

## Electronic Supplementary Information

### Portable smartphone platform based on fluorescent carbon quantum dots derived from biowaste for on-site detection of permanganate

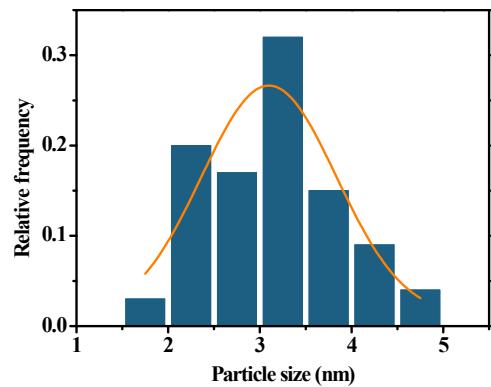
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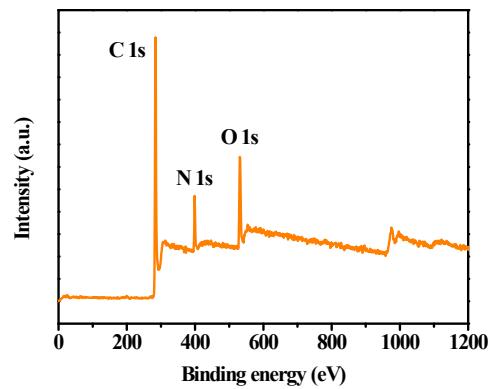
<sup>b</sup> Sichuan Yalian Hydrogen Technology Co., Ltd, Chengdu 610000, PR China.

\* Corresponding author.

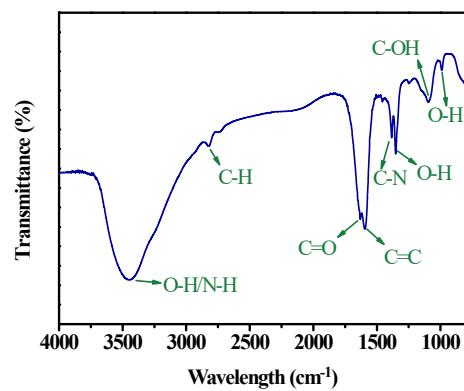
E-mail address: w-fu17@tsinghua.org.cn



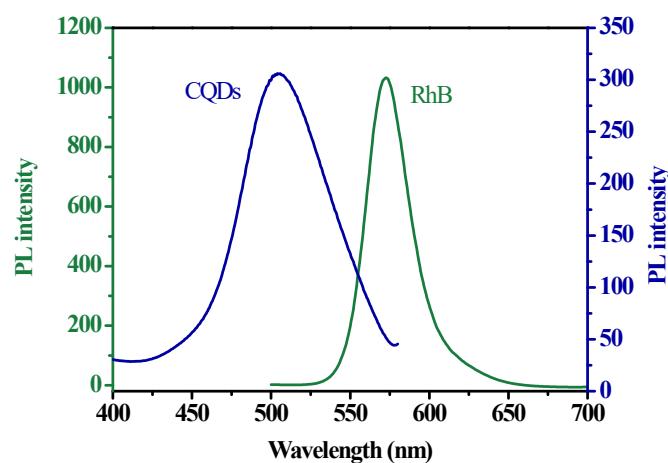
**Fig. S1** Size distribution of CQDs (Average diameter= $3.14 \pm 0.71$  nm)



