

Supporting Information

Enhancing the Photocatalytic Hydrogen Production Performance of CdS by Introducing Co-catalyst CoTPPBr₄ (7,8,17,18- Tetrabromo-5,10,15,20-tetraphenylporphyrin)

Zong Wang^a, Xinxin Wang^{a,b}, Kelai Chen^a, Haojun Yin^a, Huangsheng Su^a, Yundang Wu^b,

Chunlin Ni^{a,*}, Wei Liu^{a,*}

^a College of Materials and Energy, South China Agricultural University, Guangzhou 510642, China

^b National-Regional Joint Engineering Research Center for Soil Pollution Control and Remediation in South China, Guangdong Key Laboratory of Integrated Agro-environmental Pollution Control and Management, Institute of Eco-environmental and Soil Sciences, Guangdong Academy of Sciences, Guangzhou 510650, China

* Corresponding author. Chunlin Ni, Wei Liu

Tel.: +86 20 87025180; Fax: +86 20 87024123.

E-mail: niclchem@scau.edu.cn, liuwei97@scau.edu.cn

Table S1 Comparison between the materials prepared in this study and those reported in literature.

Photocatalyst	Co-catalysts	Sacrificial reagent	Light source	Activity (mmol h ⁻¹ g ⁻¹)	Ref.
CdS	Cd-TCPP	TEOA	300 W Xe lamp($\lambda > 420$ nm)	3.15	1
CdS	Fe-MOF-525 _{2,3}	TEOA	300 W Xe lamp($\lambda > 420$ nm)	3.638	2
CdS	Ni ₂ P	0.35 M Na ₂ S/0.25 M Na ₂ SO ₃	300 W Xe lamp($\lambda > 420$ nm)	16.02	3
CdS	ZnO/ZnS	0.35 M Na ₂ S/0.25 M Na ₂ SO ₃	300 W Xe lamp($\lambda > 420$ nm)	2.64	4
CdS	CoP	0.35 M Na ₂ S/0.25 M Na ₂ SO ₃	300 W Xe lamp($\lambda > 420$ nm)	15.74	5
CdS	MoS ₂ /Ti ₃ C ₂	Lactic acid	300 W Xe lamp	14.1	6
CdS NRs	25-CDs	0.5 M Na ₂ S/0.5 M Na ₂ SO ₃	300 W Xe lamp	1.076	7
CdS	Sr ₂ MgSi ₂ O ₇ :(Eu,Dy)	0.35 M Na ₂ S/0.25 M Na ₂ SO ₃	300 W Xe lamp	18.07	8
CdS	ZnNi-C	0.35 M Na ₂ S/0.25 M Na ₂ SO ₃	LED lamp($\lambda > 420$ nm)	15.7	9
CdS	Pyr-GDY-CeO ₂ -15%	Lactic acid	PCX-50B	21.16	10
CdS	NiWO ₄	Lactic acid	300 W Xe lamp($\lambda > 420$ nm)	26.43	11
CdS	CoTPPBr ₄	Lactic acid	300 W Xe lamp	41.3	This work

1. W. Xu, J. Wang, H. Yu, P. Liu, G.R. Zhang, H. Huang and D. Mei, *Appl. Catal. B-Environ*, 2022, **308**, 121218.
2. Y. Wu, Y. Qu, C. Su, X. Yang, Y. Yang, Y. Zhang and W. Huang, *Inorg. Chem.*, 2023, **62**, 21290-21298.
3. W. Ma, D. Zheng, B. Xiao, Y. Xian, Q. Zhang, S. Wang, J. Liu, P. Wang and X. Hu, *J. Environ. Chem. Eng.*, 2022, **10**, 107822.
4. K. He, *Int. J. Hydrogen Energy*, 2024, **51**, 30-40.
5. Y. Zou, C. Guo, X. Cao, L. Zhang, T. Chen, C. Guo and J. Wang, *J. Environ. Chem. Eng.*, 2021, **9**, 106270.
6. Y. Wang, C. Liu, C. Kong and F. Zhang, *Colloid Surface A*, 2022, **652**, 129746.
7. M. Ge, H. Yin, W. Tian, H. Zhang, S. Li, S. Wang and Z. Chen, *J. Colloid Interface Sci.*, 2024, **660**, 147-156.
8. J. Wang, R. Pan, Z. Yuan, Q. Hao, X. Niu, R. Wang, J. Ye, H. Y. Yang and Y. Wu, *Chem. Eng. J.*, 2024, **481**, 148296.
9. L. Xue, Y. Shi, C. Huang, Q. Wu, B. Chen and W. Yao, *J. Colloid Interface Sci.*, 2023, **635**, 72-82.
10. Y. Liu, Z. Fan, Y. Sun, X. Guo and Z. Jin, *ACS Appl. Energy Mater.*, 2023, **6**, 9743-9755.
11. X.L. Yin, L.L. Li, G.M. Gao, Y. Lu, Q.Q. Shang, H.T. Zhao, D.C. Li and J.M. Dou, *Int. J. Hydrogen Energy*, 2022, **47**, 9895-9904.

