

Supporting information for

Graphene-based iron single-atom catalyst for electrocatalytic nitric oxide reduction: a first-principles study

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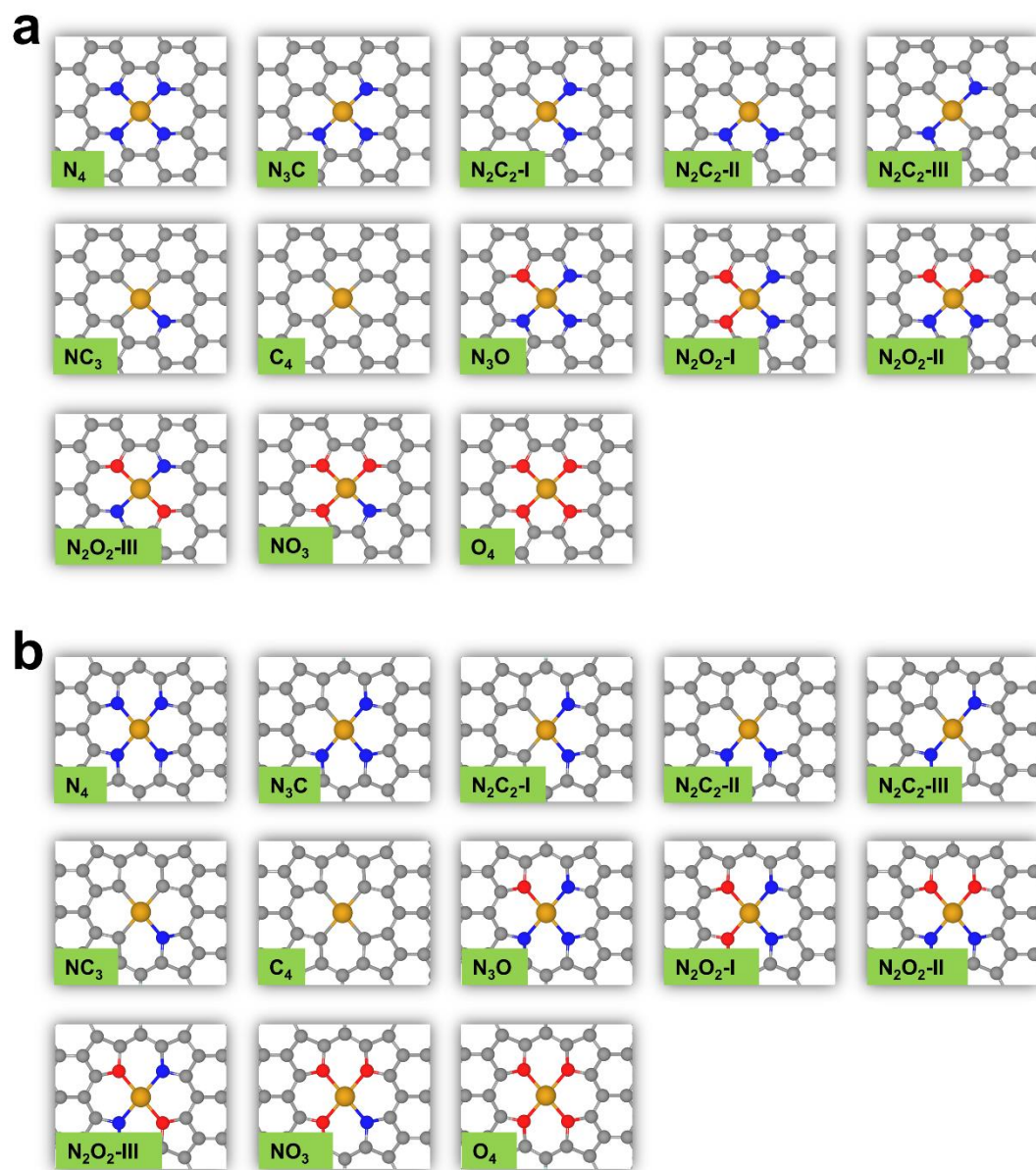


Fig. S1. Optimized atomic structures of the PD-Fe-SACs (a) and PR-Fe-SACs (b).