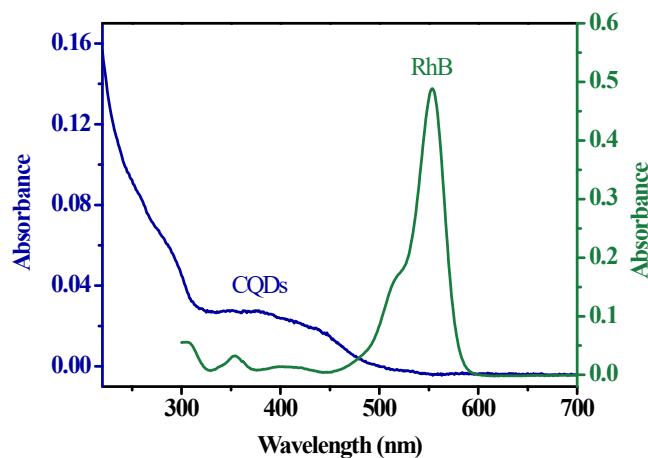
**Fig. S2** XPS survey scan of CQDs



**Fig. S3** FT-IR spectrum of CQDs.



**Fig. S4** PL spectrum of CQDs and Rhodamine B



**Fig. S5** UV-vis spectrum of CQDs and Rhodamine B (RhB)

**Table S1.** Quantum yield of the CQDs.

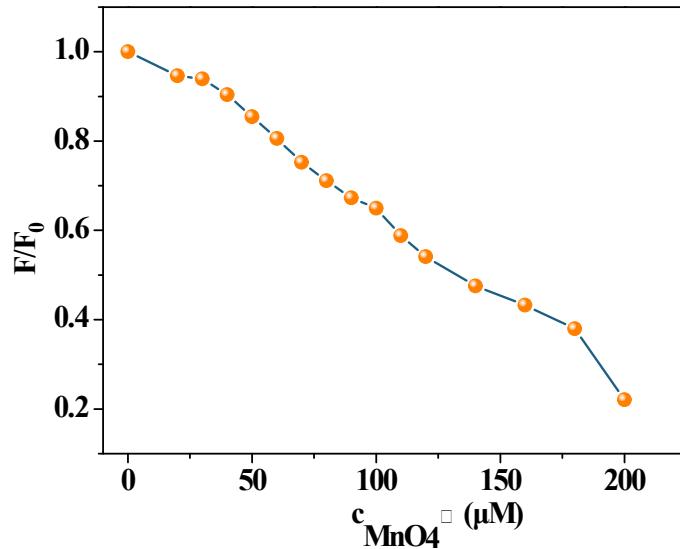
Sample	Abs. at 360nm	PL Integrated intensity	Refractive index of solvent ( $n$ )	PLQY (%)
RhB	0.0246	38790.29	1.33	31
CQDs	0.0265	17330.45	1.33	12.86

**Table S2.** A comparison of fluorescence quantum yields (QY) exhibited by CQDs.

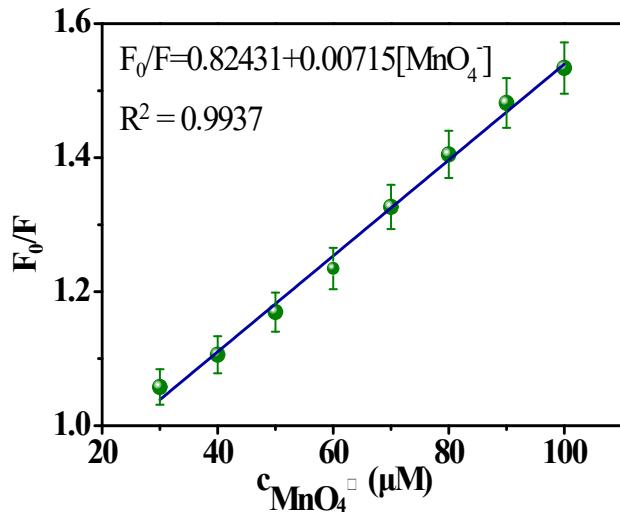
Quantum dots	QY (%)	References
Ionic liquids modified carbon quantum dots (ILs-CQDs)	18.13	[1]
Carbon quantum dots (CQDs)	11.37	[2]
Carbon quantum dots (CQDs)	16.2	[3]
Carbon quantum dots (CQDs)	10.36	[4]
N-doped carbon quantum dots (CQDs)	9	[5]
Nitrogen doped carbon quantum dots (N-CQDs)	12.33	[6]
Nitrogen doped carbon quantum dot (N-CQDs)	14.81	[7]
Nitrogen doped carbon quantum dot (N-CQDs)	9.3	[8]
Carbon quantum dots (CQDs)	3.5	[9]
Lotus stem-derived carbon dots (LS-CQDs)	0.44	[10]
Carbon quantum dots (CQDs)	14.12	[11]
CQDs	12.86	This work

