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Supporting information

Naphthyl and Biphenyl *para*-substituted POCOP-Ni(II) Pincer Complexes as Efficient Catalysts in C-S Cross-Coupling Reactions

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NMR Spectra of complexes



Figure S1. ¹H NMR (CDCl₃, 500 MHz) spectrum of complex 4





Figure S3. ³¹P{¹H} NMR (CDCl₃, 202 MHz) spectrum of complex 4



Figure S4. ¹H NMR (CDCl₃, 500 MHz) spectrum of complex 5



Figure S5. ¹³C{¹H} NMR (CDCl₃, 125 MHz) spectrum of complex 5



Figure S6. ³¹P{¹H} NMR (CDCl₃, 202 MHz) spectrum of complex 5



Figure S8. ${}^{13}C{}^{1}H$ NMR (CDCl₃, 125 MHz) spectrum of complex 6



Figure S10. ¹H NMR (CDCl₃, 500 MHz) spectrum of complex 7



Figure S11. ¹³C{¹H} NMR (CDCl₃, 125 MHz) spectrum of complex 7



Figure S12. ³¹P{¹H} NMR (CDCl₃, 202 MHz) Spectrum of complex 7



Figure S14. ¹³C{¹H} NMR (CDCl₃, 125 MHz) Spectrum of complex 8



Figure S16. ¹H NMR (CDCl₃, 500 MHz) Spectrum of complex 9



Figure S17. ¹³C{¹H} NMR (CDCl₃, 125 MHz) spectrum of complex 9



Figure S18. ³¹P{¹H} NMR (CDCl₃, 202 MHz) spectrum of complex 9

ATR-FTIR spectra of complexes



Figure S20. ATR-FTIR spectrum of complex 5







Figure S22. ATR-FTIR spectrum of complex 7



Figure S24. ATR-FTIR spectrum of complex 9

Complex	4	8	9
Empirical formula	[C ₃₁ H ₃₉ ClNiO ₄ P ₂][CHCl ₂]	[C ₆₆ H ₉₀ Cl ₂ Ni ₂ O ₈ P ₄][C ₇ H ₈]	C ₄₁ H ₂₉ Cl NiO ₄ P ₂
Formula weight	716.65	1415.71	741.74
T(K)	150(2)	199(2)	298(2)
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	$P2_1/n$	$P2_1/n$	$P2_1/n$
a (Å)	27.1448(13)	8.0438(2)	9.1360(4)
<i>b</i> (Å)	9.4225(4)	28.5592(10)	29.4452(16)
<i>c</i> (Å)	27.3977(12)	16.4261(5)	12.7283(7)
α (°)	90	90	90
β (°)	92.154(2)	102.7660(10)	94.484
γ (°)	90	90	90
$V(Å^3)$	7002.6(5)	3680.20(19)	3413.6(3)
Ζ	8	2	4
$ ho_{ m Calc}$ (Mg m ⁻³)	1.360	1.278	1.443
μ (mm ⁻¹)	0.908	0.723	0.784
<i>F</i> (000)	2992	1500	1528
Crystal size (mm ³)	0.404 x 0.258 x 0.202	0.407 x 0.221 x 0.104	0.398 x 0.202 x 0.161
Wavelength (Å)	0.71073	0.71073	0.71073
2θ range for data collection	2.286 to 25.401°	2.489 to 25.401°	2.624 to 25.369°
Index ranges	-32<=h<=28, - 11<=k<=11, -33<=l<=32	-9<=h<=9, -34<=k<=34, - 19<=l<=19	-11<=h<=11, - 31<=k<=35, -15<=l<=15
Reflections collected	49360	45956	35264
Independent reflections	12842 [R(int) = 0.0281]	6769 [R(int) = 0.0540]	6251 [R(int) = 0.0304]
Data/restraints/par ameters	12842 / 0 / 773	6769 / 235 / 508	6251 / 0 / 442
Goodness-of-fit (GOF) on <i>F</i> ²	1.170	1.068	1.116
Final <i>R</i> indices $[I \ge 2\sigma(I)]$	R1 = 0.0592, wR2 = 0.1280	R1 = 0.0502, wR2 = 0.0971	R1 = 0.0436, wR2 = 0.0882
Final <i>R</i> indices [all data]	R1 = 0.0701, wR2 = 0.1333	R1 = 0.0768, wR2 = 0.1090	R1 = 0.0561, wR2 = 0.0948
Largest difference peak/hole (e Å- ³)	0.948 and -0.732	0.496 and -0.350	0.383 and -0.224

Table S1. Crystallographic data of compounds 4, 8 and 9.

Bond	C-Ni (Å)	P-Ni (Å)	Ni-Cl (Å)	C-Ni-Cl (°)	P-Ni-P (°)
4	1.885(3)	2.1623(11)	2.2030(10)	178.96(12)	164.23(4)
	1.884(4)	2.1649(11)	2.2022(10)	178.56(13)	164.26(4)
		2.1587(11)			
		2.1640(12)			
8	1.882(3)	2.1870(9)	2.2049(9)	179.03(10)	163.37(4)
		2.1934(9)			
9	1.880(3)	2.1533(8)	2.1829(8)	176.79(8)	163.83(3)
		2.1680(7)			

Table S2. Selected distances and angles of compounds 4, 8 and 9.

 Table S3. Principal interactions in the molecular structure of compounds.

Compound	Interaction	Distance (Å)	Distance (Å) D…A	Angle (°)	Symmetry Operation
		D-X···A		D-X···A	
4	H41-Cg	2.996	3.736	131.72	¹ / ₂ -x, ¹ / ₂ +y, 1.5-z
	H16-Cg	2.921	3.643	129.87	1-x, 1-y, 1-z
	H58-O8	2.537	3.322	140.16	-1/2+x, ½-y, -1/2+z
	H64B-O8	2.288	3.262	167.45	x, y, z
	H63A-O4	2.264	3.157	149.47	x, y, z
8	H21C-O4	2.618	3.488	147.93	-x, 1-y, 1-z
	H17B-Cg	3.60	3.835	137.04	1-x, 1-y, 1-z
9	H4-Cg	3.261	3.954	133.05	-x, 1-y, 2-z
	H6-Cg	2.969	3.802	149.75	1-x, 1-y, 1-z
	H37-Cl	2.907	3.530	125.56	1-x, 1-y, 1-z
	H24-Cg	2.814	3.633	147.55	¹ / ₂ -x, -1/2+y, 1.5-z