

Supporting Information

Metal Decoration of Si Particles via High-Energy Milling for Lithium-ion Battery Anodes

Khryslyn G. Araño^{1}, Beth L. Armstrong², Robert L. Sacchi¹, Matthew S. Chambers¹, Chun-Sheng Jiang³, Joseph Quinn⁴, Harry M. Meyer III¹, Anton W. Tomich⁵, Amanda Musgrove¹, Steven Lam¹, Elena Toups¹, Chongmin Wang⁴, Christopher S. Johnson⁵, Gabriel M. Veith^{1*}*

¹Chemical Sciences Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831,
United States

²Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge,
Tennessee 37831, United States

³Materials Science Center, National Renewable Energy Laboratory, Golden, Colorado 80401,
United States

⁴Environmental Molecular Sciences Laboratory, Pacific Northwest National Laboratory,
Richland, Washington 99354, United States

⁵Chemical Science and Engineering Division, Argonne National Laboratory, 9700 South Cass
Avenue, Argonne Illinois 60439-4837, United States

Corresponding authors email address:

Khryslyn G. Araño: khryslyn.arano@gmail.com

Gabriel M. Veith: veithgm@ornl.gov

XRD Analysis of the Si Powder

Table S1. Starting models used in Rietveld refinement.

Model	Reference
Si	Többers <i>et al.</i> ¹
Ag	Swathi <i>et al.</i> ²
Al	Didier <i>et al.</i> ³
Fe	Kohlhaas <i>et al.</i> ⁴
Ni	Conway & Prior <i>et al.</i> ⁵
Ti	Wasilewski <i>et al.</i> ⁶
Y	Spedding <i>et al.</i> ⁷

Table S2. Fitting parameters for the Rietveld refinement of the metal-decorated Si powders.

Material	R _{wp}	χ^2
Si ^{Ag}	9.996%	2.642
Si ^{Al}	8.520%	2.385
Si ^{Fe}	2.867%	1.323
Si ^{Ni}	3.616%	2.095
Si ^{Ti}	10.137%	2.486
Si ^Y	9.145%	2.933

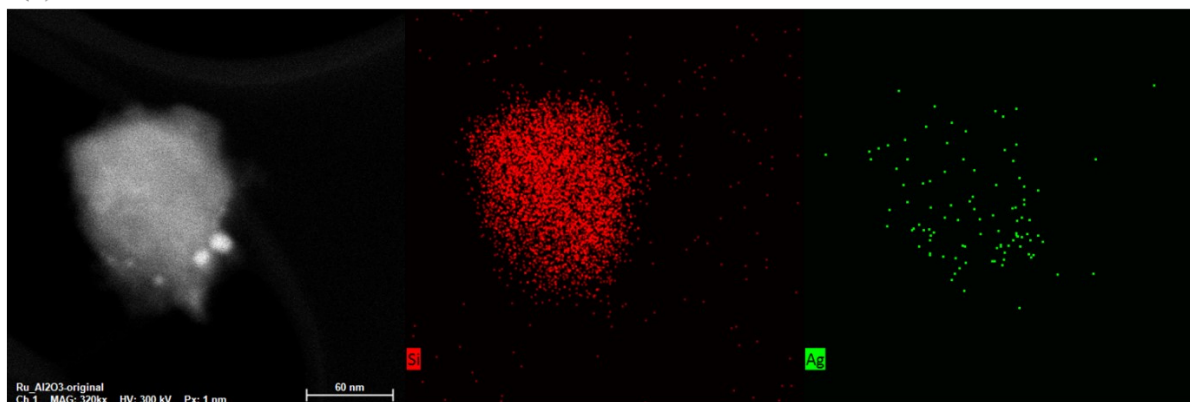
Table S3. Si^M cell parameters.

