

SUPPORTING INFORMATION FOR
Fe₃O₄@C Magnetite Nanocomposite: An Artificial Peroxidase Nanozyme for
Development of A Colorimetric Glucose Biosensor

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Table SI. 1. Mass of chemicals and reagents for FeC samples preparations

| Sample code | Mass ratio of Fe ₃ O ₄ :glucose | Fe@Cl ₃ .6H ₂ O (gam) | Fe(NH ₄) ₂ (SO ₄) ₂ .6H ₂ O (gam) | Glucose (gam) | Note |
|--------------------------------------|---|---|--|---------------|--|
| Fe ₃ O ₄ @C100 | 1:0 | 5,95 | 4,31 | 0 | Neat Fe ₃ O ₄ |
| Fe ₃ O ₄ @C101 | 1:1 | 5,95 | 4,31 | 2,5 | Carbon coated Fe ₃ O ₄ |
| Fe ₃ O ₄ @C102 | 1:2 | 5,95 | 4,31 | 5,0 | Carbon coated Fe ₃ O ₄ |
| Fe ₃ O ₄ @C103 | 1:3 | 5,95 | 4,31 | 7,5 | Carbon coated Fe ₃ O ₄ |
| Fe ₃ O ₄ @C105 | 1:5 | 5,95 | 4,31 | 12,5 | Carbon coated Fe ₃ O ₄ |
| Fe ₃ O ₄ @C107 | 1:7 | 5,95 | 4,31 | 17,5 | Carbon coated Fe ₃ O ₄ |
| Fe ₃ O ₄ @C110 | 1:10 | 5,95 | 4,31 | 25,0 | Carbon coated Fe ₃ O ₄ |

Table SI.2. Detected glucose concentration in reference sample 0.55mM glucose and diluted the 5% intravenous glucose solutions using the developed biosensor

| Samples | $A_{652\text{nm}}$ | Glucose concentration, mM | | |
|----------------|--------------------|---------------------------------------|---------------|------------------------|
| | | Detected concentration ^(*) | Diluted ratio | Original concentration |
| Refence sample | 0.331 | 0.562 | 1 | 0.562 |
| Diluted1/300 | 0.480 | 0.966 | 300 | 289.90 |
| Diluted1/200 | 0.528 | 1.095 | 200 | 219.09 |
| Diluted1/100 | 0.793 | 1.808 | 100 | 180.80 |

Note: ^(*) Extracted from the calibration curve;

Table SI.3. Detected glucose concentration in real samples (reference sample and 10 folds diluted human sera: (b) serum #1; (c) serum #2; (d) serum #3) using the developed biosensor

| Samples | $A_{652\text{nm}}$ | Glucose concentration, mM | | |
|----------------|--------------------|---------------------------------------|--------------------|------------------------|
| | | Detected concentration ^(*) | Diluted ratio(s) | Original concentration |
| Refence sample | 0.322 | 0.545 | N/A ^(*) | 0.545 |
| Serum #01 | 0.416 | 0.7943 | 1/10 | 7.943 |
| Serum #02 | 0.268 | 0.3965 | 1/10 | 3.965 |
| Serum #03 | 0.298 | 0.4772 | 1/10 | 4.77 |

Note: ^(*) Extracted from the calibration curve;

^(**) N/A: no application;