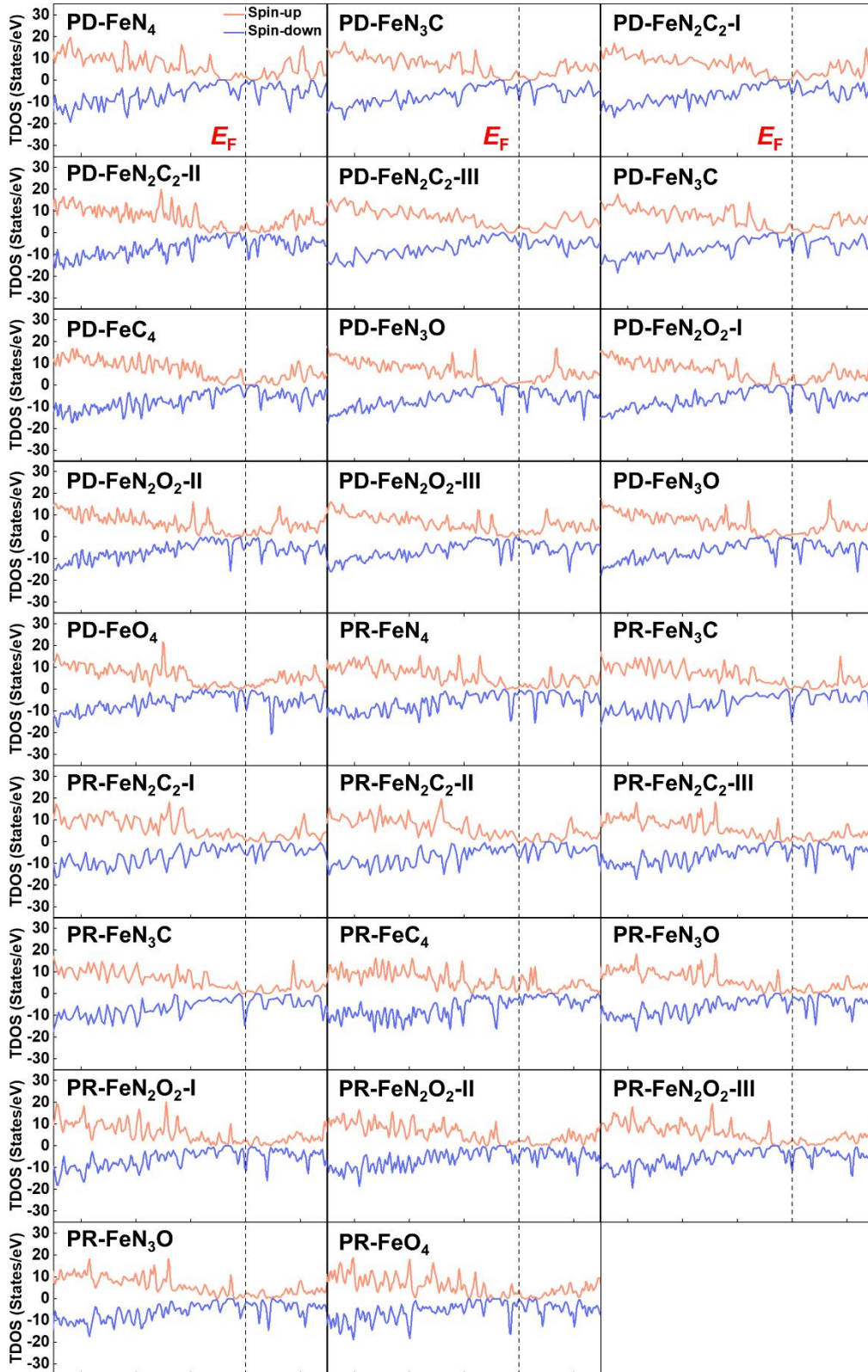


Fig. S2. Calculated total density of states (TDOS) of the PD-Fe-SACs and PR-Fe-SACs. The Fermi level (E_F) is set to 0 eV.

