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## **Supplementary Information**

Preparation of inorganic molten salt composite phase change materials and study on their electrothermal conversion properties

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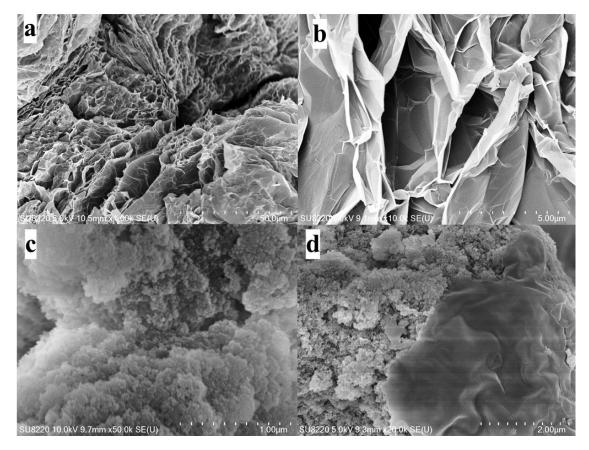


Fig. S1 The micromorphology of each substance. (a, b) EG; (c)  $LiNO_3$ - $KCl/SiO_2$ ; (d)  $LiNO_3$ - $KCl/SiO_2$ /EG.

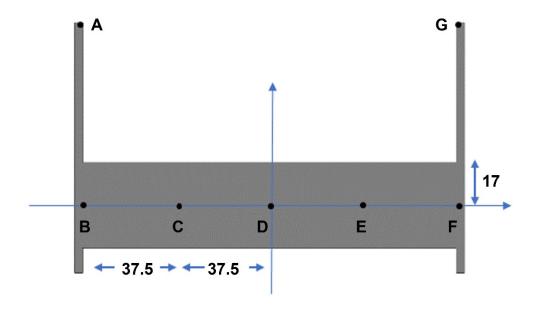


Fig. S2 Temperature monitoring point distribution.

Table S1 Physical properties of materials.

	graphite electrode	CPCM block
specific heat (J g <sup>-1</sup> K <sup>-1</sup> )	0.71	1.45
thermal conductivity (W m <sup>-1</sup> K <sup>-1</sup> )	121	5.2

**Table S2** Parameter setting values of each component under simulated working conditions

	graphite electrode	potting compound	CPCM block
specific heat capacity / J g <sup>-1</sup> K <sup>-1</sup>	0.71	1	The function of the 6 wt% curve in Figure 2b.
density / g cm <sup>3</sup>	1.8	1.3	1.8
thermal conductivity / W m <sup>-1</sup> K <sup>-1</sup>	121	0.4	5.2
heat generation density /W m <sup>-3</sup>	Fig. S3 Graphite electrode heat generation density function	0	Fig. S3 Heat generation density function of the CPCM block

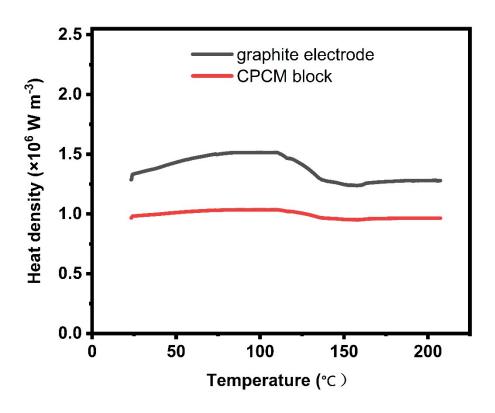


Fig. S3 The heating density curve of graphite electrode and composite PCM

block as a function of temperature.

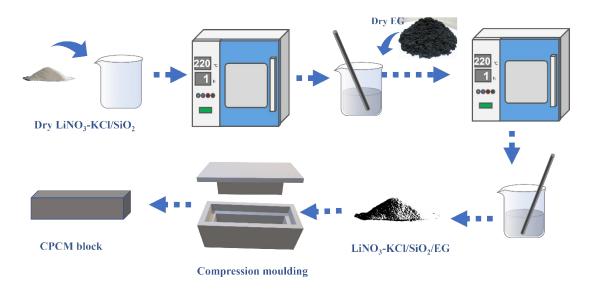


Fig. S4 Composite phase change material preparation process diagram.

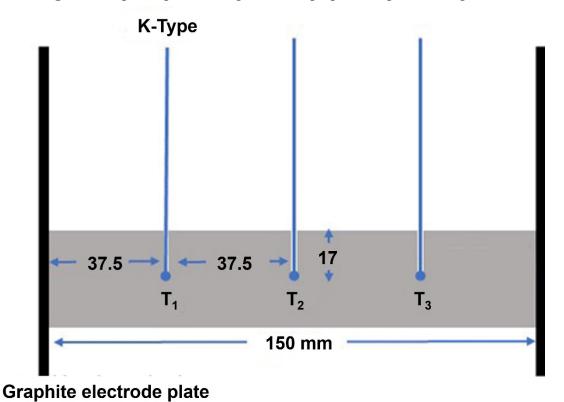
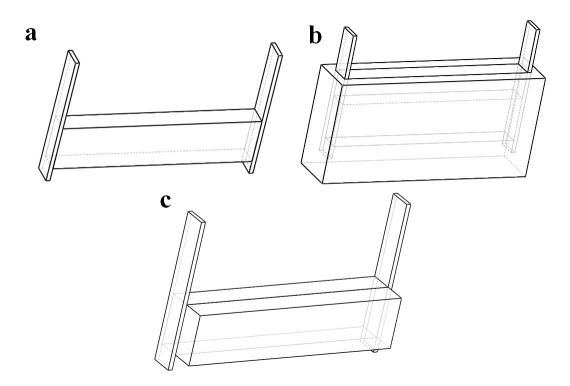


Fig. S5. The distribution of temperature measurement points.



**Fig. S6.** Three-dimensional graphics of EThCM. (a) Before potting; (b) After potting; (c) Indirect heating.