

Supplementary Information

Improvement of chlorine evolution stability and activity of $\text{RuO}_2\text{-TiO}_2/\text{IrO}_2\text{-Ta}_2\text{O}_5$ electrode with low iridium content by alternating coating thermal decomposition method

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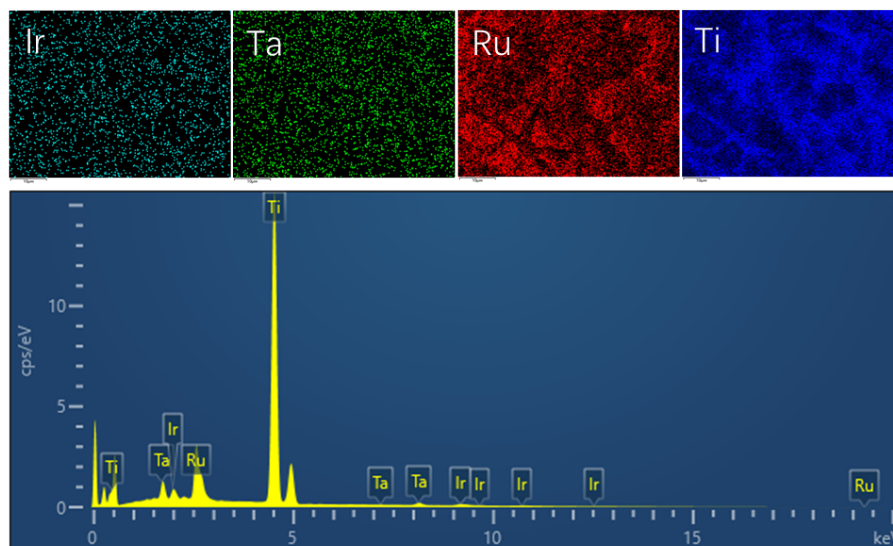


Figure S1 The SEM mapping of RuTi/LC-IrTa electrode.

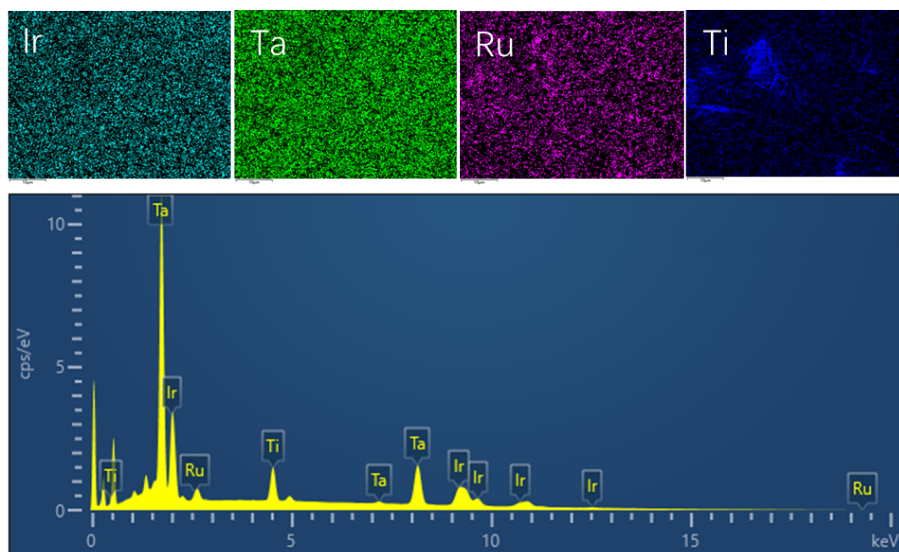


Figure S2 The SEM mapping of RuTi/HC-IrTa electrode.