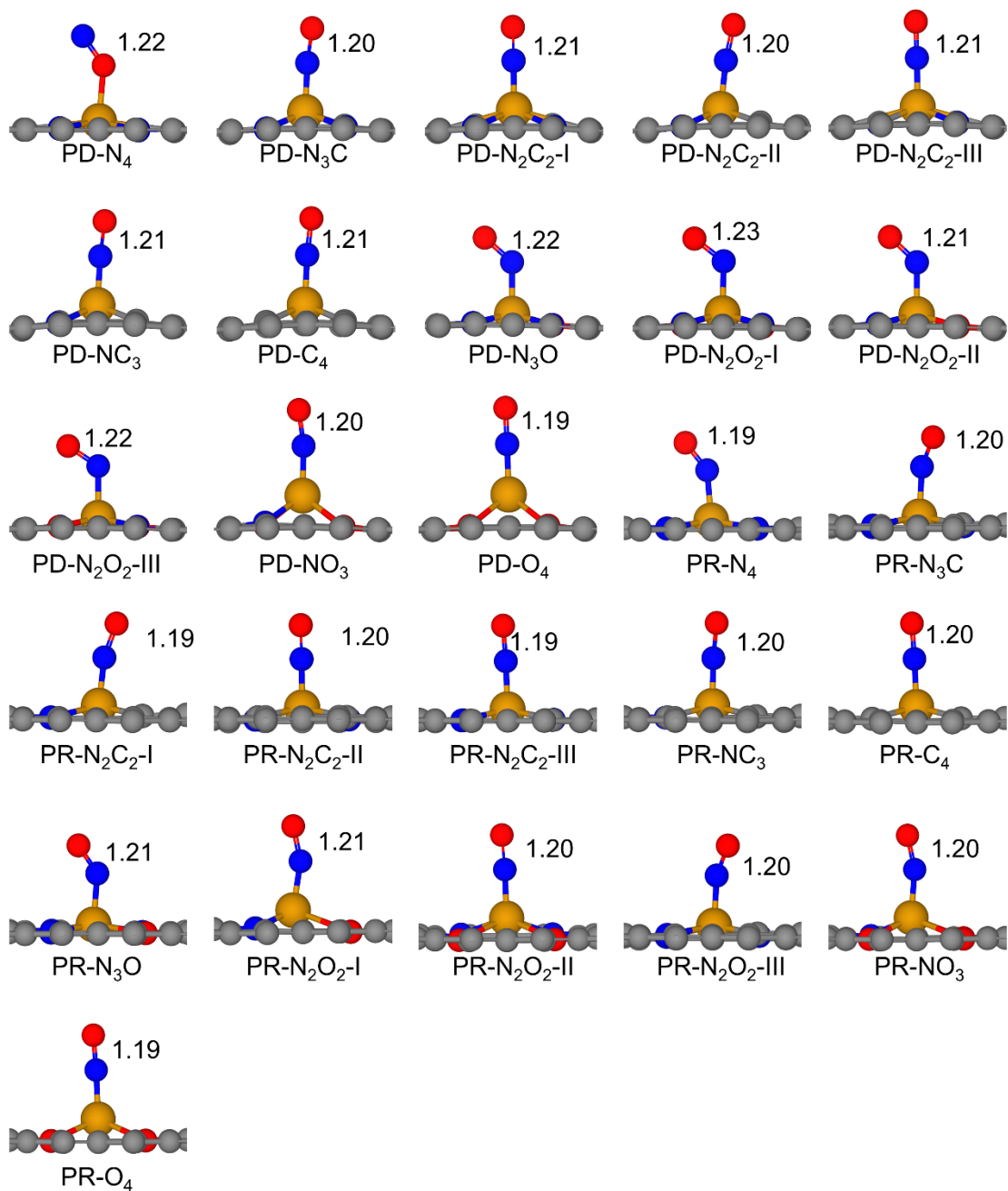