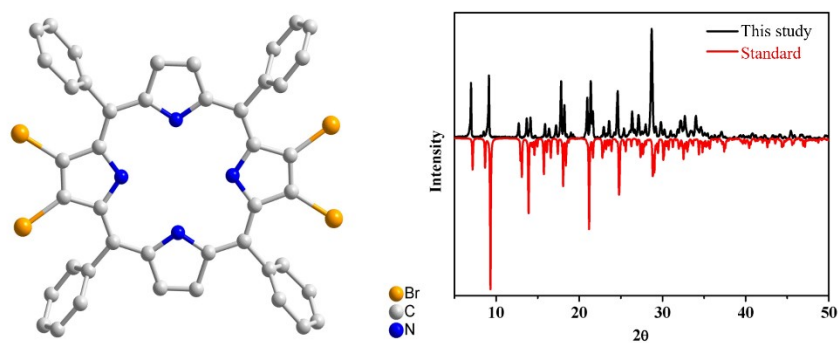


Figure S1. The structure, standard XRD spectrum, and XRD data obtained for H_2TPPBr_4 in this study.

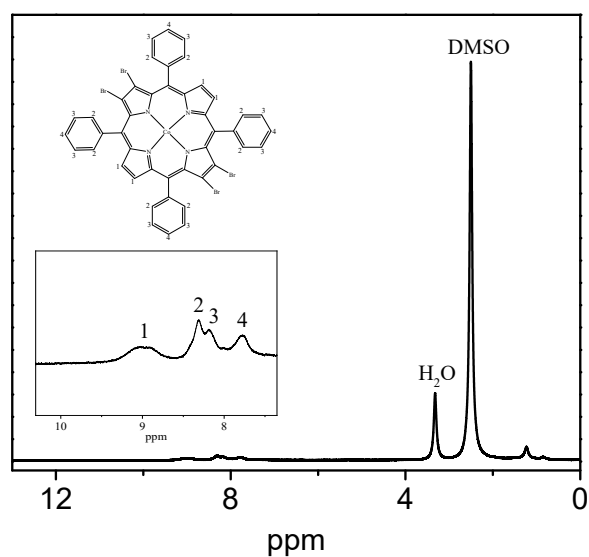
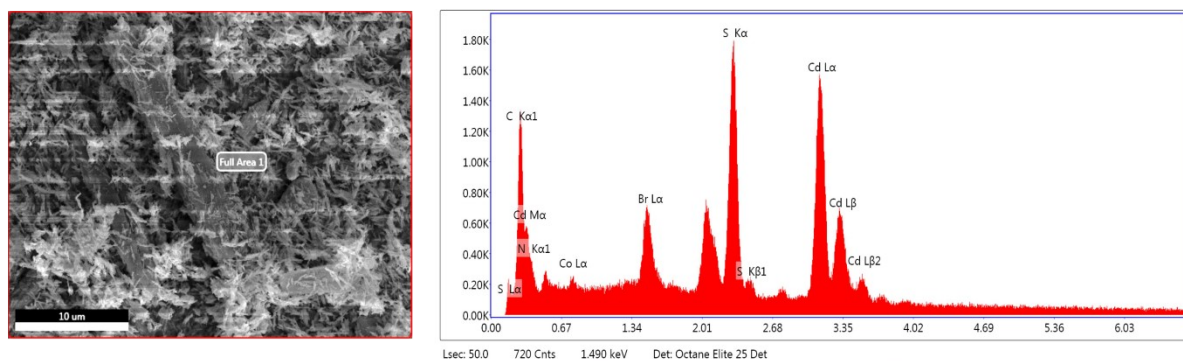


Figure S2. The NMR hydrogen spectrum data of CoTPPBr_4 .



Element	Weight %	Atomic %	Error %
C	14.63	49.47	10.16
N	0.43	1.23	49.01
Co	0.64	0.44	30.63
Br	5.55	2.82	6.59
S	19.41	24.59	4.02
Cd	59.35	21.45	4.72

Figure S3. Energy dispersive X-ray spectroscopy (EDX) analysis of the composite.

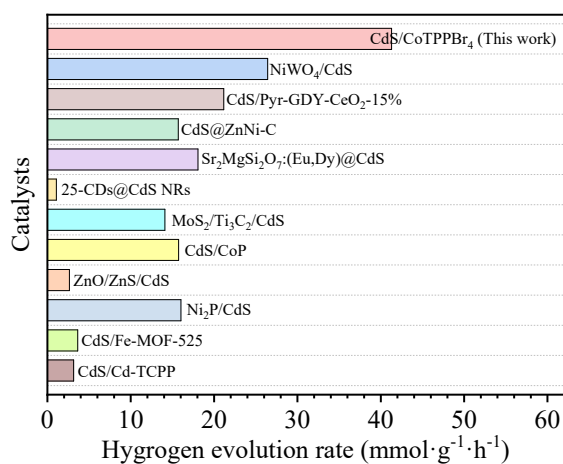


Figure S4. Comparison between the materials prepared in this study and those reported in literature.