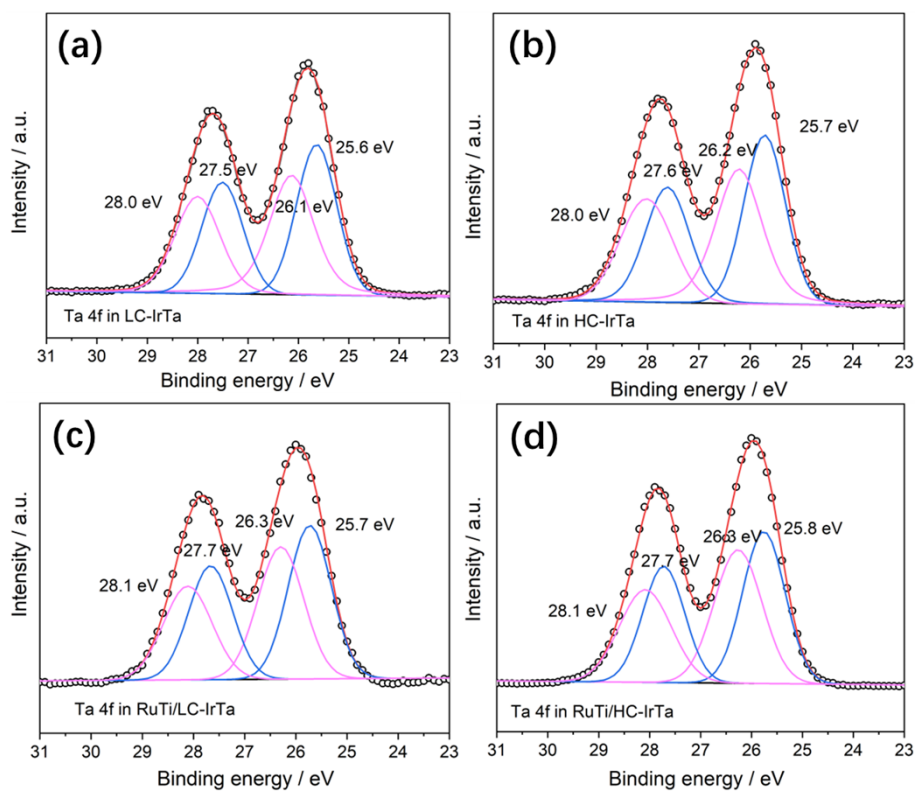


Figure S3 The XPS core-level spectra of Ta 4f obtained from LC-IrTa(a), HC-IrTa(b), RuTi/LC-IrTa(c) and RuTi/HC-IrTa(d).

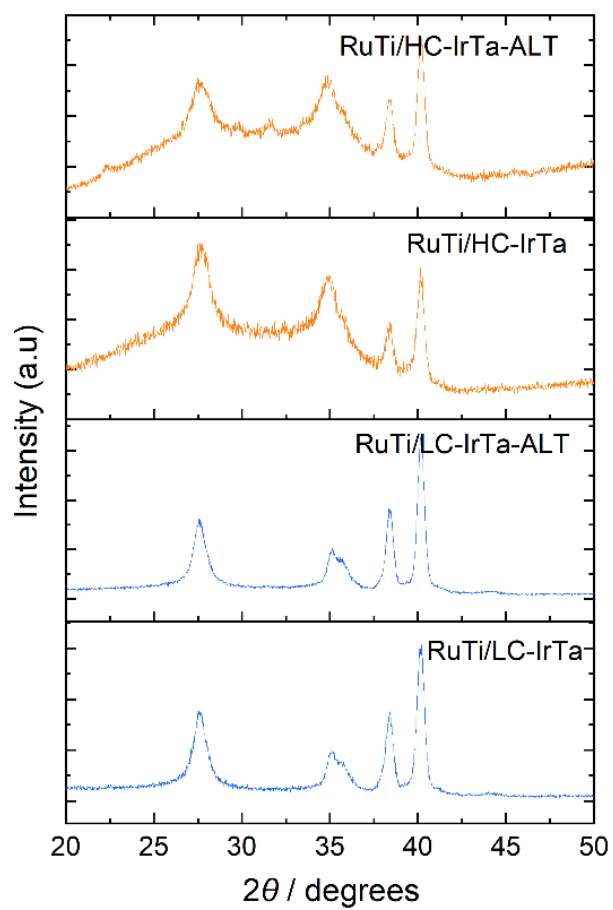


Figure S4 The X-ray diffraction patterns of RuTi/LC-IrTa and RuTi/HC-IrTa electrodes before and after ALT.

