

1 Electronic Supplementary information to

2 **Millifluidic magnetophoresis-based chip for age-specific fractionation:**  
3 **Evaluating the impact of age on metabolomics and gene expression in**  
4 **yeast**

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8 **A Results and Discussion**

9 **A Magnetic nanoparticle characterization.**

10 Achieving continuous high separation selectivities necessitates the understanding of concentration and time  
11 dependence of the linker-protein@particle agglomerate formation (see Section Results A). Therefore, the  
12 nanoparticles are characterized in the following. The magnetic nanoparticles (MNP) (ethylenediaminetetraacetic  
13 acid (EDTA)- functionalized silica shell iron oxide nanoparticles) utilized for labeling the yeast cell's bud scars, as  
14 previously described<sup>1</sup>, exhibit an iron oxide core, enabling magneto-responsive behavior. Superconducting quantum  
15 interference device (SQUID) measurements, presented in Fig. S1 a, reveal no remanence and superparamagnetic  
16 behavior, having a saturation magnetization of 13.13 emu g<sup>-1</sup>. X-ray diffraction (XRD) measurements reveal that the  
17 iron oxide nanoparticles (IONs) exhibit a diameter of 13.11 ± 2.07 nm (Fig. S1 b). Further, the XRD analysis confirms  
18 the typical reflections ((220), (311), (400), (511), and (440)) for magnetite with its cubic crystal structure<sup>2</sup>. These  
19 iron oxide cores, evident in transmission electron microscopy (TEM) images in Fig. S1 c, are encapsulated by a silica  
20 shell that precludes the IONs from agglomerating, as previously shown<sup>3,4</sup>. For these particles to bind with the  
21 histidine tag of the linker-protein, an EDTA surface was introduced via an amide bond<sup>1,5</sup>. Although this surface is  
22 not discernible in the TEM images, it is detectable in the thermogravimetric analysis (TGA). TGA results corroborate  
23 the presence of different coatings. Fig. S1 d illustrates the weight loss across temperatures for the different coating  
24 steps. The initial weight loss step, associated with the desorption of surface-bound water, is evident at temperatures  
25 below 100°C. The iron oxide core is stable up to 700°C, with negligible weight loss. For the coatings, the weight loss  
26 between 100 and 700°C is attributed to the decomposition of functional groups, such as the carboxyl groups in the  
27 EDTA-functionalized particles. However, the thermal stability for the ION@Si and ION@Si@NH<sub>2</sub> particles is higher  
28 and leaves more residual<sup>6,7</sup>. The overall EDTA content is calculated to be 15.33 wt%. The Fourier-transform infrared  
29 spectroscopy (FT-IR) data in Fig. S1 e confirms the different particle functionalization steps. The Fe-O-Si vibration, at

30 592  $\text{cm}^{-1}$ , verifies the binding of the silica to the ION core, besides the Si-O-Si stretching vibration at 1080  $\text{cm}^{-1}$  and  
31 the Si-O vibration at 794  $\text{cm}^{-1}$ . The successful functionalization with amine groups is confirmed by the 1488 and 1638  
32  $\text{cm}^{-1}$  bands, which are associated with the N-H bending vibrations. This is indicative of the effective incorporation of  
33 (3-aminopropyl)triethoxysilane (APTES) <sup>8</sup>. EDTA was coated on the particles in an additional synthesis step, proved  
34 by the asymmetrical and symmetrical  $\text{COO}^-$  stretching vibrations, located at 1570 – 1610  $\text{cm}^{-1}$  and 1350 – 1450  $\text{cm}^{-1}$ ,  
35 respectively. Furthermore, the disappearance of the N-H bending vibration at 1488  $\text{cm}^{-1}$  signifies the successful  
36 establishment of an amide bond between APTES and EDTA <sup>5</sup>. The primary particle size from TEM analysis is 120.62  
37  $\pm$  16.86 nm (compare Fig. S1 c and f). After synthesis, dynamic light scattering (DLS) measurements yielded a  
38 hydrodynamic diameter of 2297.67  $\pm$  850.78 nm ( $\zeta$  = -8.22  $\pm$  0.39 mV, pH = 9.1) for the ION@Si@NH<sub>2</sub> precursor and  
39 179.01  $\pm$  11.99 nm ( $\zeta$  = 21.10  $\pm$  0.62 mV, pH = 4.89) for the EDTA-functionalized MNPs with a polydispersity index of  
40 0.17  $\pm$  0.02 (Fig. S1 g). These results confirm the colloidal stability and monomodal distribution of the built  
41 ION@Si@EDTA particles (the term 'Si' refers to the silica shell). Moreover, the hydrodynamic diameter can be  
42 preserved for up to 28 days, as confirmed by an agglomeration study in Fig. S1 h. After this period, the hydrodynamic  
43 diameter measures 176.60  $\pm$  32.55 nm, with a zeta potential of -29.26  $\pm$  1.53 mV. The EDTA-functionalized surface  
44 enables the binding of the linker-protein via the His-tag. Therefore, this surface is saturated with nickel (Ni) ions,  
45 achieving maximum binding capacities of 4  $\text{mg}_{\text{Ni}} \text{g}_{\text{Particle}}^{-1}$  (Fig. S1 i), corresponding to 0.40 nickel ions per  $\text{nm}_{\text{Particle}}^2$ ,  
46 comparable to Fraga García et al. <sup>9</sup>. The binding kinetics occurs within minutes (Fig. S1 j), a finding that aligns with  
47 previous studies <sup>10</sup>, and the built EDTA@Ni chelate complex remains sTab. for several days, as no leaching was  
48 observed over five days, presented in Fig. S1 k. The differences in surface charges for the precursor particles, the  
49 EDTA-functionalized ones, and those saturated with nickel ions were investigated across a range of pH values in  
50 water. The study reveals that the isoelectric point (IEP) of the ION@Si@NH<sub>2</sub> is at pH 8.8 and decreases to pH 6.8 for  
51 the ION@Si@EDTA (Fig. S1 l). The presence of loaded nickel ions slightly reduces the IEP to 7.6. The particles are  
52 colloiddally sTab. ( $\zeta > |20 \text{ mV}|$ ) or unsTab. ( $-20 \text{ mV} < \zeta < 20 \text{ mV}$ ), dependent on the surface charge. This pH-dependent  
53 agglomeration directly influence the sedimentation velocity under a magnetic field gradient, as Fig. S1 m shows.  
54 After the synthesis, the ION@Si@NH<sub>2</sub> particles exhibit a pH of 9.1, close to the IEP, leading to rapid sedimentation  
55 in a magnetic field. Contrarily, the ION@Si@EDTA exhibit a pH of 4.89 after the synthesis, further away from their  
56 IEP. Thus, they tend to sediment slower in the magnetic field due to the formation of smaller agglomerates. These  
57 findings emphasize the importance of carefully controlling the agglomeration behavior for the subsequent  
58 magnetophoretic fractionation process (Fig. S1 m).  
59 Consequently, adhering to the established particle dispersion protocol for each iteration of fractionation  
60 experimentation became important. The MNPs, stored at a temperature of 4°C under a nitrogen atmosphere,  
61 exhibit an average hydrodynamic diameter of 306.73  $\pm$  7.96 nm, accompanied by a polydispersity index of 0.32  $\pm$   
62 0.1. The MNP solution underwent ultrasonication to disperse larger agglomerates, which diminishes the  
63 polydispersity index to 0.13  $\pm$  0.01 ( $d_{\text{hyd}} = 226.57 \pm 0.85 \text{ nm}$ ). Consequently, this ensures the use of relatively  
64 homogeneous agglomerates for the nickel loading procedure. Post-loading, the hydrodynamic diameter increases  
65 to approximately 1000 nm, as the nickel-loaded MNPs were resuspended in the buffer utilized for incubation with  
66 the linker-protein labeled yeast cells (Fig. S1 n).  
67 The process of nickel loading is notably time-intensive, requiring approximately two hours. Improving the efficiency  
68 of the overall labeling process, we assessed whether nickel-loaded particles could be stored for several days and  
69 subsequently reused. This involved monitoring the hydrodynamic diameter over a three-day storage period, as  
70 presented in Fig. S1 o. The findings suggest that a consistent hydrodynamic diameter cannot be maintained, as the  
71 distribution indicates a reduction in the hydrodynamic diameter throughout the storage. This reduction implies that,  
72 even during storage, the agglomerates persist in achieving equilibrium within the buffer. Consequently, the MNPs  
73 were freshly loaded with nickel ions before each fractionation iteration to ensure reproducible agglomerate sizes.

