

Supporting Information

Abundant heterointerfaces in CoS₂/MoS₂ nanosheet array electrocatalysts for enhanced oxygen evolution reaction

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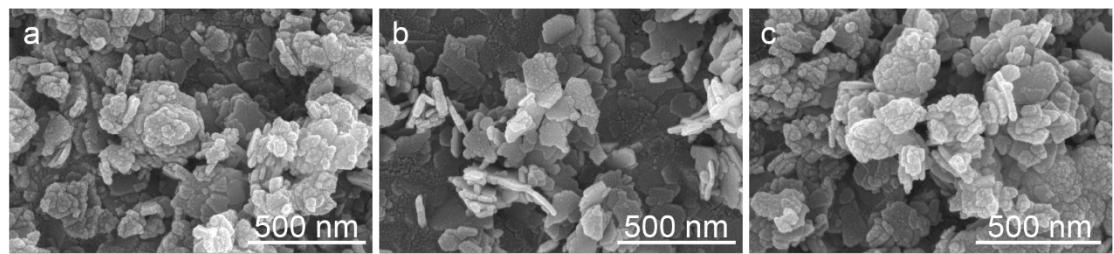


Fig. S1 SEM images of CoS₂/MoS₂ with different Co/Mo ratios: (a) CoS₂/MoS₂-1:2; (b) CoS₂/MoS₂-1:1; (c) CoS₂/MoS₂-2:1

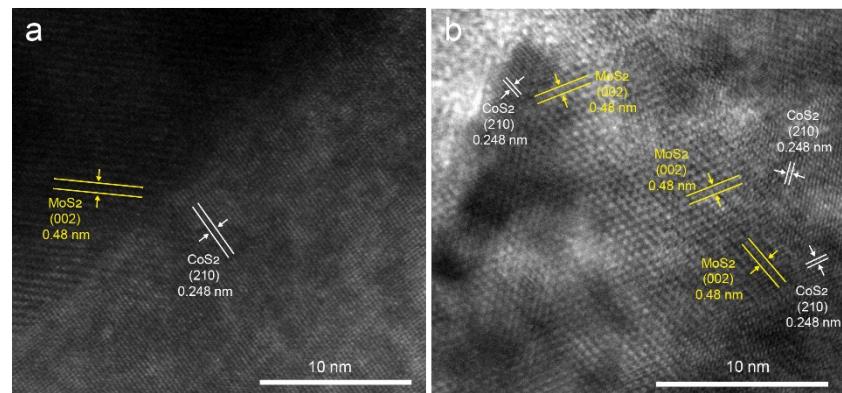


Fig. S2 HRTEM images of the CoS₂/MoS₂-1:1 nanosheets

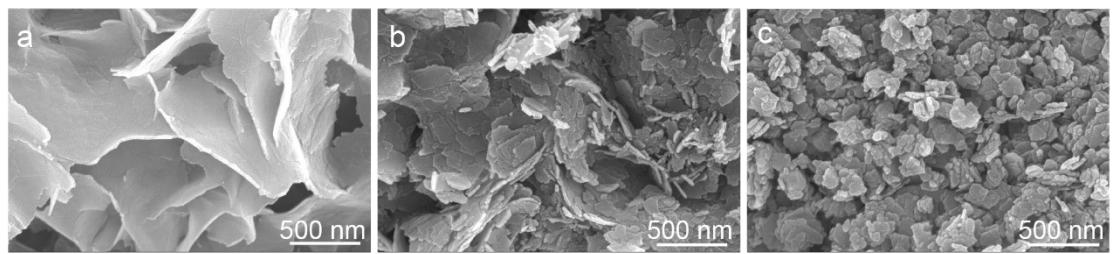


Fig. S3 SEM images of $\text{CoS}_2/\text{MoS}_2$ -1:1 obtained at different hydrothermal time: (a) 2 h; (b) 4 h; (c) 6 h

