

## Supplementary Information (SI)

### Tunable electronic and optical properties of Janus Al<sub>2</sub>M<sub>2</sub>ClBr (M=O, S) monolayers as UV photodetectors applications

Yujin Liu<sup>a</sup>, Xinguo Ma<sup>\*ab</sup>, Tian Xie<sup>a</sup>, Yijing Ren<sup>a</sup>, Jinyi Zhu<sup>a</sup>, Nan Ma<sup>c</sup>, Jingjing Lu<sup>a</sup>

Jeongmin Hong<sup>\*d</sup>,

<sup>a</sup> School of Science, Hubei University of Technology, Wuhan 430068, China. E-mail: [maxg@hbust.edu.cn](mailto:maxg@hbust.edu.cn)

<sup>b</sup> State Key Laboratory of Advanced Technology for Float Glass, Bengbu Glass Industrial Design and Research Institute, Bengbu, 233030, China.

<sup>c</sup> Key Laboratory of Inorganic Functional Materials and Devices, Chinese Academy of Sciences, Shanghai 201899, China. E-mail: [manan@mail.sic.ac.cn](mailto:manan@mail.sic.ac.cn)

<sup>d</sup> EECS, UC Berkeley, Berkeley, CA, USA. Email: [jehong@berkeley.edu](mailto:jehong@berkeley.edu)

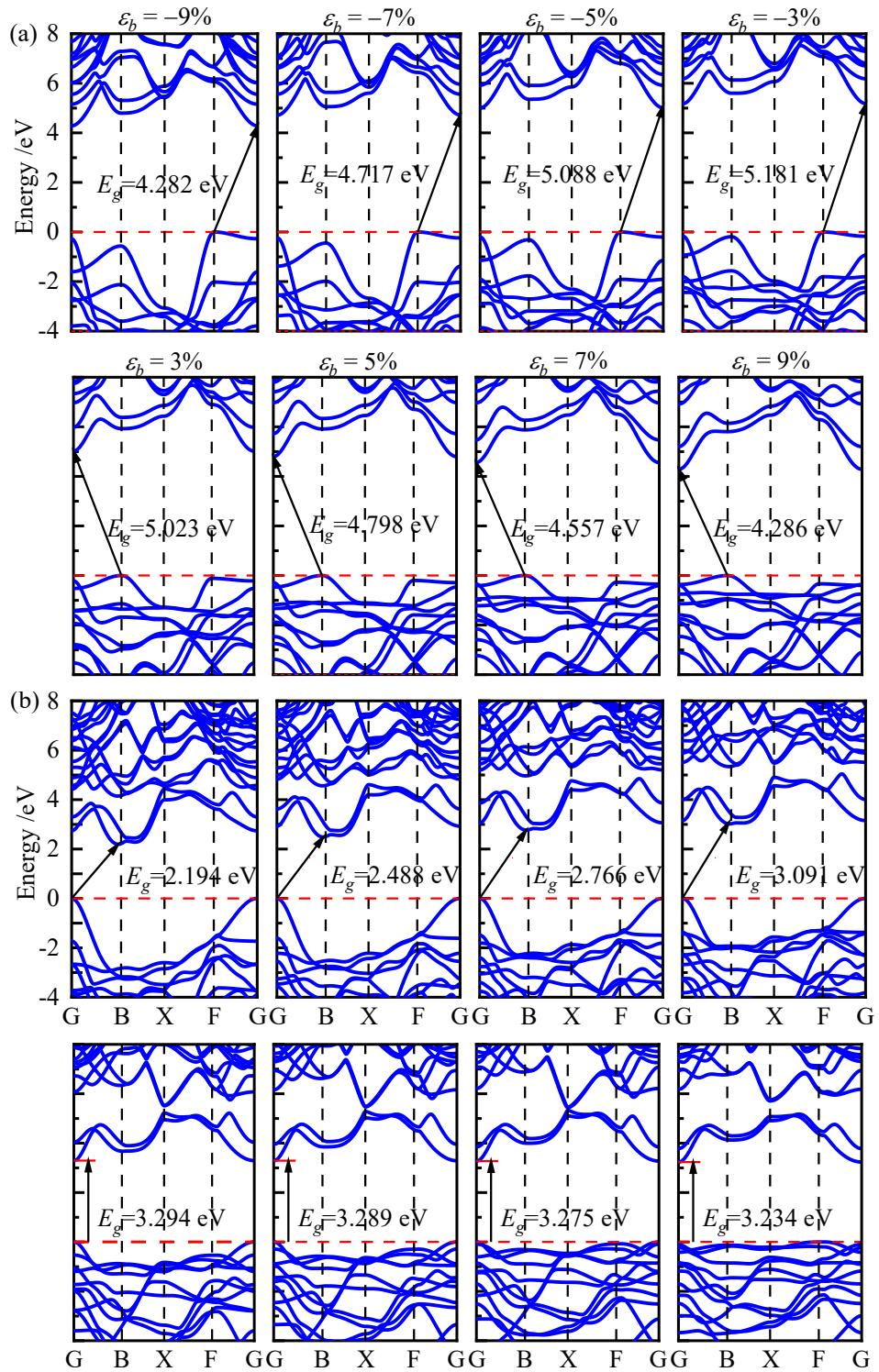
**\*Corresponding Authors, E-mail:** maxg@hbust.edu.cn (X. Ma), jehong@berkeley.edu

