

Supporting information

Proceeding of catalytic water splitting on Cu/Ce@g-C₃N₄ photocatalysts: an exceptional approach for sun light driven hydrogen generation†

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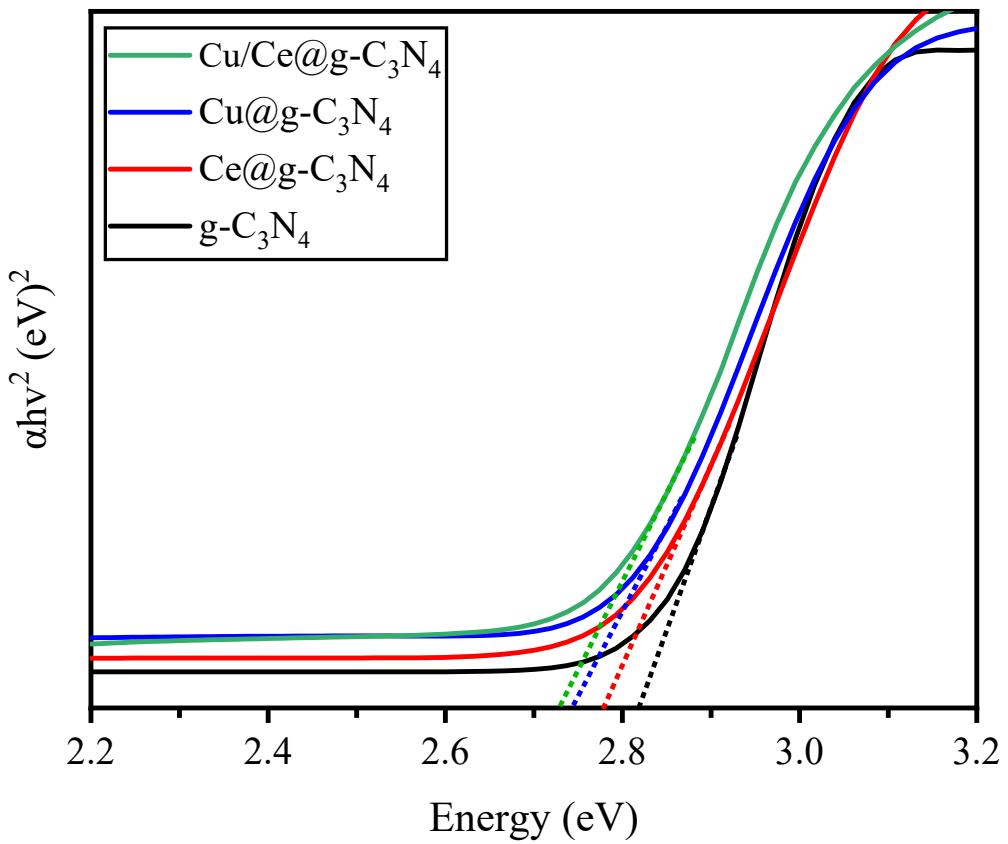


Figure S1: Bandgap energies of g-C₃N₄, Ce@g-C₃N₄, Cu@g-C₃N₄, Cu/Ce@g-C₃N₄ catalysts calculated using Tauc plot method.