## 74 **B Homogenization of magnetically labeled yeast cells during the fractionation process**

75 Maintaining a consistent sample concentration flowing into the chip was essential in optimizing the magneto-  
76 responsive fractionation process since the magnetic dipole moment modulates the magnetophoretic force acting  
77 upon the labeled yeast cells. This moment is a function of both concentration and agglomerate size, as described in  
78 Equation 1.

79 The brewing industry uses top-fermenting *S. cerevisiae* and bottom-fermenting *S. pastorianus* var. *carlsbergensis*  
80 for ale and lager production. Both encompass strains that flocculate or remain single cells, influencing the aroma  
81 formation and reuse after fermentation <sup>11,12</sup>. The sedimentation rate is primarily influenced primarily by the strain's  
82 flocculation characteristics within minutes <sup>13,14</sup>. Although the strain employed in this study was non-flocculating,  
83 consisting mainly of single cells or budding cells (Fig. 2 f and S2 a), they tend to sediment due to buoyancy depending  
84 on the cell (agglomerate) size (Fig. S2 c). Fig. S3 a illustrates the sedimentation behavior of yeast cells within the  
85 sample syringe before their introduction into the millifluidic chip. Notably, the yeast cells exhibited significant  
86 sedimentation without any dispersion method (when stationary). Within 20 min, they settled to a proportion 0.27  
87  $\pm$  0.4 of their initial count, leading to an inhomogeneous yeast concentration during fractionation. Consequently, a

88 dispersion method ensures homogeneity throughout the fractionation process. Previous studies used conventional  
 89 magnetic stirring for mixing inside the syringe <sup>15</sup>, but these were incompatible due to their interference with the  
 90 MNPs. Therefore, alternative dispersion strategies were explored. This sphere facilitates direct mechanical mixing  
 91 inside the syringe when positioned on the rocker. As presented in Fig. S3 a, the shaker does not significantly mitigate  
 92 sedimentation, especially at higher mixing rates, which induce turbulences within the chip. The rocker results in  
 93 better dispersion but with variability, maintaining between a proportion of  $0.77 \pm 0.12$  and  $0.93 \pm 0.15$  of the initial  
 94 yeast cell number. Introducing a sphere within the syringe emerges as the most efficient approach. With the sphere's  
 95 motion enabled by the teetering rocker, yeast cells are uniformly dispersed, preserving nearly 100 % of the initial  
 96 yeast cell number. Importantly, this method does not comprise yeast cell viability:  $75 \pm 14$  % of cells are viable pre-  
 97 insertion and  $79 \pm 13$  % post-process without affecting the magnetic labeling of the yeast cells (Fig. S2 b).

### 98 C Magnetic aggregation and magnetically induced convection

99 The cooperative motion is quantified by the dimensionless aggregation parameter  $N^*$ , influenced by the volume  
 100 fraction  $\Phi_0$  in the solution, and the magnetic coupling parameter  $\Gamma$  are expressed as,

$$N^* = \sqrt{\Phi_0 e^{(\Gamma-1)}} \quad (S1)$$

$$\Gamma = \frac{\mu_0 m^2}{2\pi d^3 k_B T} \quad (S2)$$

101 Where  $\mu_0 = 4\pi \times 10^{-7} \text{ N A}^{-2}$  is the magnetic permeability of free space,  $\vec{m}$  the induced magnetic dipole moment,  $k_B$   
 102  $= 1.38 \times 10^{-23} \text{ J K}^{-1}$  is Boltzmann's constant and  $T$  is the absolute temperature. In line with the observations made by  
 103 Leong et al., field-induced self-assembly of particles is not anticipated to occur when the dimensionless aggregation  
 104 parameter is  $N^* < 1$ .

