

## Supporting Information

### Fluorescence sensing for chloride ions in water based on wavelength shifts of hexane-dispersed CsPbBr<sub>3</sub>@silicone

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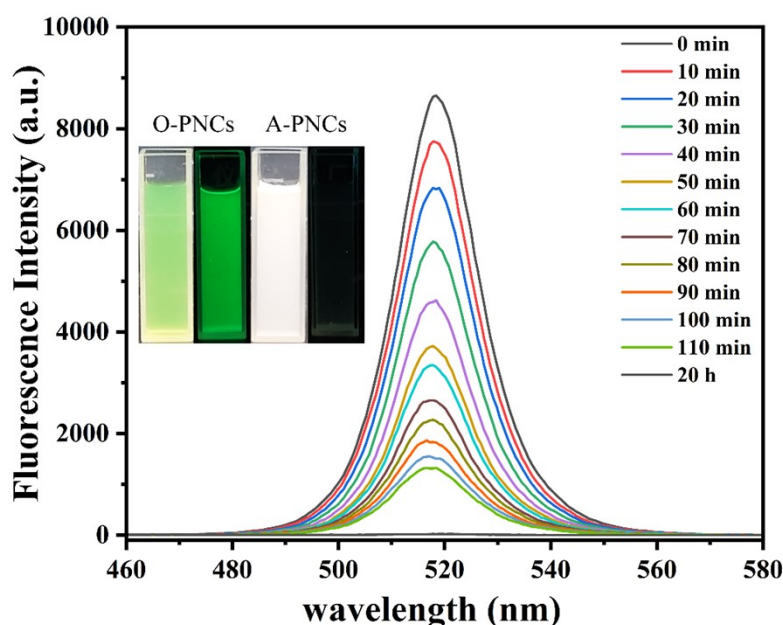
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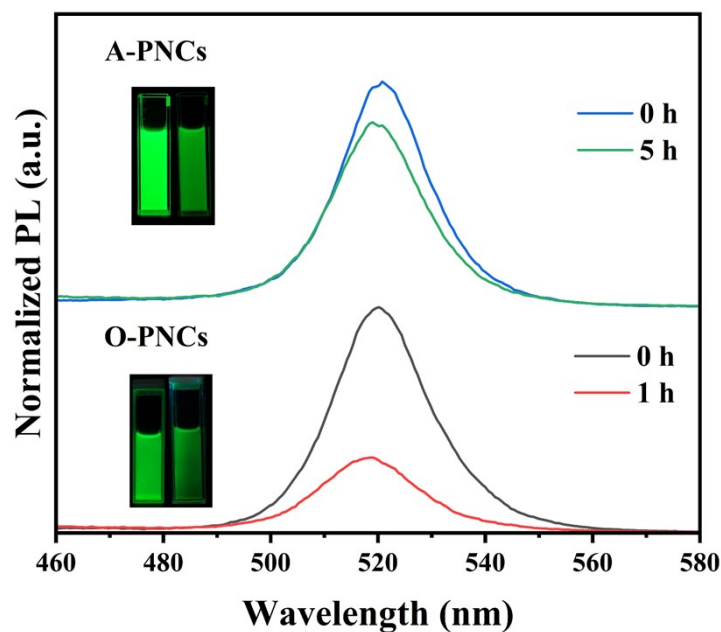
### Measurement of Cl<sup>-</sup> concentration in sweat samples by traditional argentometric method

Due to the low sample volume, the collected sweat was diluted 10 times before testing. The 3 mL sample solution and 120  $\mu$ L K<sub>2</sub>CrO<sub>4</sub> (Indicator) were added in a 10 mL conical flask. The AgNO<sub>3</sub> standard solution (5.000 mM) was slowly added to the conical flask through a burette and the solution was stirred. When the color of the solution changed from clear yellow to stable red, the volume of AgNO<sub>3</sub> was recorded. The concentration of Cl<sup>-</sup> in the sample was obtained from:

$$C_{\text{Cl}^-} = \frac{5.000 \times V_{(\text{AgNO}_3)}}{3} \text{ mM}$$



**Figure S1** Fluorescence emission spectra of O-PNCs in water, inset show solution colors under natural light and 365 nm UV light, with O-PNCs on the left side and A-PNCs on the right side.



**Figure S2** Fluorescence emission spectra of O-PNCs and A-PNCs in ethanol, the inset shows solution colors under 365 nm UV light.

**Table S1.** An overview of recently reported nanomaterial-based optical methods for the determination of  $\text{Cl}^-$  in aqueous phases.

| Materials used                                         | Solvents for nanomaterials | Fluorescence sensing type | Visual resolution | Response time | LOD                | Ref       |
|--------------------------------------------------------|----------------------------|---------------------------|-------------------|---------------|--------------------|-----------|
| GQD                                                    | Water                      | Intensity based           | Low               | 5 min         | 10 mM              | [15]      |
| Supramolecular network                                 |                            | Intensity based           | Low               |               | 24.9 $\mu\text{M}$ | [16]      |
| $\text{CsPbBr}_3$                                      | Hexane                     | Wavelength based          | High              | 5 min         | 3.0 mM             | [24]      |
| $\text{CsPbBr}_3\text{-SiO}_2$                         | Ethanol                    | Wavelength based          | High              | 15 min        | 0.05 mg/g          | [27]      |
| $\text{CsPbBr}_3\text{-OPA}$                           | Water                      | Wavelength based          | High              | 5 min         | 0.34 mM            | [28]      |
| $\text{CsPbBr}_3@\text{CsPb}_2\text{Br}_5$             | Water                      | Wavelength based          | High              | 20 s          | 3.2 $\mu\text{M}$  | [29]      |
| $\text{CsPbBr}_3\text{-ACD}$                           | Ethanol                    | Wavelength based          | High              | 1 min         | 3.2 $\mu\text{M}$  | [30]      |
| DUT-101 and $\text{Ag}^+/\text{Eu}^{3+}@\text{UIO-67}$ |                            | Intensity based           | High              |               | 0.1 mM             | [53]      |
| $\text{CsPbBr}_3\text{-APTMS}$                         | Hexane                     | Wavelength based          | High              | 1 min         | 0.188 mM           | This work |

**Table S2.** Detection of Cl<sup>-</sup> in sweat based on traditional argentometric method.

| sample | Sample<br>volume (ml) | AgNO <sub>3</sub><br>volume<br>(ml) | AgNO <sub>3</sub><br>concentration<br>(mM) | Sample Cl <sup>-</sup><br>concentration<br>(mM) | RSD<br>(%) |
|--------|-----------------------|-------------------------------------|--------------------------------------------|-------------------------------------------------|------------|
| 1      | 3.000                 | 2.256                               | 5.055                                      | 3.801                                           | 0.35       |
| 2      | 3.000                 | 1.467                               | 5.055                                      | 2.472                                           | 0.51       |