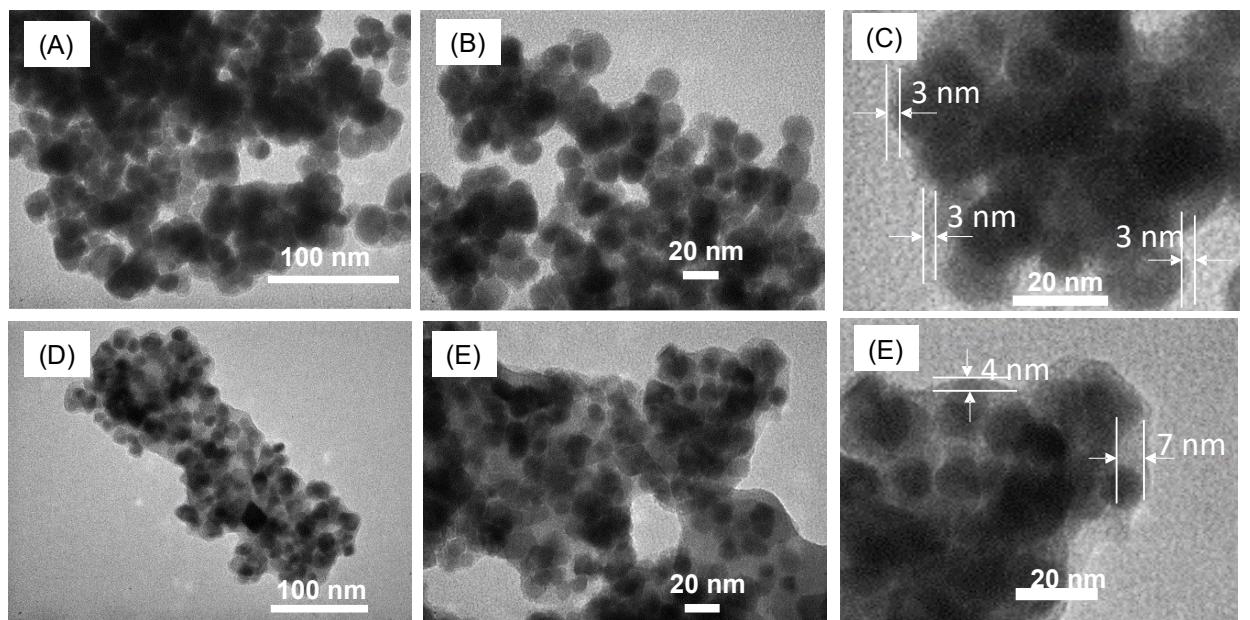


Figure SI1. TEM images of (A, B, C) $\text{Fe}_3\text{O}_4@\text{C}102$ and (D, E, F) $\text{Fe}_3\text{O}_4@\text{C}107$

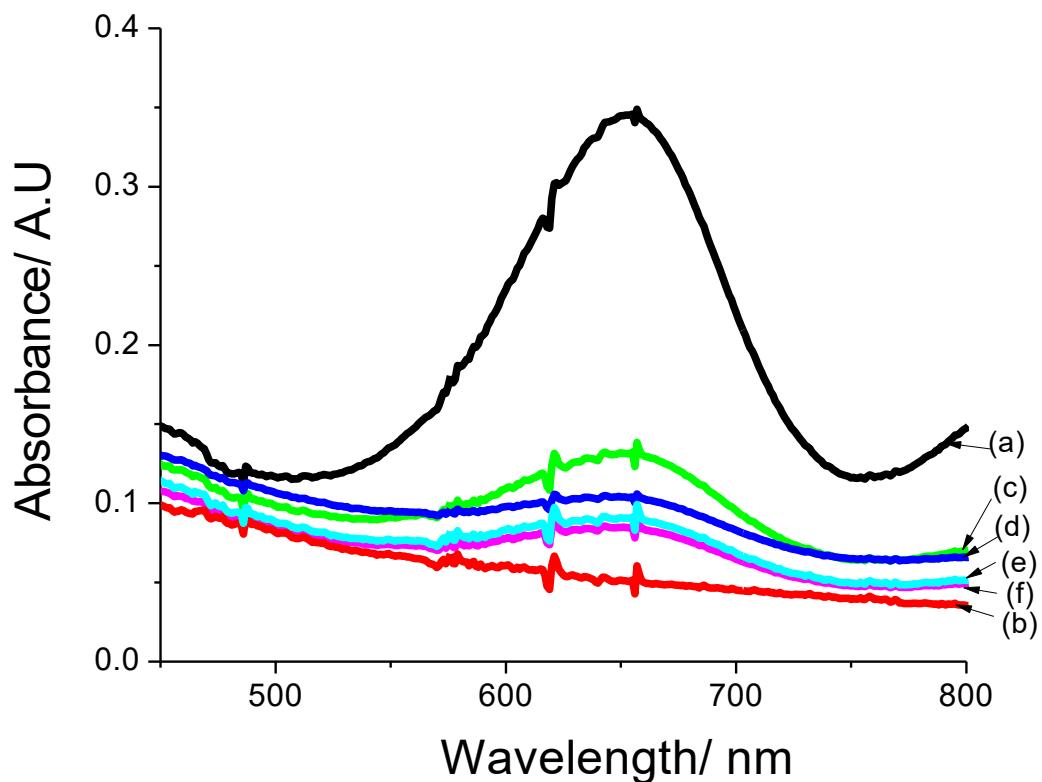


Figure SI.2. UV-vis spectra developed biosensors with diffirent substrates were used: (a) glucose; (b) acid ascorbic, (c) galactose, (d) sacarose and (e) saccarose. Conentration of all substrates were 0. 2 mM;

