

Supporting information

Regulation of closed pores in hard carbon for enhanced electrochemical sodium storage

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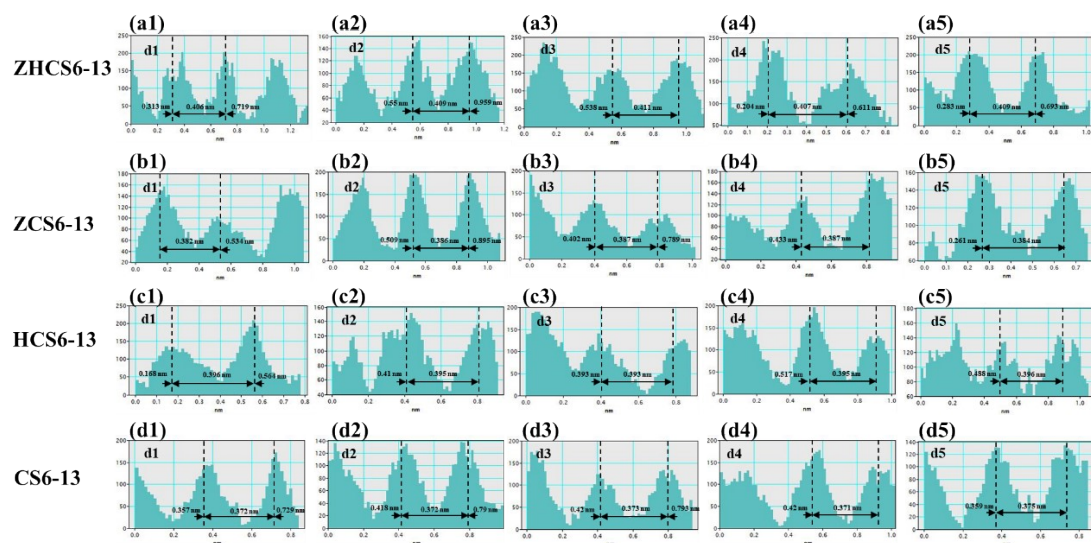


Fig. S1. The measured interlayer distance from TEM of (a1-a5) ZHCS6-13, (b1-b5) ZCS6-13, (c1-c5) HCS6-13, and (d1-d5) CS6-13.

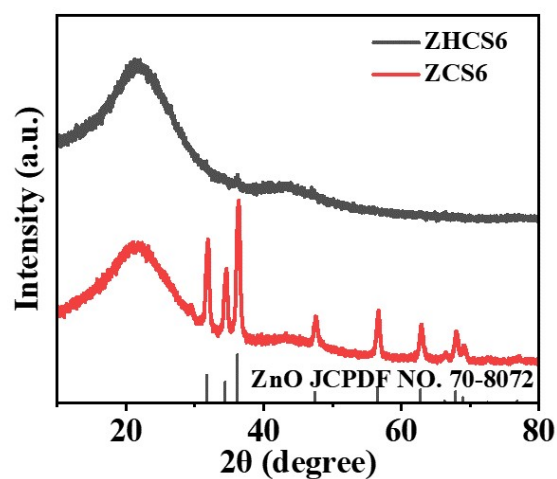


Fig. S2. XRD patterns of ZHCS6 and ZCS6.

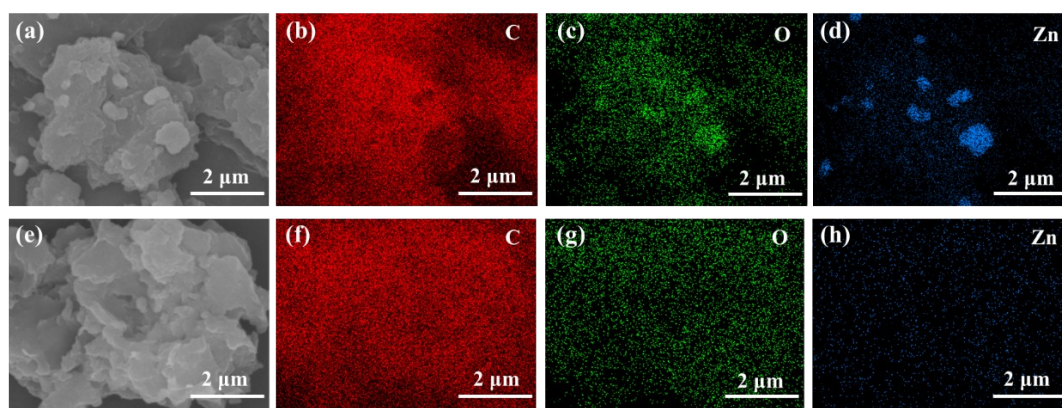


Fig. S3. SEM image and the corresponding EDS mappings of (a-d) ZCS6, (e-h) ZHCS6.