Sample	Si cell parameter (Å)
Si ^{Ag}	5.4302(3)
Si ^{Al}	5.43059(14)

Si^{Fe}	5.43299(16)
Si^{Ni}	5.4302(3)
Si^{Ti}	5.4330(2)
Si^{Y}	5.4299(3)
Si	5.4311(5)

Scanning Electron Microscopy

(a)



(b)

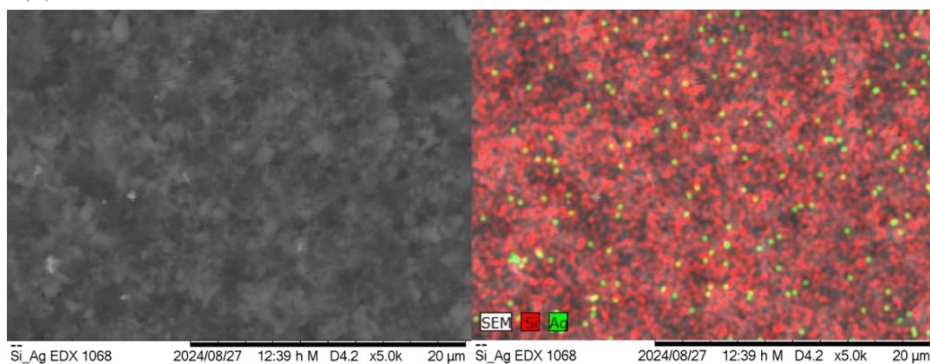


Figure S1. a) TEM images of the Si^{Ag} particle and b) surface SEM-EDX images of the pristine Si^{Ag} electrode.

Scanning Spreading Resistance Microscopy (SSRM) of the Electrodes

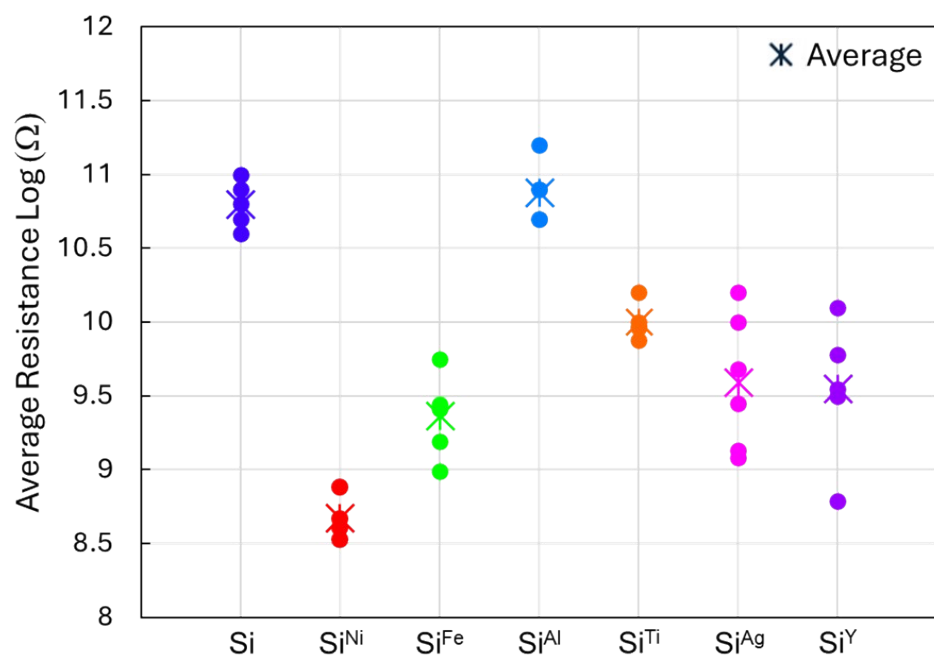


Figure S2. Average resistance of the electrodes prepared from the baseline Si and metal-decorated Si powders obtained from SSRM-AFM measurements.