**Table S3.** Comparison of the performance of different methods for  $\text{MnO}_4^-$  detection.

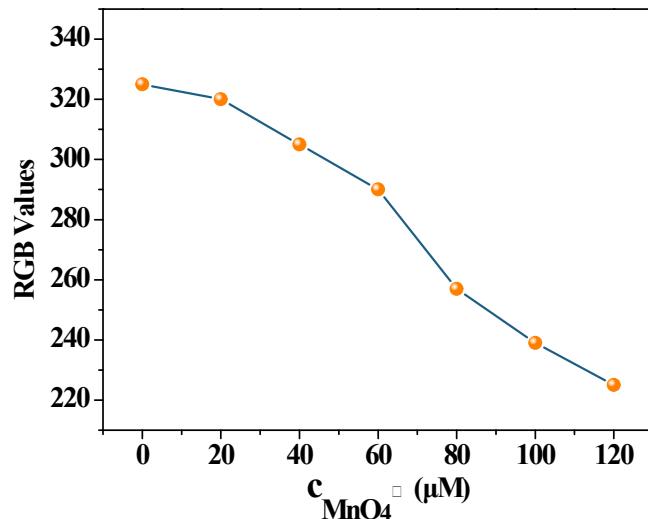
Fluorescent probes	LOD	References
ILs-CQDs	0.09 $\mu\text{M}$	[1]
CQDs	0.06 $\mu\text{M}$	[4]
Luminescent lanthanide metal-organic frameworks (Ln-MOFs)	3.53 $\mu\text{M}$	[12]
Red emission carbon dots (RCDs)	14.5 nM	[13]
Manganese-doped carbon dots (Mn-CDs)	0.66 $\mu\text{M}$	[14]
Covalent organic framework (COF)	0.01 mM	[15]
$[\text{Zn}(2,2'\text{-bipy})(\text{ppa})(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}$	6.73 $\mu\text{M}$	[16]
$[\text{Co}(\text{NPDC})(\text{bpee})] \cdot \text{DMF} \cdot 2\text{H}_2\text{O}$	1.50 $\mu\text{M}$	[17]
$\{\text{[Eu}_2\text{Na}(\text{Hpdbb})(\text{pdubb})_2(\text{CH}_3\text{COO})_2\} \cdot 2.5\text{DMA}\}_n$	5.99 $\mu\text{M}$	[18]
Carbon nanospheres (CNs)	0.72 $\mu\text{M}$	[19]
CQDs	3.31 $\mu\text{M}$	This work



**Fig. S6** Relationship between fluorescence intensity and different  $\text{MnO}_4^-$  concentrations.



**Fig. S7** Linear plots of different concentrations of  $\text{MnO}_4^-$  versus  $F_0/F$ .



**Fig. S8** The relationship between RGB values and  $\text{MnO}_4^-$  concentrations

**Table S4.** Recoveries of  $\text{MnO}_4^-$  in water samples based on smartphone sensing platform ( $n = 3$ ).

Sample	Added ( $\mu\text{M}$ )	Found ( $\mu\text{M}$ )	Recovery (%)	RSD (%)
<b>Tap water</b>	0	NF	-	-
	40	39.35	98.38	4.02
	70	68.34	97.63	4.94
	80	82.78	103.48	3.56
<b>River water</b>	0	NF	-	-
	40	39.76	99.40	3.69
	70	68.73	98.19	4.27
	80	84.01	105.01	5.19

Note: NF=Not found

## References

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