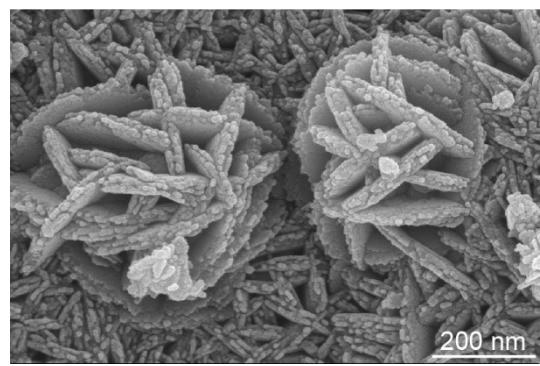


Fig. S4 SEM image of Co/Mo oxide with Co/Mo ratio of 1:1

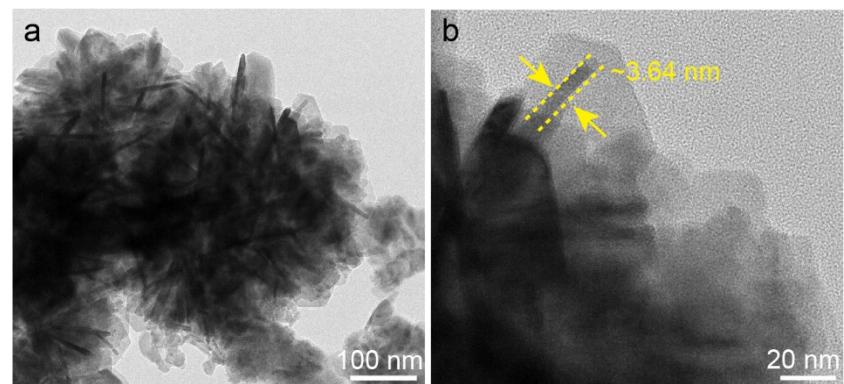


Fig. S5 TEM images of Co/Mo oxide with Co/Mo ratio of 1:1

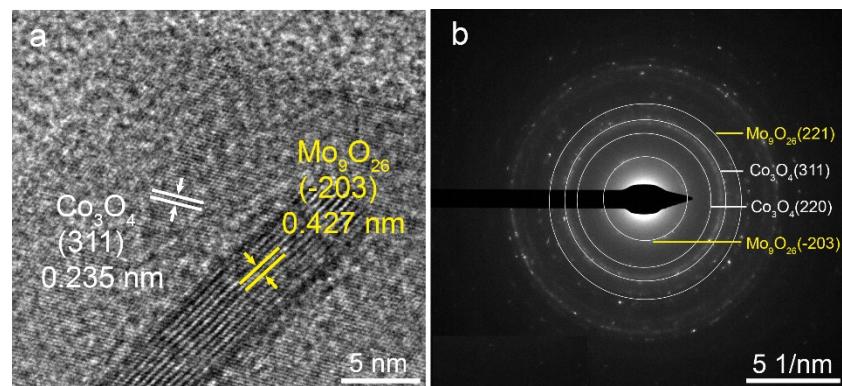


Fig. S6 HRTEM and SAED images of Co/Mo oxide with Co/Mo ratio of 1:1

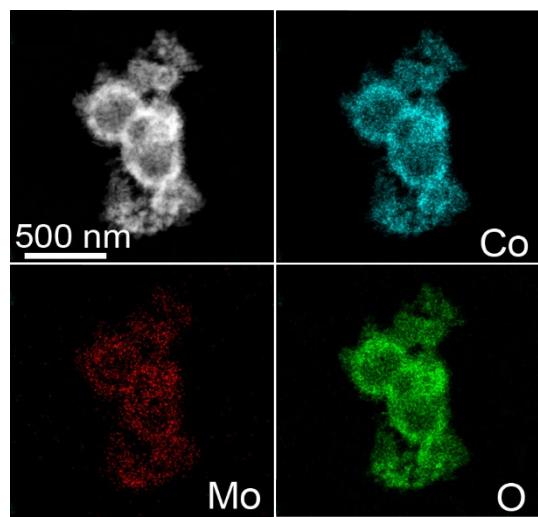


Fig. S7 Elemental mappings of Co/Mo oxide with Co/Mo ratio of 1:1

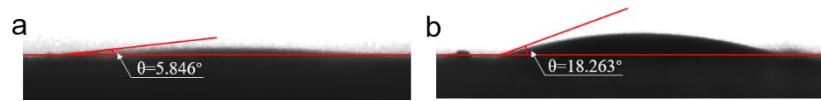


Fig. S8 Contact angle testing: (a) $\text{CoS}_2/\text{MoS}_2$ -1:1; (b) Co/Mo oxide

