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# Supporting Information

## Efficient Interfacial Optimization of NiO<sub>x</sub>-based Perovskite Solar Cells Via a Butterfly-Structured Semiconductor

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#### 1 Characterization.

The molecular structure was characterized utilizing a Bruker AV400 Spectrometer operating at 400 MHz and 100 MHz for <sup>1</sup>H NMR and <sup>13</sup>C NMR, respectively. UV-vis absorption spectra were recorded with a Shimadzu UV-2600 absorption spectrophotometer. Fluorescence emission spectra were recorded with a Hitachi F-4500 FL Spectrophotometer. Differential scanning calorimetry (DSC) was performed on a NETZSCH DSC 200 F3 differential scanning calorimeter with a scanning from - 40 °C to 300 °C at a rate of 10 °C min<sup>-1</sup> under a nitrogen atmosphere. Ultraviolet photoelectron spectroscopy (UPS) was carried out using a Thermo ESCALAB XI+ analysis system.

Hole mobility was evaluated using the space-charge-limited current (SCLC) method with a device configuration of ITO/PEDOT:PSS/HTMs/Ag.<sup>1, 2</sup> The hole mobility was calculated according to the formula below:

$$J = 9\mu\varepsilon_0\varepsilon_r V^2/(8d^3)$$

Here,  $\varepsilon_r \approx 3$  represents the average dielectric constant of the film,  $\varepsilon_0$  is the vacuum permittivity of the free space (8.85\*10<sup>-12</sup> F/m),  $\mu$  is the carrier mobility, *d* signifies the thickness of HTM film, and *V* stands for the applied voltage.

The SEM images were obtained by a JEOL JSM-7800 scanning electron microscope. The XPS spectra were acquired with a Kratos X-ray photoelectron spectrometer (AXIS Ultran HAS), which is equipped with monochromatic Al K $\alpha$  (1,486.6 eV) and non-monochromatic HeI (21.22 eV) sources. X-ray diffraction (XRD) characterization was performed using a Bruker D8 X-ray diffractometer

(Bruker D8 Advance diffractometer, Germany). Steady-state photoluminescence (PL) and time-resolved photoluminescence (TRPL) were measured through Spectrofluorometer FS5 with excitation at 500 nm.

The trap-state density  $(N_t)$  was obtained using a SCLC method by collecting the *I-V* characteristics of devices under dark conditions.<sup>3</sup> The  $N_t$  was calculated using the formula below:

$$N_{\rm t} = 2\varepsilon_0\varepsilon_{\rm r}V_{\rm TFL}/{\rm q}L^2$$

in which  $V_{\text{TFL}}$  stands for the trap filling limit voltage,  $\varepsilon_0$  represents the vacuum permittivity of the free space that is generally specified as  $8.85 \times 10^{-12}$  F m<sup>-1</sup>;  $\varepsilon_r$  signifies the material dielectric constant (~32). *L* is the thickness of perovskite films, and *q* is  $1.6 \times 10^{-19}$  C.

*J-V* curves were recorded by a Keithley 2400 source/meter under AM 1.5 illustration (100 mW cm<sup>-2</sup>), calibrated with a standard Si photodiode detector.

#### **2** Device Fabrication

A structure of ITO/NiO<sub>x</sub>/Y34/MAPbI<sub>3</sub>/PC<sub>60</sub>BM/Bathocuproine (BCP)/Ag was utilized for the fabrication of perovskite solar cells (PSCs). The cleaned ITO glass substrates were processed in an ultraviolet Ozone cleaner for 30 min. NiO<sub>x</sub> nanoparticles were dispersed in deionized water at a concentration of 20 mg/mL. The freshly prepared NiO<sub>x</sub> ink was spin-coated onto the ITO substrate at 2000 rpm for 30 s, and then annealed at 120 °C for 15 min in ambient air. Subsequently, Y34 in CB with a concentration of 5 mg mL<sup>-1</sup> was spin-coated onto the substrate at 3000 rpm for 30 s and annealed at 100 °C for 10 min. The perovskite layers were fabricated through a one-step spin-coating procedure. PbI<sub>2</sub> (466 mg) and MAI (159 mg) were added into 1 mL mixed solution of DMSO (71 uL) and DMF (639 uL), and further stirred for 30 min at 60 °C. Afterwards, the filtered precursor solution was spin-coated on the NiO<sub>x</sub> layer at 4000 rpm for 25 s. 110 uL of CB was poured on the spinning substrate. The substrates were heated at 100 °C for 10 min, followed by spin-coating a solution of PC<sub>61</sub>BM (20 mg/mL in CB) at 3000 rpm for 30 s. Subsequently, a BCP solution (0.5 mg/mL in IPA) was deposited at 4000 rpm for 40 s as a buffer layer. Finally, the electrode Ag with a thickness of 100 nm was coated on the top of BCP film using thermal evaporation.





