

**Supplementary Information**

**Enhanced electrochemical sensing of methyl parathion using  
AgNPs@IL/GO nanocomposites in aqueous matrices**

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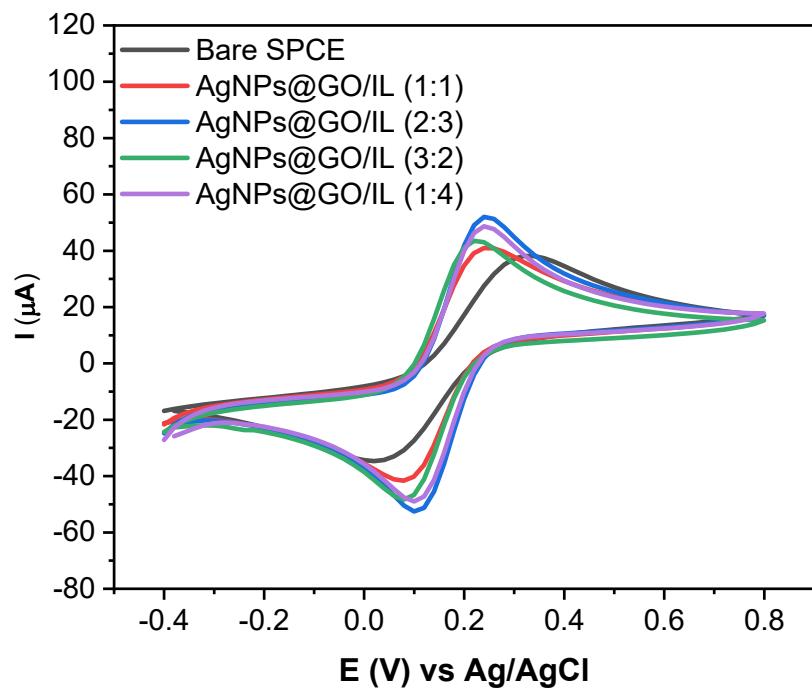
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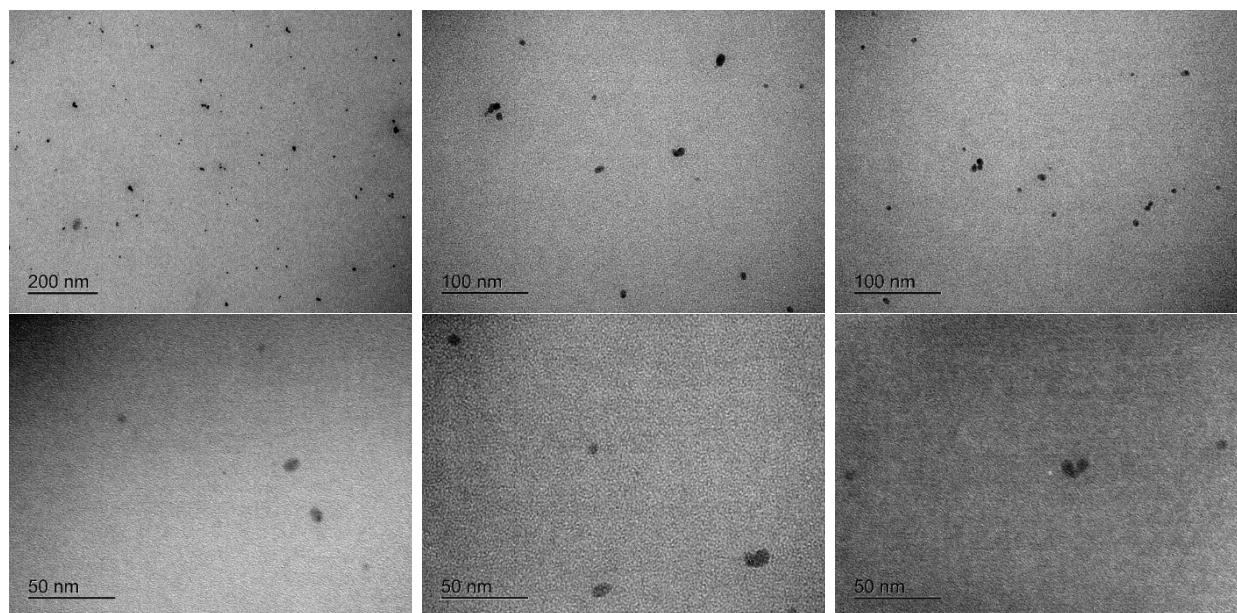
