

Supporting Information

Photophysicochemical and Electrochemical Properties of Pyrene-BODIPY

Platforms

İpek Ömeroğlu^a, Baybars Köksoy^b, Dilek Öztürk^a, Lubna Salah^c, Saad Maksheed^c, Mahmut Durmuş^{a*}

^a*Gebze Technical University, Department of Chemistry, 41400, Gebze, Kocaeli, Türkiye*

^b*Bursa Technical University, Department of Chemistry, 16130, Bursa, Türkiye*

^c*Kuwait University, Department of Chemistry, 5969, Safat 13060, Kuwait*

**Corresponding Author: Mahmut Durmuş*

Email: durmus@gtu.edu.tr

Content

1. Characterization of BODIPY compounds (1-5).....	S5
1.1. 4,4'-Difluoro-8-(phenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (1).....	S5
1.2. 4,4'-Difluoro-8-(4-iodophenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (2).....	S7
1.3. 4,4'-Difluoro-8-(phenyl)-2,6-diiodo-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (3).....	S10
1.4. 4,4'-Difluoro-8-(iodophenyl)-2,6-diiodo-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (4).....	S12
1.5. 4,4'-Difluoro-8-(4-pyrenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (5).....	S15
2. Characterization of novel pyrene-BODIPY compounds (5*, 6, and 7).....	S16
3. Electrochemical characterizations.....	S24
4. X-Ray study of compound 6.....	S25
References.....	S28

Table Caption

Table S1. Crystal structure and crystal data of BODIPY compound 6.....S26

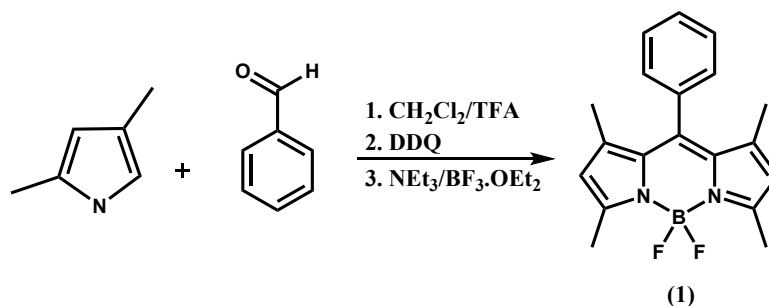
Figure Caption

Fig. S1. ¹ H-NMR spectrum of compound 1 in CDCl ₃	S5
Fig. S2. ¹³ C-NMR spectrum of compound 1 in CDCl ₃	S6
Fig. S3. ¹¹ B-NMR spectrum of compound 1 in CDCl ₃	S6
Fig. S4. ¹⁹ F-NMR spectrum of compound 1 in CDCl ₃	S7
Fig. S5. ¹ H-NMR spectrum of compound 2 in CDCl ₃	S8
Fig. S6. ¹³ C-NMR spectrum of compound 2 in CDCl ₃	S8
Fig. S7. ¹¹ B-NMR spectrum of compound 2 in CDCl ₃	S9
Fig. S8. ¹⁹ F-NMR spectrum of compound 2 in CDCl ₃	S9
Fig. S9. ¹ H-NMR spectrum of compound 3 in CDCl ₃	S10
Fig. S10. ¹³ C-NMR spectrum of compound 3 in CDCl ₃	S11
Fig. S11. ¹¹ B-NMR spectrum of compound 3 in CDCl ₃	S11
Fig. S12. ¹⁹ F-NMR spectrum of compound 3 in CDCl ₃	S12
Fig. S13. ¹ H-NMR spectrum of compound 4 in CDCl ₃	S13
Fig. S14. ¹³ C-NMR spectrum of compound 4 in CDCl ₃	S13
Fig. S15. ¹¹ B-NMR spectrum of compound 4 in CDCl ₃	S14
Fig. S16. ¹⁹ F-NMR spectrum of compound 4 in CDCl ₃	S14
Fig. S17. ¹ H-NMR spectrum of compound 5 in CDCl ₃	S16
Fig. S18. ¹³ C-NMR spectrum of compound 5 in CDCl ₃	S16
Fig. S19. ¹¹ B-NMR spectrum of compound 5 in CDCl ₃	S17
Fig. S20. ¹⁹ F-NMR spectrum of compound 5 in CDCl ₃	S17
Fig. S21. HRMS spectrum of compound 5♦.....	S18

Fig. S22. HRMS spectrum of compound 6.....	S18
Fig. S23. HRMS spectrum of compound 7.....	S19
Fig. S24. ¹ H-NMR spectrum of compound 5♦ in CDCl ₃	S19
Fig. S25. ¹³ C-NMR spectrum of compound 5♦ in CDCl ₃	S20
Fig. S26. ¹³ C-NMR spectrum of compound 6 in CDCl ₃	S20
Fig. S27. ¹³ C-NMR spectrum of compound 7 in DMSO-d ₆	S21
Fig. S28. ¹¹ B-NMR spectrum of compound 5♦ in CDCl ₃	S21
Fig. S29. ¹¹ B-NMR spectrum of compound 6 in CDCl ₃	S22
Fig. S30. ¹¹ B-NMR spectrum of compound 7 in DMSO-d ₆	S22
Fig. S31. ¹⁹ F-NMR spectrum of compound 5♦ in CDCl ₃	S23
Fig. S32. ¹⁹ F-NMR spectrum of compound 6 in CDCl ₃	S23
Fig. S33. ¹⁹ F-NMR spectrum of compound 7 in DMSO-d ₆	S24
Fig. S35. Cyclic voltammograms at different scan rates of synthesized BODIPY compounds (1-7) in DCM with 0.1 M TBAPF ₆ as supporting electrolyte.	S24
Fig. S35. Square wave voltammograms of synthesized BODIPY compounds (1-7) in DCM with 0.1 M TBAPF ₆ as supporting electrolyte, scan rate 0.05 V/s.....	S25
Fig. S36. π-π interaction of the solid-state crystal form of iodinated pyrene-BODIPY compound 6.....	S27

1. Characterization of BODIPY compounds (1-5)

1.1. 4,4'-Difluoro-8-(phenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (1)



An orange solid. Chemical formula $C_{19}H_{19}BF_2N_2$, (780 mg, 51%). FT-IR (ν , cm^{-1}): 3100-3025 (Aromatic-CH), 2955-2851 (Aliphatic-CH), 1462 (B-N). 1H NMR (500 MHz, $CDCl_3$) δ , ppm: 7.51-7.50 (m, 3H, Ar-H), 7.31-7.30 (m, 2H, Ar-H), 6.00 (s, 2H, Pyrrole-H), 2.58 (s, 3H, -CH₃), 1.40 (s, 3H, -CH₃). ^{13}C NMR (125 MHz, $CDCl_3$) δ , ppm: 155.46, 143.19, 141.76, 135.03, 131.46, 130.03, 129.15, 128.96, 127.97, 121.21, 14.34. ^{11}B NMR (160 MHz, $CDCl_3$) δ , ppm: 0.79 (t, J : 66.21 Hz). ^{19}F NMR (470 MHz, $CDCl_3$) δ , ppm: -146.29 (q, J : 98.68 Hz).

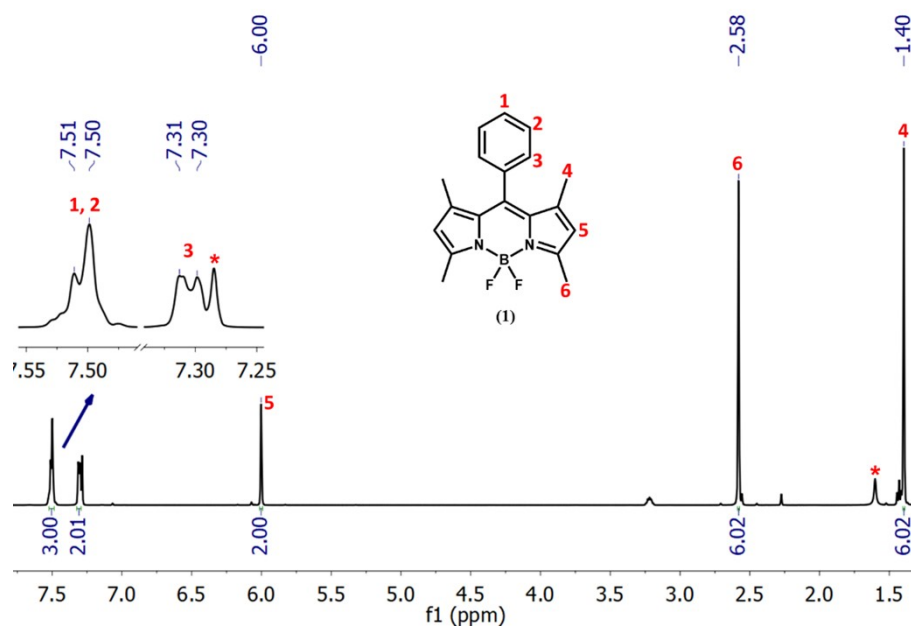


Fig. S1. 1H -NMR spectrum of compound **1** in $CDCl_3$.

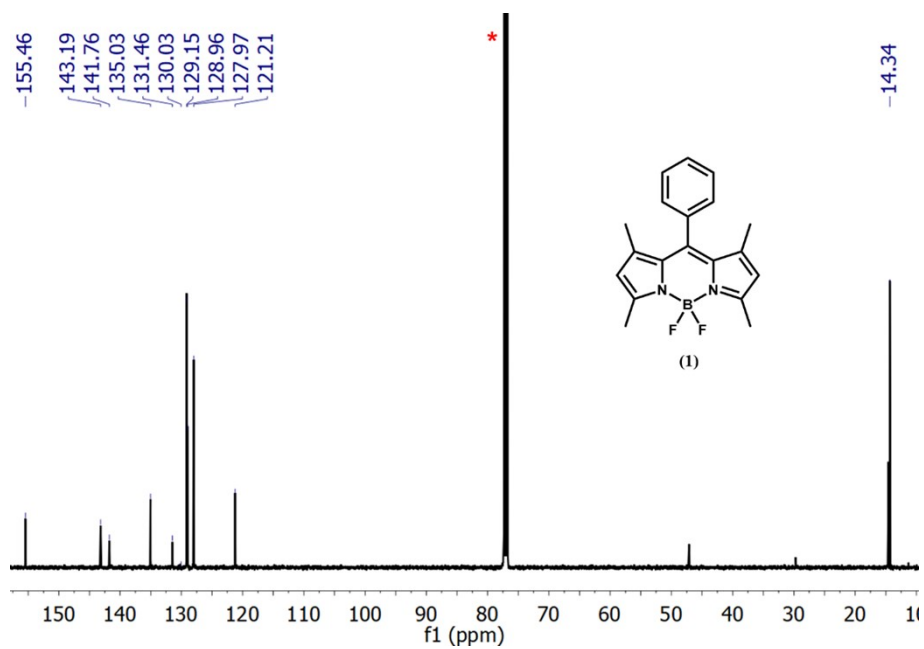


Fig. S2. ¹³C-NMR spectrum of compound **1** in CDCl₃.