105 The second dimensionless number, characterizing the motion of MNPs when exposed to a magnetic field, is the  
 106 magnetic Grashof number  $Gr_m$ , accounting for the induced convective motion, given by

$$Gr_m = \frac{\rho \nabla B \left( \frac{\partial M}{\partial c} \right)_H (c_s - c_\infty) L_c^3}{\eta^2} \quad (S3)$$

107 Where  $\rho$  is the density,  $\vec{\nabla} B$  is the magnetic field gradient within the magnetic field strength  $B$ ,  $M$  is the volumetric  
 108 magnetization in solution.  $c_s$  is the particle concentration at the collection plane,  $c_\infty$  is the one in the sample  
 109 solution, and  $L_c$  is the characteristic length of the system subjected to magnetophoresis, which is the chip width in  
 110 our case.

111 The following values are used for calculating the magnetic Grashof number and the aggregation parameter:

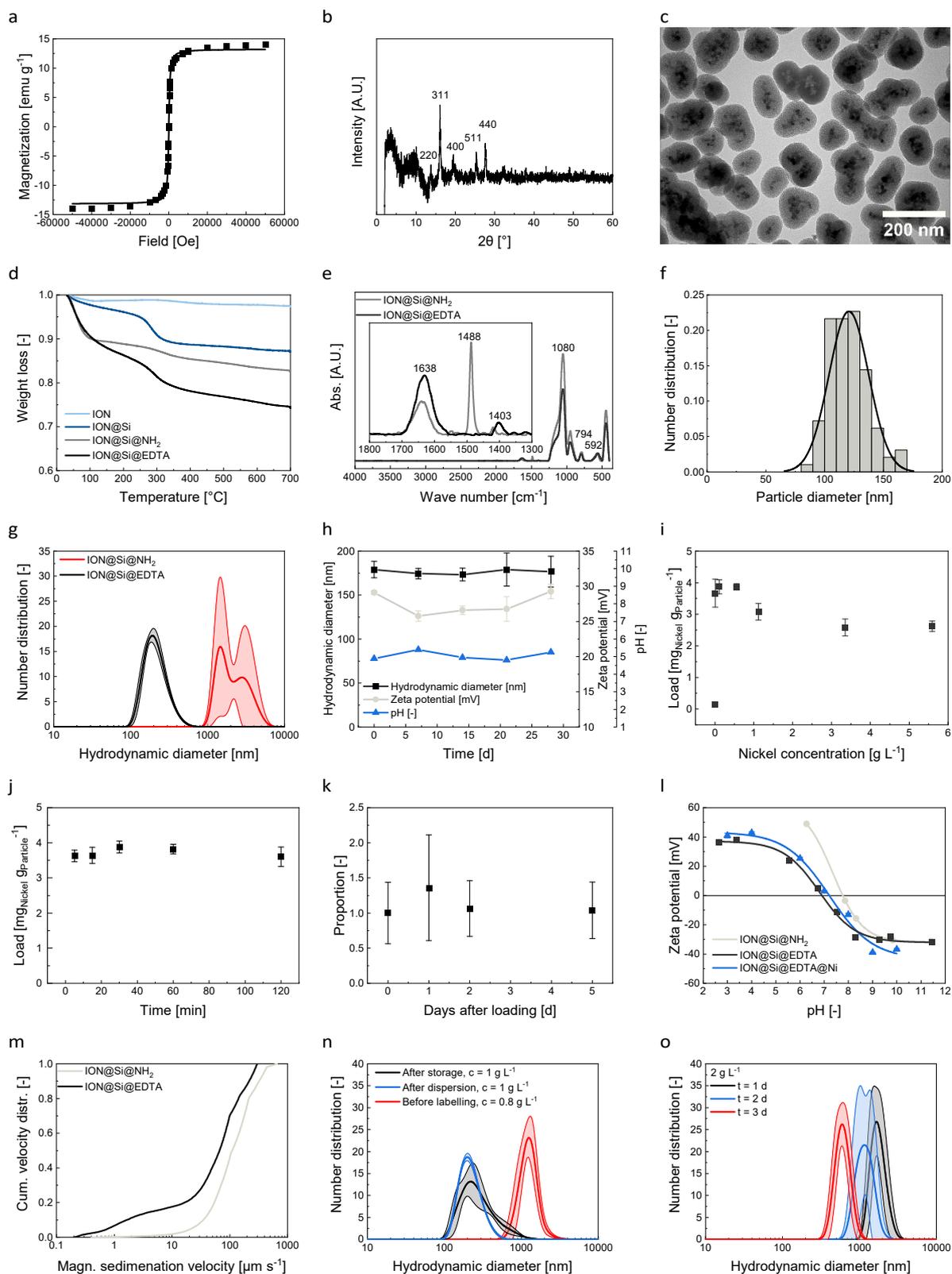
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Magnetization at 0.38 T [A m <sup>-1</sup> ]	28450.88
Particle diameter $d$ [m]	9.68x10 <sup>8</sup>
Particle volume [m <sup>3</sup> ]	4.75x10 <sup>22</sup>
Volume all particles in system [m <sup>3</sup> ]	3.89x10 <sup>6</sup>
Density particles, 22°C [g L <sup>-1</sup> ]	2800.80
Density water, 22°C [g L <sup>-1</sup> ]	997.77
Density system $\rho$ [g L <sup>-1</sup> ]	998.03
Particle volume fraction $\phi$	0.00015
Coupling parameter $\Gamma$	1.36x10 <sup>28</sup>
Magnetic field gradient at collection surface $\nabla B$ [T m <sup>-1</sup> ]	4.82
Volumetric magnetization of solution $M$ [(A m <sup>-1</sup> ) (g L <sup>-1</sup> ) <sup>-1</sup> ]	10.16
Particle concentration at collection plane $c_s$ [g L <sup>-1</sup> ]	0
Particle concentration of bulk solution $c_\infty$ [g L <sup>-1</sup> ]	0.4
Characteristic length of system $L_c$ [m]	0.00313
Viscosity $\eta$ [kg ms <sup>-1</sup> ]	0.00010
Chip volume [m <sup>3</sup> ]	0.02721