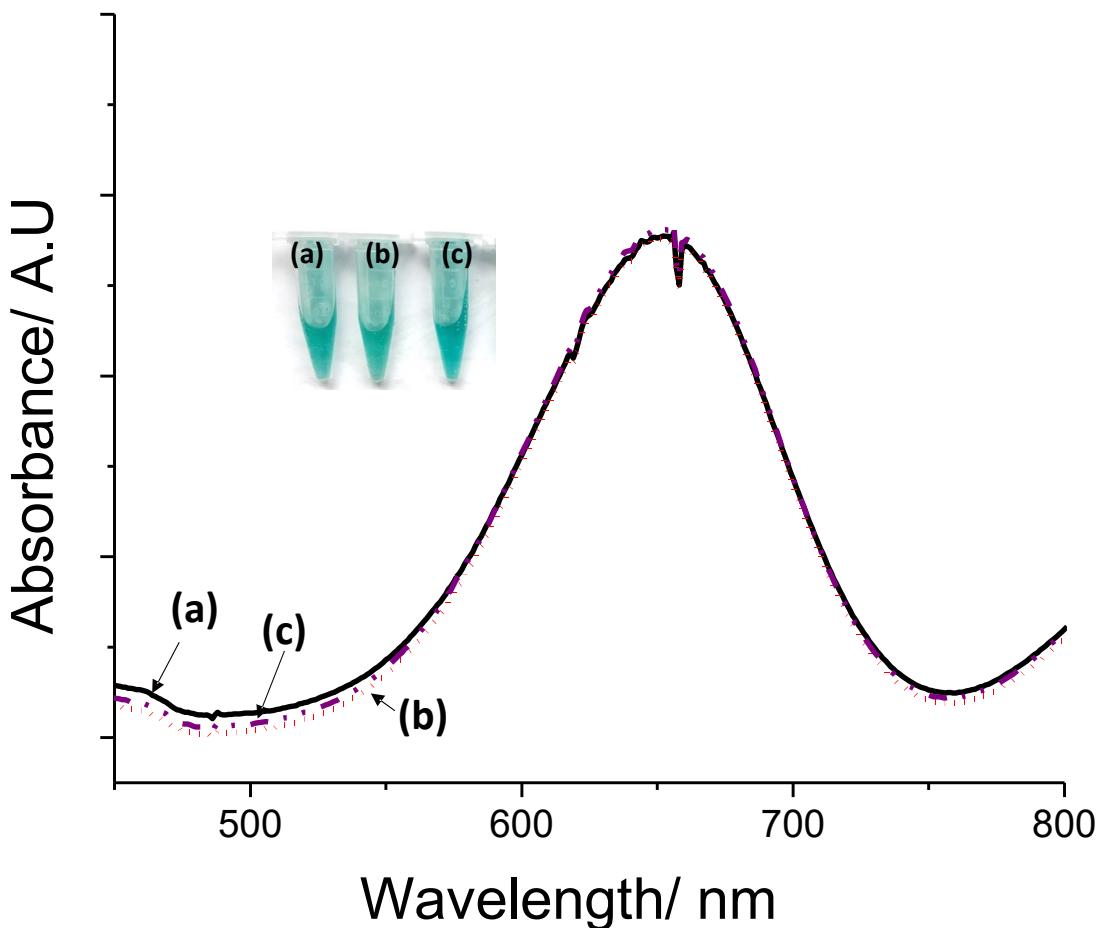


Figure SI.3. UV-vis spectra developed H_2O_2 sensor with $\text{Fe}_3\text{O}_4@\text{C}103$ nanozyme with different storage time: (a) as synthesized, (b) 6 months and (c) 1 year. Storage conditions: $\text{Fe}_3\text{O}_4@\text{C}103$ nanozyme was kept in vaccumm bag and at 4°C (in a fridge).