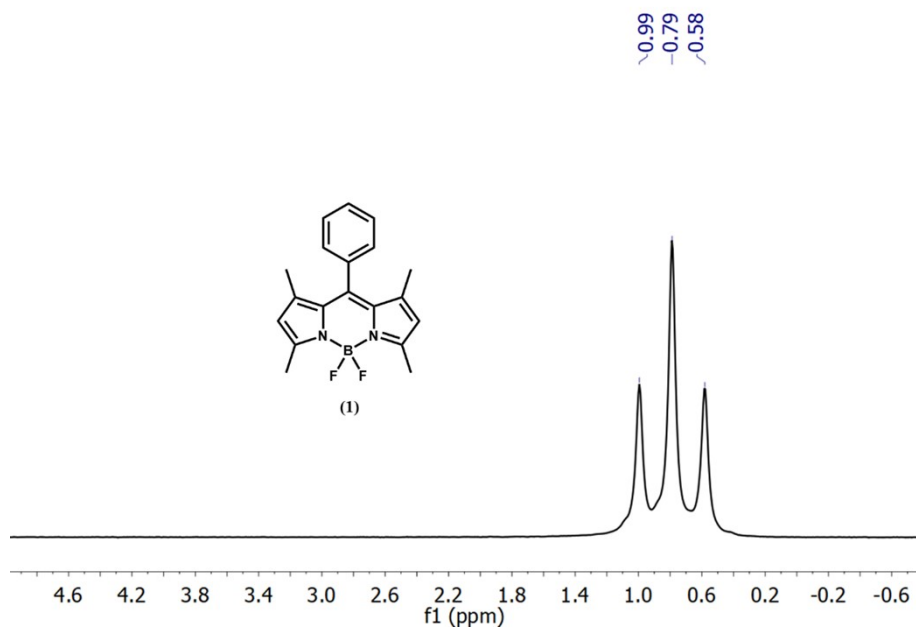


Fig. S3. ¹¹B-NMR spectrum of compound **1** in CDCl₃.

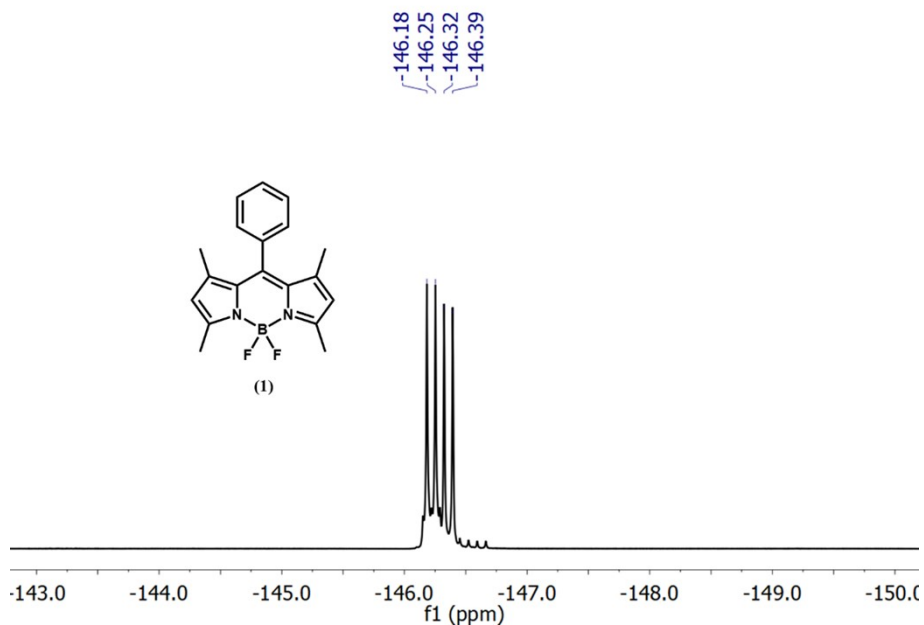
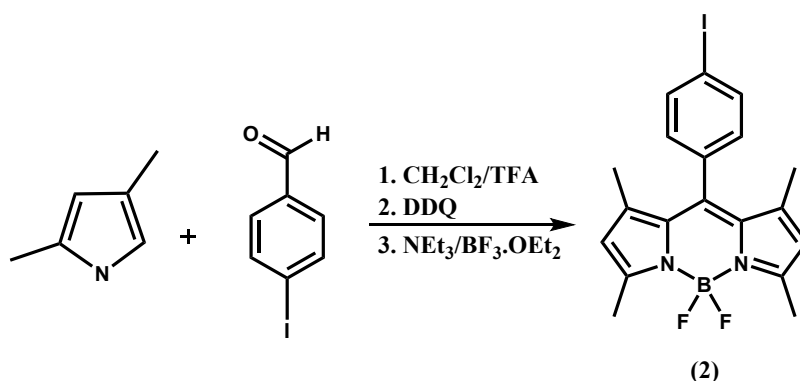


Fig. S4. ^{19}F -NMR spectrum of compound **1** in CDCl_3 .

1.2. 4,4'-Difluoro-8-(4-iodophenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (**2**)



An orange solid. Chemical formula $\text{C}_{19}\text{H}_{18}\text{BF}_2\text{IN}_2$, (285 mg, 33%). FT-IR (ν , cm^{-1}): 3107-3050 (Aromatic-CH), 2959-2852 (Aliphatic-CH), 1466 (B-N). ^1H NMR (500 MHz, CDCl_3) δ , ppm: 7.87 (d, 2H, Ar-H, J :7.65 Hz), 7.07 (d, 2H, Ar-H, J :7.63 Hz), 6.01 (s, 2H, Pyrrole-H), 2.57 (s, 6H, $-\text{CH}_3$), 1.44 (s, 6H, $-\text{CH}_3$). ^{13}C NMR (125 MHz, CDCl_3) δ , ppm: 155.92, 142.95, 140.10, 138.38, 134.59, 131.16, 129.99, 121.46, 94.76, 14.67. ^{11}B NMR (160 MHz, CDCl_3) δ , ppm: 0.73 (t, J : 65.80 Hz). ^{19}F NMR (470 MHz, CDCl_3) δ , ppm: -146.26 (q, J : 98.84 Hz).

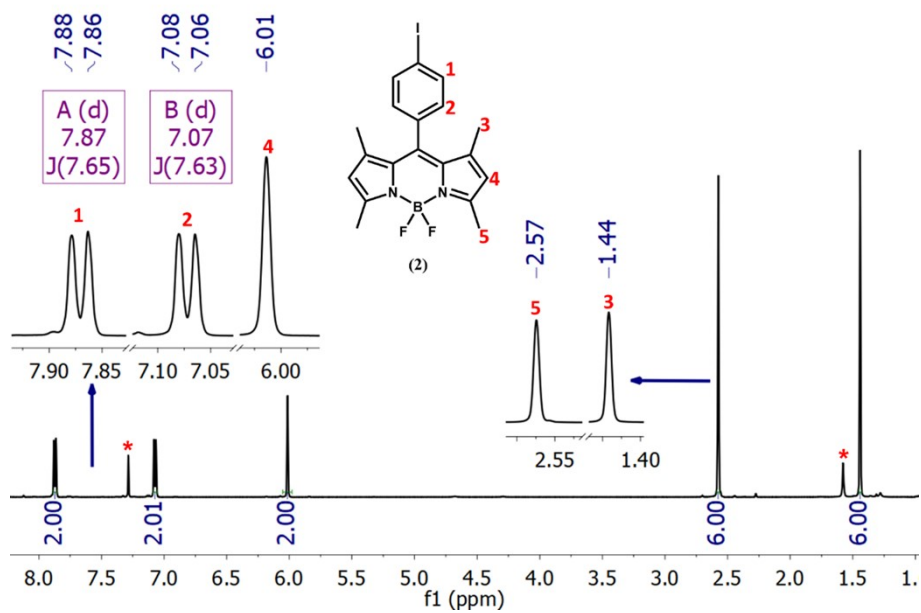


Fig. S5. ¹H-NMR spectrum of compound **2** in CDCl₃.

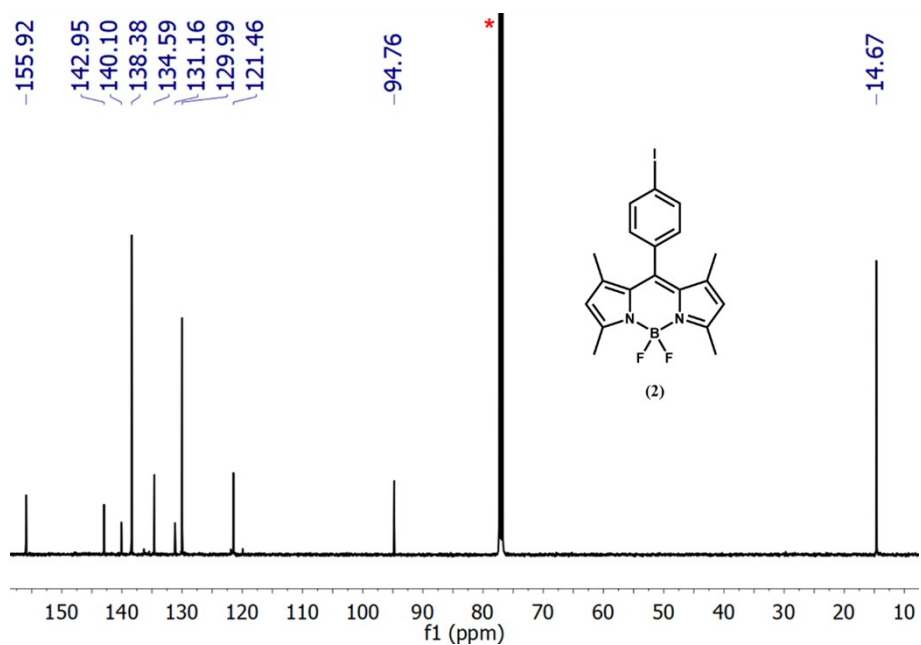


Fig. S6. ¹³C-NMR spectrum of compound **2** in CDCl₃.

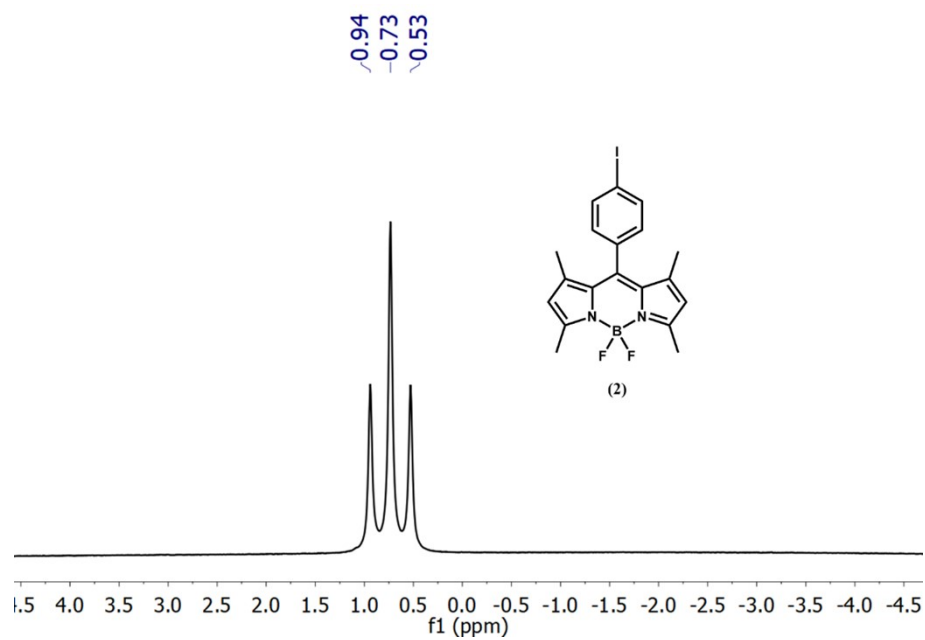


Fig. S7. ^{11}B -NMR spectrum of compound **2** in CDCl_3 .