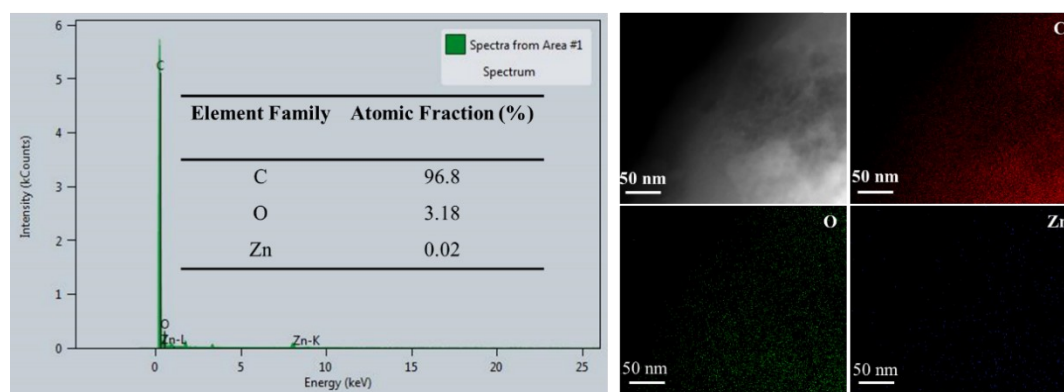


Fig. S4. TEM-EDS spectra and EDS mapping of ZHCS6-13.

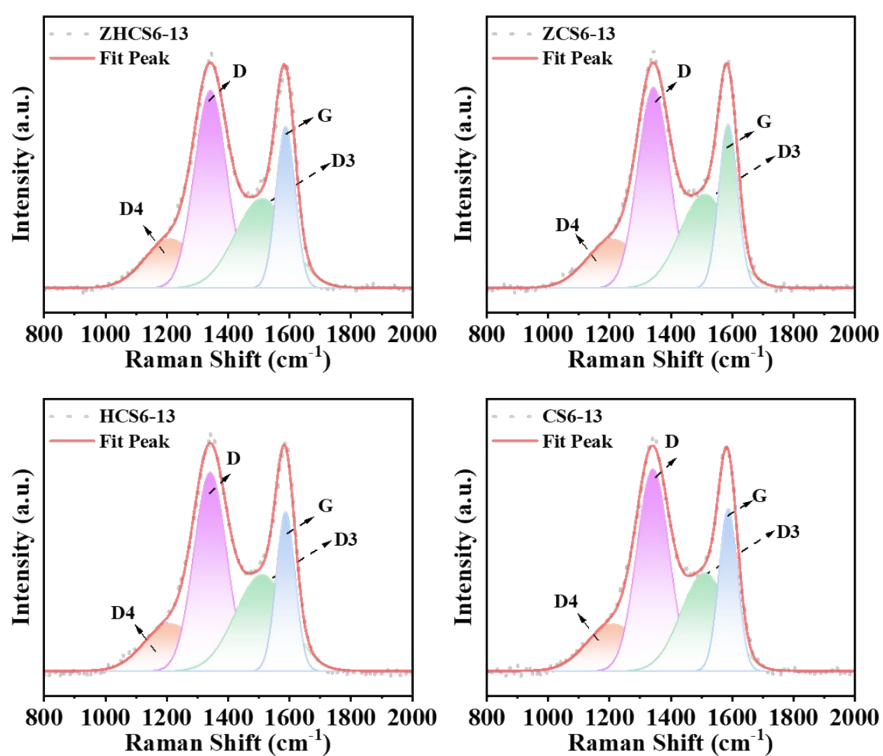


Fig. S5. Fitted Raman spectra of ZHCS6-13, ZCS6-13, HCS6-13, CS6-13.

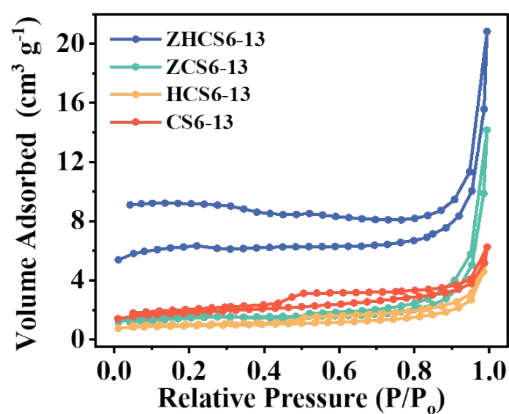


Fig. S6. N₂ adsorption-desorption isotherms.