Electrochemical Tests of the Si^M Electrodes

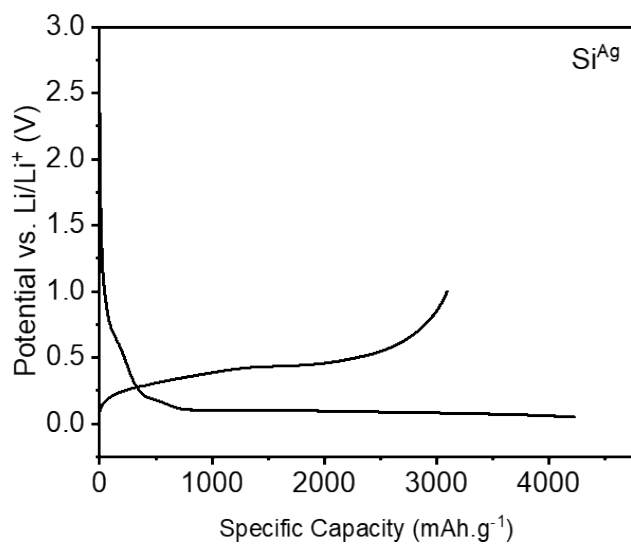


Figure S3. First cycle half-cell charge-discharge profile of Si^{Ag}.

