Supporting Information to

On the Structural Evolution of Nanoporous Optically Transparent CuO Photocathodes upon Calcination for Photoelectrochemical Applications

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Figure S1. Flow chart representing the novel synthesis route starting from dissolution of the precursors over the dip-coating process to the applied temperature profile for the preparation of nanocrystalline and nanoporous CuO thin films.



Figure S2. Bright-field TEM images of A) CuO-400, B) CuO-500, C) CuO-600, and D) CuO-750. Arrows indicating CuO crystallites and areas representing the pores. The inset in A) shows a SAED pattern of the area. The pattern is indexed applying the crystallographic data for monoclinic CuO.



Figure S3. Energy dispersive X-ray spectroscopy (EDS) applied for A) CuO-400, B) CuO-500, C) CuO-600, and D) CuO-750 presenting the detected signals of Cu, O, and C.



Figure S4. Representative profilometry measurements of mesoporous CuO-400 (left) and CuO-600 (right) thin films.

Table S1: Crystallite sizes and full width at half maximum (FWHM) values of CuO thin films calcinated at four different temperatures and calculated by Williamson-Hall method.

Calcination temperature	400 °C	500 °C	600 °C	750 °C
$\Lambda(2\theta)$ (110) (202) and (004)	1.09°, 1.33°,	0.88°, 1.17°,	0.80°, 1.06°,	0.82°, 0.94°,
-(10), (202), and (004)	and 1.61°	and 1,40°	and 1.23°	and 1.29°
Crystallite size	8.9 nm	11.9 nm	12.2 nm	13.7 nm



Figure S5. XPS survey spectrum of the mesoporous CuO-400, CuO-500, CuO-600, and CuO-750 thin films. The additional In3d and Sn3d emission lines for CuO-750 originate from the ITO substrate.



Figure S6. Absorbance spectra of the mesoporous CuO-400, CuO-500, CuO-600, and CuO-750 thin films measured in transmission mode.





Figure S7. A) Intermittent-light voltammetry for three consecutive measurements at CuO-600 (black 1st, red 2nd, and blue 3rd cycle). B) SEM picture of CuO-600 sample after photoelectrochemical characterization (i.e., subsequential chopped-light voltammetry and Mott-Schottky analysis).



Figure S8. IPCE spectra of CuO-400, CuO-500, and CuO-600 thin films measured at 400 mV vs. RHE in in $1.0 \text{ M} \text{ Na}_2\text{SO}_3 / 0.2 \text{ M} \text{ Na}_2\text{SO}_4$ electrolyte.



Figure S9. Photocurrent decay curve for CuO thin film in $1.0 \text{ M} \text{ Na}_2\text{SO}_3 / 0.2 \text{ M} \text{ Na}_2\text{SO}_4$ electrolyte under AM 1.5 irradiation and at 0.3 V vs RHE.



Figure S10. Intermittent-light voltammetry for CuO-600 thin films analyzed in oxygen (violet) and nitrogen (orange) purged aqueous $0.2 \text{ M} \text{ Na}_2\text{SO}_4$ electrolyte (containing a phosphate buffer and 1 M Na₂SO₃).

Wavelength [nm]	FWHM [nm]	P [Wm ⁻²]	
1008	44	139	
968	35	98	
933	46	114	
913	37	153	
867	32	221	
844	25	221	
804	23	222	
774	22	201	
727	20	214	
708	21	269	
663	17	132	
656	56	322	
630	14	544	
593	12	184	
567	63	689	
530	22	450	
516	22	519	
478	15	811	
454	13	958	
422	10	946	
410	15	797	
395	10	390	
384	8	408	
369	8	293	

Table S2: Wavelengths, full width at half maximum (FWHM), and power of the LEDs accessed in thetunable light source.



Figure S11. Linear sweep voltammetry (LSV) scans under chopped illumination (AM 1.5) for CuO-400, CuO-500 and CuO-600 photocathodes in 0.01 M MVCl₂/0.05 M sodium phosphate buffer (pH = 7) with continuous N_2 bubbling.



Figure S12. Photocurrent decay curve for CuO-600 thin film under AM 1.5 Illumination at 0.4 V vs. RHE in 0.01 M MVCl₂/ 0.05 M sodium phosphate buffer (pH = 7) with continuous N₂ bubbling.

Table S3. Average film thickness determined by measuring three different spots on the CuO thin films composed of 1, 2, 3, 4, and 5 layers.

Number of layers	1	2	3	4	5
Thickness [nm]	170 ± 20	300 ± 20	380 ± 20	430 ± 20	490 ± 20



Figure S13. Three consecutive CLV measurements conducted on a five-layered CuO-600 sample.



Figure S14. Heating protocol for the preparation of mesoporous CuO thin films calcinated at 600 °C.