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Electronic Supplementary Information for

A gold nanoparticles-loaded MoS₂ nanosheets with peroxidase-like and pyranose oxidase-like activities for bio-enzyme-free visual detection of glucose, xylose and galactose

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1. Main materials and apparatus

Pure xylose, glucose, galactose, maltose, fructose, sucrose and sodium citrate $(C_6H_5O_7Na_3 2H_2O)$ were purchased from Sangon Biotech Co., Ltd (Shanghai, China). The sodium borohydride (NaBH₄), acetic acid (CH₃COOH), sodium acetate (CH₃CO₂Na 3H₂O), trichloroacetic acid, acetonitrile and chloroform were obtained from Aladdin Company (Shanghai, China). 3,3',5,5'-Tetramethylbenzidine (TMB) were purchased from Shanghai Titan Technology Co., Ltd (China). All reagents are of analytical grade or ultrapure grade, and were used without any further purification. The water used in the experiment is Milli-Q water (18.2 M Ω ·cm), which was prepared with Millipore pure system.

A Talos F200S G2 field transmission electron microscopy (TEM) (FEI, Netherlands) together with an energy dispersive X-ray spectroscopy (EDX) was used to obtain TEM images and EDX mapping. The X-ray diffraction (XRD) analysis was carried out with a Ultima IV X-ray diffractometer (Rigaku, Japan). The UV-visible spectrometry was performance with an Infinite 200 Pro microplate reader (Tecan, Switzerland). An Esca Lab 250 X-ray photoelectron spectrometer (XPS, Thermo Fisher Scientific Company, USA) was used for the X-ray photoelectron spectroscopy (XPS) analysis.

Table S1: Comparison on the apparent Michaelies-Menten constant (K_m) andmaximum reaction rate (V_{max}) among AuNPs@MoS2 dual-activenanozyme, HRP and GOx.

	Substrate	K _m (mM)	V _{max} (mol/L s)
AuNPs@MoS ₂	H ₂ O ₂	1.31	1.36×10 ⁻⁶
HRP	H_2O_2	1.01	2.66×10 ⁻⁶
AuNPs@MoS ₂	TMB	0.044	5.60×10 ⁻⁸
HRP	TMB	0.079	1.27×10 ⁻⁷
AuNPs@MoS ₂	Glucose	0.448	5.80×10 ⁻⁷
GOx	Glucose	0.646	9.87×10 ⁻⁷

Table S2: Analytical performances of the developed bio-enzyme-free colorimetricmethods for the detection of glucose, xylose and galactose based onAuNPs@MoS2

Analyte	Equation	Linear range	Visual	Spectrometry	RSD (n=5)
			LOD	LOD	
Glucose	A ₆₅₂ =0.2699×C + 0.1178	0.0 - 0.7 mM	0.3 mM	11.0 µM	4 % (detecting 0.3 mM
					glucose)
Xylose	$A_{652}=0.3825 \times C + 0.1077$	0.0 - 0.7 mM	0.3 mM	8.0 μΜ	3% (detecting 0.3 mM
					xylose)
Galactose	$A_{652}=0.6187 \times C + 0.0980$	0.0 - 0.5 mM	0.2 mM	5.0 μΜ	2% (detecting 0.2 mM
					galactose)

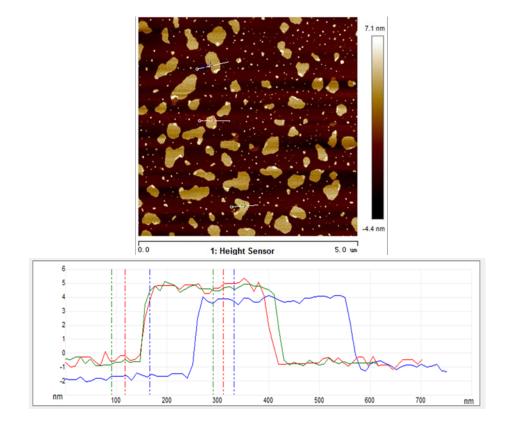


Figure S1: The AFM analysis of the MoS_2 nanosheets, which used to prepare AuNPs@MoS₂ nanozyme.