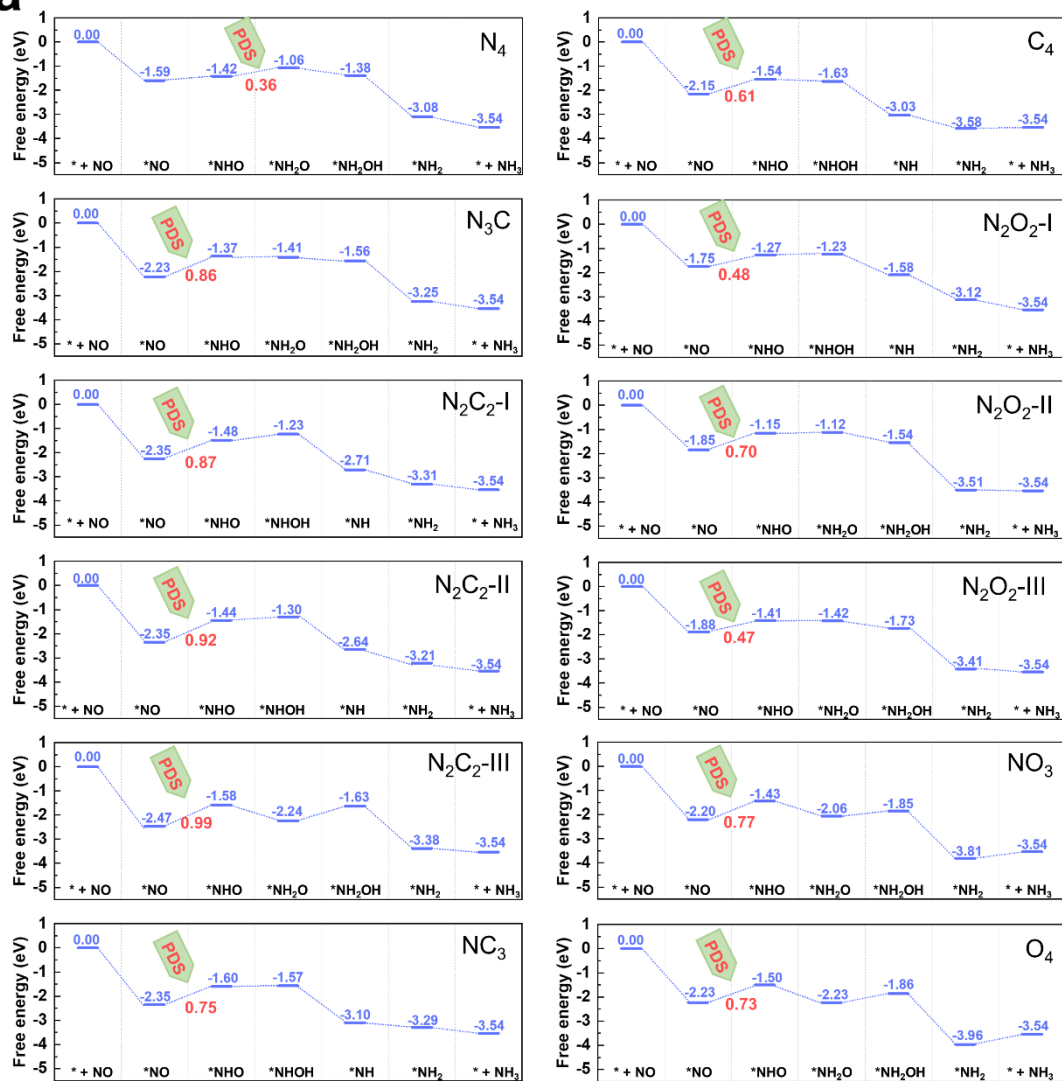