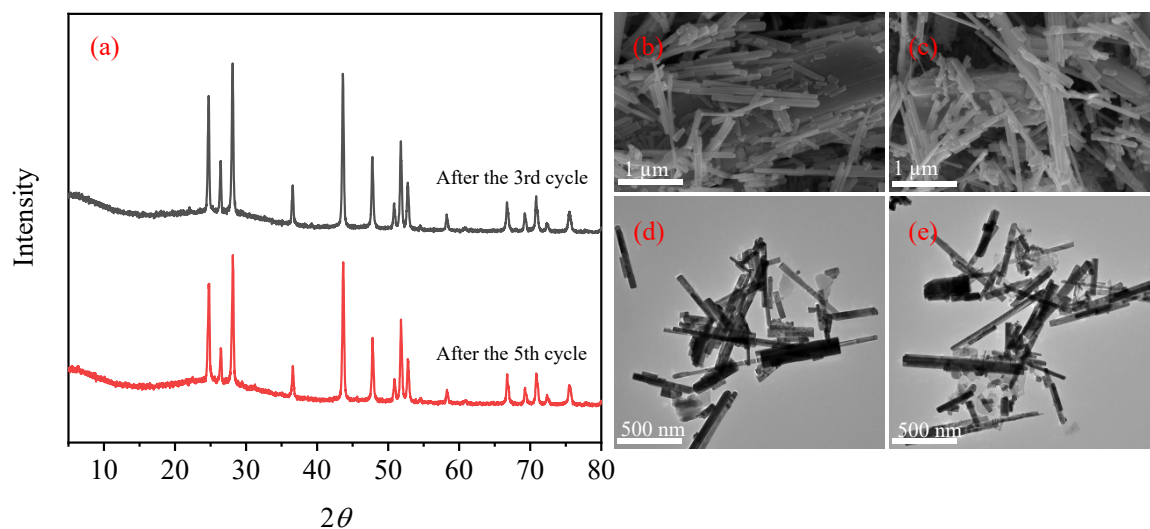


Figure S5. The XRD, SEM, and TEM characterization of the materials after 3 and 5 cycles of reaction. (a) XRD; (b) SEM after 3 cycles; (c) SEM after 5 cycles; (d) TEM after 3 cycles; (e) TEM after 5 cycles.

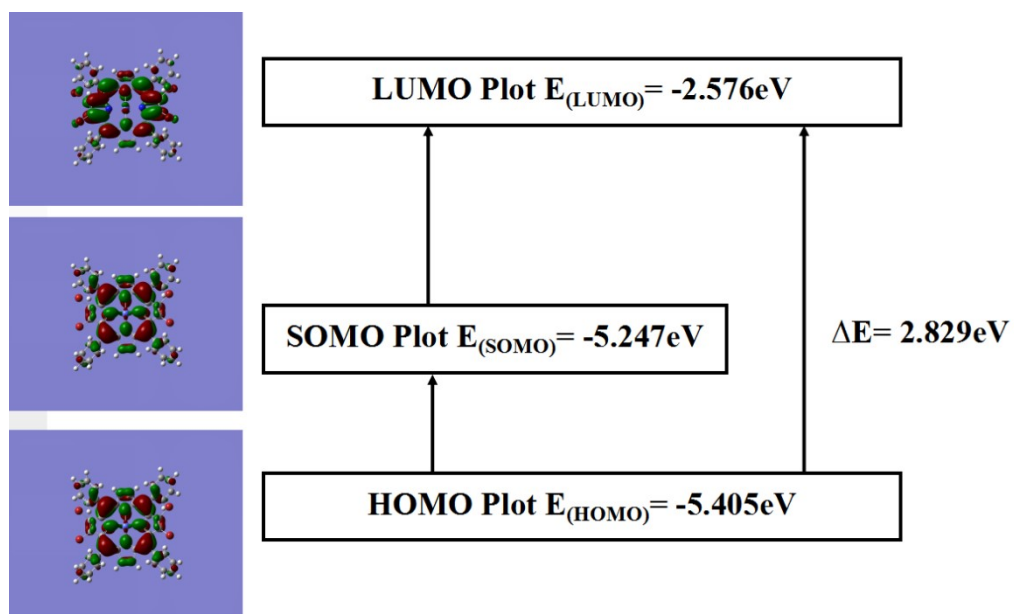


Figure S6 The DFT calculation of the CoTPPBr₄