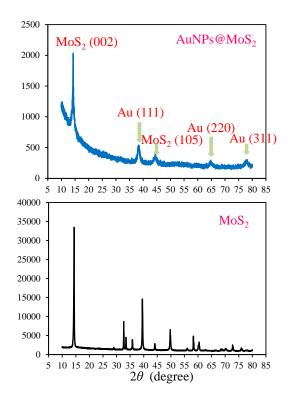


Figure S2: The XRD spectra of MoS_2 used to prepare MoS_2 nanosheets and the prepared AuNPs@MoS₂ nanozyme.

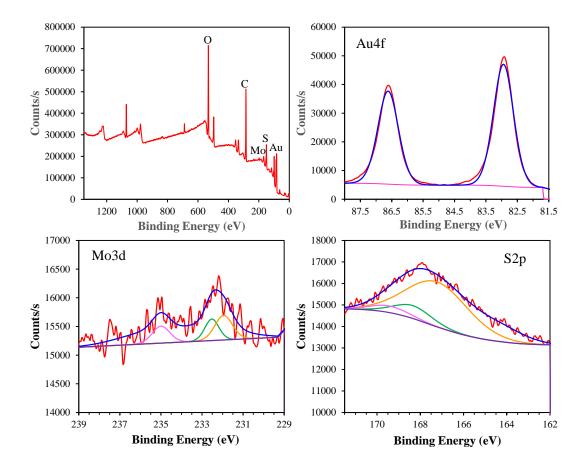


Figure S3: The X-ray photoelectron spectroscopy (XPS) analysis of the prepared AuNPs@MoS₂ nanozyme.

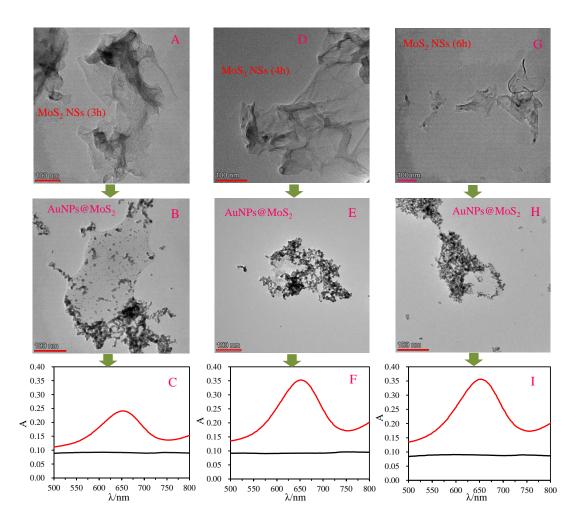


Figure S4: The TEM image of the prepared MoS₂ nanosheets under different time of ultra-sonication and the corresponding AuNPs@MoS₂ nanozyme, and the dual enzyme-like activities characterization of the prepared AuNPs@MoS₂ nanozyme. 3 hours (A, B, C), 4 hours (D, E, F) and 5 hours (G, H, I). The dual enzyme-like activities characterization was performed by directly using AuNPs@MoS₂ to catalyze the tandem reactions of glucose oxidation and H₂O₂-mediated oxidation of TMB, according to the procedure of 2.4 in experimental section. The concentration of glucose is 2.0 mM.

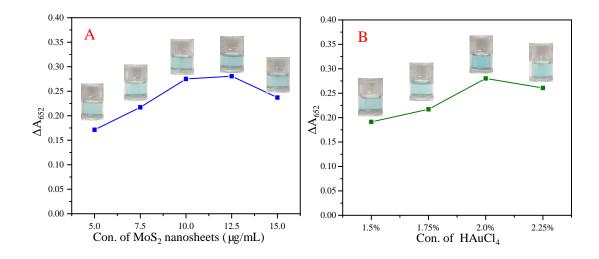


Figure S5: The effect of MoS₂ nanosheet concentration on the dual enzyme-like activities of the prepared AuNPs@MoS₂ nanozyme (A); the effect of HAuCl₄ concentration on the dual enzyme-like activities of the prepared AuNPs@MoS₂ nanozyme (B). The dual enzyme-like activities characterization was performed by directly using AuNPs@MoS₂ to catalyze the tandem reactions of glucose oxidation and H₂O₂-mediated oxidation of TMB, according to the procedure of 2.4 in experimental section. The concentration of glucose is 2.0 mM.