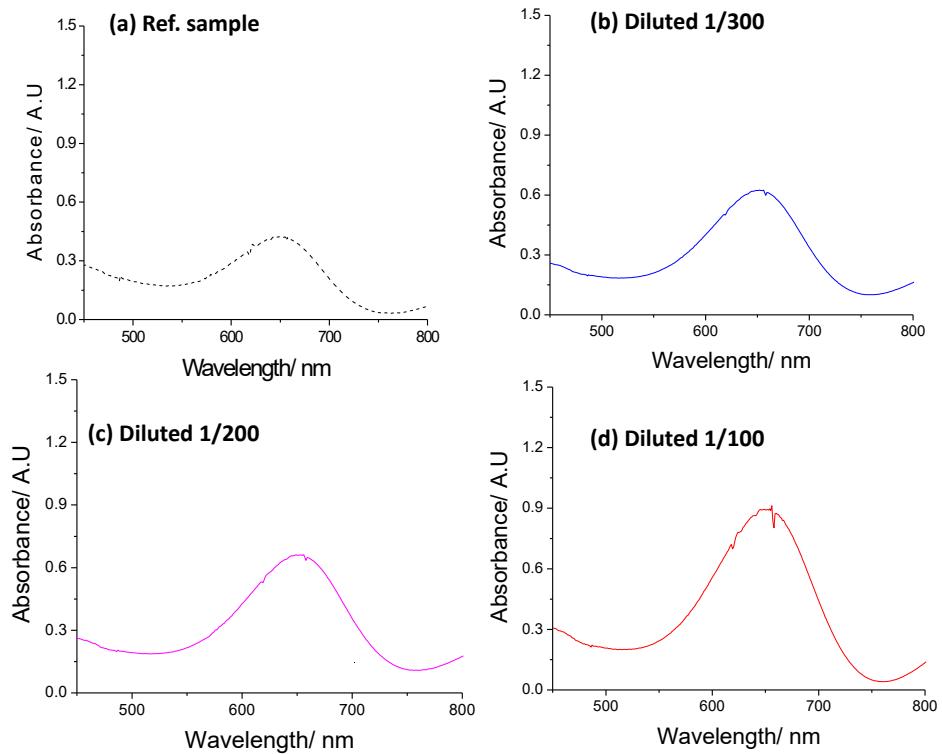


Figure SI.4. UV-vis spectra of the developed biosensor was tested with real samples: (a) control sample, (b-d) the 5% intravenous glucose solution was diluted by D.I water with dilution ratios v/v of (b) 1:300, (c) 1:200 and (d) 1:100, respectively.

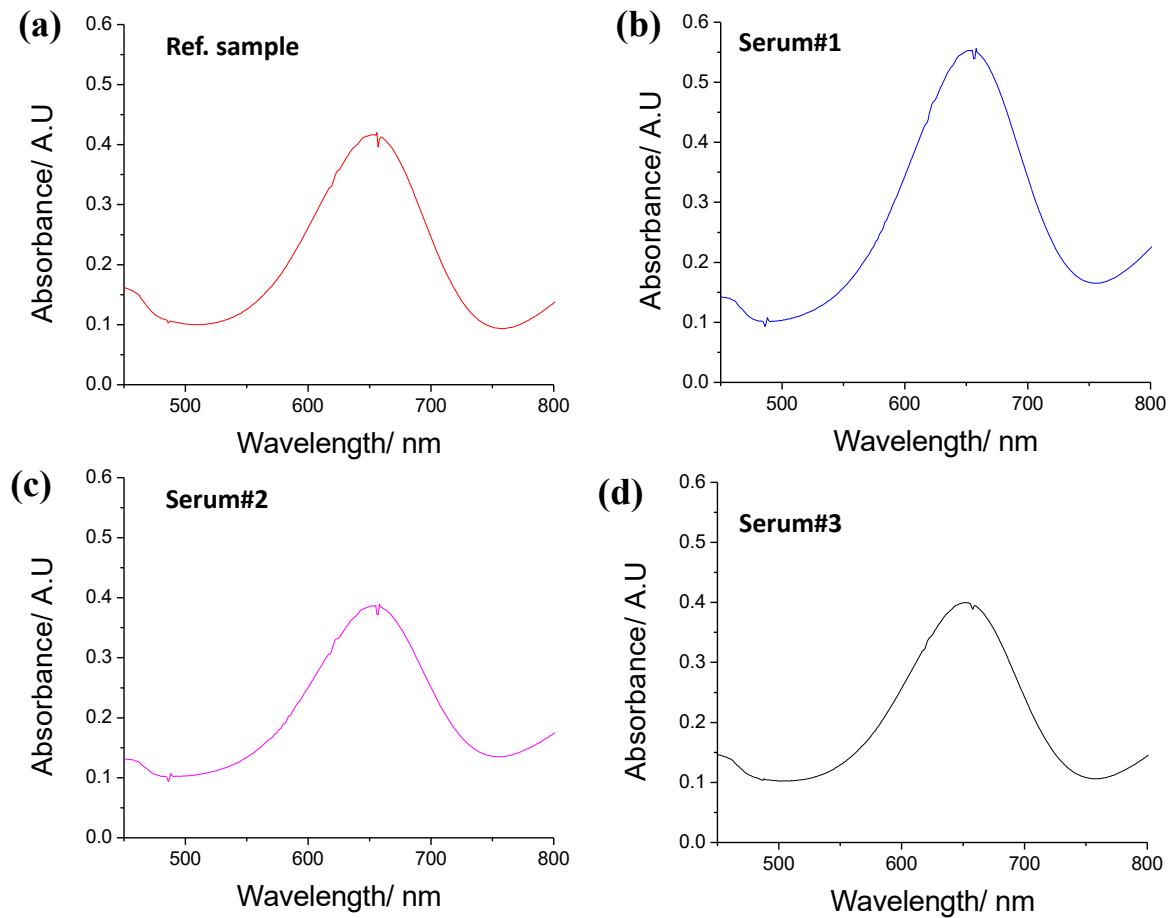


Figure SI.5. Response UV-vis spectra of the developed biosensor tested with real samples: (a) reference sample; (c-d) 10 folds diluted human serum sample (b) serum #1; (c) serum #2; (d) serum #3