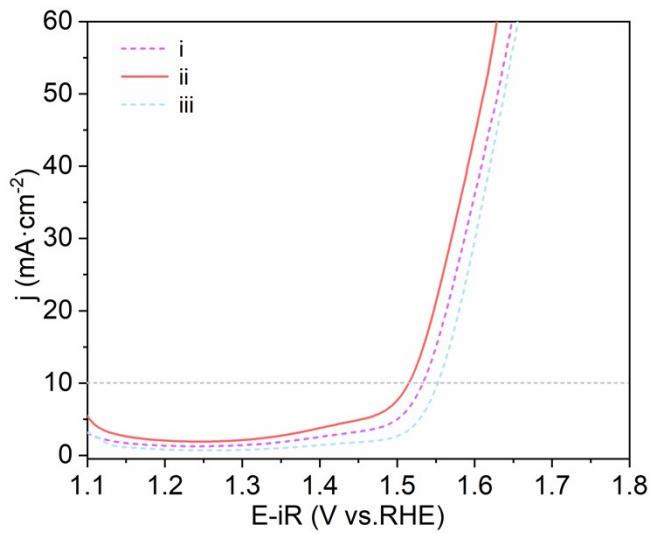


Fig. S9 LSV curves of $\text{CoS}_2/\text{MoS}_2$ -1:1 with different loads

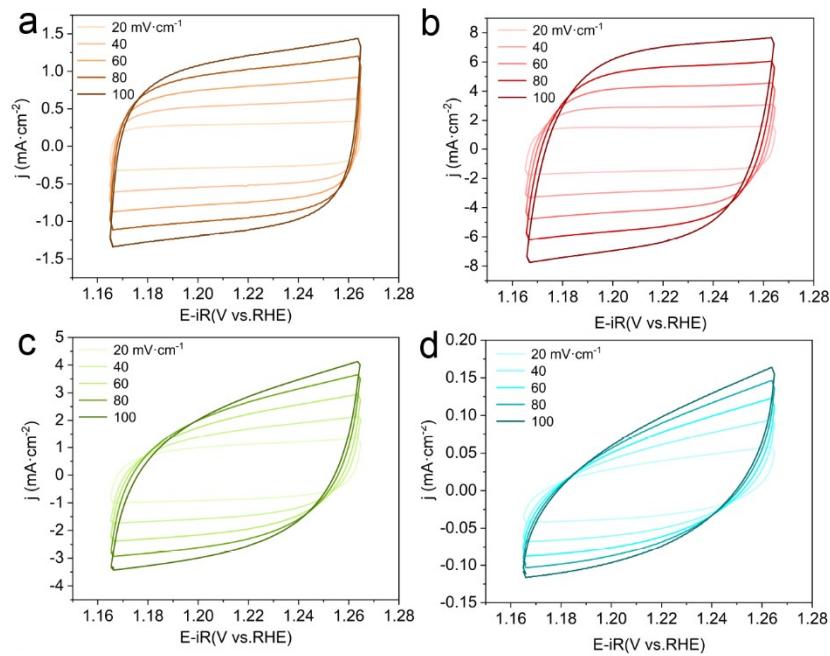


Fig. S10 CVs at different scan rates of in a potential window where no Faradaic processes occur (1.16–1.26 V vs. RHE) for: (a) CoS₂/MoS₂-1:2, (b) CoS₂/MoS₂-1:1, (c) CoS₂/MoS₂-2:1, (d) Co/Mo oxide

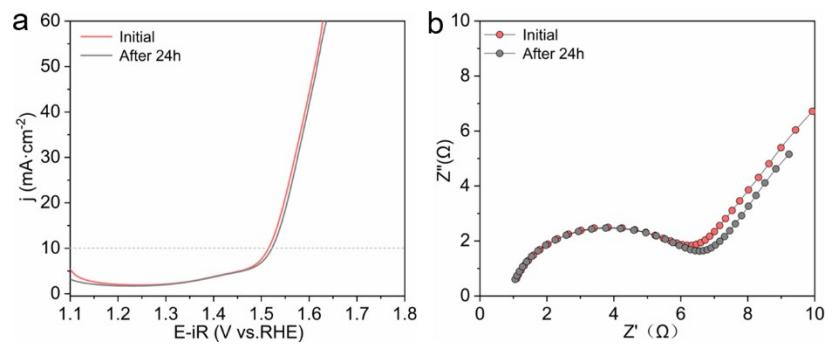


Fig. S11 (a) LSV curves, (b) Nyquist plots of $\text{CoS}_2/\text{MoS}_2$ -1:1 before and after stability test