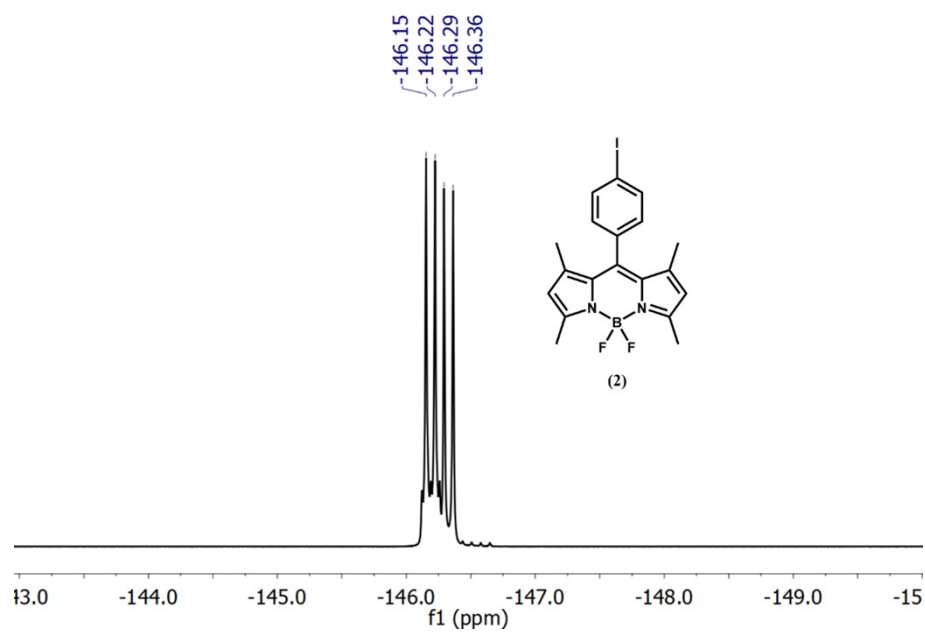
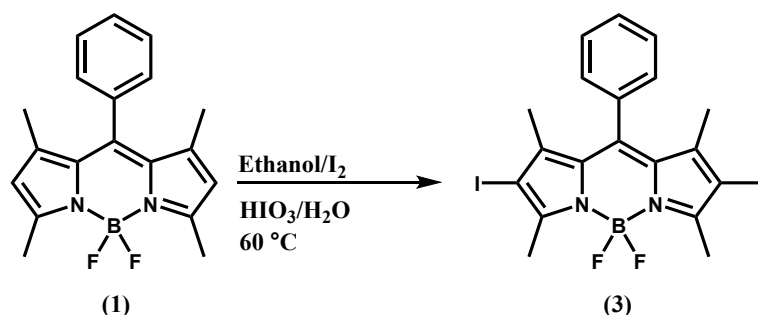


Fig. S8. ^{19}F -NMR spectrum of compound **2** in CDCl_3 .

1.3. 4,4'-Difluoro-8-(phenyl)-2,6-diiodo-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (3)



A dark red solid. Chemical formula $\text{C}_{19}\text{H}_{17}\text{BF}_2\text{I}_2\text{N}_2$, (175 mg, 78.8%). FT-IR (ν , cm^{-1}): 3070-3018 (Aromatic-CH), 2959-2844 (Aliphatic-CH), 1479 (B-N). ^1H NMR (500 MHz, CDCl_3) δ , ppm: 7.55-7.54 (m, 3H, Ar-H), 7.28-7.27 (m, 2H, Ar-H), 2.67 (s, 6H, - CH_3), 1.41 (s, 6H, - CH_3). ^{13}C NMR (125 MHz, CDCl_3) δ , ppm: 156.80, 145.40, 141.40, 134.76, 131.32, 129.56, 127.80, 85.69, 29.72, 16.96, 16.05. ^{11}B NMR (160 MHz, CDCl_3) δ , ppm: 0.58 (t, J : 64.46 Hz). ^{19}F NMR (470 MHz, CDCl_3) δ , ppm: -145.64 (q, J : 97.09 Hz).

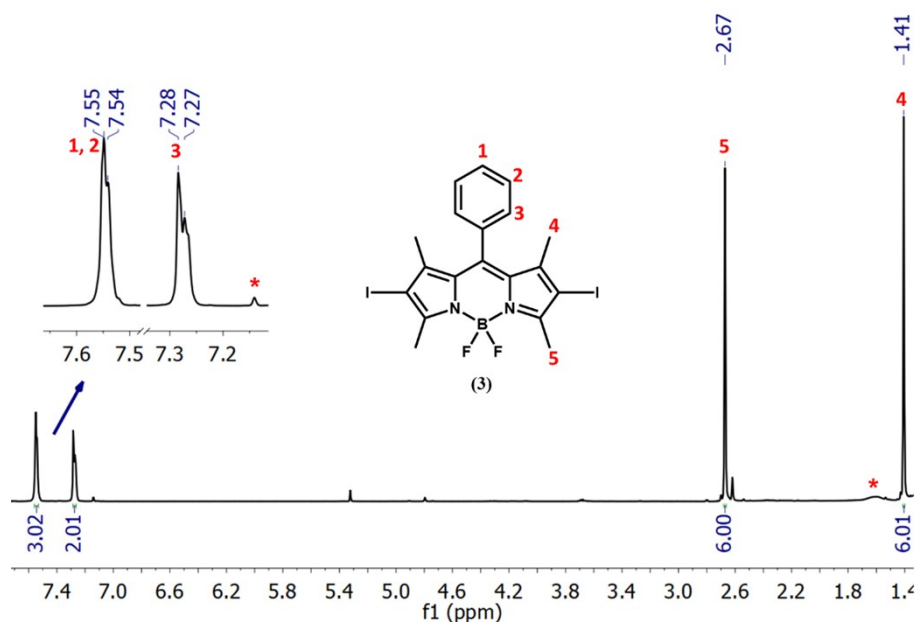


Fig. S9. ^1H -NMR spectrum of compound 3 in CDCl_3 .

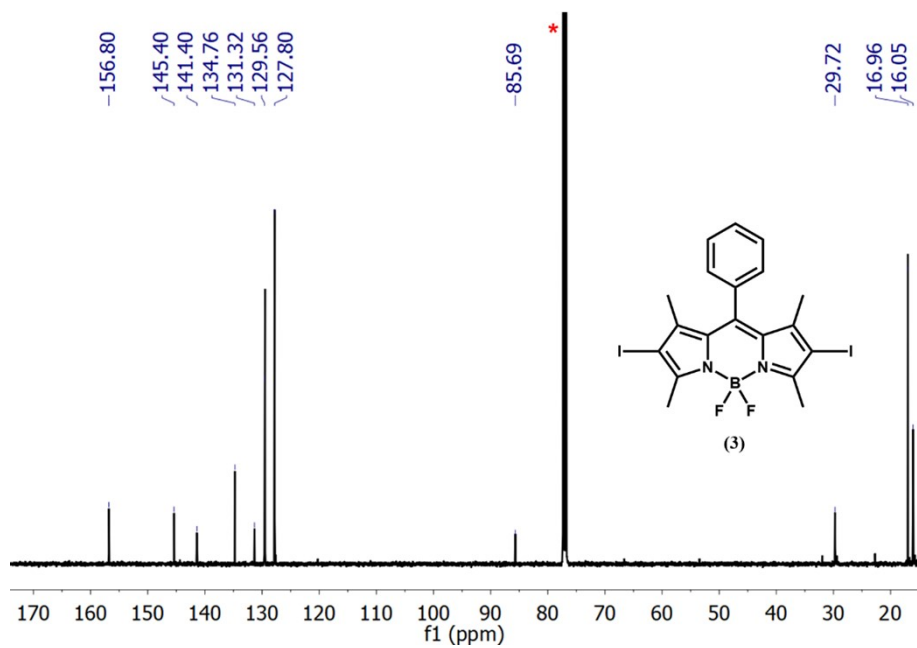


Fig. S10. ^{13}C -NMR spectrum of compound **3** in CDCl_3 .

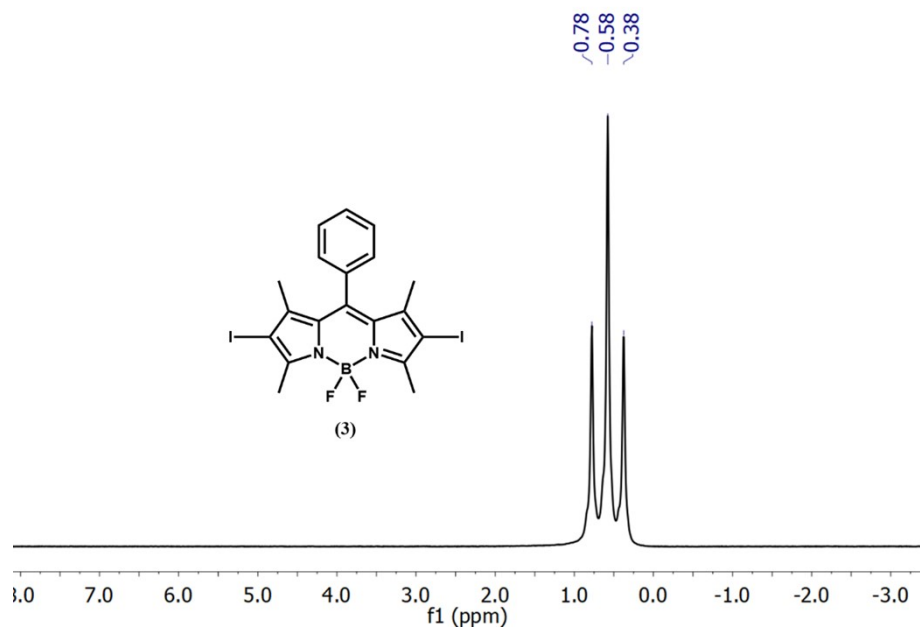


Fig. S11. ^{11}B -NMR spectrum of compound **3** in CDCl_3 .

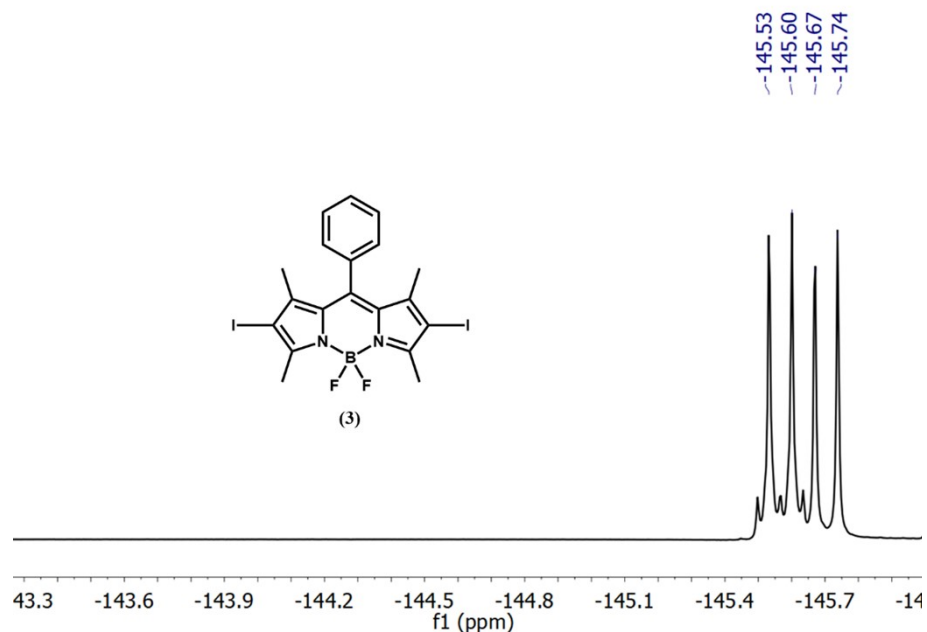
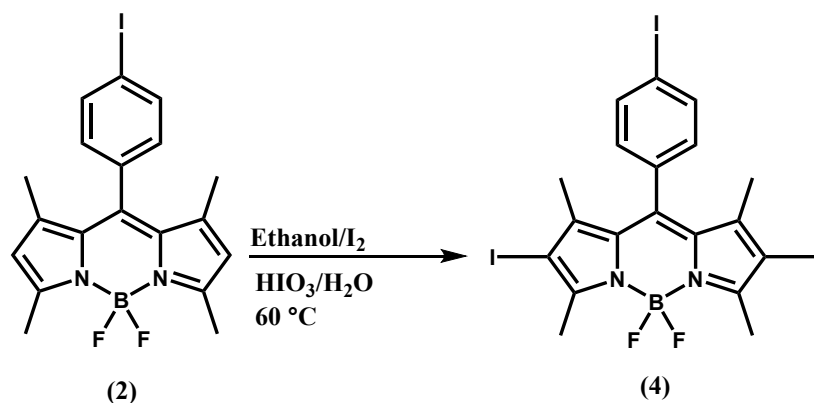


Fig. S12. ^{19}F -NMR spectrum of compound **3** in CDCl_3 .