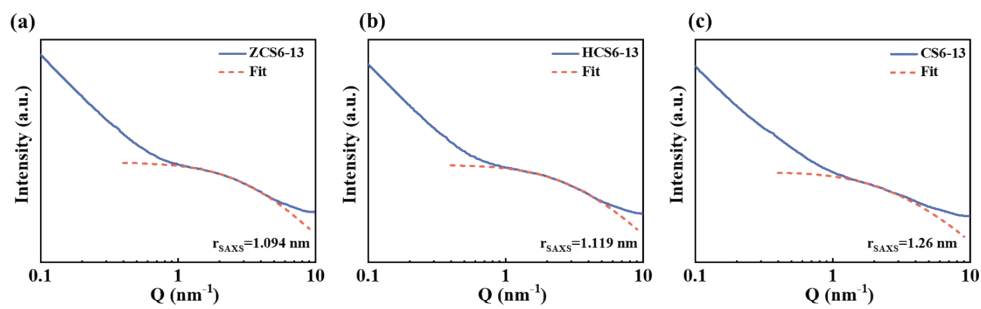


Fig. S7. (a-c) Fitted SAXS patterns of ZCS6-13, HCS6-13 and CS6-13.

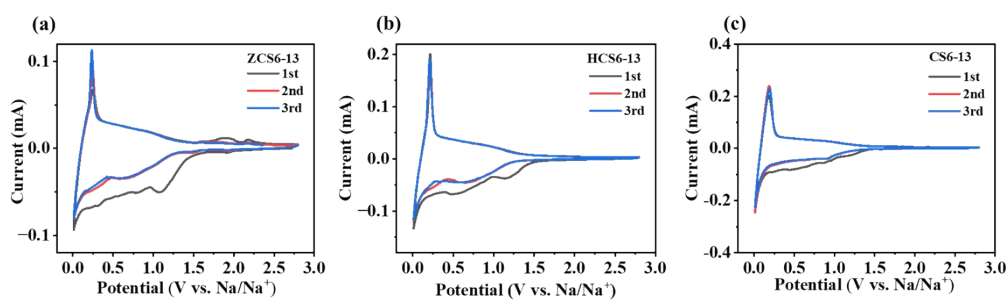


Fig. S8. (a-c) CV curves of ZCS6-13, HCS6-13 and CS6-13 at 0.1 mV s⁻¹.