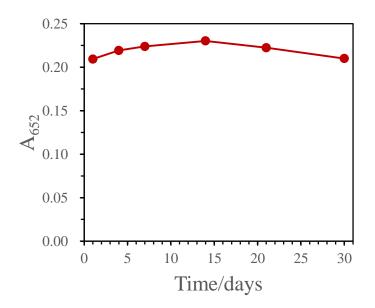


Figure S6: The variation of absorbance at 652 nm for detecting 0.5 mM glucose by using AuNPs@MoS₂, which was stored at 5 ℃ for different days. Data was obtained under the optimal selection of glucose detection method described in text.

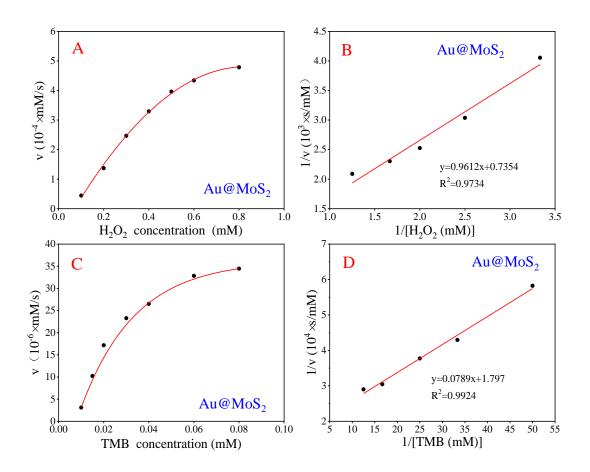


Figure S7: Steady-state kinetic analysis of AuNPs@MoS₂ as HRP mimics. The velocity of the reaction was measured by using 50 μL of AuNPs@MoS₂ in acetic acid/sodium acetate buffer (200 mM, pH 4.6) with a total volume of 200 μL. A: The time-reaction velocity variation of H₂O₂ under 0.1 mM of TMB; B: Double reciprocal plots of velocity versus H₂O₂ concentration at 0.1 mM TMB; C: The time-reaction velocity variation of TMB under 1.0 mM of H₂O₂; D: Double reciprocal plots of velocity versus TMB concentration at 1.0 mM H₂O₂.

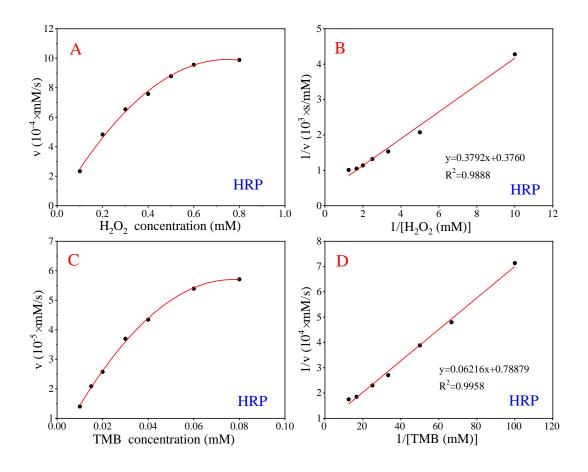


Figure S8: Steady-state kinetic analysis of natural HRP. The velocity of the reaction was measured by using 0.5 ng/mL of HRP in acetic acid/sodium acetate buffer (200 mM, pH 4.6). A: The time-reaction velocity variation of H_2O_2 under 0.1 mM of TMB; B: Double reciprocal plots of velocity versus H_2O_2 concentration at 0.1 mM TMB; C: The time-reaction velocity variation of TMB under 1.0 mM of H_2O_2 ; D: Double reciprocal plots of velocity versus TMB concentration at 1.0 mM H_2O_2 .