1.4. 4,4'-Difluoro-8-(iodophenyl)-2,6-diiodo-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (4)



A dark red solid. Chemical formula $\text{C}_{19}\text{H}_{16}\text{BF}_2\text{I}_3\text{N}_2$, (187 mg, 60%). FT-IR (ν , cm^{-1}): 3055-3037 (Aromatic-CH), 2962-2851 (Aliphatic-CH), 1454 (B-N). ^1H NMR (500 MHz, CDCl_3) δ , ppm: 7.91 (d, 2H, Ar-H, J :7.80 Hz), 7.04 (d, 2H, Ar-H, J :7.63 Hz), 2.66 (s, 6H, - CH_3), 1.45 (s, 6H, - CH_3). ^{13}C NMR (125 MHz, CDCl_3) δ , ppm: 157.7, 145.15, 139.71, 138.71, 134.31, 131.03, 129.75, 95.39, 85.98, 17.30, 16.09. ^{11}B NMR (160 MHz, CDCl_3) δ , ppm: 0.53 (t, J : 64.06 Hz). ^{19}F NMR (470 MHz, CDCl_3) δ , ppm: -145.64 (q, J : 95.91 Hz).

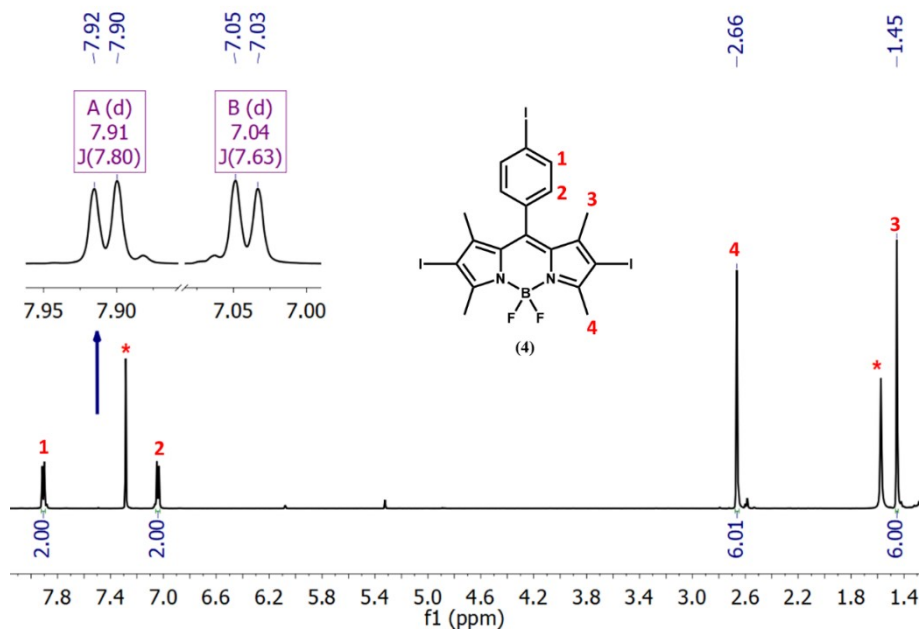


Fig. S13. ¹H-NMR spectrum of compound 4 in CDCl₃.

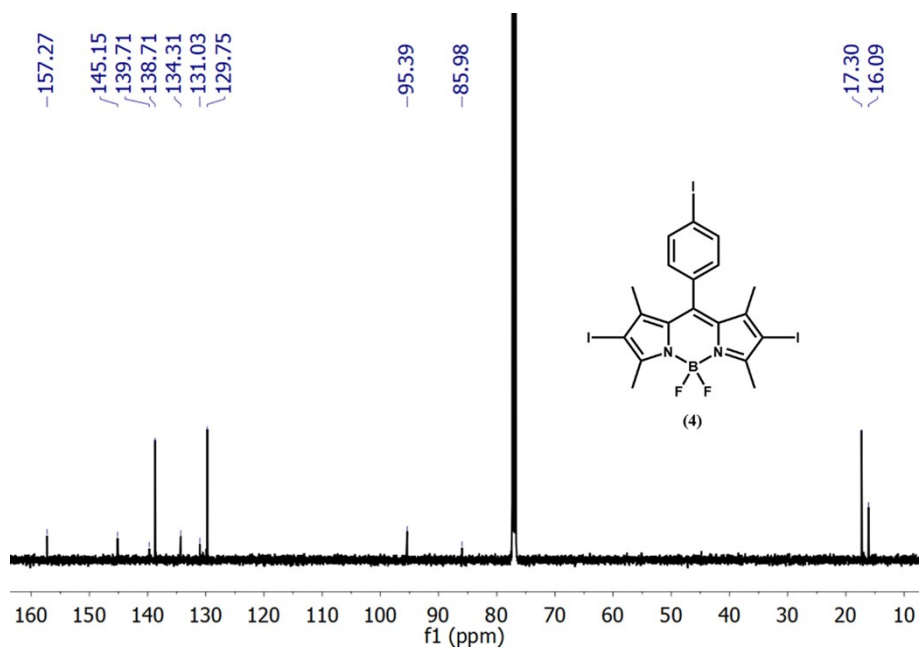


Fig. S14. ¹³C-NMR spectrum of compound 4 in CDCl₃.

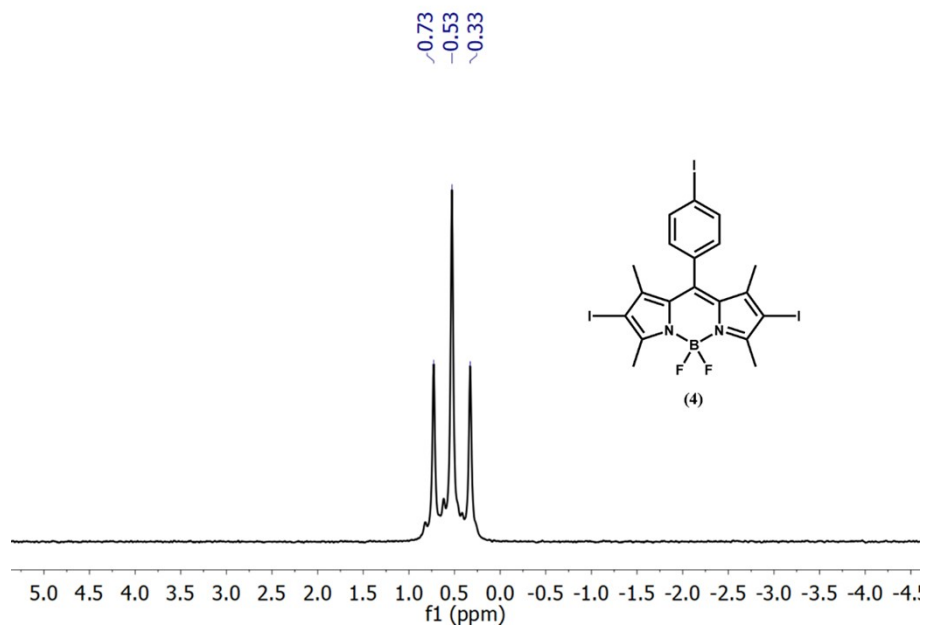


Fig. S15. ^{11}B -NMR spectrum of compound **4** in CDCl_3 .

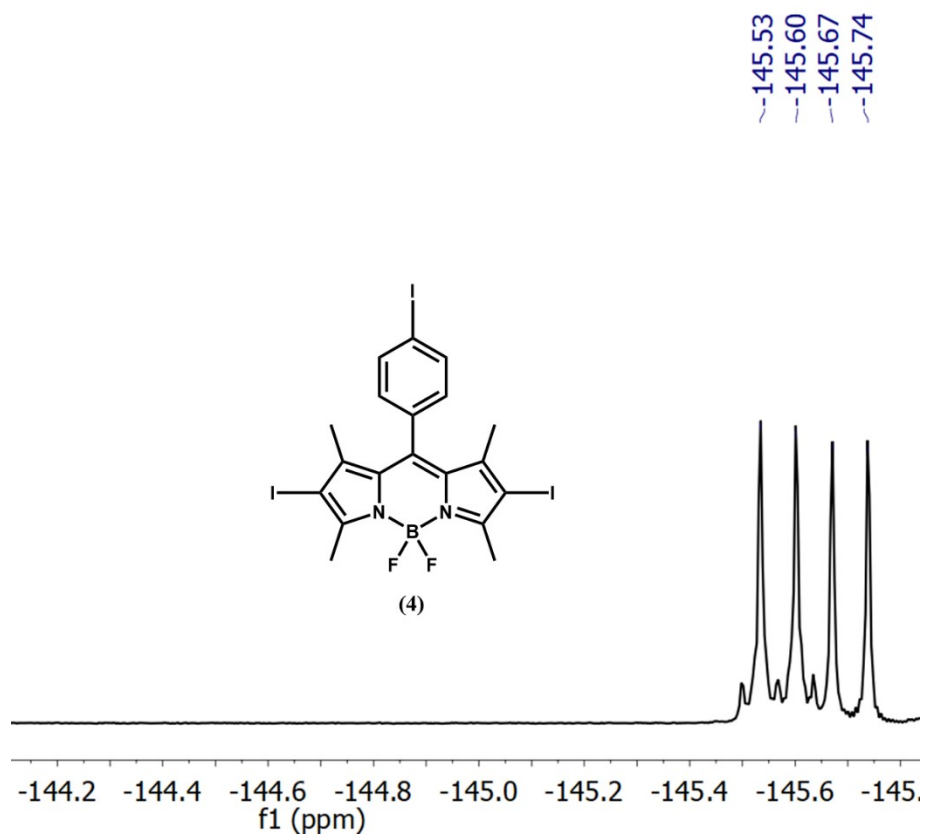
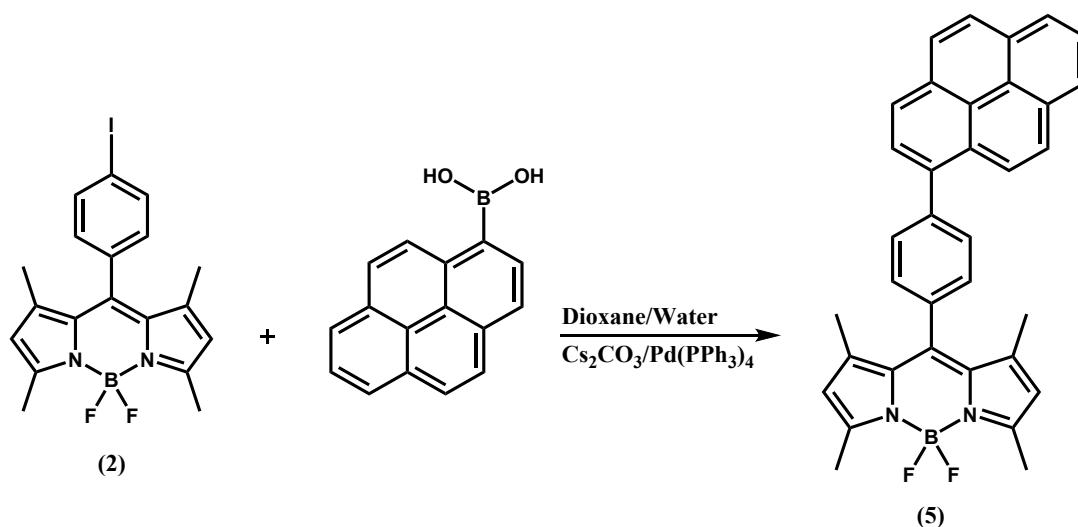


Fig. S16. ^{19}F -NMR spectrum of compound **4** in CDCl_3 .