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114

115 **B Figures and Tables**

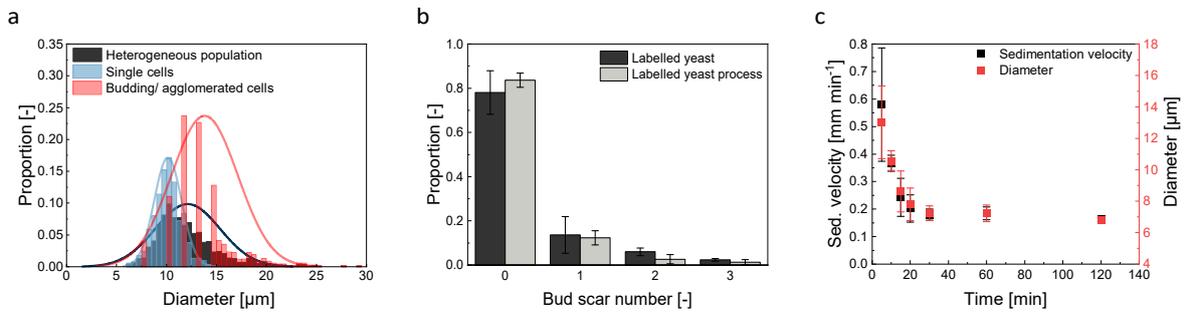


116 Fig. S1 Superconducting quantum interference device (SQUID) measurement (a), X-ray diffractogram (b), and transmission electron microscopy (TEM) images (c)  
 117 of the final ethylenediaminetetraacetic acid (EDTA)-functionalized silica-coated IONs. Thermogravimetric analysis (d) of the different coating steps (the term 'Si'  
 118 refers to the silica shell). Fourier-transform infrared (FT-IR) spectroscopy analysis of the precursor 3-aminopropyltriethoxysilan (APTES)-coated and EDTA-  
 119 functionalized IONs in black (e). Number distribution over particle diameter derived from ImageJ analysis of the TEM images for  $n = 100$  MNPs (f). Dynamic light  
 120 scattering measurement (DLS) of the precursor APTES-coated ( $\text{pH} = 9.1$ ) and final EDTA-functionalized IONs ( $\text{pH} = 4.89$ ) after synthesis in water for  $c = 1 \text{ g L}^{-1}$  (g).  
 121 Hydrodynamic diameter, zeta potential, and pH for the MNPs (the term MNP refers to the final EDTA-functionalized silica shell iron oxide nanoparticles) up to  
 122 28 days, stored under nitrogen at  $4^\circ\text{C}$  (h). Nickel binding isotherm to MNPs for various nickel concentrations ( $c_{\text{particels}} = 1 \text{ g L}^{-1}$ ) (i). Binding kinetics for nickel binding  
 123 ( $c_{\text{Nickel}} = 0.59 \text{ g L}^{-1}$ ) to the MNPs ( $c_{\text{particels}} = 1 \text{ g L}^{-1}$ ) (j). The nickel leaching study shows the proportion of the initial nickel load after several days of storage at  $4^\circ\text{C}$

124 (k). Zeta potential for the precursor APTES-coated and final EDTA-functionalized IONs and those saturated with nickel ions for different pH values ( $c_{\text{Particles}} = 1 \text{ g L}^{-1}$ ) (l). Cumulative velocity distribution over the magnetophoretic sedimentation velocity for the precursor APTES-coated (pH = 9.1) and final EDTA-functionalized  
 125  $(\text{pH} = 4.89)$  IONs in water after synthesis ( $c_{\text{Particles}} = 1 \text{ g L}^{-1}$ ) (m). DLS measurement of the MNP solution in water after storage (black), after dispersion (blue) before  
 126 the nickel loading, and before labeling, but after nickel loading (red) in the necessary concentrations of the protocol (n). Number distribution of the DLS  
 127 measurement for a storage study of the nickel-loaded MNPs in buffer over three days (o). Error bars or shaded areas represent the standard deviation of a  
 128 triplicate measurement.  
 129

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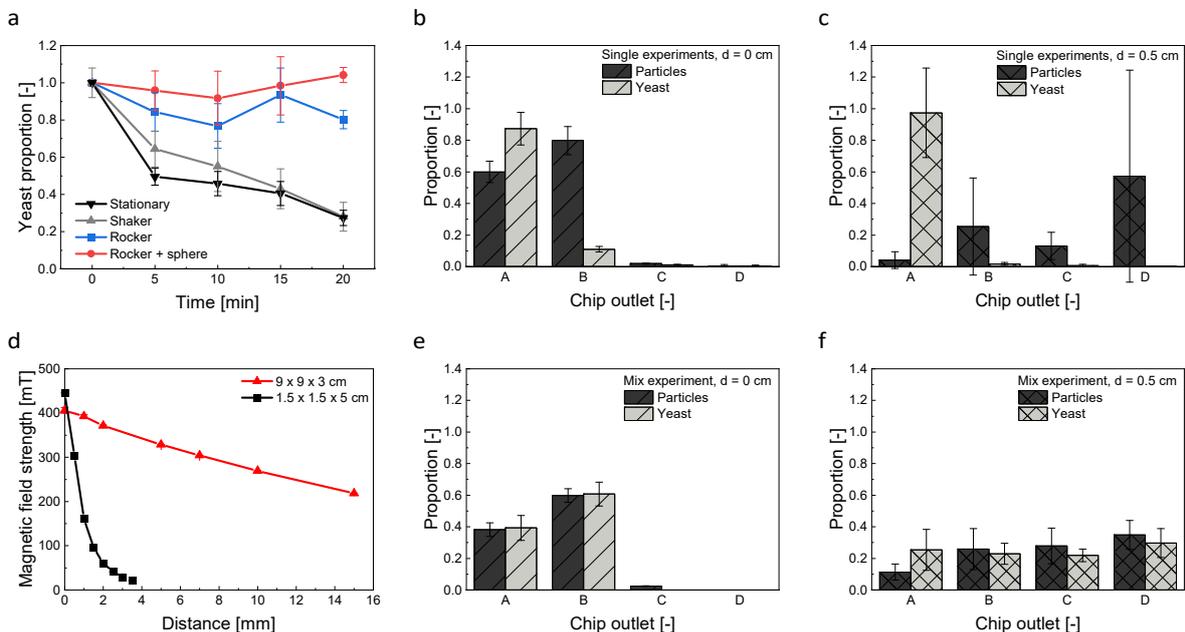
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132 Fig. S2 Yeast cell diameter of a heterogeneous population derived from ImageJ analysis of  $n = 1168$  cells containing 45 % single cells and 55 % budding/  
 133 agglomerated cells (a). Age distribution of a magnetically labeled heterogeneous yeast culture derived from cytometric analysis ( $n_{\text{min}} = 20000$  counts,  $n = 3$ , error  
 134 bars representing the standard deviation of the triplicate measurement) before the process (black) and in during the process with the dispersion via the rocker  
 135 and the sphere (b). Sedimentation velocity and corresponding yeast cell (agglomerate) diameter over time at 410 nm (c). Error bars represent the standard  
 136 deviation of a triplicate measurement.

