

## Supporting Information

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4 **Direct synthesis of 2-ethylhexanal via *n*-butanal aldol**  
5 **condensation-selective hydrogenation reaction integration over a**  
6 **highly stable Pd/TiO<sub>2</sub> catalyst**

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8 Meng Li, Diangen Zhou, Hualiang An\*, Shanshan Dong\*, Xinqiang Zhao and Yanji Wang  
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10 Hebei Provincial Key Laboratory of Green Chemical Technology and Efficient Energy Saving,  
11 Tianjin Key Laboratory of Chemical Process Safety, School of Chemical Engineering and  
12 Technology, Hebei University of Technology, Tianjin 300130, China  
13 \*Corresponding author. E-mail addresses: anhl@hebut.edu.cn (Hualiang An); dss@hebut.edu.cn  
14 (Shanshan Dong)

Table S1 Textural property of Pd/TiO<sub>2</sub> with different Pd loading

Pd loading /wt.%	Specific surface area /m <sup>2</sup> .g <sup>-1</sup>	Pore volume /cm <sup>3</sup> .g <sup>-1</sup>	Average pore size /nm
0.1	169.26	0.37	6.56
0.3	156.55	0.36	7.01
0.5	146.62	0.36	7.41
0.7	139.15	0.35	7.56
0.9	131.70	0.33	7.88

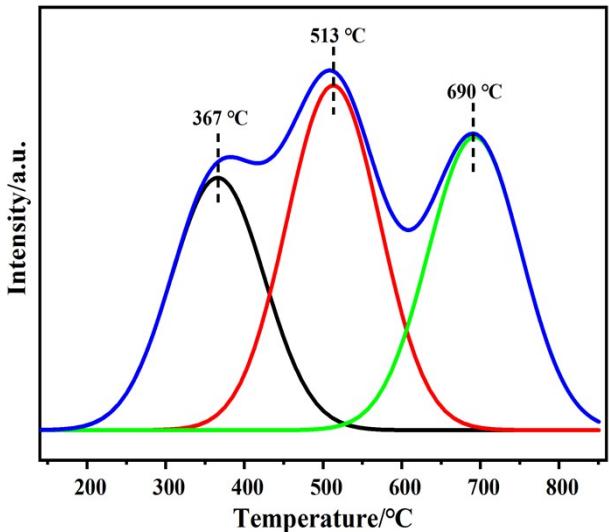
**Table S2** Effect of calcination temperature on catalytic performance of Pd/TiO<sub>2</sub>

Temperature /°C	$X_{\text{BA}}/\%$	$S_{2\text{EH}}/\%$	$S_{2\text{E}2\text{H}}/\%$
100	96.5	93.2	0
200	95.7	96.3	0
300	95.4	99.2	0
400	95.2	98.2	0
500	93.5	97.9	0

17 Reaction conditions: 180 °C; 6 h; 2 MPa; m<sub>BA</sub>:m<sub>cat</sub>=10:1.

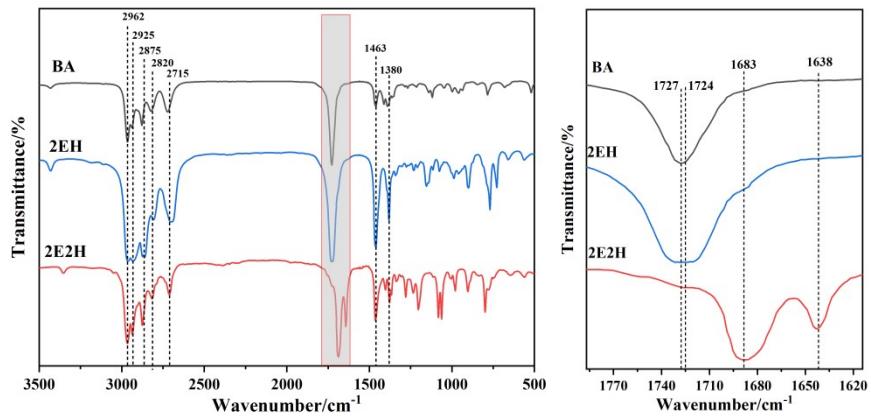
18 BA: *n*-butanal; 2EH: 2-ethylhexanal; 2E2H: 2-ethyl-2-hexenal; X: conversion; S: selectivity.

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Fig. S1 H<sub>2</sub>-TPR curves of Pd/TiO<sub>2</sub>

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Fig. S2 Infrared spectra of *n*-butanal (BA), 2-ethylhexanal (2EH) and 2-ethyl-2-hexenal (2E2H)