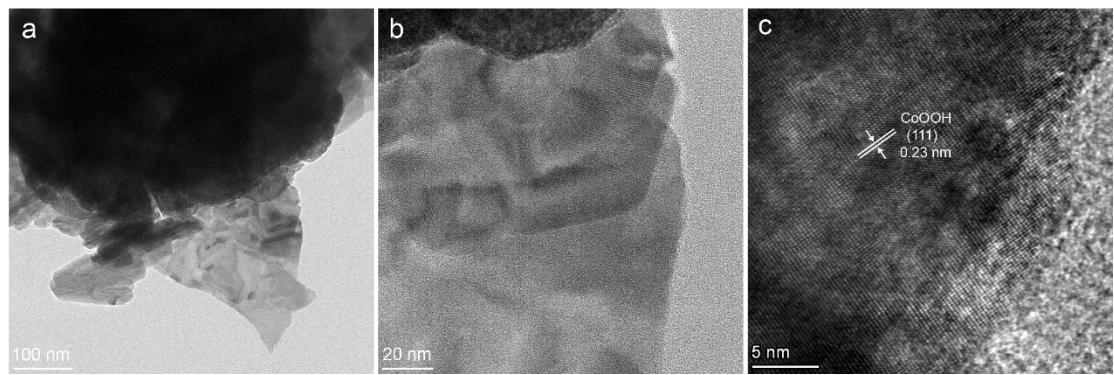


Fig. S12 TEM images of the $\text{CoS}_2/\text{MoS}_2$ -1:1 sample after reaction