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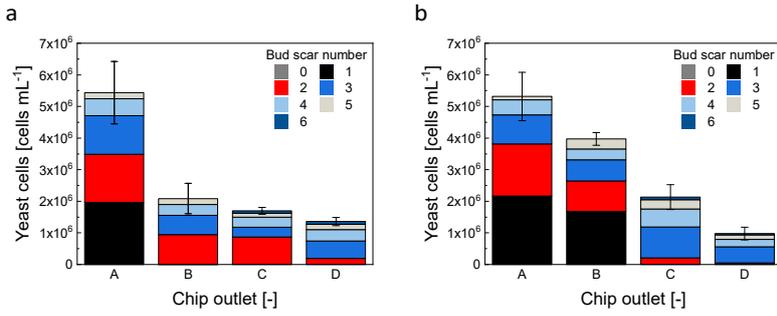


139 Fig. S3 Yeast proportion of the initial cell number of  $1.58 \times 10^7 \text{ cells mL}^{-1}$  in sample syringes over time for different dispersion methods using no dispersion method  
 140 (stationary), a shaker, a rocker, and an inserted sphere in the sample syringe positioned on the rocker to avoid sedimentation (a). Single fractionation experiments  
 141 for solely yeast and particles (b) and (c) and for yeast-particle mixture without linker-protein (e) and (f) in the rectangular geometry. Chip outlet A was the furthest  
 142 away from the magnet; chip outlet D was nearest to the magnet.  $V_{\text{Sample}} = 220 \mu\text{L min}^{-1}$ ,  $V_{\text{Buffer}} = 1840 \mu\text{L min}^{-1}$ , the magnet distance was 0 cm (a) and (d).  $V_{\text{Sample}}$   
 143  $= 220 \mu\text{L min}^{-1}$ ,  $V_{\text{Buffer}} = 920 \mu\text{L min}^{-1}$ , the magnet distance was 0.5 cm,  $c_{\text{Particle}} = 0.4 \text{ g L}^{-1}$ , cell number  $= 1.59 \times 10^7 \text{ cells mL}^{-1}$  (b) and (e). Measured magnetic field  
 144 strength using an AC/DC magnetometer (PCE Instruments) over the distance from the magnet surface at the middle for the used magnets ( $9 \times 9 \times 3 \text{ cm}$  for the  
 145 rectangular and  $1.5 \times 1.5 \times 5 \text{ cm}$  for the trapezoidal geometry) (d). All data plots include the standard deviation from a triplicate measurement.

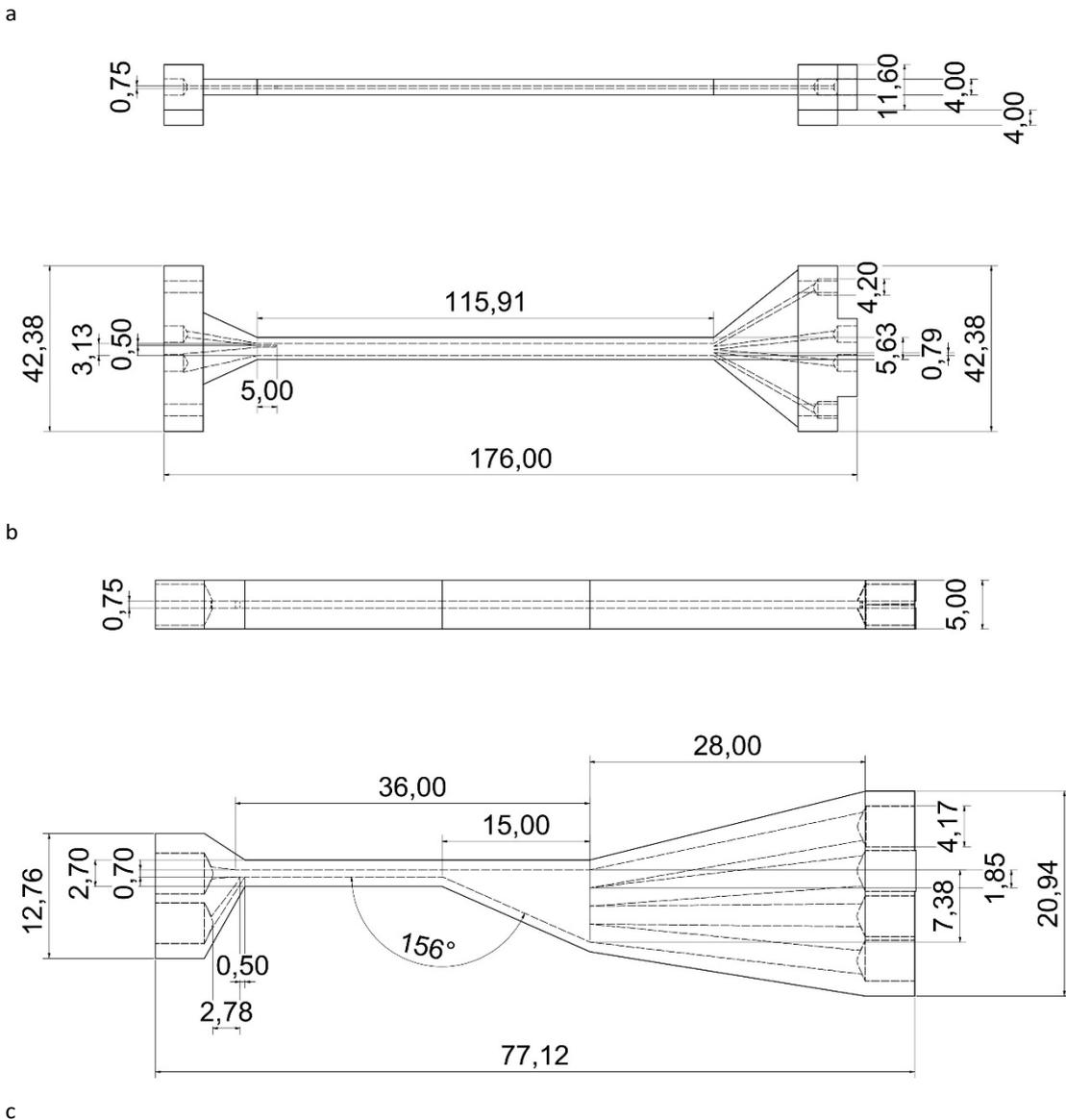
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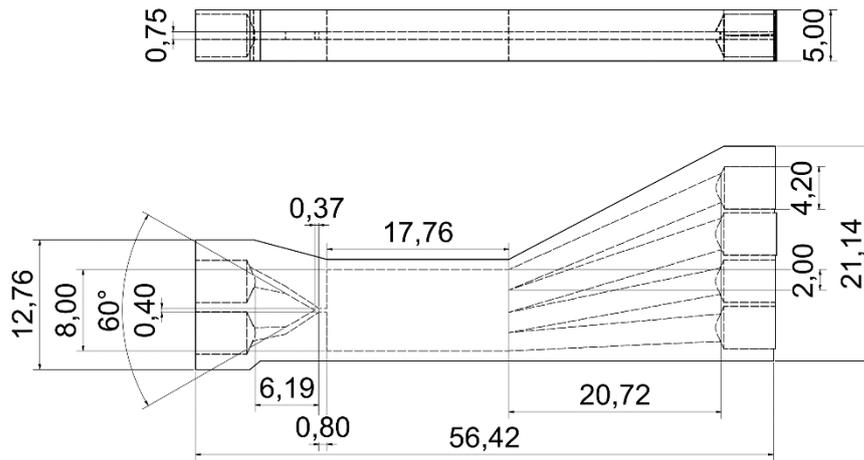
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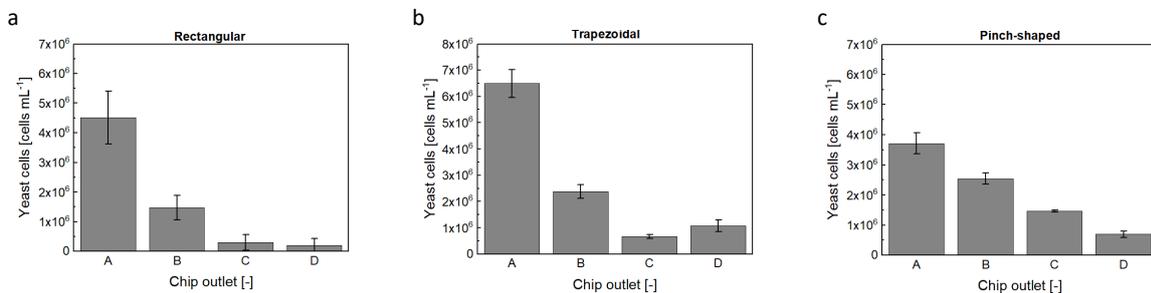