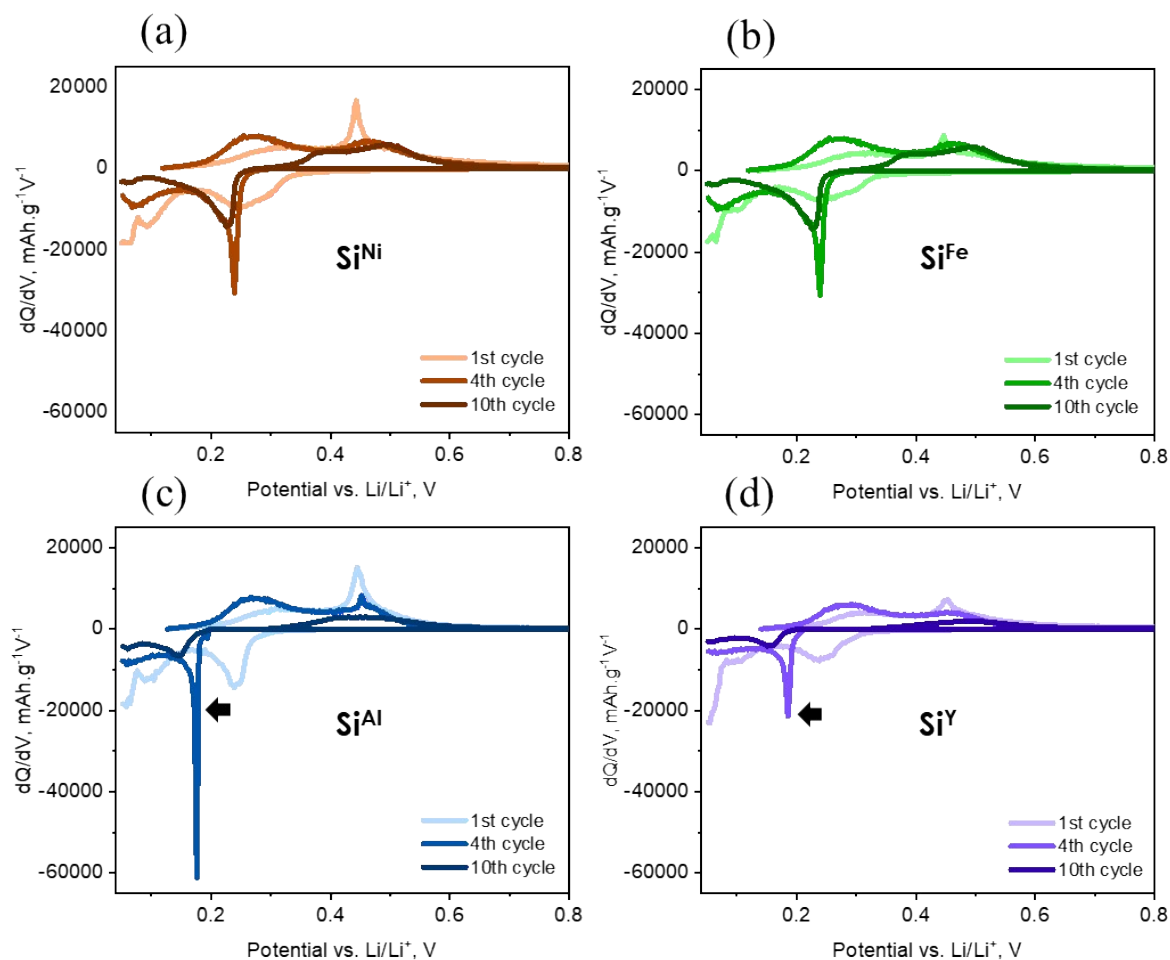


Figure S4. Voltage profiles of a) Si^{Ni} , b) Si^{Fe} , c) Si^{Al} , and d) Si^{Y} during cycling in half-cell configuration.

Table S4. Coulombic efficiencies of the Si^{M} half cells.

	20th Cycle	40th Cycle	60th Cycle	80th Cycle	100th Cycle
Baseline Si	100%	99.8%	99.2%	99.0%	99.7%
Si^{Fe}	100%	100%	100%	99.7%	99.4%
Si^{Ni}	100%	99.8%	99.6%	99.3%	99.3%
Si^{Ti}	100%	98.9%	98.5%	98.2%	98.5%
Si^{Ag}	99.7%	99.7%	99.7%	99.6%	99.5%
Si^{Al}	99.2%	100%	99.9%	99.6%	99.5%
Si^{Y}	99.6%	99.2%	99.4%	99.6%	99.8%

Electrochemical Impedance Spectroscopy Measurements in Half Cells

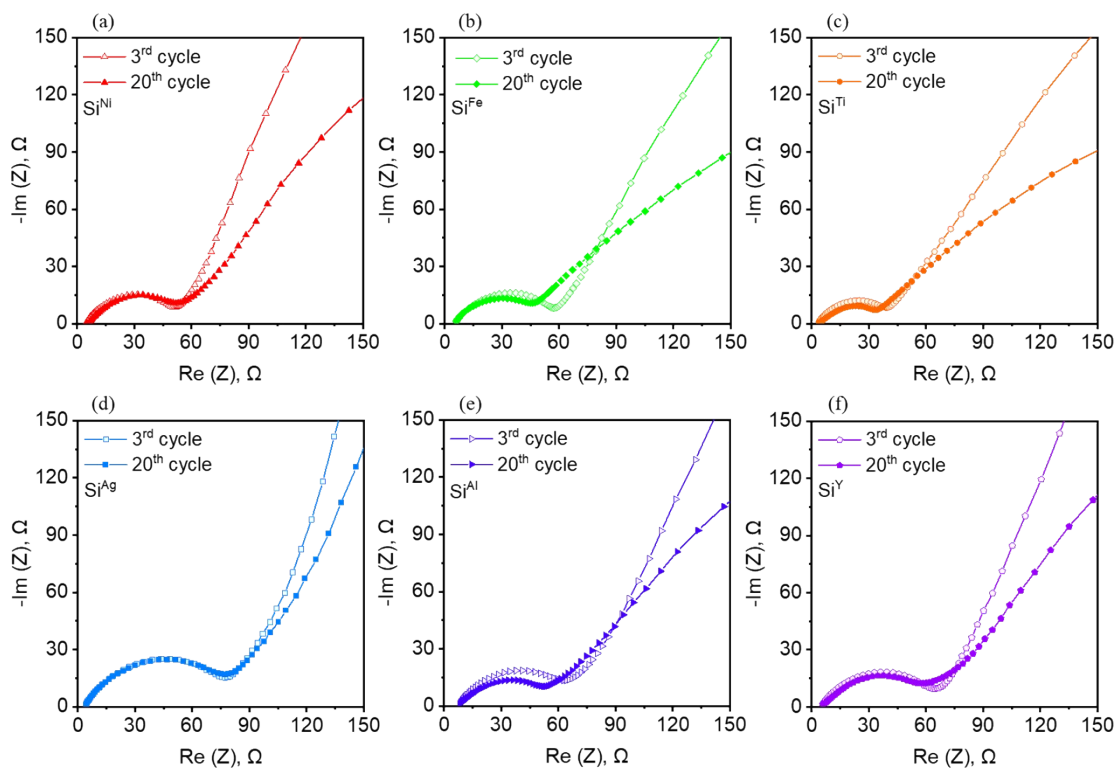


Figure S5. Nyquist profiles of a) Si^{Ni} , b) Si^{Fe} , c) Si^{Ti} , d) Si^{Ag} , e) Si^{Al} , and f) Si^{Y} at delithiated states in half-cell configuration