1.5. 4,4'-Difluoro-8-(4-pyrenyl)-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (5)



An orange solid. Chemical formula C₃₅H₂₇BF₂N₂, (93.21 mg, 80%). FT-IR (ν, cm⁻¹): 3107-3030 (Aromatic-CH), 2962-2848 (Aliphatic-CH), 1467 (B-N). ¹H NMR (500 MHz, CDCl₃) δ, ppm: 8.29-8.22 (m, 3H, Pyrene-H), 8.16-8.05 (m, 6H, Pyrene-H), 7.79 (d, 2H, Ar-H, *J*:7.62 Hz), 7.52 (d, 2H, Ar-H, *J*:7.60 Hz), 6.09 (s, 2H, Pyrrole-H), 2.63 (s, 6H, -CH₃), 1.65 (s, 6H, -CH₃). ¹³C NMR (125 MHz, CDCl₃) δ, ppm: 155.64, 143.14, 142.12, 136.67, 134.01, 131.52, 131.34, 130.94, 128.44, 128.14, 127.94, 127.73, 127.63, 127.44, 126.20, 125.43, 125.08, 125.04, 124.71, 124.68, 121.40, 14.65. ¹¹B NMR (160 MHz, CDCl₃) δ, ppm: 0.86 (t, *J*: 65.44 Hz). ⁹F NMR (470 MHz, CDCl₃) δ, ppm: -146.22 (q, *J*: 99.51 Hz).

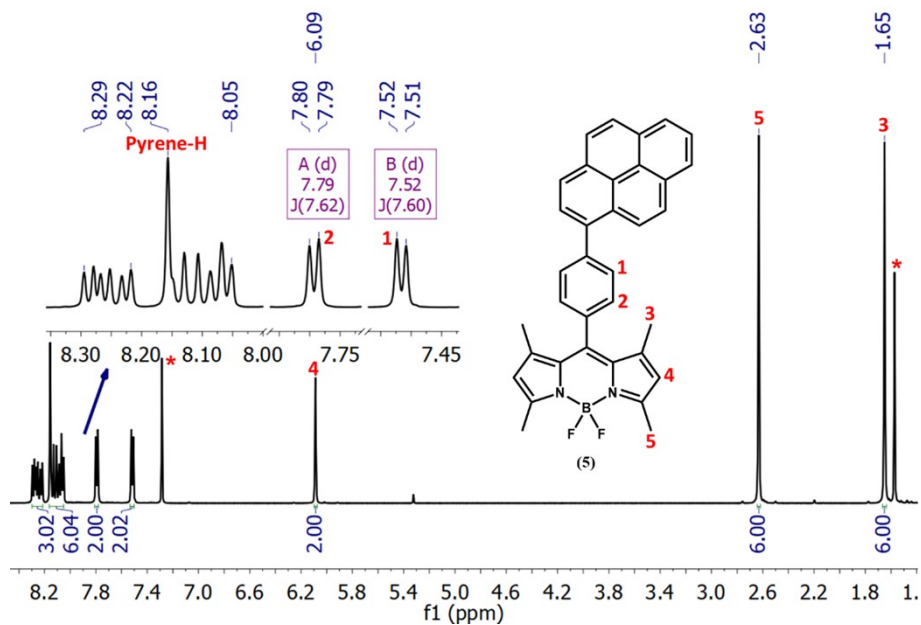


Fig. S17. ¹H-NMR spectrum of compound **5** in CDCl₃.

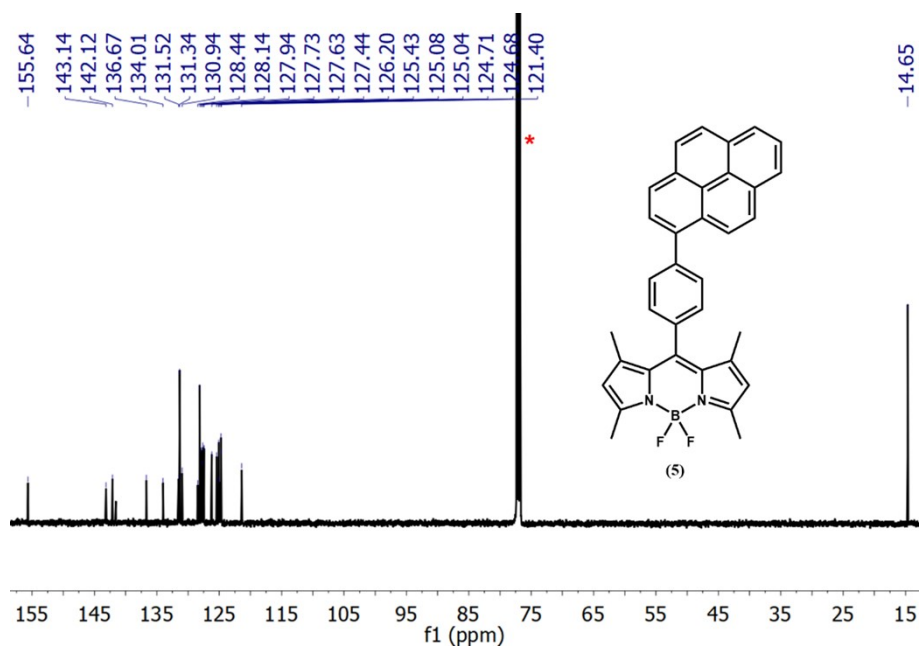


Fig. S18. ¹³C-NMR spectrum of compound **5** in CDCl₃.

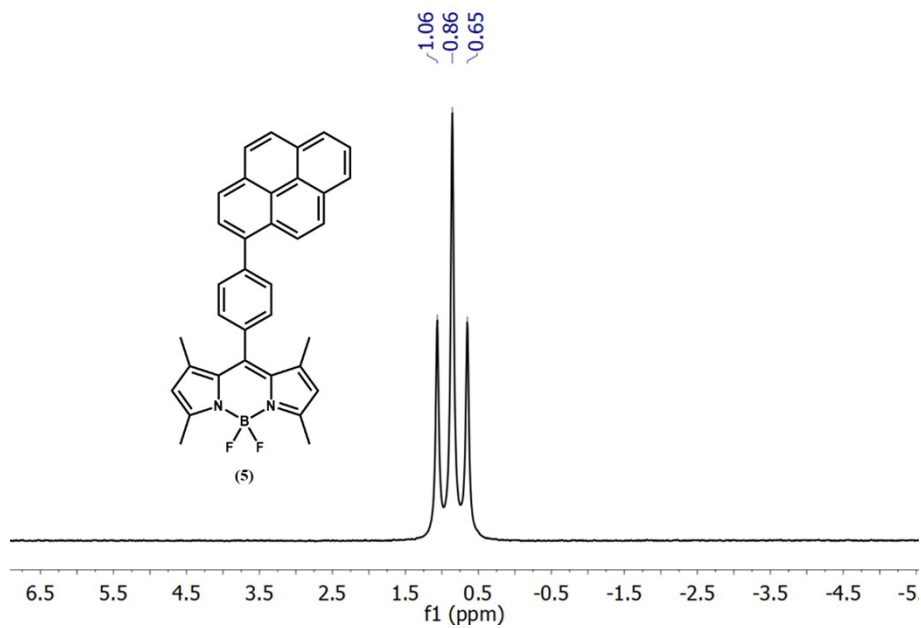


Fig. S19. ^{11}B -NMR spectrum of compound **5** in CDCl_3 .

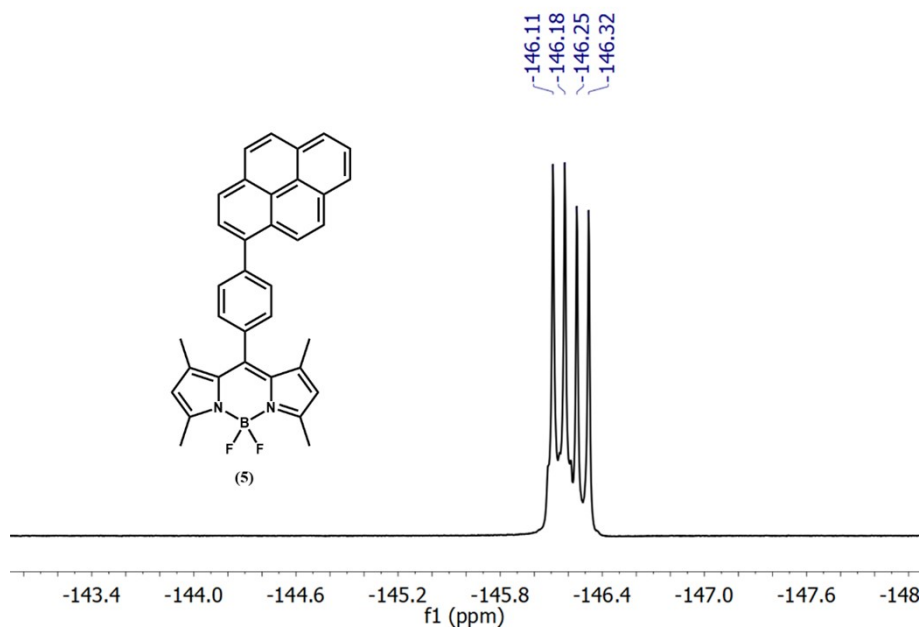


Fig. S20. ^{19}F -NMR spectrum of compound **5** in CDCl_3 .

2. Characterization of novel pyrene-BODIPY compounds (5♦, 6, and 7)

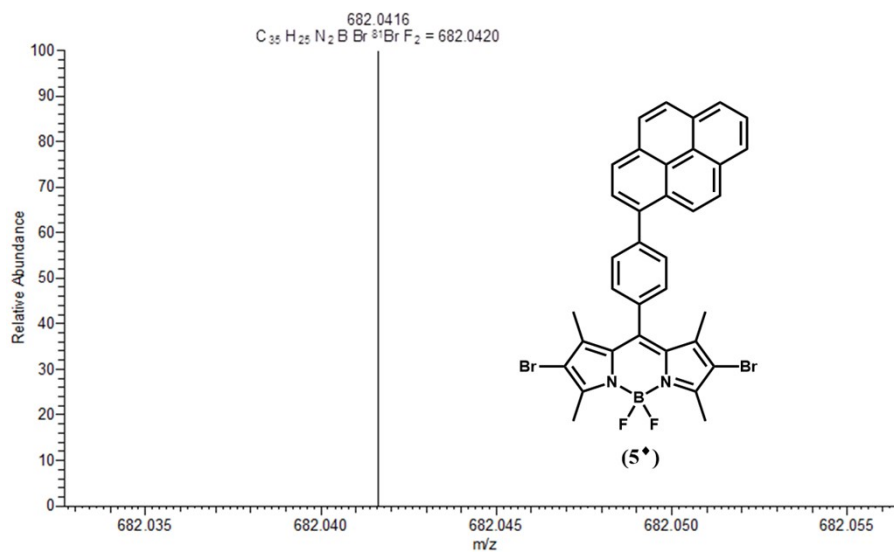


Fig. S21. HRMS spectrum of compound 5♦.