149 Fig. S4: Reproduction of the yeast cell fractionation with the pinch-shaped geometry. Yeast cell concentrations for the different fractions and corresponding bud  
 150 scar numbers are shown.  $C_{\text{particle}} = 0.4 \text{ g L}^{-1}$ , cell number =  $1.59 \times 10^7 \text{ cells mL}^{-1}$ . Error bars represent the standard deviation of the triplicate measurement. Outlet A  
 151 was the furthest away from the magnet; outlet D was the nearest to the magnet. Daughter and mother cells refer to the prior batch separation.





152 Fig. S5 Technical drawings in [mm] of the rectangular (a), trapezoidal (b), and pinch-shaped (c) chip.



153 Fig. S6: Yeast cell concentration for different fractionations with different flow channels (rectangular (a), trapezoidal (b), pinch-shaped (c)). The yeast cells were not labeled with linker-protein or MNPs. Error bars represent the standard deviation of the triplicate measurement. Outlet A was the furthest away from the magnet; outlet D was the nearest to the magnet.

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154

155 Tab. S1: Separation selectivities for the different chip geometries derived from the cumulative bud scar number distribution related to the cell number in each  
156 chip outlet. The rectangular and trapezoidal geometry fractionation was performed once, and the pinch geometry process was conducted three times.

Chip outlet	Bud scar number n	Rectangular	Trapezoidal	Pinch
A	≥ 1	0.48	0.48	1.00 ± 0.00
B	≥ 2	0.94	0.95	0.95 ± 0.02
C	≥ 3	0.96	0.40	0.97 ± 0.00
D	≥ 4	0.66	0.34	0.84 ± 0.08

157

## 158 C Methods

### 159 A Comparison between microscopic determined bud scar distribution and flow cytometric analysis.

160 To ensure the reliability of flow cytometric results on bud scar distribution in heterogeneous yeast populations and  
161 associated outlet distributions, we proposed two distinct measurement methodologies, described earlier in detail  
162 <sup>1,16,17</sup>;

163

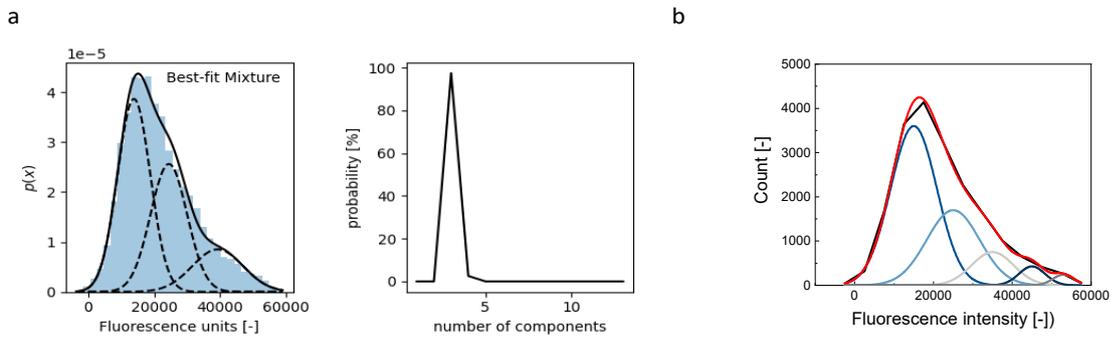
#### 164 1. Microscopy-based approach:

165 This involved manual counting of bud scars via confocal and light microscopy and subsequent predictions using the  
166 Bayes theorem. The method followed the protocol by Eigenfeld et al. <sup>17</sup>, which postulates that only 40% of the yeast  
167 cell surface was observable through microscopy. Consequently, predictions had to be made for the remaining 60%  
168 of the cell surface where bud scars were unobservable.