**Table S1** Calculated effective masses for Janus Al<sub>2</sub>M<sub>2</sub>ClBr (M=O, S) monolayers by PBE+SOC and HSE06+SOC.

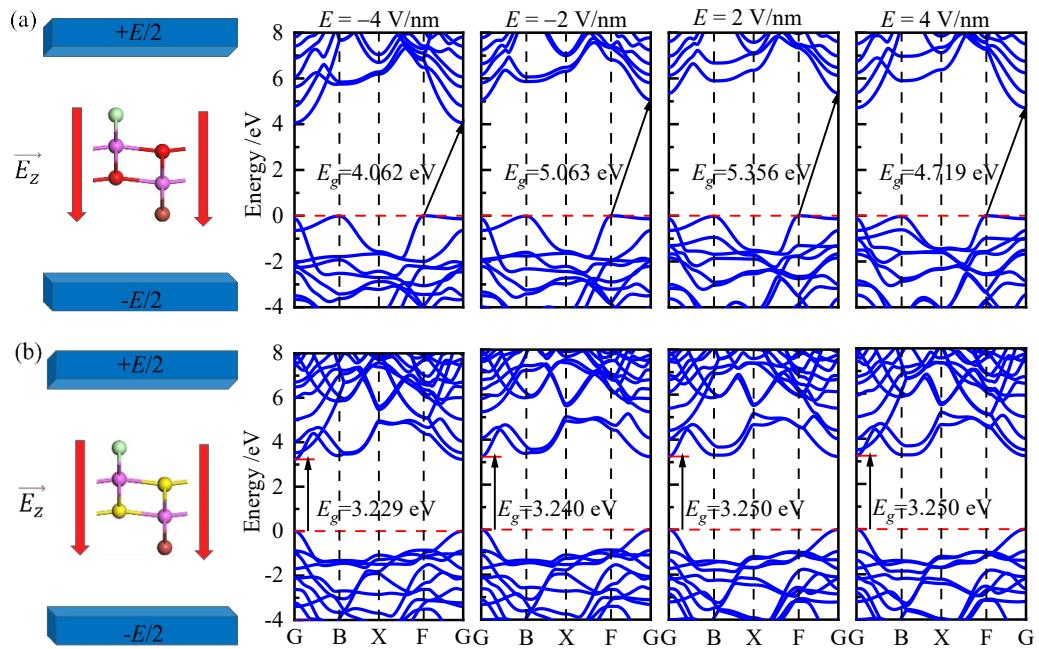
Material	Method	Direction	$m_e^*/m_0$	$m_h^*/m_0$
Al <sub>2</sub> O <sub>2</sub> ClBr	PBE+SOC	x	0.52	0.23
		y	0.34	8.08
	HSE06+SOC	x	0.58	0.23
		y	0.37	7.95
Al <sub>2</sub> S <sub>2</sub> ClBr	PBE+SOC	x	1.00	0.81
		y	0.19	0.26
	HSE06+SOC	x	0.97	0.81
		y	0.19	0.26

**Table S2** Mulliken charges and overlap populations of Janus Al<sub>2</sub>M<sub>2</sub>ClBr (M=O, S) monolayers after geometry optimization by PBE and PW91 methods.