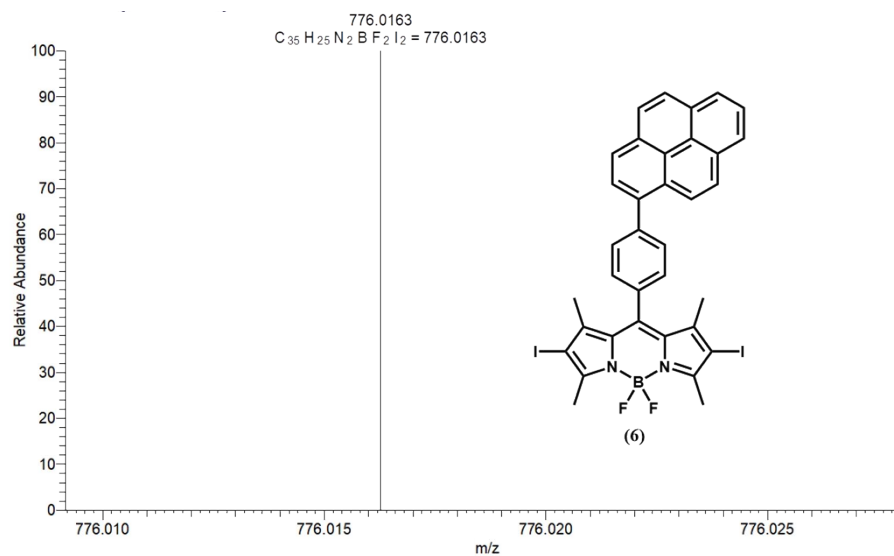


Fig. S22. HRMS spectrum of compound 6.

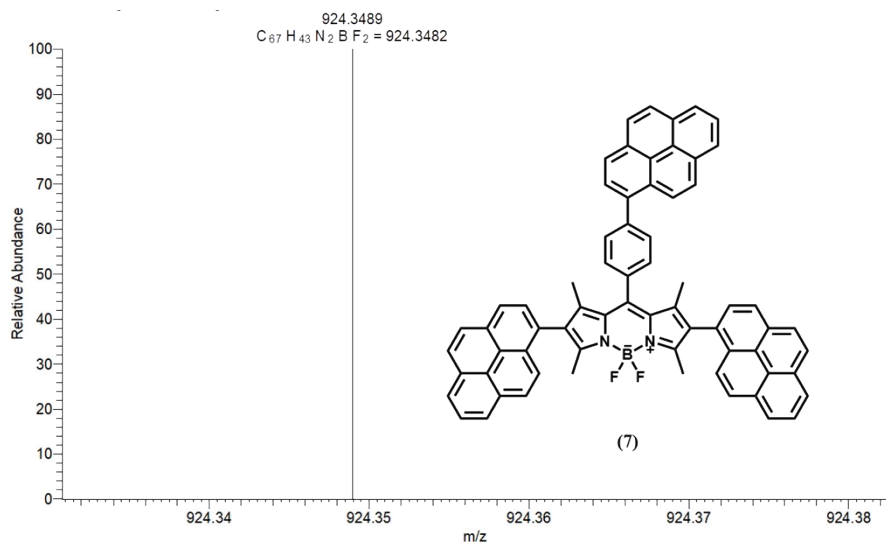


Fig. S23. HRMS spectrum of compound 7.

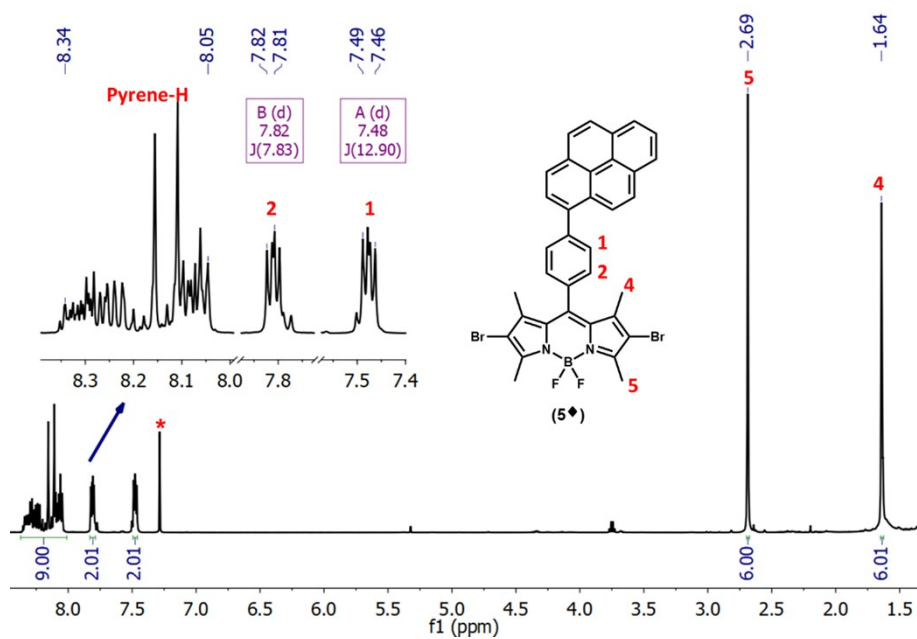


Fig. S24. ¹H-NMR spectrum of compound 5* in CDCl₃.

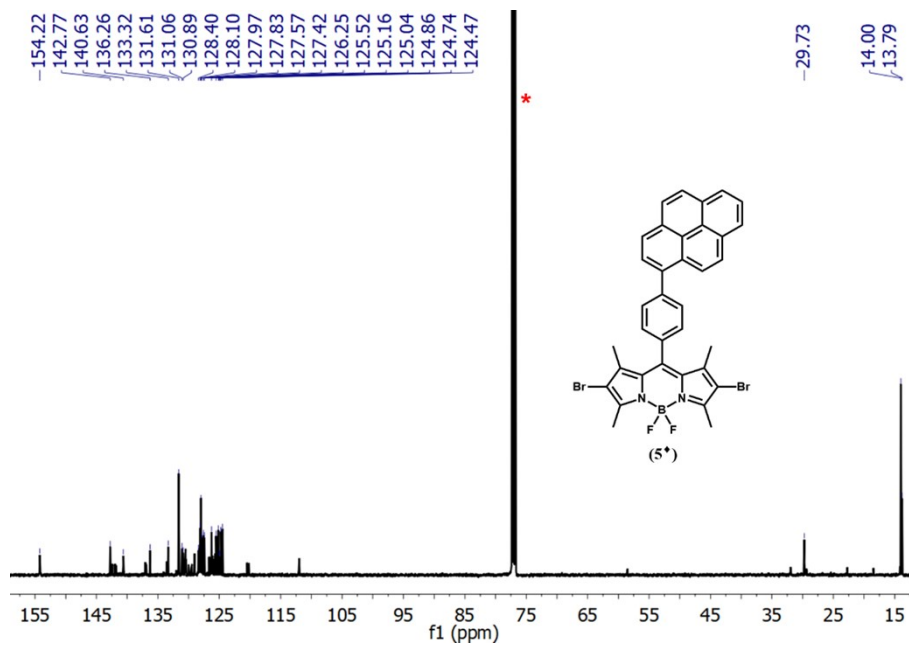


Fig. S25. ¹³C-NMR spectrum of compound **5*** in CDCl₃.

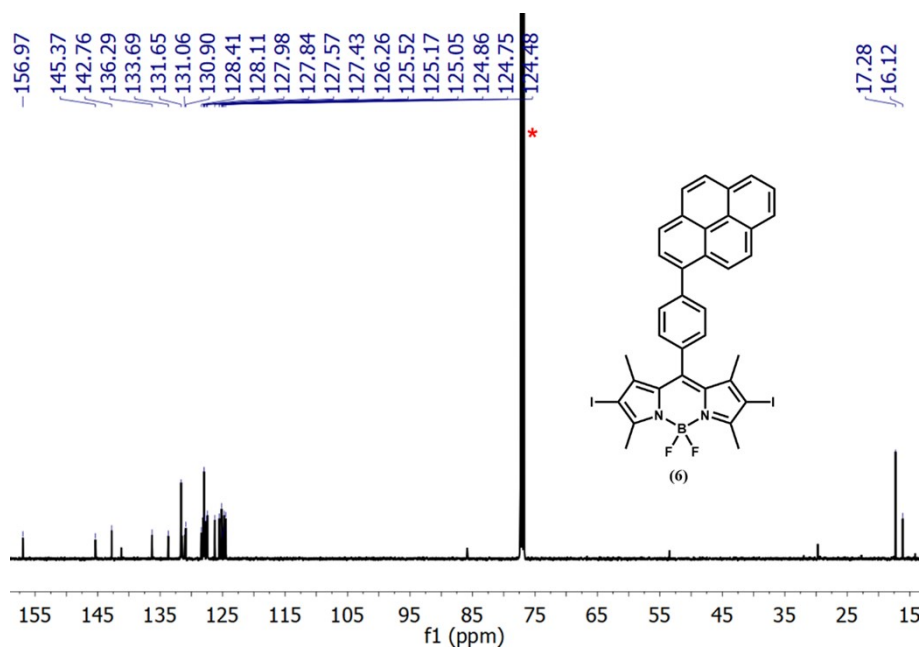


Fig. S26. ¹³C-NMR spectrum of compound **6** in CDCl₃.

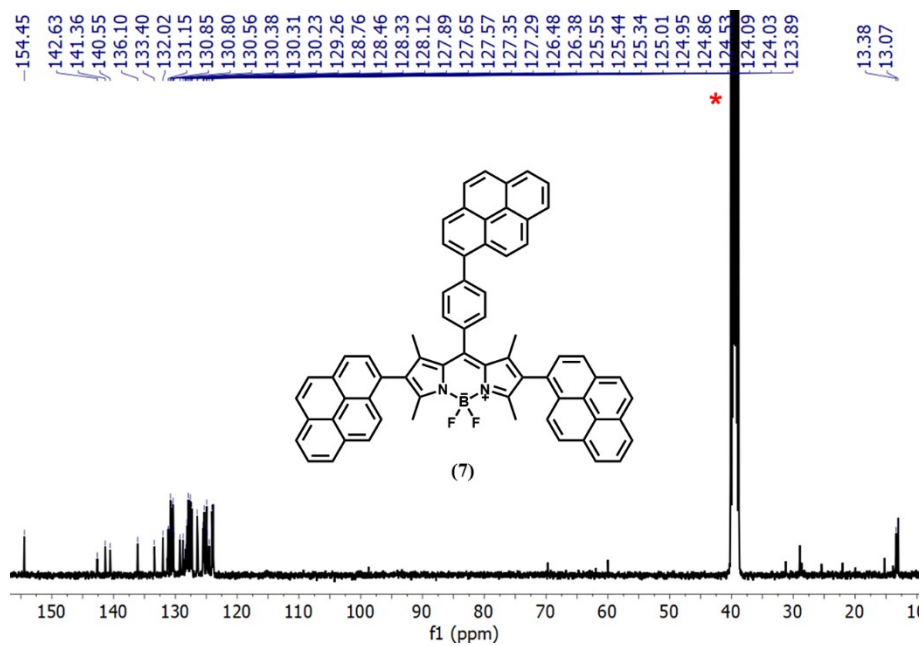


Fig. S27. ^{13}C -NMR spectrum of compound **7** in DMSO-d_6 .