#### 169 1. Flow cytometric approach <sup>16</sup>:

170 a. Outlier elimination using the R package 'mvoutlier'.

- 171 b. Autofluorescence predictions via the Random Forest algorithm.  
 172 c. Single-cell autofluorescence subtraction from each yeast cell.  
 173 d. Analysis of the resultant fluorescence signals via Gaussian histogram curve fitting (Fig. S6).  
 174 Notably, each Gaussian curve corresponds to an age cluster, symbolizing cells with an identical  
 175 count of bud scars.  
 176 This comparative study aids in validating the flow cytometric determinations against a microscopy-based reference.  
 177 The results showed no noTab. differences between the two measurement methods in terms of heterogeneous  
 178 populations (p-value = 0.9375), mother cells (p-value = 0.8125), and daughter cells (p-value = 0.5781) when a  
 179 Wilcoxon paired test was employed. Furthermore, through the automated Gaussian mixture analysis provided by  
 180 the R package 'GaussianMixture', we found p-values of 1 for both heterogeneous populations and mother cells and  
 181 a p-value of 0.4609 for daughter cells.



182 Fig. S7 Gaussian mixture analysis using R package 'GaussianMixture' (a) and using OriginLabs 2021 (R1 = 0.9995) (b) for a heterogeneous population.

### 183 B Primer sequences for quantitative polymerase chain reaction experiments

184 Tab. S2: Primer sequences of ones used to amplify the housekeeping genes.

TAF10	forward primer	TAACAACAGTCAGGCGAGAG
	reverse primer	CACCGTCAGAACAACCTTGC
KRE11	forward primer	ATTCGCCCTTGACACTGG
	reverse primer	CTCTCGGAGGTACAACCTG
UBC6	forward primer	ATGCGGCAAATACAGGTGATG
	reverse primer	TTGTTCAGCGCGTATTCTGTC

185

186 Tab. S3: Primer sequences of the tested genes.

Gene		Primer sequence
MEP2	forward primer	ACGAGGAATCCACTGCTTAC
	reverse primer	CGTCTGTGTTACCCACAATC
HSP104	forward primer	TAACTCAAGAGGCCAAGGAC
	reverse primer	TCCTTAGTGCCAGTTTGTC
HXK2	forward primer	GGCTGCCAATGCTTTGAAGG
	reverse primer	ACCGGAACCATCTTCAGCAG
HSP12	forward primer	CTCTGCCGAAAAGGCAAGG
	reverse primer	GACGGCATCGTTCAACTTGG
PHO5	forward primer	TTCAACATCACCTTGACAGAC
	reverse primer	ATTGGCATCGTAGTCCCAAG
ADH1	forward primer	GTGCTCACGGTGCATCAAC
	reverse primer	GCATACCGACAAAACGGTG
GCR1	forward primer	CAAACAACGACTCCACTAC
	reverse primer	ATCATTGGGCTCCGACTTAC

187

188 **C Minimum information for publication of quantitative real-time polymerase chain reaction experiments guidelines**

189 Tab. S4: Minimum information for publication of quantitative real-time polymerase chain reaction experiments (MIQE) checklist for authors, reviewers, and editors. All essential information (E) must be submitted with the manuscript. Desirable information  
190 (D) should be submitted if available. If using primers obtained from RTPrimerDB, information on quantitative polymerase chain reaction (qPCR) target, oligonucleotides, protocols, and validation was available from that source.

ITEM TO CHECK	IMPORTANCE	CHECKLIST
<b>EXPERIMENTAL DESIGN</b>		
Definition of experimental and control groups	E	control group: Sample A, youngest fraction
Number within each group	E	3 Biological replicates
Assay carried out by core lab or investigator's lab?	D	
Acknowledgement of authors' contributions	D	
<b>SAMPLE</b>		
Description	E	Yeast cells, separated for cell age
Volume/mass of sample processed	D	
Microdissection or macrodissection	E	not applicable
Processing procedure	E	After separation, cells were submerged in 1/3 vol. 5% phenol in abs. ethanol and frozen
If frozen - how and how quickly?	E	
If fixed - with what, how quickly?	E	5 % Phenol in absolute Ethanol, 1:3 with liquid sample, directly after cell fractionation
Sample storage conditions and duration (especially for FFPE samples)	E	Samples were stored at -80°C in 2/3 separation buffer and 1/3 vol. 5% phenol in abs. ethanol
<b>NUCLEIC ACID EXTRACTION</b>		
Procedure and/or instrumentation	E	Enzymatic cell lysis with subsequent spin column extraction
Name of kit and details of any modifications	E	Roboklon Universal RNA Kit, Roboklon GmbH, Berlin, Germany
Source of additional reagents used	D	
Details of DNase or RNase treatment	E	None, as per protocol
Contamination assessment (DNA or RNA)	E	Denaturing RNA gel electrophoresis for degradation, no-reverse-transcription control in the qPCR
Nucleic acid quantification	E	Sheet_3
Instrument and method	E	Nanodrop 2000c spectrophotometer
Purity (A260/A280)	D	
Yield	D	
RNA integrity method/instrument	E	MOPS-Gel electrophoresis
RIN/RQI or Cq of 3' and 5' transcripts	E	not applicable
Electrophoresis traces	D	
Inhibition testing (Cq dilutions, spike or other)	E	not applicable due to small sample volume