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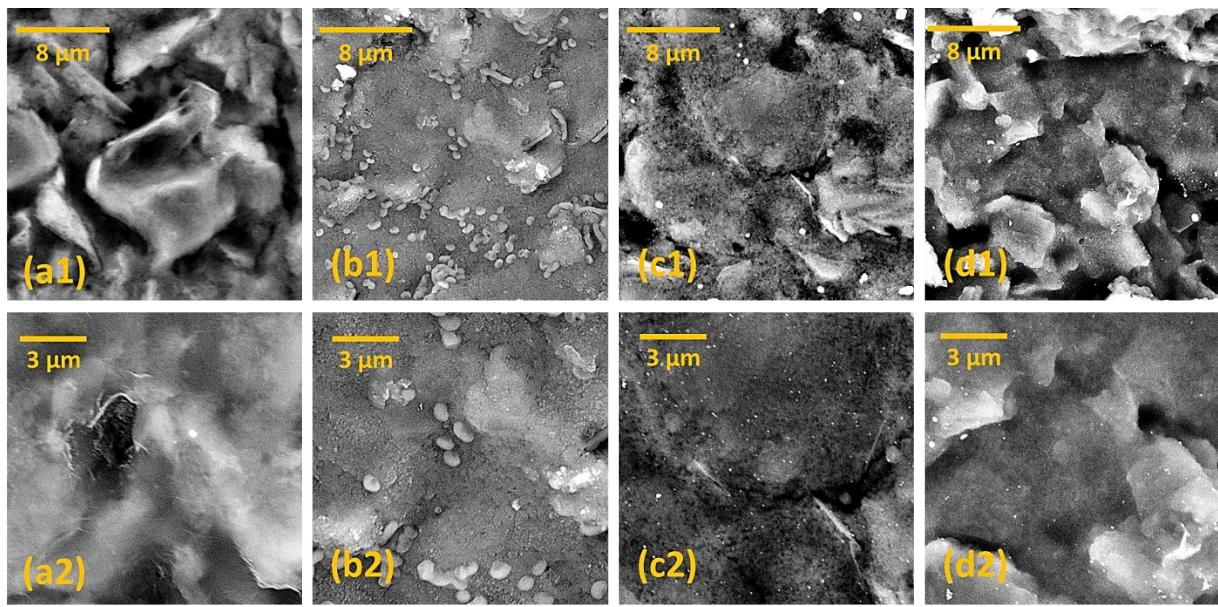
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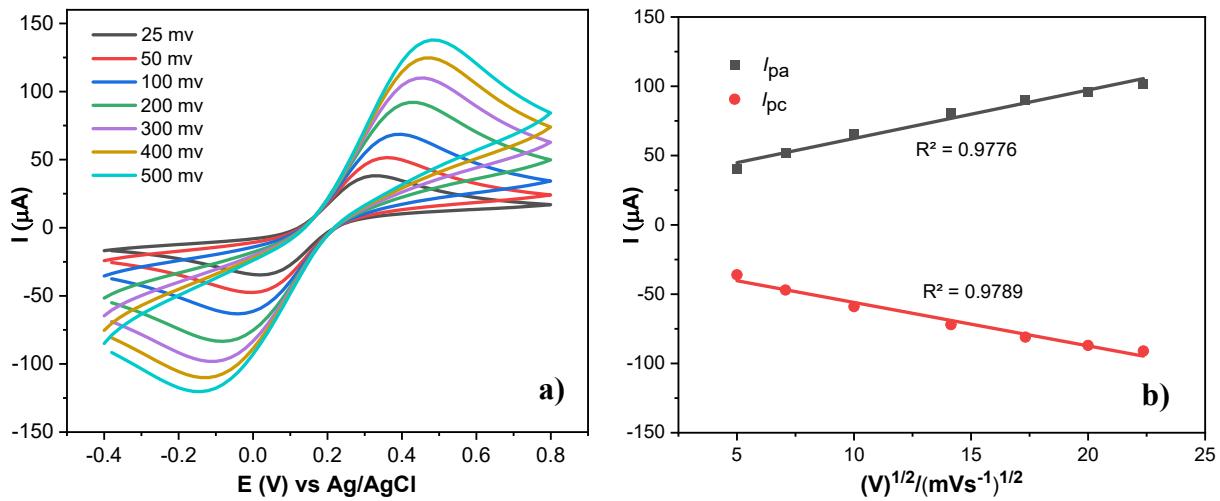
**Figure S1.** Optimization AgNP@GO/IL composite ratio in presence of  $5.0 \text{ mmol L}^{-1} [\text{Fe}(\text{CN})_6]^{3-/4-}$  in  $0.1 \text{ M KCl}$ ; scan rate  $25 \text{ mV}$ .