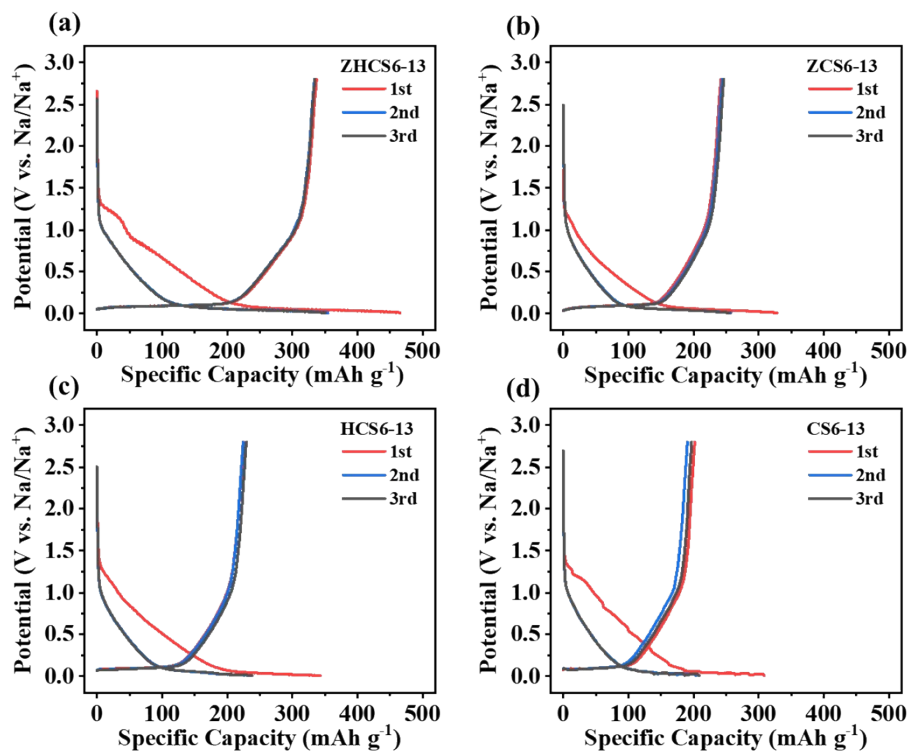


Fig. S9. Galvanostatic charge/discharge profiles of (a) ZHCS6-13, (b) ZCS6-13, (c) HCS6-13, and (d) CS6-13 at a current rate of 25 mA g^{-1} .

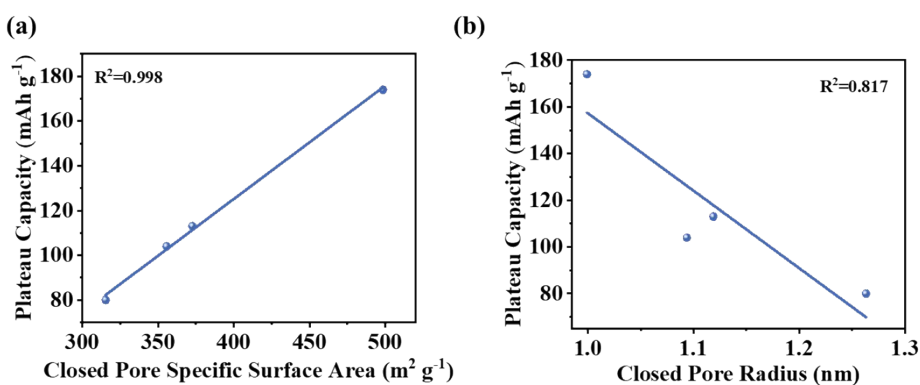


Fig. S10. The relationships between plateau capacities and closed pore specific surface area/ r_{SAXS} .

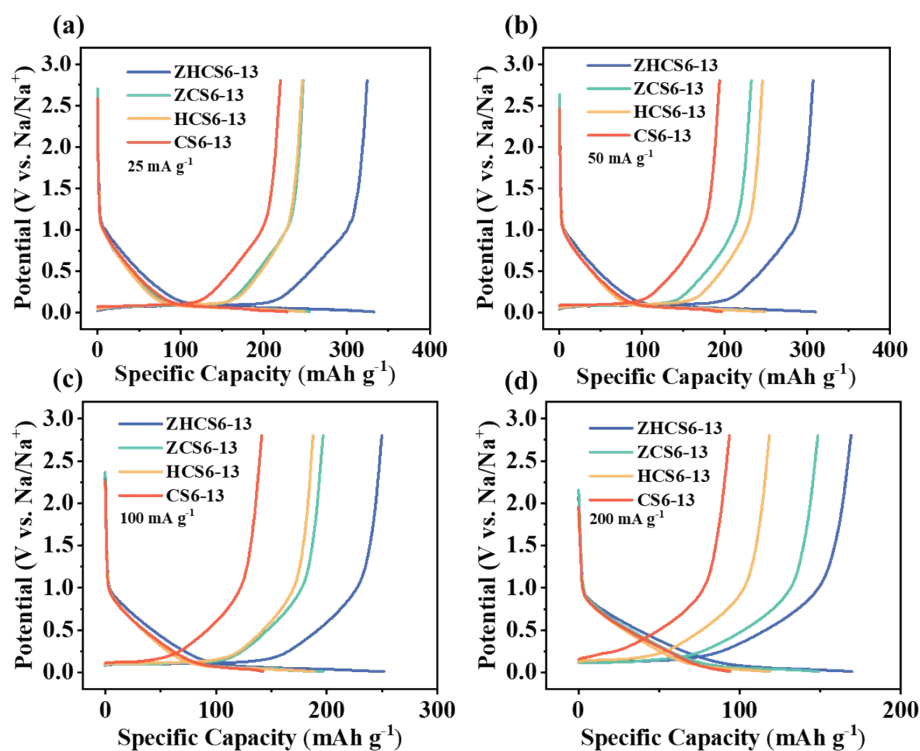


Fig. S11. GCD curves at (a) 25, (b) 50, (c) 100, and (d) 200 mA g⁻¹.

