

## Supporting Information

### ***In-situ self-assembly of molybdenum carbide and iron carbides heterostructure on N-doped carbon for efficient oxygen reduction reaction***

Sagar Ingavale<sup>1</sup>, Mohan Gopalakrishnan<sup>1</sup>, Phiralang Marbaniang<sup>2</sup>, Woranunt Lao-atiman<sup>1</sup>, Ahmad Azmin Mohamad<sup>3</sup>, Mai Thanh Nguyen<sup>4</sup>, Tetsu Yonezawa<sup>4</sup>, Anita Swami<sup>5\*</sup>, Soorathep Kheawhom<sup>1,6,7\*</sup>

<sup>1</sup> Department of Chemical Engineering, Faculty of Engineering, Chulalongkorn University, Bangkok 10330, Thailand

<sup>2</sup> Electrochemical Materials Lab, Faculty of Science (Chemistry), Ontario Tech University, Oshawa, ON L1G0C5, Canada

<sup>3</sup> Energy Materials Research Group (EMRG), School of Materials and Mineral Resources Engineering, Universiti Sains Malaysia, 14300, Nibong Tebal, Pulau Pinang, Malaysia

<sup>4</sup> Division of Materials Science and Engineering, Faculty of Engineering, Hokkaido University, Hokkaido 060-8628, Japan

<sup>5</sup> Department of Chemistry, SRM Institute of Science & Technology, Kattankulathur, Chennai 603203, India

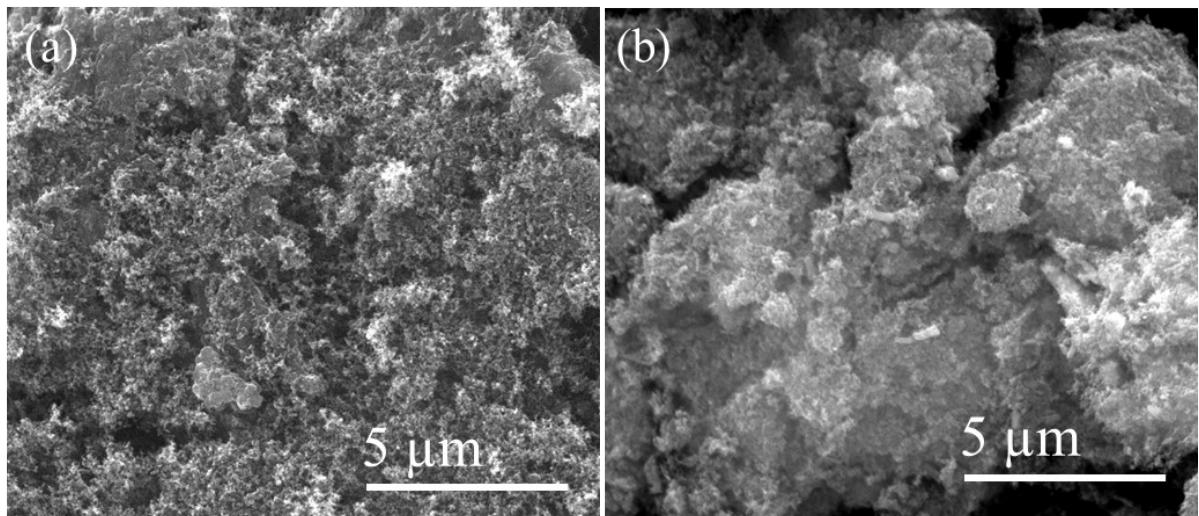
<sup>6</sup> Center of Excellence on Advanced Materials for Energy Storage, Chulalongkorn University, Bangkok 10330, Thailand

<sup>7</sup> Bio-Circular-Green-economy Technology & Engineering Center (BCGeTEC), Faculty of Engineering, Chulalongkorn University, Bangkok 10330, Thailand