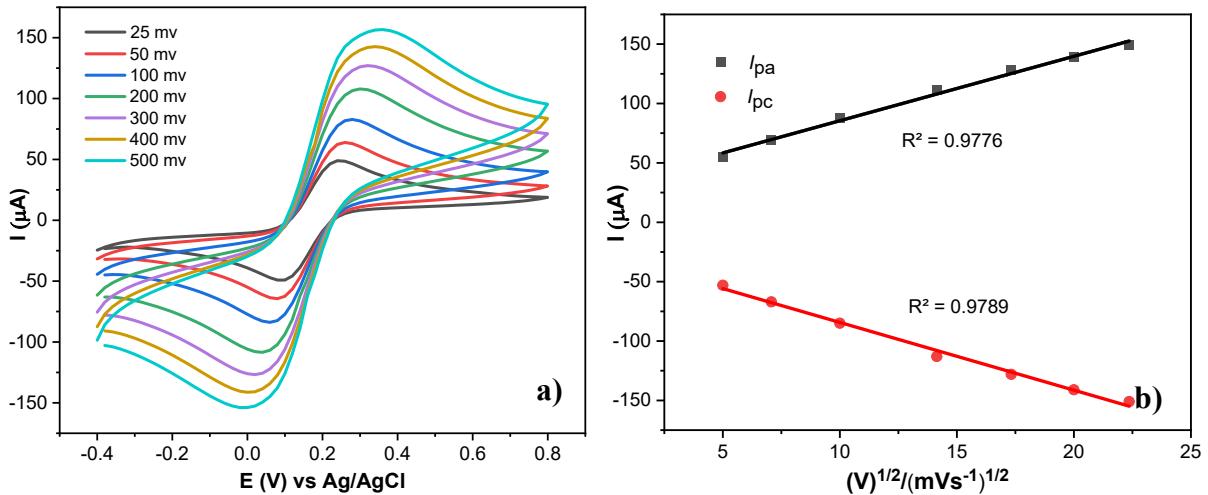
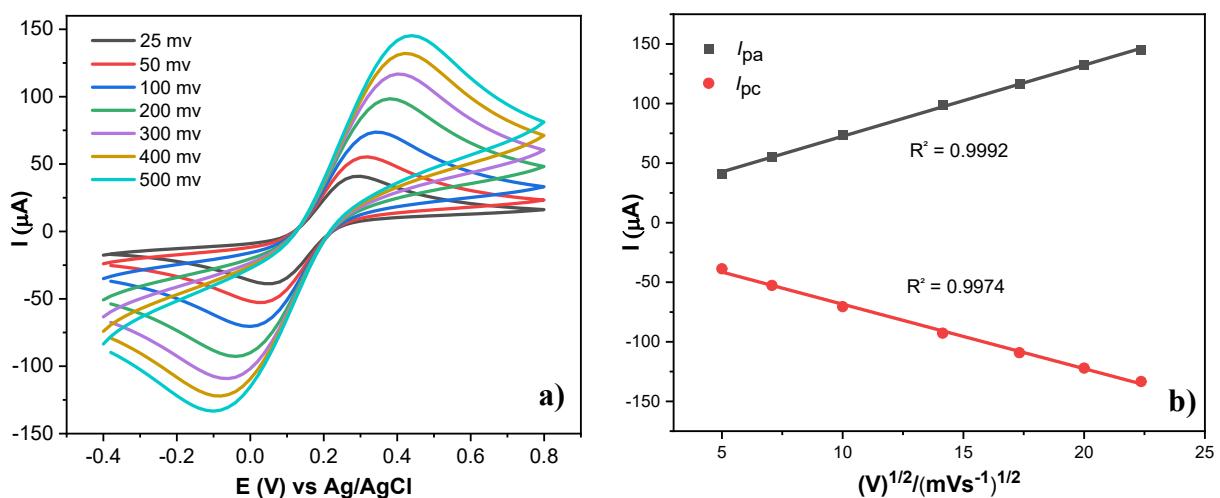
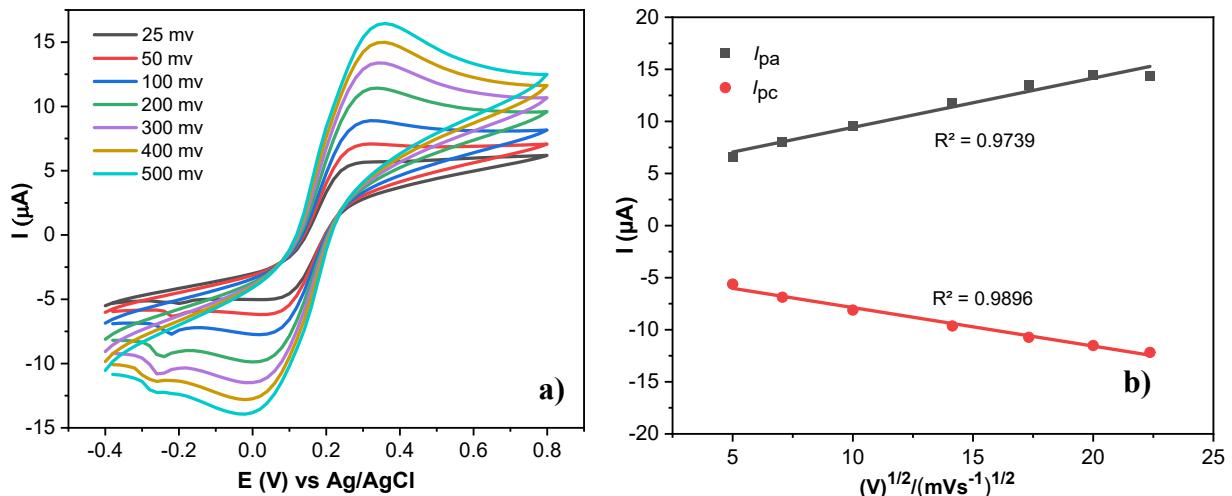
**Figure S2.** Representative TEM images of AgNPs obtained from different synthesis and different regions of the grid.