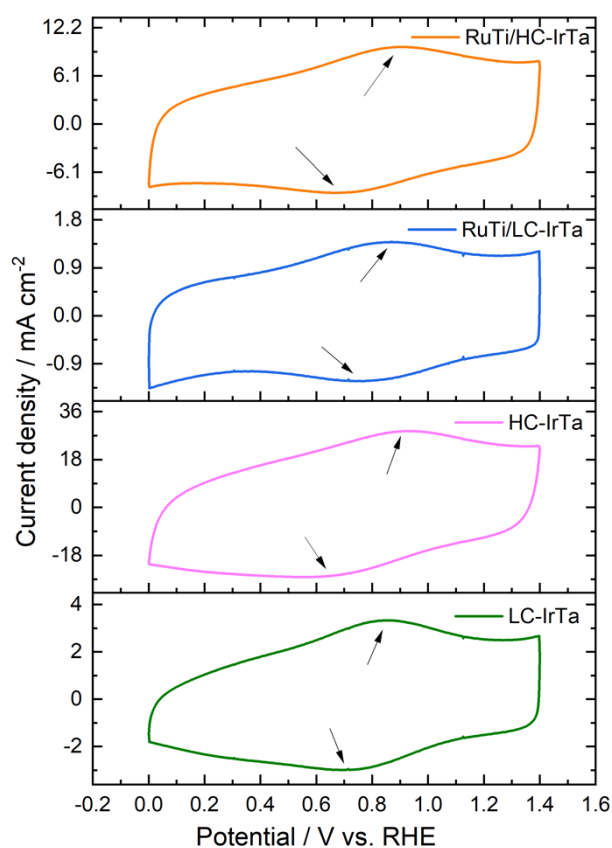
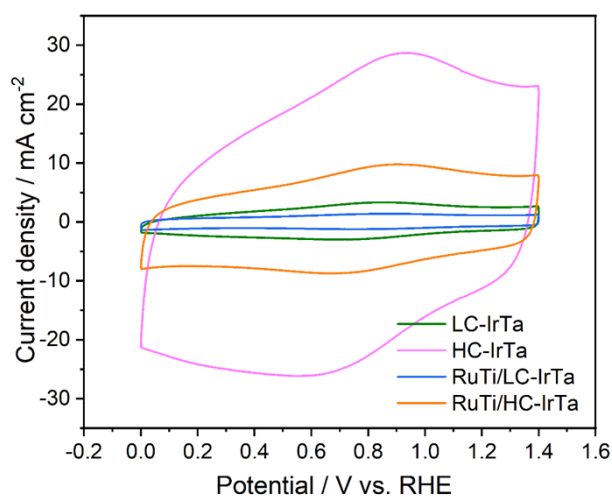


Figure S5 The cyclic voltammograms of LC-IrTa, RuTi/LC-IrTa, HC-IrTa and RuTi/HC-IrTa electrodes with a potential scan rate of 100 mV s⁻¹ in 0.5 mol L⁻¹ H₂SO₄ solution.