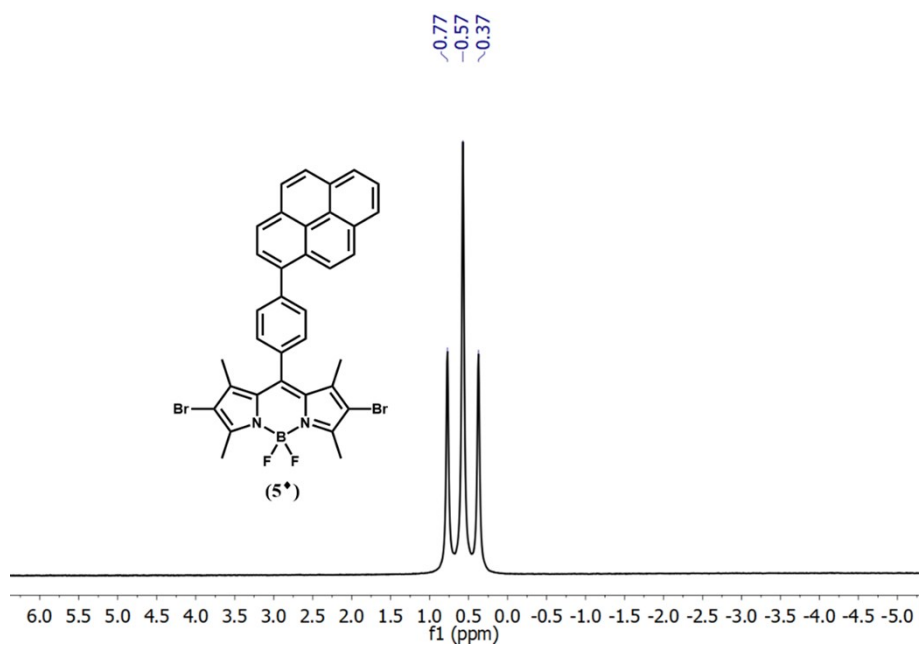


Fig. S28. ^{11}B -NMR spectrum of compound **5*** in CDCl_3 .

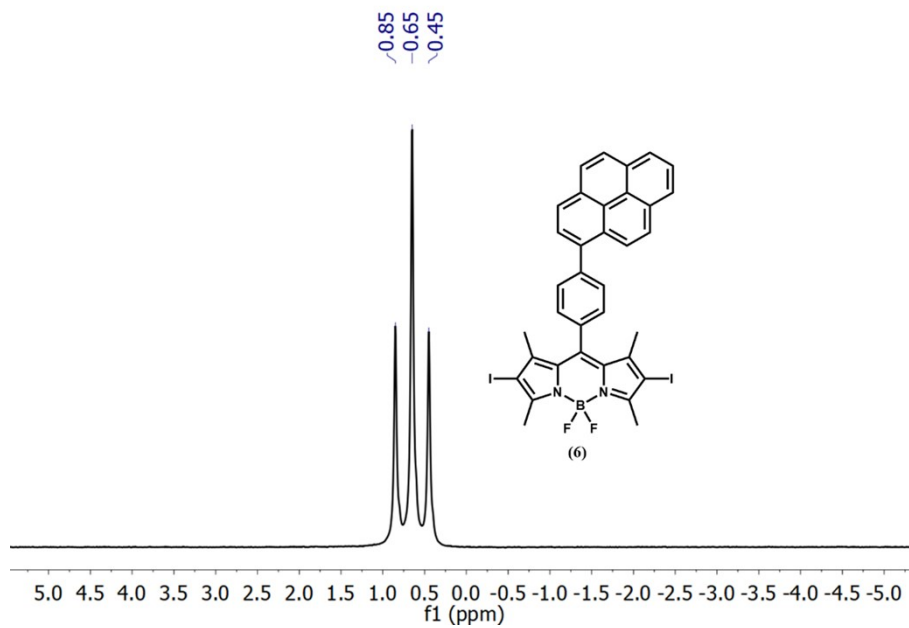


Fig. S29. ^{11}B -NMR spectrum of compound **6** in CDCl_3 .

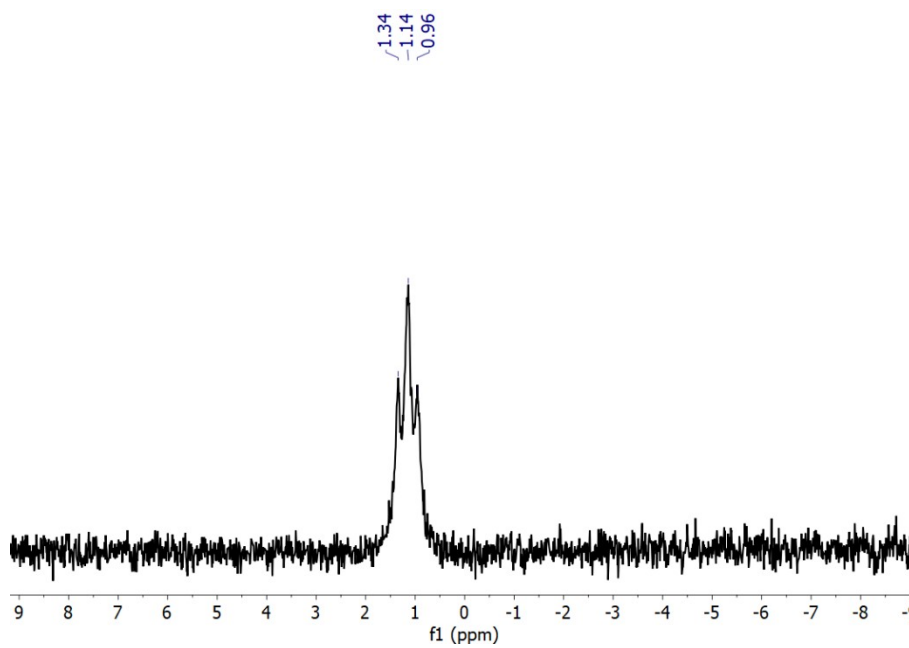


Fig. S30. ^{11}B -NMR spectrum of compound **7** in DMSO-d_6 .

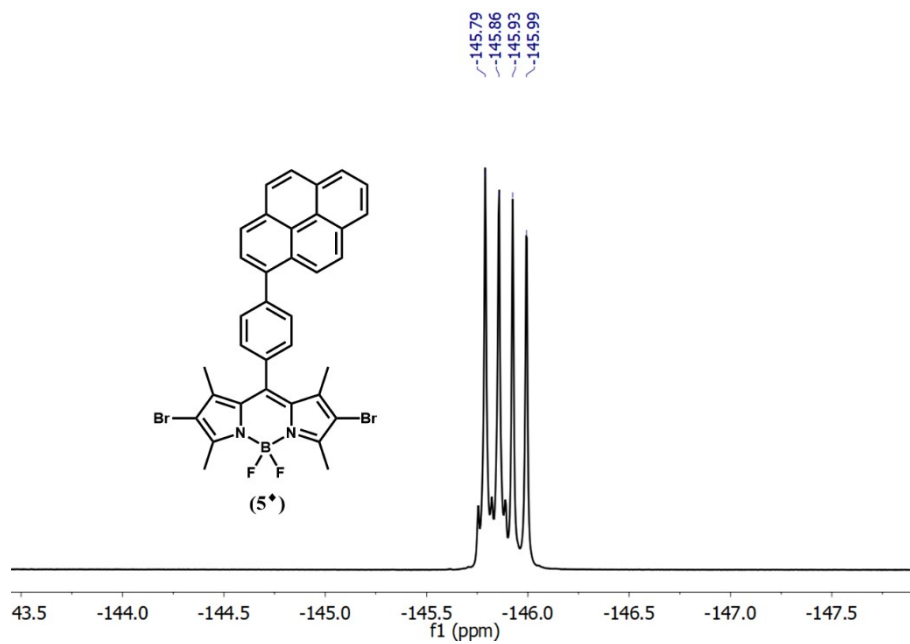


Fig. S31. ^{19}F -NMR spectrum of compound **5*** in CDCl_3 .

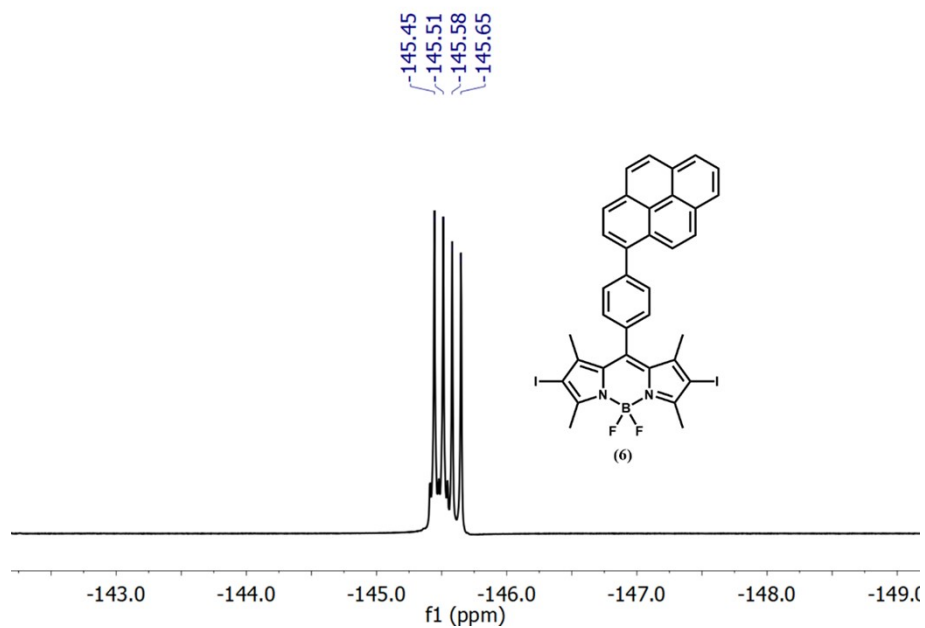


Fig. S32. ^{19}F -NMR spectrum of compound **6** in CDCl_3 .

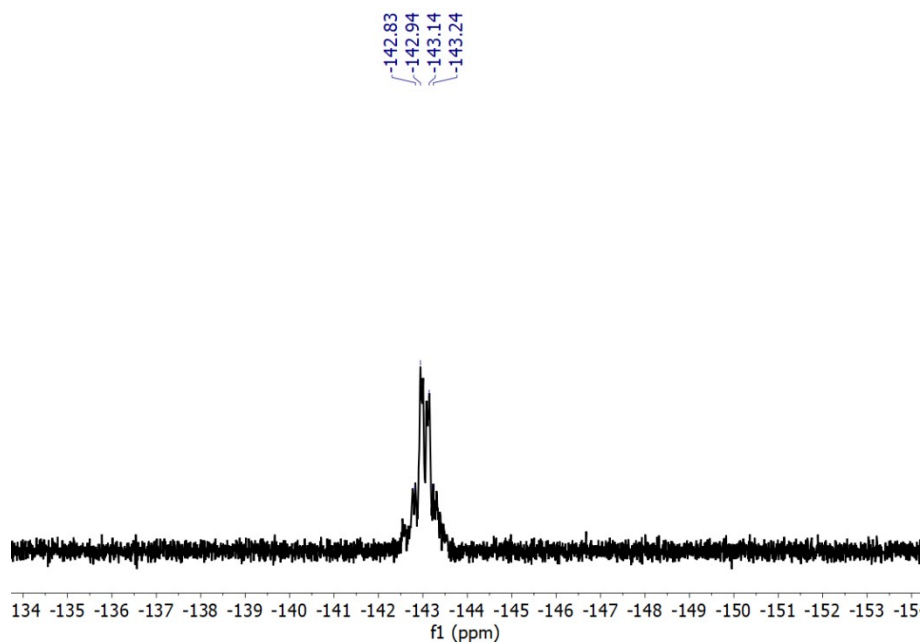


Fig. S33. ^{19}F -NMR spectrum of compound **7** in DMSO-d_6 .