Fig. S3. The most stable configurations and corresponding bond length (in Å) of NO adsorption on the PD-Fe-SACs and PR-Fe-SACs.

a

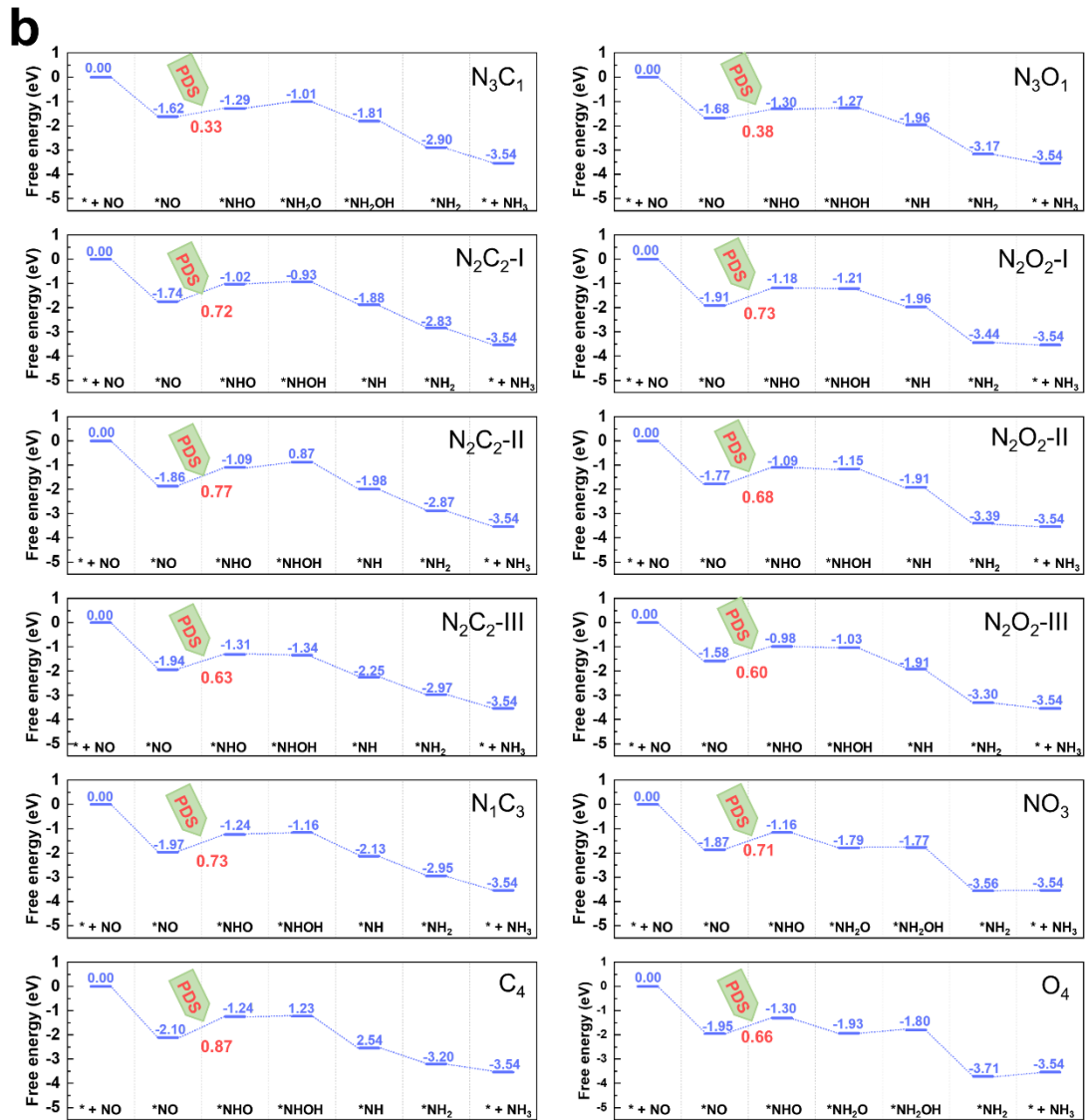


Fig. S4. NORR free energy diagram through the most favorable pathway for PD-Fe-SACs (a) and PR-Fe-SACs (b).