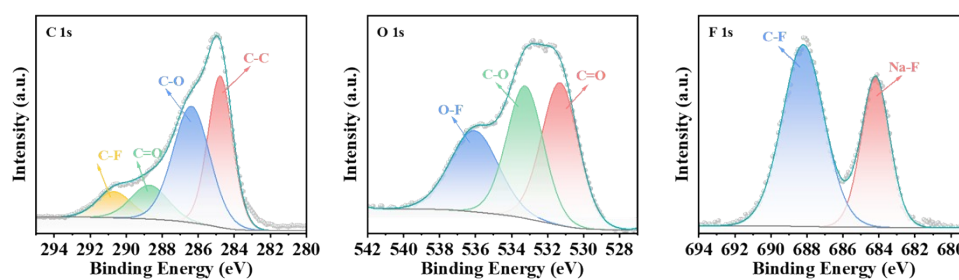


Fig. S12. Interfacial chemistry of ZHCS6-13. The high-resolution XPS spectra of (a) C 1s, (b) O 1s, and (c) F 1s collected at the ZHCS6-13 electrodes after 100 cycles at 50 mA g⁻¹.

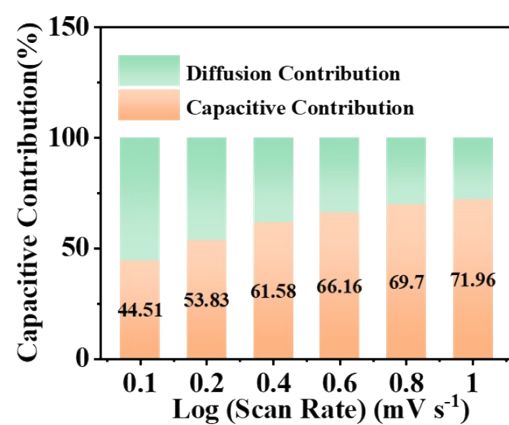


Fig. S13. Capacitive contribution percentage of ZHCS6-13 at various scan rates.

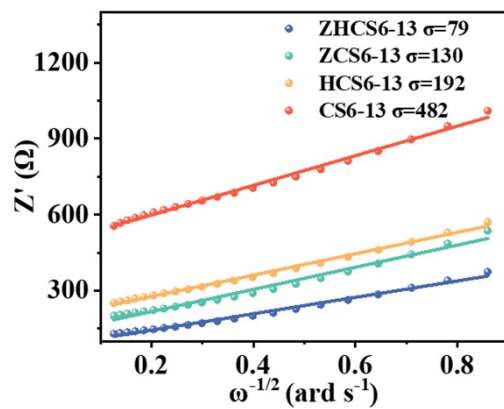


Fig. S14. The linear fits of Z' to $\omega^{-1/2}$ in the low-frequency region.

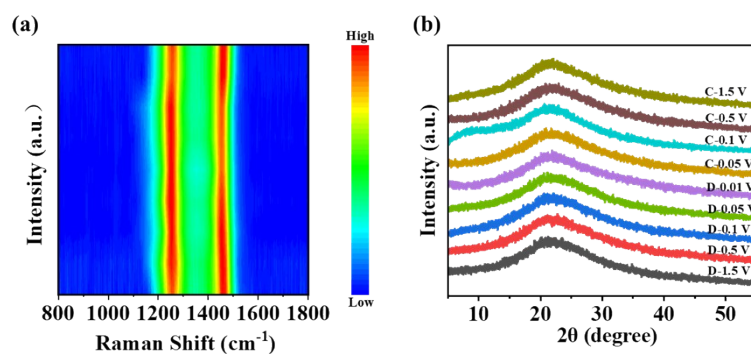


Fig. S15. (a) the ex-situ Raman contour plot and (b) the ex-situ XRD of the ZHCS6-13 electrode.

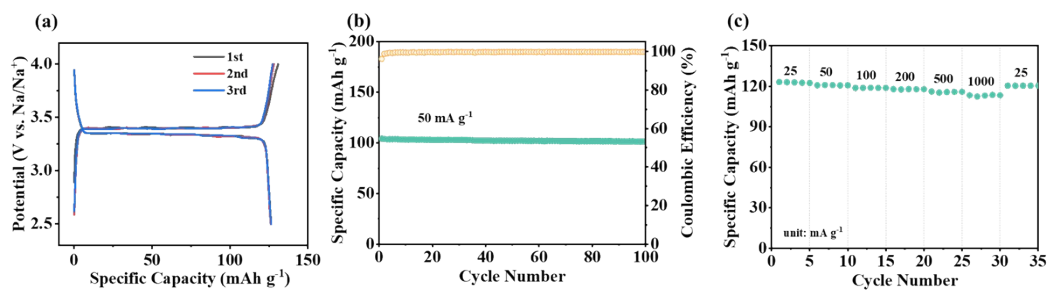


Fig. S16. (a) The GCD profiles of NVP half-cell at 25 mA g^{-1} . (b) Cycling performance and (c) rate of commercial NVP cathode in Na half-cells.