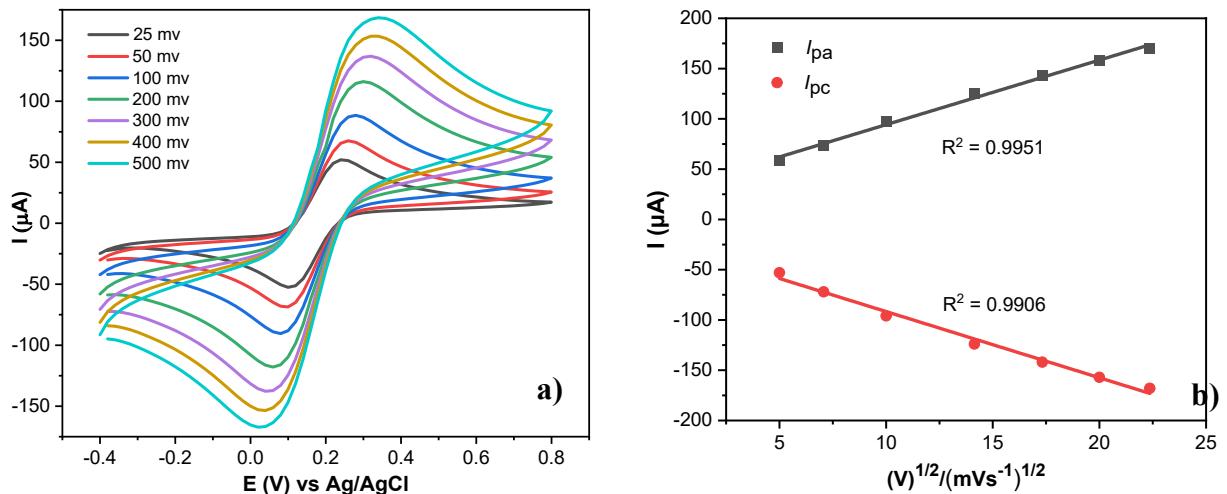


**Figure S3.** SEM images at magnifications  $\times 10,000$  (marked 1) and  $\times 20,000$  (marked 2), of the (a) GO@SPCE, (b) GO/IL@SPCE, (c) AgNPs@SPCE, and (d) AgNPs@GO/IL@SPCE. GO@SPCE and GO/IL@SPCE are covered with a Cu layer using a plasma coating procedure.

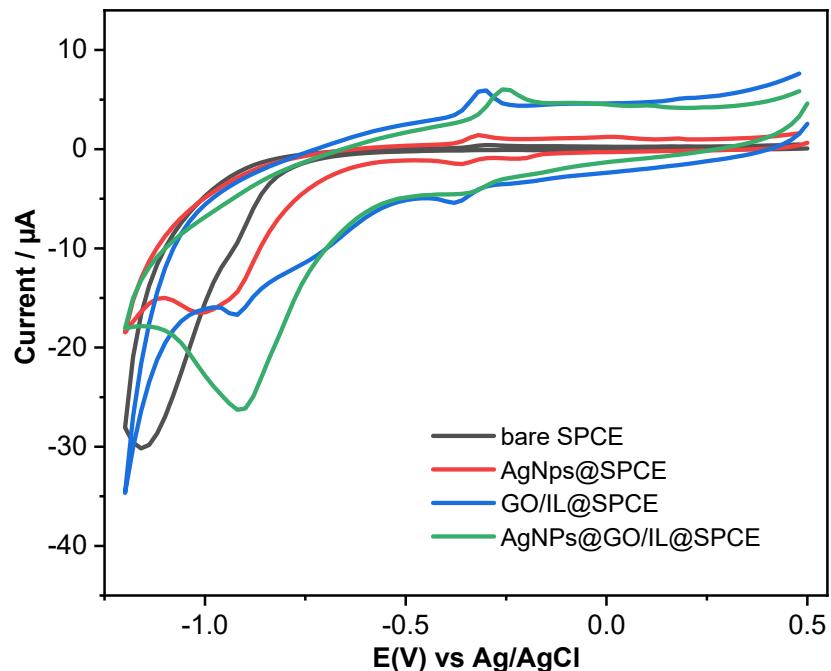


**Figure S4.** a) Cyclic voltammograms of  $5.0 \text{ mmol L}^{-1} [\text{Fe}(\text{CN})_6]^{3-/4-}$  in 0.1 M KCl obtained with a bare SPCE at different scan rates. b) Anodic current and cathodic current vs. sq. root of scan rates.