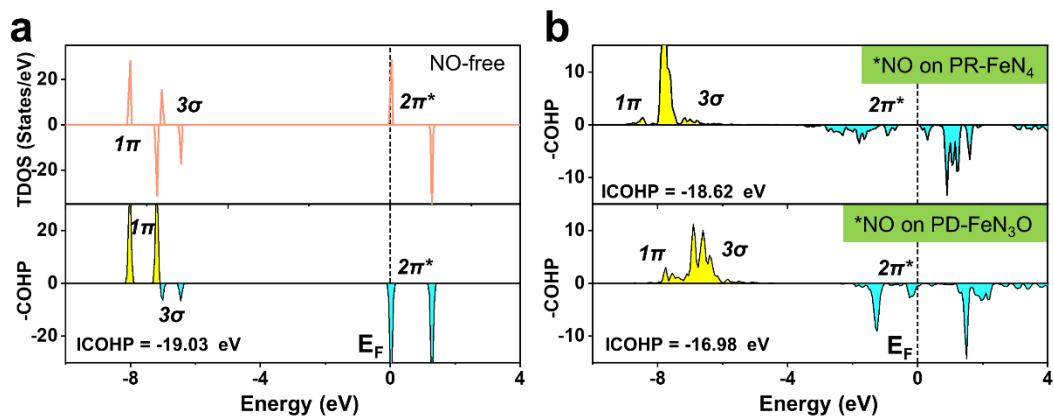


Fig. S5. (a) TDOS and corresponding COHP of free NO molecule. (b) COHP of *NO on PR-FeN₄ (up) and PD-FeN₃O (down) SACs. The Fermi level (E_F) is set to 0 eV.

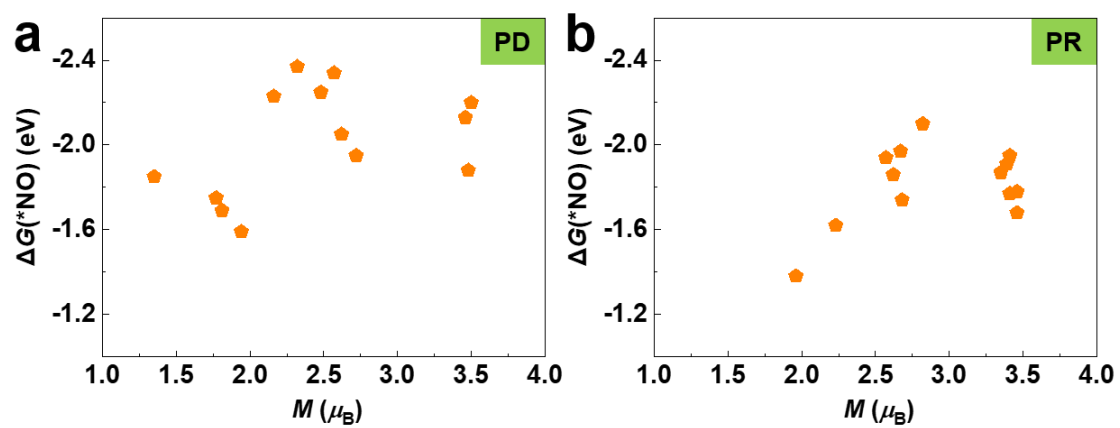


Fig. S6. The correlation between NO adsorption free energy ($\Delta G(*NO)$) and magnetic moment (M) of the Fe active center for the PD-Fe-SACs (a) and PR-Fe-SACs (b).

Table S1. Calculated zero-point energy (E_{ZPE} , in eV) and the product of temperature ($T = 298.15$ K) and entropy (S) (TS , in eV) of various species involved in the reaction pathways for NORR, where * represents the adsorption site.

