

**Table S1.** Crystal data and structure refinement for  $(\text{Bi}_{2,494}\text{Na}_{2,505})(\text{PO}_4)_3\text{Cl}$ .

Formula	$(\text{Bi}_{2,494}\text{Na}_{2,505})(\text{PO}_4)_3\text{Cl}$
Formula weight	899.25
Crystal system	hexagonal
Space group, $Z$	$P-6, 4$
$a$ , Å	9.7197(7)
$c$ , Å	13.8019(10)
$V$ , Å <sup>3</sup>	1129.2(1)
$D_x$ , g/cm <sup>3</sup>	5.29
$\mu$ , mm <sup>-1</sup>	39.62
Wavelength, Å	0.71073
$\theta$ range/degree	2.83- 30.77
Refl. collected/unique/ $R_{int}$	110703 / 2019/ 0.0546
Completeness to $\theta$ , %	0.95
Parameters	140
$GOF (S)$	1.066
$R_{all}, R_{gt}, R_{wgt}$	0.0494, 0.0415, 0.0804
$\Delta\rho_{min}/\Delta\rho_{max}, \text{e}/\text{Å}^3$	-1.652/ 2.144

$$^1 R_{gt} = \Sigma. //F_o // - /F_c // / = \Sigma /F_o / \text{ and } R_{wgt} = [\Sigma w (F_o^2 - F_c^2)^2 / \Sigma w (F_o^2)^2]^{1/2} \text{ for } F_o^2 > 2\sigma(F_o^2).$$

**Table S2.** Atomic coordinates and equivalent displacement parameters for  $(\text{Bi}_{2,494}\text{Na}_{2,505})(\text{PO}_4)_3\text{Cl}$ .

Atom	Wyckoff notation, symmetry, multiplicity	Composition				$U_{equ}$ , Å <sup>2</sup>
			$x/a$	$y/b$	$z/c$	
<b>M1</b> (Bi2)*	3j, $m$ , 0.5	Bi	0.2582(1)	0.2671(1)	0	0.019(2)
<b>M2</b> (Bi4)	2i,3, 0.333	$\text{Bi}_{0,1300}\text{Na}_{0,2033}$	0.6667	0.3333	0.3817(2)	0.0249(6)
<b>M3</b> (Bi5)	2i, 3, 0.333	$\text{Bi}_{0,15}\text{Na}_{0,1833}$	0.3333	0.6667	0.3790(2)	0.0212(5)
<b>M4</b> (Bi6)	6l, 1, 1.0	$\text{Bi}_{0,7}\text{Na}_{0,3}$	-0.0091 (1)	0.2555(1)	0.2606(1)	0.0145(2)
<b>M5</b> (Bi16)	2i, 3, 0.333	$\text{Bi}_{0,0333}\text{Na}_{0,3000}$	0.3333	0.6667	0.1299(4)	0.0191(1)
<b>M6</b> (Bi13)	3k, $m$ , 0.5	$\text{Bi}_{0,15}\text{Na}_{0,35}$	0.2521(4)	0.2579(3)	0.5	0.0470(8)
<b>M7</b> (Bi14)	2i, 3, 0.333	Na	0.6667	0.3333	0.1141(9)	0.019(3)
<b>P1</b> (P1)	6l, 1, 1.0	P	0.3795(7)	0.0272(7)	0.2504(2)	0.008(1)
<b>P2</b> (P3)	3j, $m$ , 0.5	P	0.3714(8)	-0.0275(9)	0	0.011(1)
<b>P3</b> (P4)	3j, $m$ , 0.5	P	0.5877(7)	0.6129(8)	0.5	0.014(1)
<b>Cl1</b> (Cl1)	2g,3, 0.333	Cl	0	0	0.01266(5)	0.0228(9)
<b>Cl2</b> (Cl3)	2g,3, 0.333	Cl	0	0	0.3737(5)	0.028(1)
<b>O1</b> (O2)	6l, 1, 1.0	O	0.330(2)	0.484(2)	0.2563(7)	0.030(4)
<b>O2</b> (O4)	3j, $m$ , 0.5	O	0.465(2)	0.150(2)	0	0/014(3)
<b>O3</b> (O5)	3j, $m$ , 0.5	O	0.849(2)	0.350(3)	0.5	0.034(5)
<b>O4</b> (O6)	6l, 1, 1.0	O	-0.472(2)	0.111(2)	0.2530(9)	0.032(4)
<b>O5</b> (O8)	6l, 1, 1.0	O	0.095(2)	0.350(2)	0.0891(8)	0.024(3)
<b>O6</b> (O9)	3j, $m$ , 0.5	O	0.401(2)	0.524(2)	0.5	0.019(4)
<b>O7</b> (O11)	6l, 1, 1.0	O	-0.278(2)	0.082(2)	0.1623(8)	0.029(3)
<b>O8</b> (O12)	6l, 1, 1.0	O	0.901(2)	0.637(2)	0.3348(8)	0.028(3)
<b>O9</b> (O16)	6l, 1, 1.0	O	0.651(2)	0.726(2)	0.409(9)	0.037(4)
<b>O10</b> (O17)	3j, $m$ , 0.5	O	0.482(2)	-0.105(3)	0	0.023(4)

\*In brackets are the destinations of positions in group  $P3$ .

**Table S3.** Interatomic distances for  $(\text{Bi}_{2,494}\text{Na}_{2,505})(\text{PO}_4)_3\text{Cl}$ .

<b>Bond</b>	<b>Distance, Å</b>	<b>Bond</b>	<b>Distance, Å</b>
<b>M1</b> – O17	2.13(2)	<b>M2</b> – O5x3	2.35(2)
<b>M1</b> – O11x2	2.43(1)	<b>M2</b> – O6x3	2.59(2)
<b>M1</b> – O8x2	2.44(1)	<b>M2</b> – O12x3	2.76(1)
Average	2.37	Average	2.57
<b>M3</b> -O2x3	2.44 (1)	<b>M5</b> -O2x3	2.48(2)
<b>M3</b> -O9x3	2,46(2)	<b>M5</b> -O17x3	2.65(2)
Average	2.45	Average	2.56
<b>M4</b> -O16	2.22(1)	<b>M6</b> -O9	2,. 5(2)
<b>M4</b> -O6	2.23(2)	<b>M6</b> -O12x2	2.51(1)
<b>M4</b> -O12	2.39(1)	<b>M6</b> -O16	2.62(2)
<b>M4</b> -O8	2.56(1)	Average	2.47
<b>M4</b> -O11	2.66(2)	<b>M7</b> -O4x3	2.45(1)
Average	2.41	<b>M7</b> -O6x3	2.69(2)
P1 – O11	1.49(1)	<b>M7</b> -O11x3	2.83(2)
P1 – O6	1.49(2)	Average	2.66
P1 – O12	1.52(1)	P3 – O4	1.49(2)
P1 – O2	1.53(2)	P3 – O8x2	1.57(1)
Average	1.51	P 3– O17	1.59(2)
P4– O5	1,50(2)	Average	1.55
P4– O9	1,57(2)		
P4– O16x2	1,58(1)		
Average	1.56		