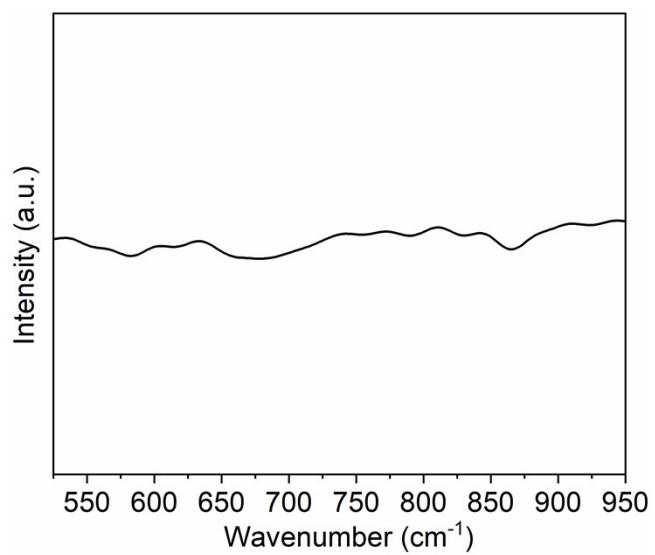


Fig. S13 FTIR spectra of $\text{CoS}_2/\text{MoS}_2$ -1:1 after reaction

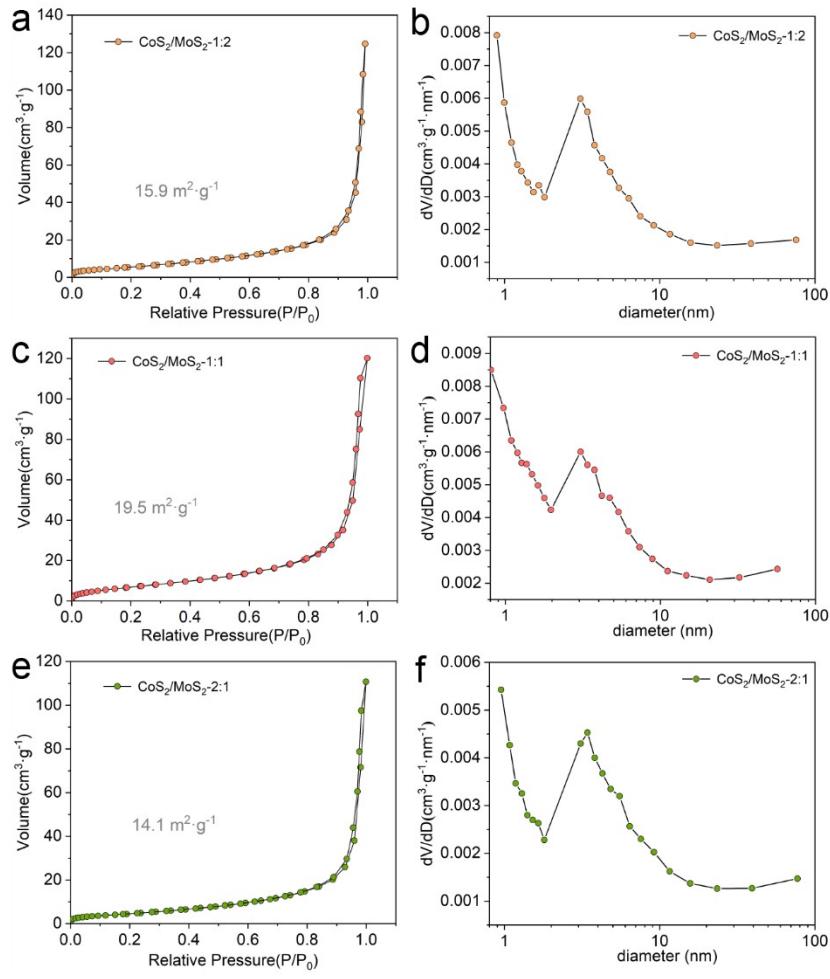


Fig. S14 (a) Nitrogen adsorption-desorption isotherm and (b) pore distribution curves of CoS₂/MoS₂-1:2; (c) Nitrogen adsorption-desorption isotherm and (d) pore distribution curves of CoS₂/MoS₂-1:1; (e) Nitrogen adsorption-desorption isotherm and (f) pore distribution curves of CoS₂/MoS₂-2:1

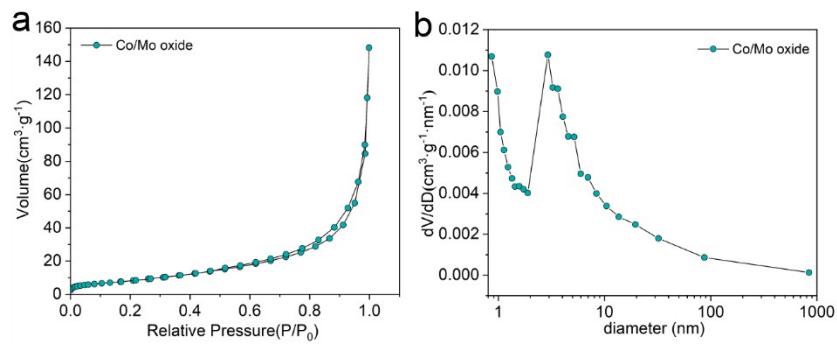


Fig. S15 (a) Nitrogen adsorption-desorption isotherm and (b) pore distribution curves of Co/Mo oxide with Co/Mo ratio of 1:1

Table S1 Mass activity of different samples under certain voltage

Sample	Mass activity($A \cdot g^{-1}$)
Co/Mo oxide	4.09
CoS ₂ /MoS ₂ -1:2	14.49
CoS ₂ /MoS ₂ -1:1	31.50
CoS ₂ /MoS ₂ -2:1	22.16

Table S2 Comparison of Tafel slope and overpotential ($10 \text{ mA}\cdot\text{cm}^{-2}$) with the electrocatalysts in literature

Materials	Electrolyte	Overpotential /mV	Tafel slope /mV·dec ⁻¹	Ref.
MoS ₂ /NiS	1.0 M KOH	350	108	[S1]
CoS ₂ /MoS ₂	1.0 M KOH	332	125	[S2]
CoS ₂ -5	1.0 M KOH	290	65.6	[S3]
CoS ₂ @MoS ₂	1.0 M KOH	332	37.5	[S4]
CoS ₂ -MoS ₂ MSHSs	1.0 M KOH	288	62.1	[S5]
CoS NF/CC	1.0 M KOH	310	73.4	[S6]
Fe-CoS ₂ /CC	1.0 M KOH	304	128	[S7]
MoS ₂ /CC	1.0 M KOH	503	131	[S8]
meso-Fe-MoS ₂ /CoMo ₂ S ₄	1.0 M KOH	290	65	[S9]
P-CoS	1.0 M KOH	340	73	[S10]
CoS ₂ /MoS ₂	1.0 M KOH	285	105.32	This work

References:

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