3. Electrochemical characterizations

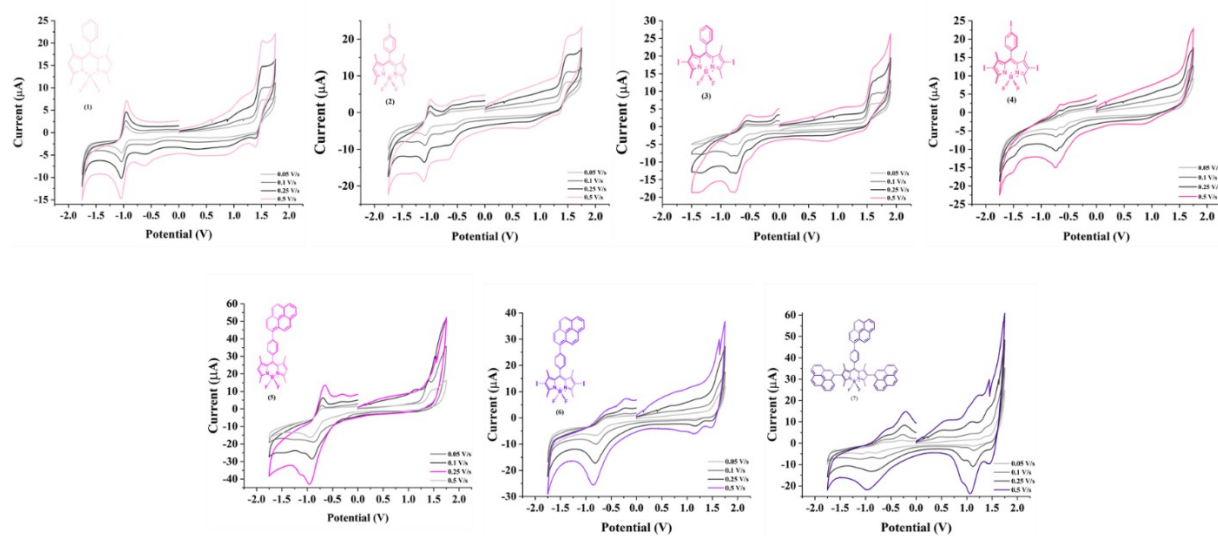


Fig. S34. Cyclic voltammograms at different scan rates of synthesized BODIPY compounds (**1-7**) in DCM with 0.1 M TBAPF₆ as supporting electrolyte.

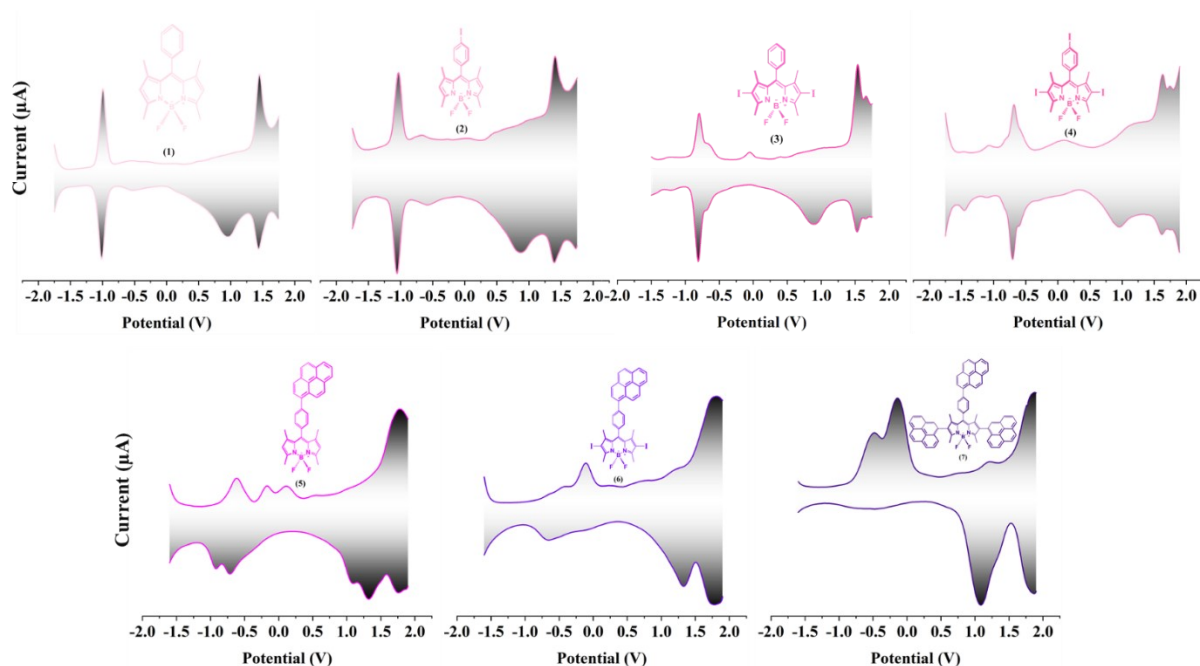


Fig. S35. Square wave voltammograms of synthesized BODIPY compounds (**1-7**) in DCM with 0.1 M TBAPF₆ as supporting electrolyte, scan rate 0.05 V/s.

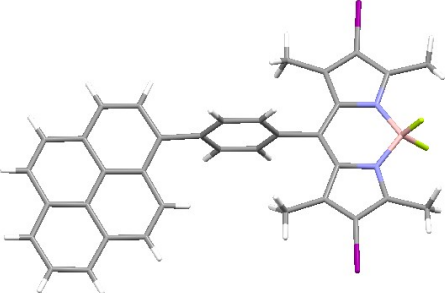
4. X-Ray study of compound 6

All starting BODIPY compounds (**1-4**) and pyrene-BODIPY (Halogen free, **5**) were not original, and their crystal structures were given in the literature before. The single crystal form of targeted pyrene-BODIPY compound **7** could not be obtained. Only compound **6** was crystallized and solved their crystal structure. The crystal structure of compound **6** was solved using single-crystal X-ray analysis and crystallographic data-refinement details given in Table S1. Suitable crystals for the X-ray diffraction study of compound **6** were obtained from recrystallization by dichloromethane-hexane solvent mixture.

The structure of compound **6** was shown in a triclinic crystal system, *P1*. As checked in the crystal structure of compound **6**, the BODIPY backbone and pyrene groups were observed at the same plane while the *meso*-phenyl group was orthogonal position to both of BODIPY center and pyrene moiety. These structural features were also compatible with the corresponding literatures including styryl-BODIPY compounds ¹⁻³.

In addition, as the solid crystal structure of compound **1** is examined, it is seen that it has a strong pi-pi interaction (Fig. S36) and the distance between these 2 pyrene groups is 3.904 Å. This interaction is caused by the effect of electron-rich pyrene groups located on the face with face each other. These interactions hold the 2 molecules in a single packaging.

Table S1. Crystal structure and crystal data of BODIPY compound **6**

Compound	
CCDC Number	2213850
Parameters	Compound 6
Empirical formula	C ₃₅ H ₂₅ BF ₂ I ₂ N ₂
Formula weight	776.18
Temperature/K	296.15
Crystal system	Triclinic
Space group	<i>P</i> 1
<i>a</i> /Å	10.350 (3)
<i>b</i> /Å	11.258 (3)
<i>c</i> /Å	15.292 (5)
α /°	75.396 (8)
β /°	75.221 (8)
γ /°	68.248 (7)
Volume/Å ³	1575.4 (7)
<i>Z</i>	2
ρ_{calc} /g/cm ³	1.636
μ /mm ⁻¹	2.04
F(000)	756
Crystal size/mm ³	0.17 × 0.05 × 0.04
Radiation	Mo K α (λ = 0.71073 Å)
2 θ range for data collection/°	1.4 to 25
Index ranges	-12 ≤ <i>h</i> ≤ 12, 13 ≤ <i>k</i> ≤ 13, -18 ≤ <i>l</i> ≤ 18
Reflections collected	11935
Independent reflections	5464 [R _{int} = 0.044, R _{sigma} = 0.0546]
Goodness-of-fit on F ²	1.02
Final R indexes [I ≥ 2 σ (I)]	R ₁ = 0.0443, wR ₂ = 0.1102
Largest diff. peak/hole	0.69, -0.77

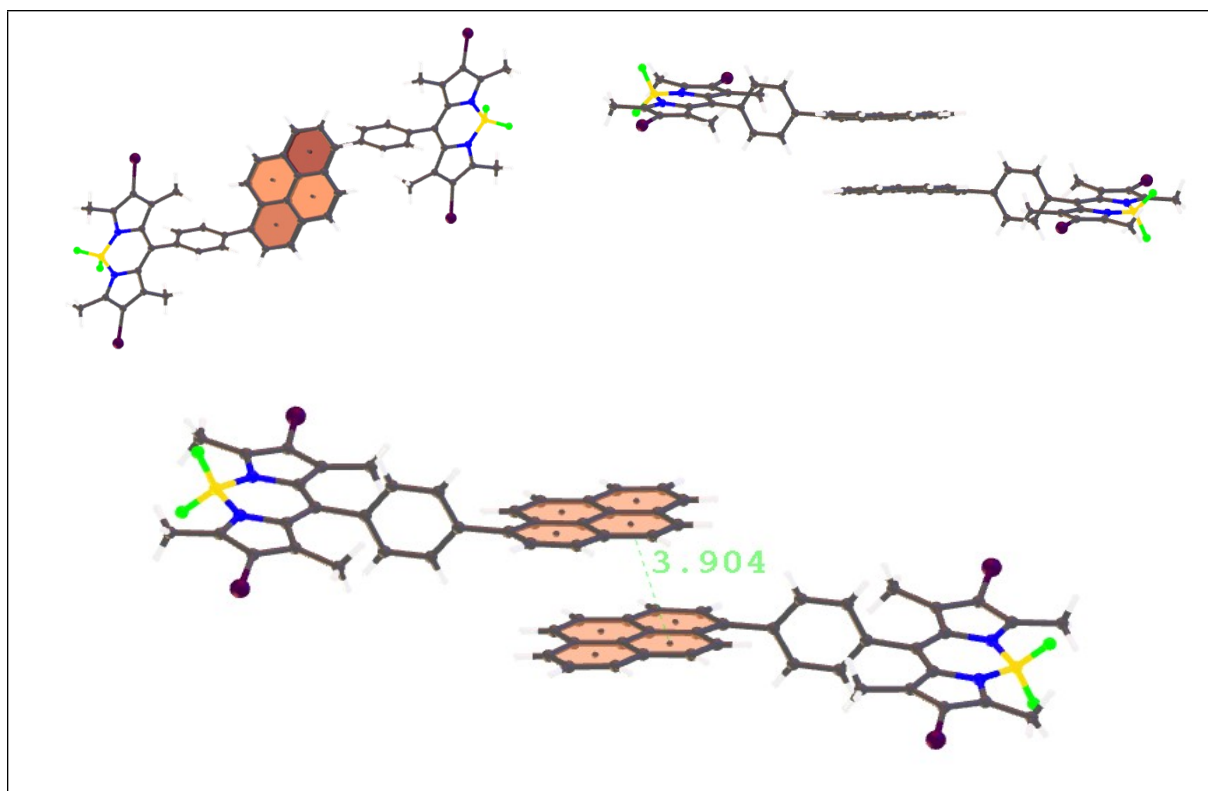


Fig. S36. π - π interaction of the solid-state crystal form of iodinated pyrene-BODIPY compound 6.

References

1. J. Tao, D. Sun, L. Sun, Z. Li, B. Fu, J. Liu, L. Zhang, S. Wang, Y. Fang and H. Xu, *Dyes and Pigments*, 2019, **168**, 166-174.
2. C. O. Obondi, G. N. Lim, P. A. Karr, V. N. Nesterov and F. D'Souza, *Physical Chemistry Chemical Physics*, 2016, **18**, 18187-18200.
3. D. Öztürk, İ. Ömeroğlu, B. Köksoy, C. Göl and M. Durmuş, *Dyes and Pigments*, 2022, **205**, 110510.