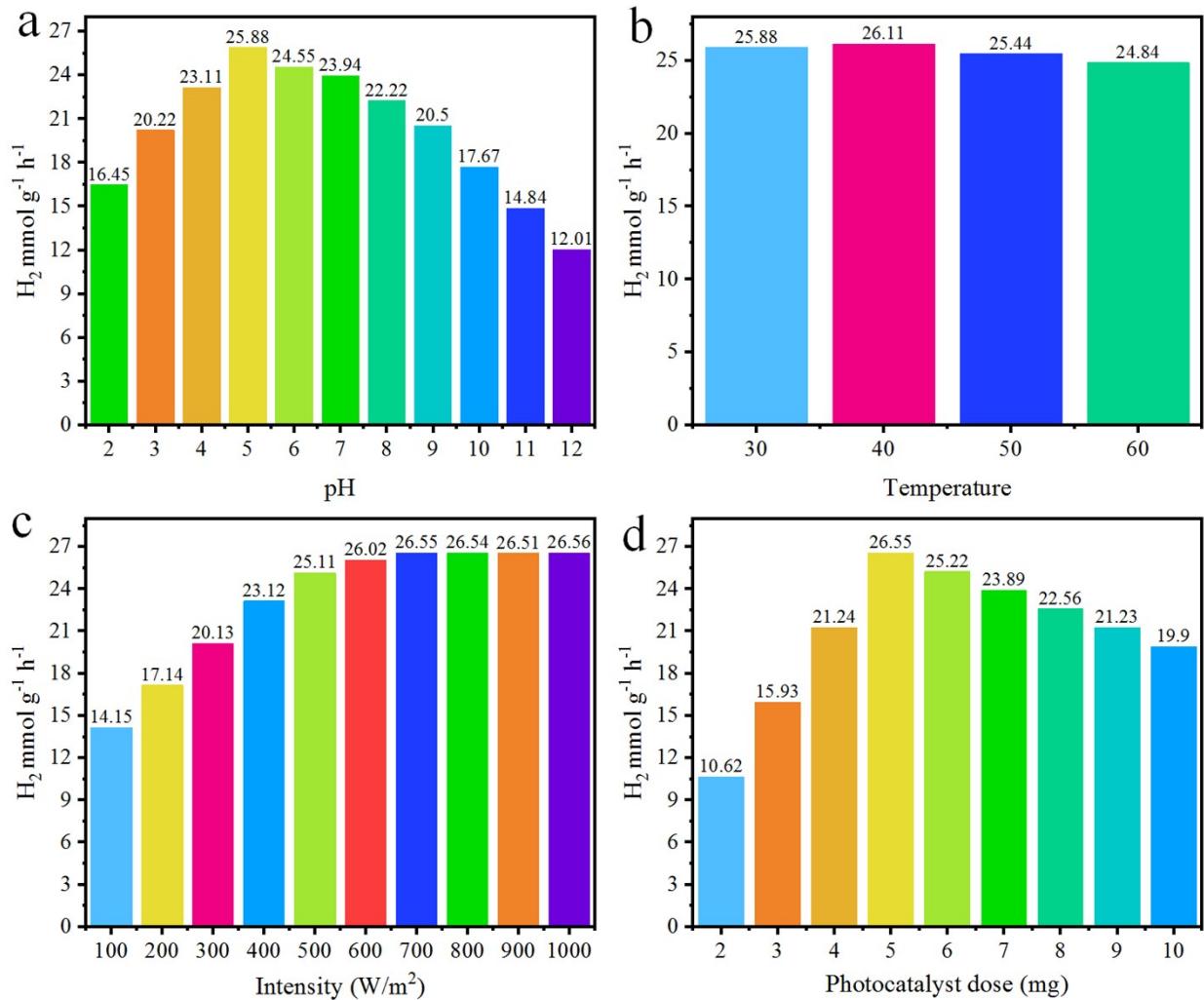


Figure S2: a) Effect of pH b) temperature c) light intensity and d) correspond to the effect of photocatalyst dose on hydrogen production rate.