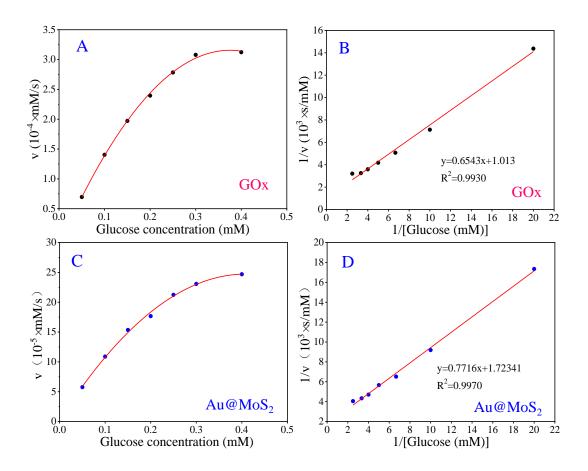


Figure S9: Steady-state kinetic analysis of natural GOx and AuNPs@MoS₂ as GOx mimics. The velocity of the reaction was measured in acetic acid/sodium acetate buffer (200 mM, pH 4.6). The concentration of HRP is 15 μg/mL, and the amount of AuNPs@MoS₂ is 60 μL in a total volume of 200 μL. A: The time-reaction velocity variation of glucose under GOx catalysis; B: Double reciprocal plots of velocity versus glucose concentration under GOx catalysis; C: The time-reaction velocity variation of glucose under AuNPs@MoS₂ catalysis; D: Double reciprocal plots of velocity versus glucose for velocity versus glucose concentration under GOX catalysis; D: Double reciprocal plots of velocity versus glucose for velocity versus glucose concentration under AuNPs@MoS₂ catalysis.

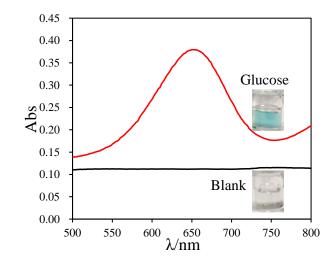


Figure S10: The photographs and absorption spectra of TMB solution in the presence of AuNPs@MoS₂ nanozyme together with glucose and AuNPs@MoS₂ without glucose (blank). The concentration of glucose is 2.0 mM.

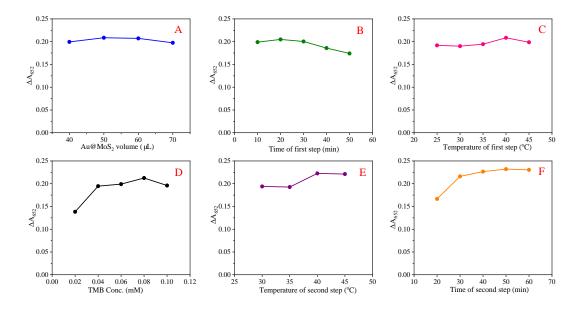


Figure S11: The effect of AuNPs@MoS₂ volume (A), the reaction time (B) and temperature (C) between AuNPs@MoS₂ and glucose, TMB concentration (D), and the reaction temperature (E) and time (F) between AuNPs@MoS₂ and TMB on the sensitivity of the method for the detection of glucose. Data were obtained under the optimal selection described in text except that will be optimized. The concentration of glucose is 1.0 mM.

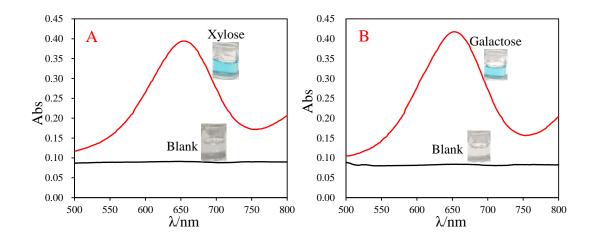


Figure S12: The photographs and absorption spectra of TMB solution in the presence of AuNPs@MoS₂ nanozyme together with xylose (A) or galactose (B), and AuNPs@MoS₂ without xylose or galactose (blank in A and B). The concentration of xylose or galactose is 2.0 mM.

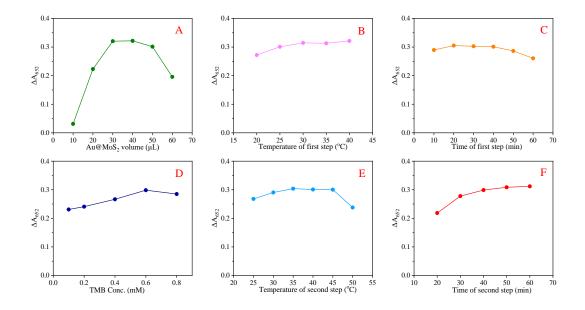


Figure S13: The effect of AuNPs@MoS₂ volume (A), the reaction temperature (B) and time (C) between AuNPs@MoS₂ and xylose or galactose, TMB concentration (D), and the reaction temperature (E) and time (F) between AuNPs@MoS₂ and TMB on the sensitivity of the method for the detection of xylose or galactose. Data were obtained under the optimal selection described in text except that will be optimized. The concentration of xylose or galactose is 2.0 mM.