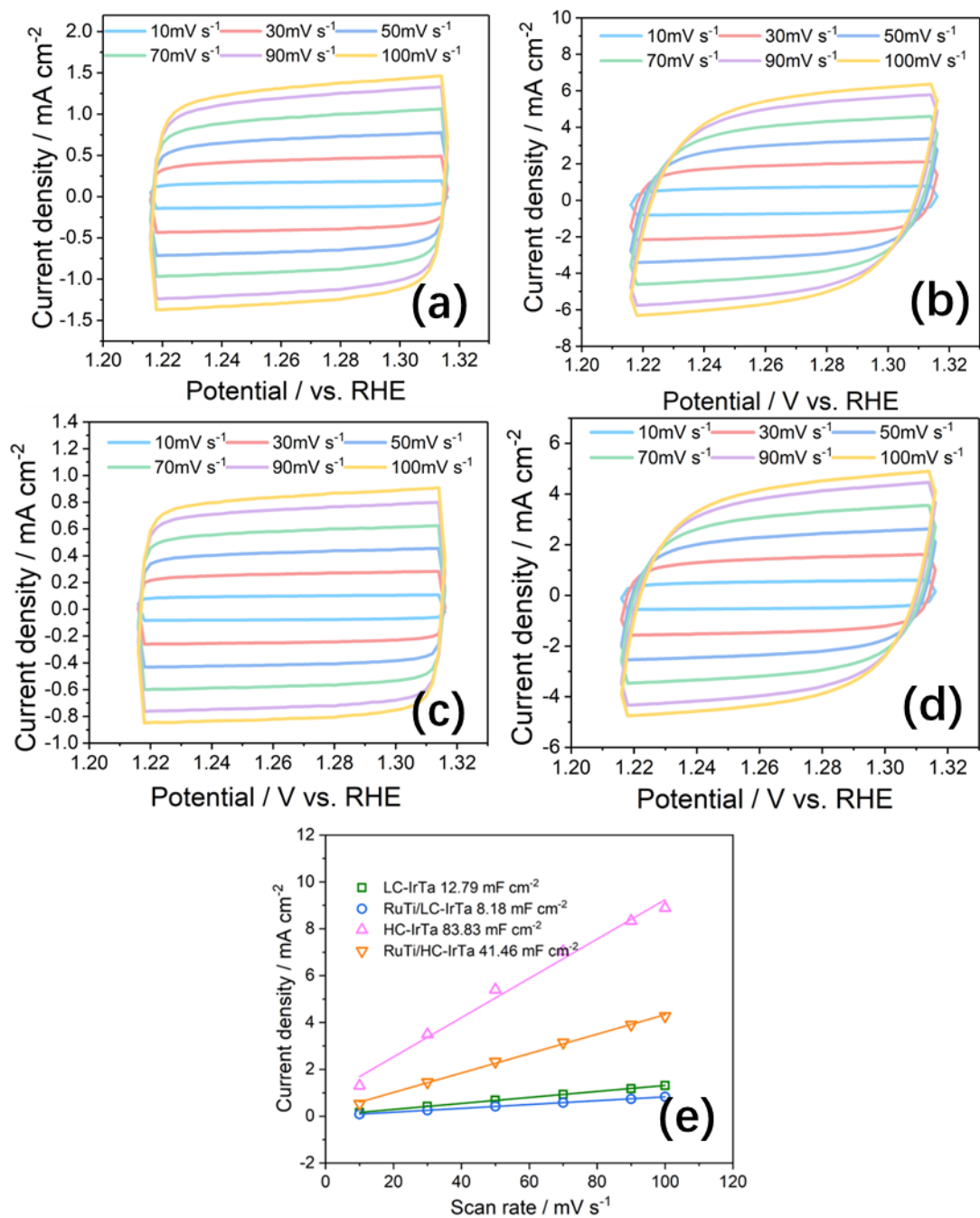


Figure S6 The cyclic voltammograms of LC-IrTa (a), HC-IrTa (b), RuTi/LC-IrTa (c) and RuTi/HC-IrTa (d) electrodes with different scan rate at potential of 1.216 ~ 1.316 V. Double-layer capacitances of LC-IrTa, HC-IrTa, RuTi/LC-IrTa and RuTi/HC-IrTa electrodes (e).