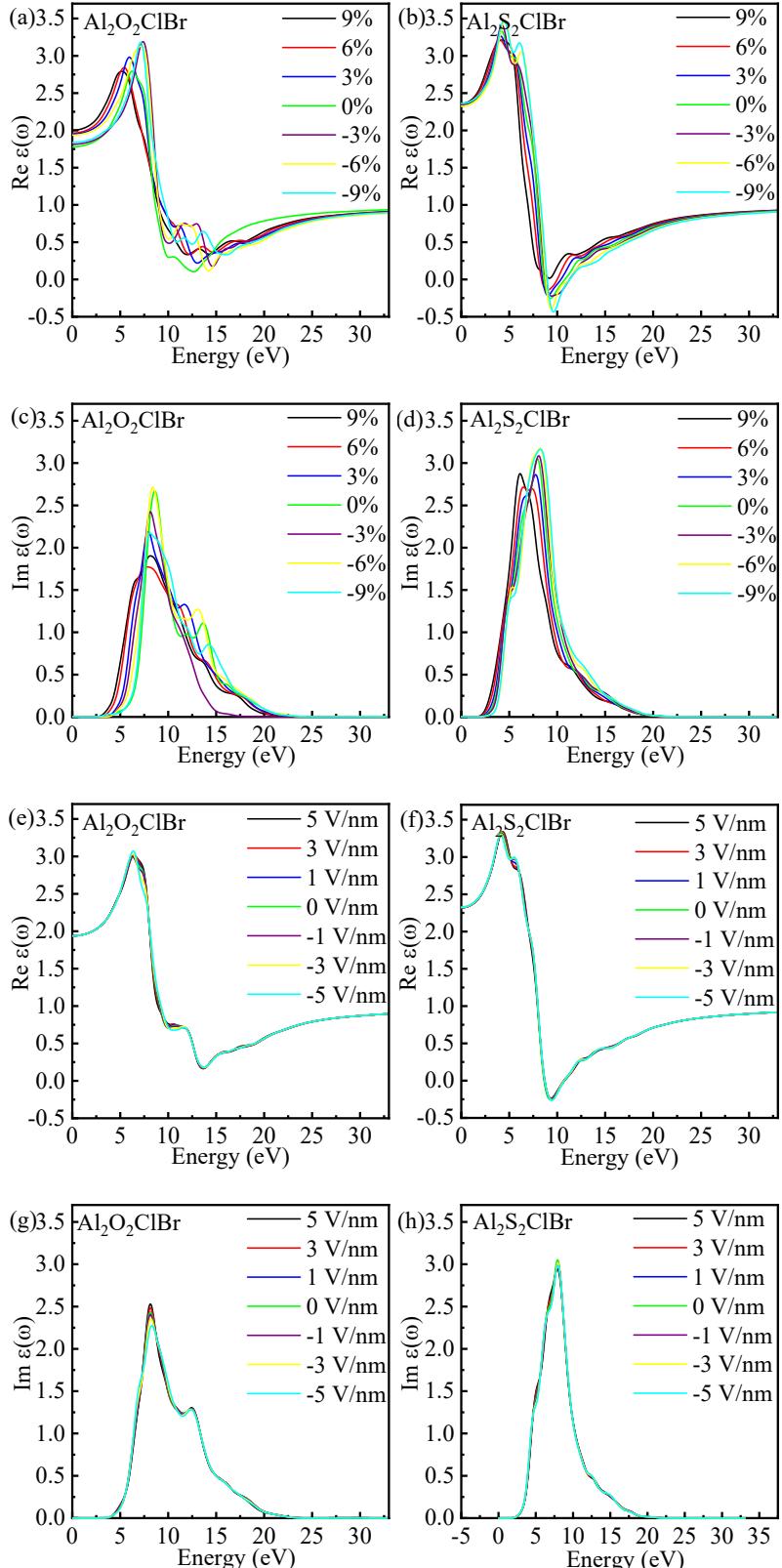
method	materials	species	charge	bonds	population
PBE	Al <sub>2</sub> O <sub>2</sub> ClBr	Al <sub>1</sub>	1.35	O <sub>1</sub> - Al <sub>1</sub>	0.56
		Al <sub>2</sub>	1.50	O <sub>2</sub> - Al <sub>2</sub>	0.55
		O <sub>1</sub>	-1.08	O <sub>2</sub> - Al <sub>1</sub>	0.59
		O <sub>2</sub>	-1.08	O <sub>1</sub> - Al <sub>2</sub>	0.59
		Cl <sub>1</sub>	-0.41	Al <sub>2</sub> - Cl <sub>1</sub>	0.71
		Br <sub>1</sub>	-0.28	Al <sub>1</sub> - Br <sub>1</sub>	0.73
	Al <sub>2</sub> S <sub>2</sub> ClBr	Al <sub>1</sub>	0.84	S <sub>1</sub> - Al <sub>1</sub>	0.54
		Al <sub>2</sub>	1.01	S <sub>2</sub> - Al <sub>2</sub>	0.54
		S <sub>1</sub>	-0.61	S <sub>2</sub> - Al <sub>1</sub>	0.97
		S <sub>2</sub>	-0.60	S <sub>1</sub> - Al <sub>2</sub>	0.97
		Cl <sub>1</sub>	-0.40	Al <sub>2</sub> - Cl <sub>1</sub>	0.63
		Br <sub>1</sub>	-0.24	Al <sub>1</sub> - Br <sub>1</sub>	0.66
PW91	Al <sub>2</sub> O <sub>2</sub> ClBr	Al <sub>1</sub>	1.35	O <sub>1</sub> - Al <sub>1</sub>	0.56
		Al <sub>2</sub>	1.51	O <sub>2</sub> - Al <sub>2</sub>	0.55
		O <sub>1</sub>	-1.09	O <sub>2</sub> - Al <sub>1</sub>	0.59
		O <sub>2</sub>	-1.09	O <sub>1</sub> - Al <sub>2</sub>	0.59