Figure S1. <sup>1</sup>H NMR spectra of Y34 (CDCl<sub>3</sub>).



Figure S2. <sup>13</sup>C NMR spectra of Y34 (CDCl<sub>3</sub>).



Figure S3. Cyclic voltammetry curves of (a) Y34 and (b) ferrocene in  $CH_2Cl_2$  solution.



Figure S4. The statistics of PSCs based on NiO<sub>x</sub> and NiO<sub>x</sub>/Y34. (a)  $V_{OC}$  and (b)  $J_{sc}$ .



Figure S5. Trap density measured utilizing the SCLC method.

### 4 Tables.

Table S1. Electrochemical and optical data of Y34.

HTM	$\lambda_{int}/\mathrm{nm}$	$E_g/eV^{(a)}$	$E_{ m HOMO}$ /V <sup>(b)</sup>	$E_{\rm HOMO}$ /eV <sup>(c)</sup>	$E_{\rm LUMO}/{\rm eV^{(d)}}$
Y34	441	2.81	0.76	-5.26	-2.45

<sup>(a)</sup>  $E_g = 1240/\lambda_{int}$ .  $\lambda_{int}$  is the intersection point of normalized UV-visible absorption curve and the fluorescence emission curve.

<sup>(b)</sup> The redox potential obtained from the CV plots and it was standardized with ferrocene.

<sup>(c)</sup>  $E_{\text{HOMO}} vs (\text{Ag/AgCl}) (\text{eV}) = -(E_{\text{HOMO}}^{(b)} + 4.5].$ <sup>(d)</sup>  $E_{\text{LUMO}} (\text{eV}) = E_{\text{HOMO}}^{(c)} + E_{\text{g}} (\text{eV}).$ 

HTMs	$V_{ m oc}$ <sup>a</sup> /V	$J_{\rm sc}$ <sup>a</sup> /mA cm <sup>-2</sup>	FF <sup>a</sup> /%	PCE a/%
NiO <sub>x</sub>	$1.054 \pm 0.007$	22.66±0.197	72.75±1.673	17.37±0.488
NiO <sub>x</sub> /Y34	1.075±0.005	22.78±0.079	75.03±0.591	18.36±0.174

Table S2. Calculated result of repeatability test.

Table S3. Fitted carrier lifetime of perovskite films measured by TRPL measurements.

Sample	$\tau_1 / ns$	A1/%	$\tau_2$ / ns	A2/%	$ au_{\rm avg}/{ m ns}$
MAPbI <sub>3</sub>	28.98	91.61	321.65	4.39	176.53
NiO <sub>x</sub>	21.11	90.69	207.25	9.31	114.55
NiO <sub>x</sub> /Y34	12.92	90.24	143.21	9.75	83.94

Note: The average lifetime can be calculated with the equation of:  $\tau_{avg} = (A_1\tau_1^2 + A_2\tau_2^2)/(A_1\tau_1 + A_2\tau_2)$ , where parameters  $A_1$  and  $A_2$  are the amplitude fraction foreach decay component, and  $\tau_1$  and  $\tau_2$  represent the time constant of the two types of decay.

Table S4. Fitted XRD ratio of perovskite films deposited on different substrates.

Sample	ITO/NiO <sub>x</sub> /Perovskite	NiO <sub>x</sub> /Perovskite/Y34	ITO/Perovskite
FWHM	0.29826	0.26693	0.34139

## 4 References.

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