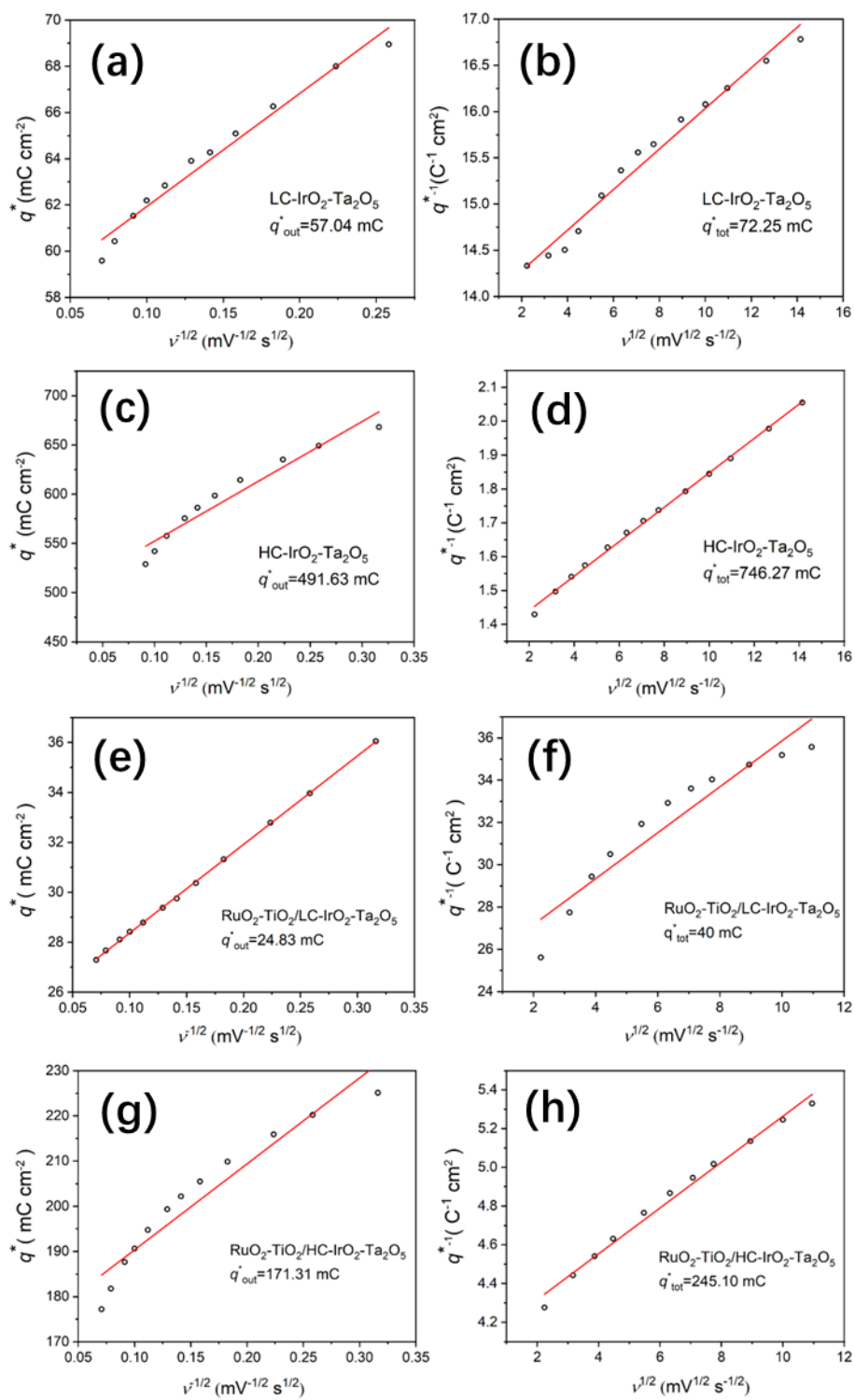


Figure S7 The total surface charges (q_{tot}^*) and outer surface charges (q_{out}^*) of LC-IrTa (a-b), HC-IrTa (c-d), RuTi/LC-IrTa (e-f) and RuTi/HC-IrTa electrodes (g-h).