Table S1. Structure parameters of hard carbon materials.

Sample	D ₀₀₂ (nm)	L _a (nm)	L _c (nm)	A _D /A _G	S _{BET} (m ² g ⁻¹)	r _{SAXS} (nm)	S _{closed pore} (m ² g ⁻¹)
ZHCS6-13	0.4	4.25	0.99	2.05	12.700	0.999	499
ZCS6-13	0.39	4.35	1.00	2.03	5.890	1.094	356
HCS6-13	0.39	3.95	0.96	2.14	3.467	1.119	373
CS6-13	0.37	4.20	0.98	2.13	6.626	1.260	315

(1) D₀₀₂: Interlayer distances calculated by XRD.

(2) The Scherrer equations were used to calculate the average graphite domain length (L_a) and the graphite crystallite thickness (L_c)¹. The specific expressions are as follows:

$$L_a = \frac{1.84\lambda}{B_{100}\cos\theta_{100}} \quad (\text{equation S1})$$

$$L_c = \frac{0.89\lambda}{B_{002}\cos\theta_{002}} \quad (\text{equation S2})$$

S2)

Where λ is the wavelength of the X-rays (0.15418 nm), B_{100} and B_{002} are the full width at half maximum (FWHM) value of the (100) and (002) peaks, and $2\theta_{100}$ and $2\theta_{002}$ are the diffraction angle.

(3) The mid-angle region of the SAXS data (1 to 10 nm⁻¹) was fitted based on the semi-empirical Teubner-Strey model, to obtain more detailed structural information about the internal pores structure^{2, 3}. The specific expressions are as follows:

$$I_{mp} = I_0 \frac{1}{1 + C_1 Q^2 + C_2 Q^4} \quad (\text{equation S3})$$

S3)

Where I_0 , C_1 and C_2 are tunable constants, the following structural parameters can be further calculated.

$$r_{\text{saxs}} = \sqrt{5C_1} \quad (\text{equation S4})$$

$$S_{\text{closed pore}} = I_0 \frac{1}{C_2 2\pi(\Delta\text{SLD})^2} \quad (\text{equation S5})$$

$$\Delta\text{SLD} = \rho_{\text{graphite}} \frac{d_{002}}{d_{002}^{\text{graphite}}} \left(\frac{d_{100}}{d_{100}^{\text{graphite}}} \right) \quad (\text{equation S6})$$

Where r_{saxs} is the average pore radius. $S_{\text{closed pore}}$ are the total specific surface areas (including open and closed pores). ΔSLD is the structural density, where d_{002} is the interlayer distance and d_{100} is in-plane distance. The fitting results are shown in Table S1.

Table S2. Equivalent circuit fitting calculation results of EIS.

Sample	$R_s (\Omega)$	$R_{ct} (\Omega)$
ZHCS6-13	3.91	113
ZCS6-13	4.04	188
HCS6-13	2.29	245
CS6-13	4.00	590

Table S3. Electrochemical performance of hard carbon anodes prepared from different biomass precursors for SIBs.

Biomass precursor	ICE (%)	Reversible capacity (mA h g ⁻¹)	Plateau capacity (<0.1V vs. Na ⁺ /Na) (mA h g ⁻¹)	Current density (mA g ⁻¹)	Reference
ZHCS6-13	73	340	174	50	This work
NDC-1000 (walnut shell)	-	257	157	50	4
HC-W (walnut shells)	66	315	175	50	5
P1000 (pistachio shell)	74.9	225	-	10	6
HC-1000 (Bagasse)	73.1	331.3	156.3	25	7
C-1300 (Lignin)	69	260.55	179.87	50	8
LS1200 (lotus seedpod)	50.4	328.8	174.4	50	9
CL-HC (cellulose and lignin)	69	259	160.06	100	10
CPM-1100-A (peat moss)	57.5	298	157.9	50	11
CCL (corn cob lignin)	62.8	292	167	50	12
GDHC-1300 (garlic)	50.7	260	148	50	13
HHT (Lotus leaves)	66	288	126	0.2 C	14

Reference

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