Species	E_{ZPE}	TS
*NO	0.19	0.09
*NHO	0.49	0.15
*NOH	0.48	0.14
*NH ₂ O	0.81	0.19
*NHOH	0.79	0.17
*NH ₂ OH	1.12	0.22
*NH	0.34	0.07
*NH ₂	0.67	0.11
*NH ₃	1.04	0.14
*N ₂ O ₂	0.38	0.24
*N ₂ O ₂ H	0.68	0.27
*N ₂ O	0.31	0.21
*NO	0.19	0.09
*NHO	0.49	0.15
*NOH	0.48	0.14
*NH ₂ O	0.81	0.19
N ₂ O	0.29	0.68
NH ₃	0.89	0.60
N ₂	0.15	0.59
H ₂	0.27	0.40
H ₂ O	0.56	0.67

Table S2. Key parameters of the studied Fe-SACs, including spin magnetic moment (M , in μ_B), net charge (Q , in e), the positive value means the electrons are transferred from the Fe atom to supports, and the d -band center (ε_d , in eV) of the Fe atom.

Systems	M	Q	ε_d
PD-N ₄	1.94	1.40	-1.39
PD-N ₃ C	2.16	1.38	-1.08
PD-N ₂ C ₂ -I	2.48	1.32	-1.06
PD-N ₂ C ₂ -II	2.57	1.37	-0.95
PD-N ₂ C ₂ -III	2.32	1.35	-0.90
PD-NC ₃	2.62	1.31	-1.24
PD-C ₄	2.72	1.29	-1.29
PD-N ₃ O	1.81	1.42	-1.32
PD-N ₂ O ₂ -I	1.77	1.53	-1.30
PD-N ₂ O ₂ -II	1.35	1.33	-1.12
PD-N ₂ O ₂ -III	3.48	1.18	-1.28
PD-NO ₃	3.50	1.17	-1.19
PD-O ₄	3.46	1.38	-1.14
PR-N ₄	1.96	1.34	-1.46
PR-N ₃ C	2.23	1.25	-1.36
PR-N ₂ C ₂ -I	2.68	1.19	-1.21
PR-N ₂ C ₂ -II	2.62	1.45	-1.13
PR-N ₂ C ₂ -III	2.57	1.38	-1.26
PR-NC ₃	2.67	1.35	-1.13
PR-C ₄	2.82	1.44	-1.06
PR-N ₃ O	3.46	1.42	-1.36
PR-N ₂ O ₂ -I	3.39	1.35	-1.16
PR-N ₂ O ₂ -III	3.41	1.40	-1.20
PR-N ₂ O ₂ -III	3.46	1.35	-1.24
PR-NO ₃	3.35	1.39	-1.11
PR-O ₄	3.41	1.44	-1.28

Table S3. N-O bond length of *NO ($d_{\text{N-O}}$, in Å) and the number of electrons transferred from the SACs to *NO on the studied Fe-SACs ($\Delta Q(*\text{NO})$, in e).

Systems	$d_{\text{N-O}}$	$\Delta Q(*\text{NO})$
PD-N ₄	1.22	0.53
PD-N ₃ C	1.20	0.62
PD-N ₂ C ₂ -I	1.21	0.64
PD-N ₂ C ₂ -II	1.20	0.63
PD-N ₂ C ₂ -III	1.21	0.65
PD-NC ₃	1.21	0.61
PD-C ₄	1.21	0.56
PD-N ₃ O	1.22	0.54
PD-N ₂ O ₂ -I	1.23	0.56
PD-N ₂ O ₂ -II	1.21	0.57
PD-N ₂ O ₂ -III	1.22	0.57
PD-NO ₃	1.20	0.64
PD-O ₄	1.19	0.63
PR-N ₄	1.19	0.29
PR-N ₃ C	1.20	0.37
PR-N ₂ C ₂ -I	1.19	0.35
PR-N ₂ C ₂ -II	1.20	0.41
PR-N ₂ C ₂ -III	1.19	0.54
PR-NC ₃	1.20	0.53
PR-C ₄	1.20	0.57
PR-N ₃ O	1.21	0.36
PR-N ₂ O ₂ -I	1.21	0.53
PR-N ₂ O ₂ -III	1.20	0.39
PR-N ₂ O ₂ -III	1.20	0.36
PR-NO ₃	1.20	0.45
PR-O ₄	1.19	0.46