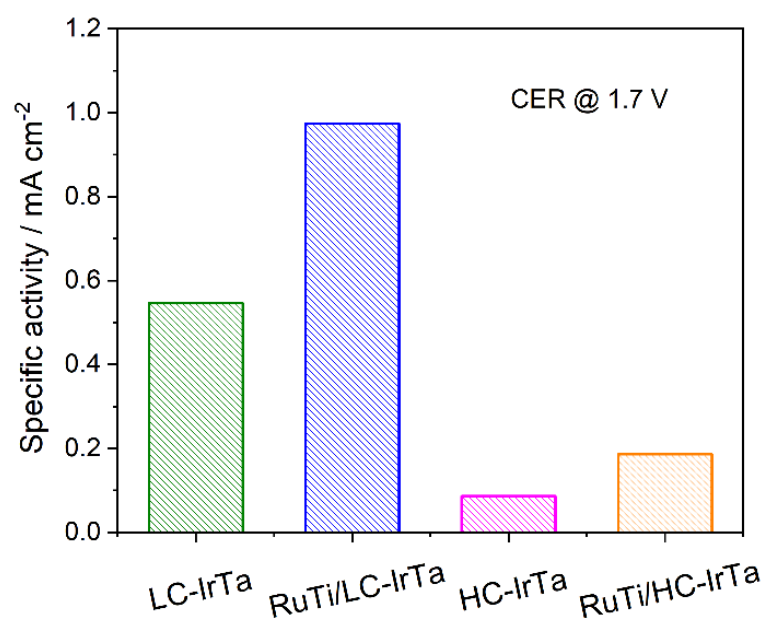


Figure S8 Specific activities of CER of LC-IrTa, RuTi/LC-IrTa, HC-IrTa and RuTi/HC-IrTa electrodes.

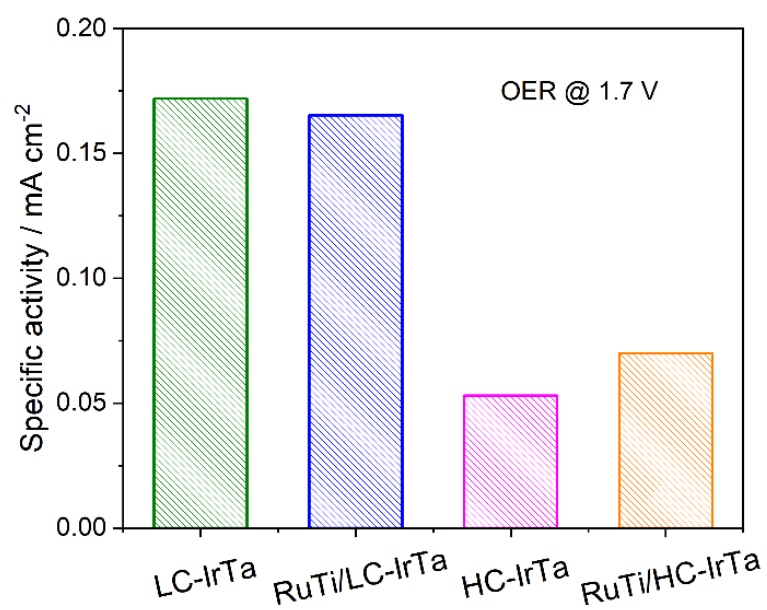


Figure S9 Specific activities of OER of LC-IrTa, RuTi/LC-IrTa, HC-IrTa and RuTi/HC-IrTa electrodes.