REVERSE TRANSCRIPTION		
Complete reaction conditions	E	Thermo Scientific™ Maxima H Minus Reverse Transcriptase, according to manufacturer protocol, with 200 U of enzyme per reaction
Amount of RNA and reaction volume	E	1849.5 ng/μL, Samples 6: 810 ng/μL, Samples 7: 483.3 ng/μL; reaction volume 20 μL
Priming oligonucleotide (if using GSP) and concentration	E	NEB Oligo dT VN
Reverse transcriptase and concentration	E	Thermo Scientific™ Maxima H Minus Reverse Transcriptase, 200 U of enzyme per reaction
Temperature and time	E	30 min at 50°C
Manufacturer of reagents and catalog numbers	D	
Cqs with and without RT	D*	
Storage conditions of cDNA	D	
qPCR TARGET INFORMATION		
If multiplex, efficiency and LOD of each assay.	E	not applicable
Sequence accession number	E	Sheet_2
Location of amplicon	D	
Amplicon length	E	Sheet_2
<i>In silico</i> specificity screen (BLAST, etc.)	E	Primer BLAST, <a href="https://www.ncbi.nlm.nih.gov/tools/primer-blast/">https://www.ncbi.nlm.nih.gov/tools/primer-blast/</a>
Pseudogenes, retropseudogenes, or other homologs?	D	
Sequence alignment	D	
Secondary structure analysis of amplicon	D	
Location of each primer by exon or intron (if applicable)	E	only exons targeted
What splice variants were targeted?	E	not applicable
qPCR OLIGONUCLEOTIDES		
Primer sequences	E	Sheet_2
RTPrimerDB Identification Number	D	
Probe sequences	D**	
Location and identity of any modifications	E	not applicable
Manufacturer of oligonucleotides	D	
Purification method	D	
qPCR PROTOCOL		

Complete reaction conditions	E	Biozym Blue S'Green qPCR Mix Separate ROX
Reaction volume and amount of cDNA/DNA	E	10 µL
Primer (probe), Mg++ and dNTP concentrations	E	Primer conc. 400 nM each
Polymerase identity and concentration	E	Manufacturer Master Mix
Buffer/kit identity and manufacturer	E	Biozym Blue S'Green qPCR Mix Separate ROX, Biozym Scientific GmbH
Exact chemical constitution of the buffer	D	
Additives (SYBR Green I, DMSO, etc.)	E	SYBR Green
Manufacturer of plates/tubes and catalog number	D	
Complete thermocycling parameters	E	Initial denaturation: 2 min, 95°C; 45x 5 sec, 95°C, 30 sec, 60°C
Reaction setup (manual/robotic)	D	
Manufacturer of qPCR instrument	E	Roche LightCycler 480 II

#### qPCR VALIDATION

Evidence of optimization (from gradients)	D	
Specificity (gel, sequence, melt, or digest)	E	Melting curve, single peaks for all reactions
For SYBR Green I, Cq of the NTC	E	37.11
Standard curves with slope and y-intercept	E	
PCR efficiency calculated from slope	E	
Confidence interval for PCR efficiency or standard error	D	
r2 of standard curve	E	
Linear dynamic range	E	
Cq variation at lower limit	E	
Confidence intervals throughout range	D	
Evidence for limit of detection	E	
If multiplex, efficiency and LOD of each assay.	E	not applicable

#### DATA ANALYSIS

qPCR analysis program (source, version)	E	LightCycler 480 Software release 1.5.0; qbase+, V3.4
Cq method determination	E	Proprietary Roche LighCycler
Outlier identification and disposition	E	according to qbase+
Results of NTCs	E	

Justification of number and choice of reference genes	E	according to geNorm V and geNorm M
Description of normalization method	E	Modified Pfaffl Method, according to Hellemans et al., implemented in qbase+
Number and concordance of biological replicates	D	
Number and stage (RT or qPCR) of technical replicates	E	technical duplicates in the qPCR
Repeatability (intra-assay variation)	E	not applicable
Reproducibility (inter-assay variation, %CV)	D	
Power analysis	D	
Statistical methods for result significance	E	One-way ANOVA
Software (source, version)	E	Biogazelle qbase+, V3.4
Cq or raw data submission using RDML	D	

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192 \*: Assessing the absence of DNA using a no RT assay was essential when first extracting RNA. Once the sample had been validated as RDNA-free, the inclusion of a no-RT control was  
193 desirable but no longer essential.

194 \*\*: Disclosure of the probe sequence was highly desirable and strongly encouraged. However, since not all commercial pre-designed assay vendors provide this information, it cannot  
195 be an essential requirement. Use of such assays was advised against.

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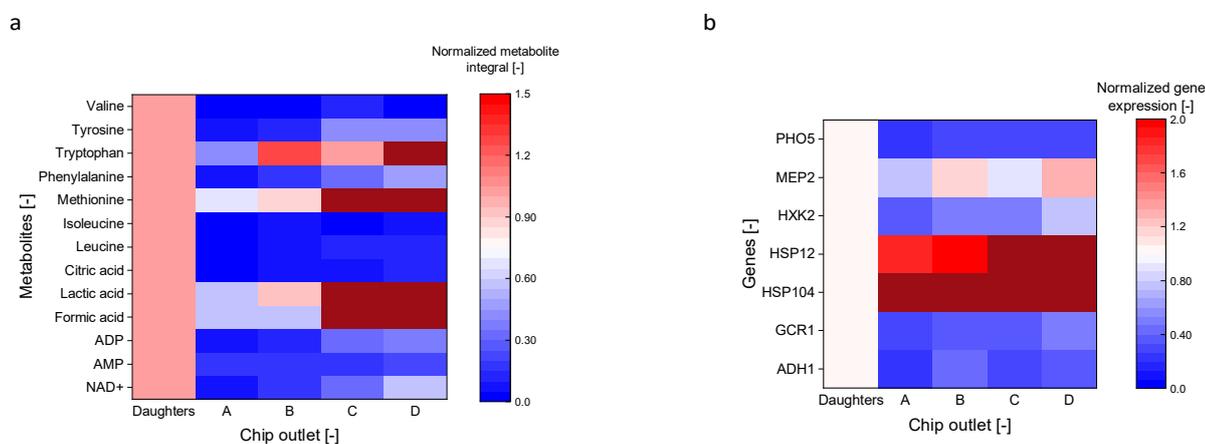
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## 204 D Data analysis of the metabolome and gene expression

205 In Fig. 5, the metabolome and gene expression data are normalized to outlet A of the pinch shaped fractionation, and not to the  
 206 daughter cells. This distinction is essential, as the daughter cells were separated prior to the millifluidic fractionation to avoid the  
 207 magnetically induced co-migration. Consequently, direct comparisons between the separated daughter cells and the older  
 208 fractionated cells are not feasible for two reasons. First, the daughter cells were not fractionated in the chip and, therefore, did  
 209 not undergo 'sorting stress.' Second, they were fixed at varying time points due to the multi-step experimental process.



211 Fig. S8 Heat map depicting the normalized integral area of selected metabolites (a) and gene expression (b) against the cell fraction. Normalized to the daughter cells. Data  
 212 included from a triplicate measurement.

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