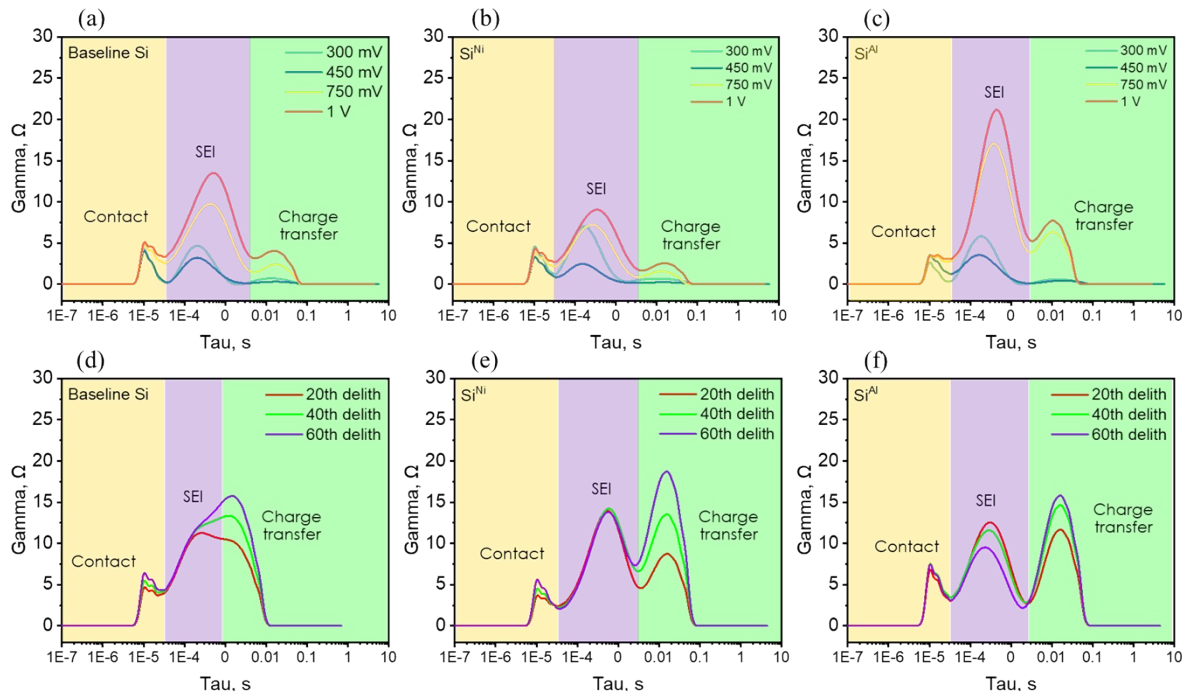


Figure S6. Distribution of relaxation times (DRT) plots at different potentials during the first delithiation of a) baseline Si, b) Si^{Ni} , and c) Si^{Al} , and DRT plots during cycling at delithiated states (delithiated at 1V) of d) baseline Si, e) Si^{Ni} , and f) Si^{Al} .

XPS Analysis of the Electrodes

Table S5. Surface composition (at %) of the baseline Si electrode at different lithiation and delithiation potentials

	Li	F	O	C	Si	P	N
250 mV	38.5	29.2	18.7	11.5	1.1	0.8	0.1
100 mV	32.8	35.2	9.6	17.9	3	1	0.6
50 mV	37.1	35.9	11	12.9	2	0.9	0.2
300 mV	36.2	36.7	9.8	14.5	1.7	0.8	0.3
450 mV	37.7	35.1	13.2	12.8	0	0.8	0.3
750 mV	34.9	20.1	25	18.9	0	0.8	0.1
1.5 V	36	23	21.5	18.3	0	0.9	0.1