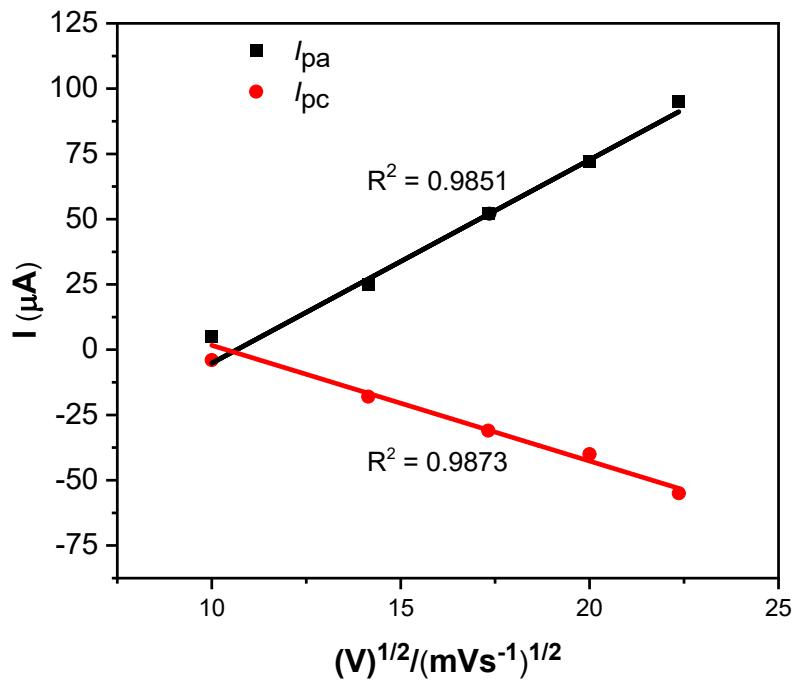




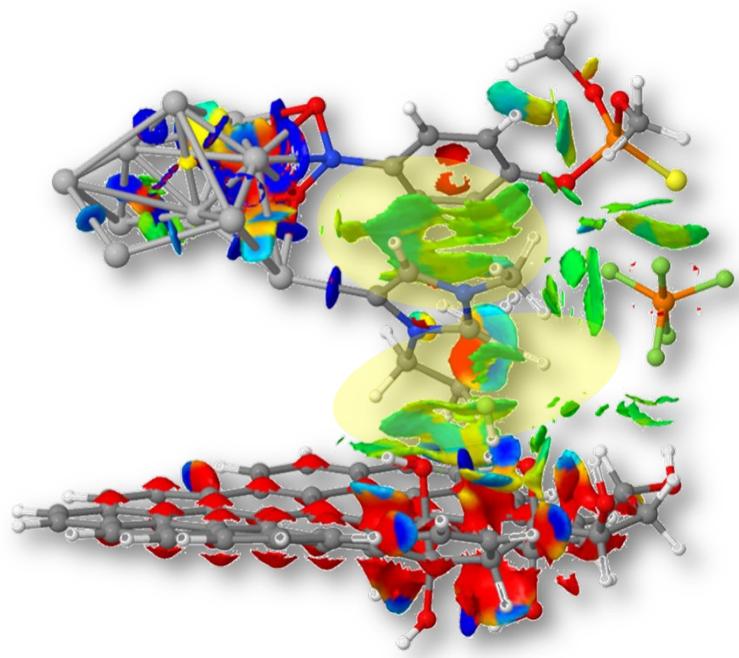
**Figure S8.** a) Cyclic voltammograms of  $5.0 \text{ mmol L}^{-1} [\text{Fe}(\text{CN})_6]^{3-/-4-}$  in  $0.1 \text{ M KCl}$  obtained with AgNPs@GO/IL@SPCE at different scan rate. b) Anodic current and cathodic current vs. sq. root of scan rates.



**Figure S9.** Electrochemical behavior of MP at different electrodes ( $20 \mu\text{mol L}^{-1}$  MP; in  $0.1 \text{ M PBS pH 7}$ ; Scan rate  $100 \text{ mV s}^{-1}$ )



**Figure S10.** Anodic current and cathodic current vs. sq. root of scan rates of MP at AgNPs@GO/IL@SPCE (E5) electrode (Ox<sub>1</sub> and R<sub>2</sub> processes).



**Figure S11.** Non-covalent interaction domains in the (MP + Ag@GO/IL) system.