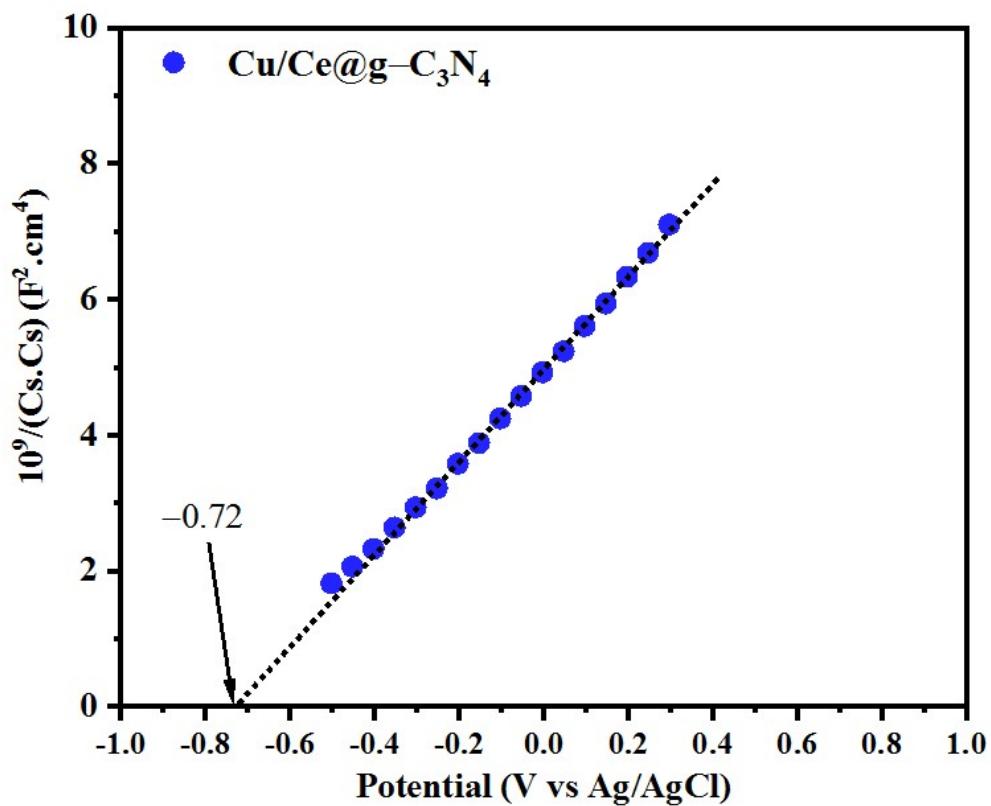


Figure S3: Mott-schottky plot of $\text{Cu/Ce@g-C}_3\text{N}_4$ catalysts.

Table S1: Crystalline sizes and band gap energies of as-synthesized catalysts.

Photocatalysts	2θ	D(nm)	Band gap (eV)
g-C ₃ N ₄	27.21	28.25	2.82
Cu@g-C ₃ N ₄	27.21	27.88	2.78
Ce@g-C ₃ N ₄	27.20	27.95	2.74
Cu/Ce@g-C ₃ N ₄	27.20	27.45	2.72

Table S2: The hydrogen evolution performances of as-synthesized catalysts.

Sr. No.	Photocatalyst	H ₂ mmol g ⁻¹ h ⁻¹
1	g-C ₃ N ₄	4.34± 0.04
2	2%Cu@g-C ₃ N ₄	9.37± 0.37
3	2%Ce@g-C ₃ N ₄	11.91± 0.62
4	2%Cu/Ce@g-C ₃ N ₄	23.94± 0.70

Table S3: Effect of pH on hydrogen evolution performances of Cu/Ce@g-C₃N₄ photocatalysts.

Sr. No.	pH	H ₂ mmol g ⁻¹ h ⁻¹
1	2	16.45
2	3	20.22
3	4	23.11
4	5	25.88
5	6	24.55
6	7	23.94
7	8	22.22
8	9	20.5
9	10	17.67
10	11	14.84
11	12	12.01

Table S4: Effect of temperature on photocatalytic activity of Cu/Ce@g-C₃N₄ photocatalysts.

Sr. No.	Temperature (°C)	H ₂ mmol g ⁻¹ h ⁻¹
1	30	25.88
2	40	26.11
3	50	25.44
4	60	24.84

Table S5: Effect of light intensity on photocatalytic activity of Cu/Ce@g-C₃N₄ photocatalysts.

Sr. No.	Light intensity (W/m ²)	H ₂ mmol g ⁻¹ h ⁻¹
1	100	14.15
2	200	17.14
3	300	20.13
4	400	23.12
5	500	25.11
6	600	26.02
7	700	26.55
8	800	26.54
9	900	26.51
10	1000	26.56

Table S6: Effect of catalyst dosage on photocatalytic activity of Cu/Ce@g-C₃N₄ photocatalysts.

Sr. No.	Catalyst dose (mg)	H ₂ mmol g ⁻¹ h ⁻¹
1	2	10.62
2	3	15.93
3	4	21.24
4	5	26.55
5	6	25.22
6	7	23.89
7	8	22.56
8	9	21.23
9	10	19.9