Table S6. Surface composition (at %) of the Si^{Ni} electrode at different lithiation and delithiation potentials

	Li	F	O	C	Si	P	N	Ni
250 mV	25.9	28.9	13.8	22.6	3.8	3.4	1.4	0.1
100 mV	31.6	35.5	11.6	16.1	2.1	2.4	0.6	0.1
50 mV	34.7	19.7	24.8	19.6	0.1	0.7	0.1	0
300 mV	35.6	26.6	18.5	17.6	0.3	1.2	0.2	0

450 mV	35.5	22.5	22.1	18.9	0	0.7	0.2	0
750 mV	37.3	23.8	20.7	17.2	0	0.8	0.1	0
1.5 V	36.5	21.7	22.7	18.2	0	0.8	0.1	0

Table S7. Surface composition (at %) of the Si^{Al} electrode at different lithiation and delithiation potentials

	Li	F	O	C	Si	P	N	Al
250 mV	24.3	30.8	12.2	26.8	2.6	0.9	1.3	1.2
100 mV	37.1	38.7	8.7	12.5	2.2	0.0	0.6	0.3
50 mV	38.0	33.8	14.9	12.3	0.1	0.0	0.7	0.1
300 mV	36.1	32.9	12.9	17.0	0.0	0.0	0.7	0.2
450 mV	28.7	14.1	21.0	35.3	0.0	0.5	0.2	0.1
750 mV	36.3	23.6	22.3	17.2	0.0	0.3	0.1	0.2
1.5 V	34.1	16.0	28.2	20.9	0.0	0.5	0.1	0.1

Electrochemical Performance of Si^M in Full Cells

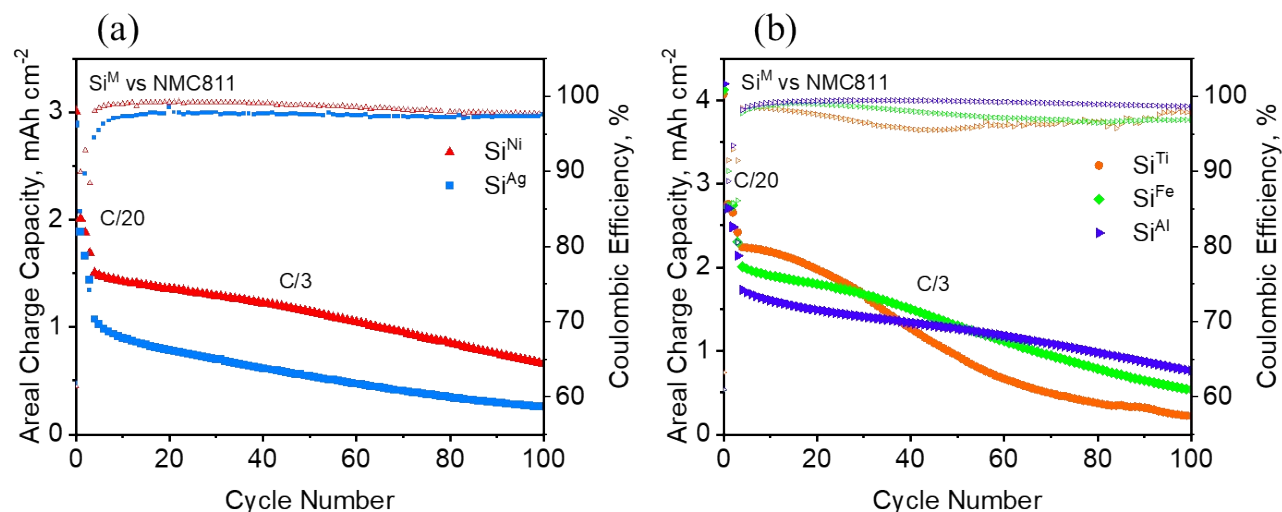


Figure S7. Full cell cycling behavior of select Si^M with NMC811 with an areal loading of a) 2.59 mAh cm⁻² and b) 4.0 mAh cm⁻²

References:

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