		Cl <sub>1</sub>	-0.41	Al <sub>2</sub> - Cl <sub>1</sub>	0.71
		Br <sub>1</sub>	-0.28	Al <sub>1</sub> - Br <sub>1</sub>	0.73
	Al <sub>2</sub> S <sub>2</sub> ClBr	Al <sub>1</sub>	0.84	S <sub>1</sub> - Al <sub>1</sub>	0.54
		Al <sub>2</sub>	1.01	S <sub>2</sub> - Al <sub>2</sub>	0.54
		S <sub>1</sub>	-0.61	S <sub>2</sub> - Al <sub>1</sub>	0.97
		S <sub>2</sub>	-0.60	S <sub>1</sub> - Al <sub>2</sub>	0.97
		Cl <sub>1</sub>	-0.40	Al <sub>2</sub> - Cl <sub>1</sub>	0.63
		Br <sub>1</sub>	-0.24	Al <sub>1</sub> - Br <sub>1</sub>	0.66



**Fig. S1.** Band structures of Janus (a)  $\text{Al}_2\text{O}_2\text{ClBr}$  and (b)  $\text{Al}_2\text{S}_2\text{ClBr}$  monolayers under different strain  $\varepsilon_b$ .



**Fig. S2.** Band structures of Janus (a) Al<sub>2</sub>O<sub>2</sub>ClBr and (b) Al<sub>2</sub>S<sub>2</sub>ClBr monolayers under different electric field  $E$ .



**Fig. S3.** The changes of the real part of the dielectric and the imaginary part of the dielectric of Janus  $\text{Al}_2\text{O}_2\text{ClBr}$  and  $\text{Al}_2\text{S}_2\text{ClBr}$  monolayers under different biaxial strain  $\varepsilon_b$  (a-d) electric field  $E$  (e-h).