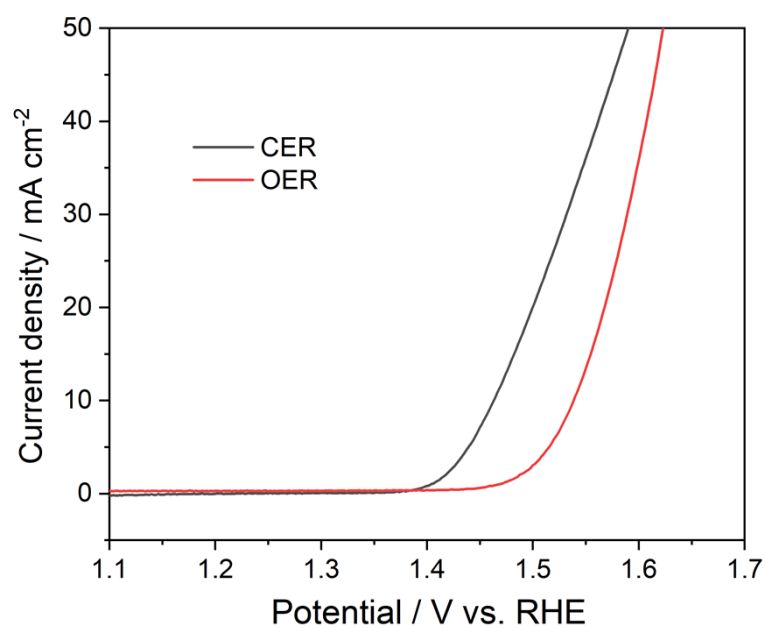


Figure S10 The CER apparent activities in 4.0 M NaCl (pH = 1.0) and the OER apparent activities in 0.5 M H₂SO₄ + 1.33 M Na₂SO₄ of RuTi-HC-IrTa electrode.

Table S1 The loading of Ir, Ta and Ru of RuTi, LC-IrTa, HC-IrTa, RuTi/LC-IrTa, RuTi/HC-IrTa and RuTi-HC-IrTa electrodes by XRF.

Electrodes	Ir/ $\mu\text{g cm}^{-2}$	Ta/ $\mu\text{g cm}^{-2}$	Ru/ $\mu\text{g cm}^{-2}$	Ir:Ta/mol:mol
RuTi	---	---	180.3	---
LC-IrTa	56.3	65.2	---	45:55
HC-IrTa	680.4	846.0	---	43:57
RuTi/LC-IrTa	30.3	35.8	62.1	44:56
RuTi/HC-IrTa	373.6	463.2	82.5	43:57
RuTi-HC-IrTa	350.5	440.3	97.1	43:57

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Table S2 ALs and SLs of RuTi, LC-IrTa, HC-IrTa, RuTi/LC-IrTa, RuTi/HC-IrTa and RuTi-HC-IrTa electrodes.

Electrodes	Accelerated	service
	Lifetime (AL) / h	Lifetime (SL) / years
	($i_a = 2 \text{ A cm}^{-2}$)	($i_n = 0.2 \text{ A cm}^{-2}$)
RuTi	5	0.06
LC-IrTa	38	0.43
HC-IrTa	245	2.80
RuTi/LC-IrTa	178	2.03
RuTi/HC-IrTa	602	6.87
RuTi-HC-IrTa	41	0.47

