- 53 K. Mizrahi Rodriguez, S. Lin, A.X. Wu, G. Han, J. J. Teesdale, C. M. Doherty and Z. P. Smith, *Angew. Chem. Int. Ed.*, 2021, **60**, 6593-6599.
- 54 S. S. He, X. Jiang, S. W. Li, F. T. Ran, J. Long and L. Shao, *AIChE J.*, 2020, **66**, e16543.
- 55 M. Carta, R. Malpass-Evans, M. Croad, Y. Rogan, J. C. Jansen, P. Bernardo, F. Bazzarelli and N. B. McKeown, *Science*, 2013, **339**, 303-307.
- 56 T. Yoshioka, K. Kojima, R. Shindo and K. Nagai, *J. Appl. Polym. Sci.*, 2017, **134**, 44569.
- 57 X. H. Ma, M. A. Abdulhamid and I. Pinnau, *Macromolecules*, 2017, **50**, 5850-5857.
- 58 Y. G. Liu, S. H. Li, X. F. Zhang, H. O. Liu, J. S. Qiu, Y. S. Li and K. L. Yeung, *Inorg. Chem. Commun.*, 2014, **48**, 77-80.