$$SL = AL (i_{AL} / i_{SL})^n, \quad n=2.0$$

Table S3 Comparison of ALs of RuTi/HC-IrTa and previously reported CER catalysts

Electrode	Loadin g/mg cm ⁻²	Electrolyte	Current density/ A cm ⁻²	AL/h	Ref.
RuO ₂ :TiO ₂ (30:70)		0.5 M NaCl	1.2	1.8	<i>Catal. Sci. Technol.</i> , 2018, 8, 4957-4974.
RuO ₂ :TiO ₂ (30:70)	0.3	0.5 M NaCl	0.7	31	<i>J. Electroanal. Chem.</i> , 2005, 579, 67-76.
RuO ₂ :TiO ₂	2	2.0 M H ₂ SO ₄	1.5	16	<i>J. Appl. Electrochem.</i> , 1997, 27, 1283-1289.
Ir _{0.3} Ru _{0.3} Ti _{0.4} O ₂	2	3.0 M H ₂ SO ₄	2	9	<i>Prog. Nat. Sci. Mater.</i> , 2014, 24, 134-141.
TiO ₂ -RuO ₂ -IrO ₂ - SnO ₂ (60:25:5:10)		0.5 M NaCl	1	9	<i>J. Sol-Gel Sci. Techn.</i> , 2019, 89, 553-561.
Ru _{0.25} Ir _{0.25} Ti _{0.5} O ₂	4	1.0 M H ₂ SO ₄	2	100	<i>Ionics</i> , 2013, 19, 113- 125.
40IrO ₂ -30RuO ₂ - 15Ta ₂ O ₅ -15TiO ₂	1.5	1.0 M Na ₂ SO ₄	0.1	200	<i>J. Mater. Eng. Perform.</i> , 2019, 28, 5595-5604.
38IrO ₂ -36RuO ₂ - 21TiO ₂ -5Ta ₂ O ₅		1.0 M NaCl	2	98	<i>Corros. Sci.</i> , 2023, 214, 111005.
36RuO ₂ -38IrO ₂ - 26TiO ₂		1.0 M NaCl	2	120	<i>Corros. Sci.</i> , 2022, 206, 110481.
IrO ₂ -RuO ₂ -SiO ₂	1.5	0.5 M H ₂ SO ₄	2	175	<i>Corros. Sci.</i> , 2020, 170, 108662.
Ta ₂ O ₅ -IrO ₂ (20:80)	1.35	1.0 M NaCl + 2.0 M HClO ₄	2	272	<i>Mater. Sci. Appl.</i> , 2011, 2, 237-243.
IrO ₂ -Ta ₂ O ₅ (70:30)	0.8	0.5 M H ₂ SO ₄	2	799	<i>Corros. Sci.</i> , 2002, 44, 1655-1668.
IrO ₂ -Ta ₂ O ₅ (70:30)		0.5 M H ₂ SO ₄	4.24	93	<i>Electrochim. Acta</i> , 2018, 263, 353-361.

IrO ₂ -Ta ₂ O ₅ (70:30)		1.0 M H ₂ SO ₄	2	450	<i>Corros. Sci.</i> , 2003, 45, 2729–2740.
Ta ₂ O ₅ -IrO ₂ -rGO		1.0 M H ₂ SO ₄	2	787	<i>J. Electrochem.Soc.</i> , 2020 167 146506
IrO ₂ -Ta ₂ O ₅		1.0 M H ₂ SO ₄ + 0.5 M Na ₂ SO ₄	2	108	<i>Sur. Coat. Tech.</i> , 2017, 320, 270-278.
IrO ₂ (35%)-Ta ₂ O ₅ (30%)-SiO ₂ (35%)		0.5 M H ₂ SO ₄	1.5	230	<i>Mater. Chem. Phys.</i> , 2022, 285, 126086.
RuTi/HC-IrTa	0.37	0.5 M H₂SO₄		602	This work

Table S4 Fitting parameters of equivalent circuits of LC-IrTa, HC-IrTa, RuTi/LC-IrTa, RuTi/HC-IrTa and RuTi-HC-IrTa electrodes

Reaction	Electrodes	$R_s/\Omega \text{ cm}_2$	$R_{ct}/\Omega \text{ cm}^2$
CER	LC-IrTa	0.41	0.28
	HC-IrTa	0.39	0.22
	RuTi/LC-IrTa	0.41	0.33
	RuTi/HC-IrTa	0.38	0.46
OER	LC-IrTa	0.41	3.37
	HC-IrTa	0.40	0.46
	RuTi/LC-IrTa	0.43	3.54
	RuTi/HC-IrTa	0.39	1.89