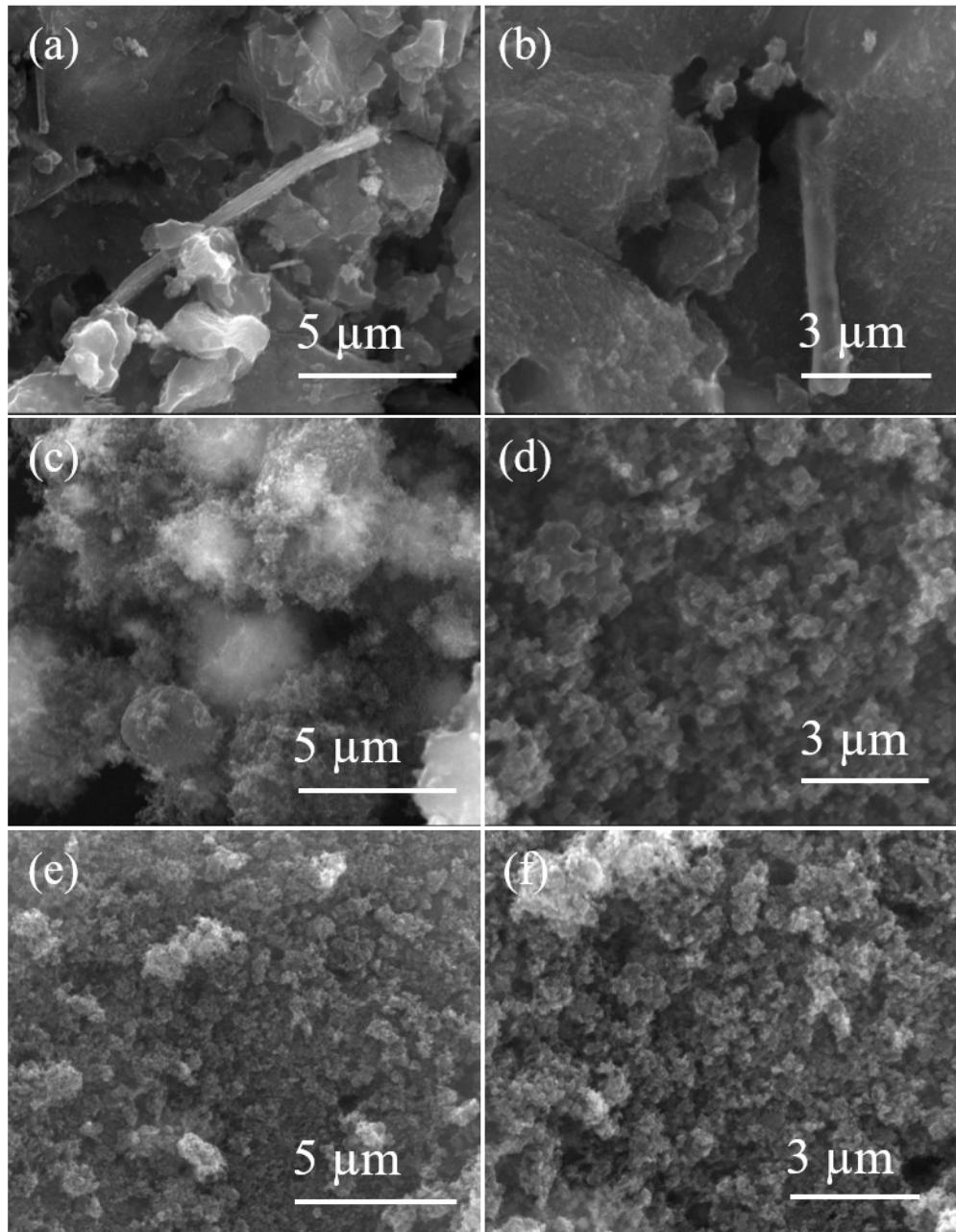
\*Corresponding author Email: [swamians@srmist.edu.in](mailto:swamians@srmist.edu.in) (Anita Swami), [soorathep.k@chula.ac.th](mailto:soorathep.k@chula.ac.th) (Soorathep Kheawhom)

## Characterization

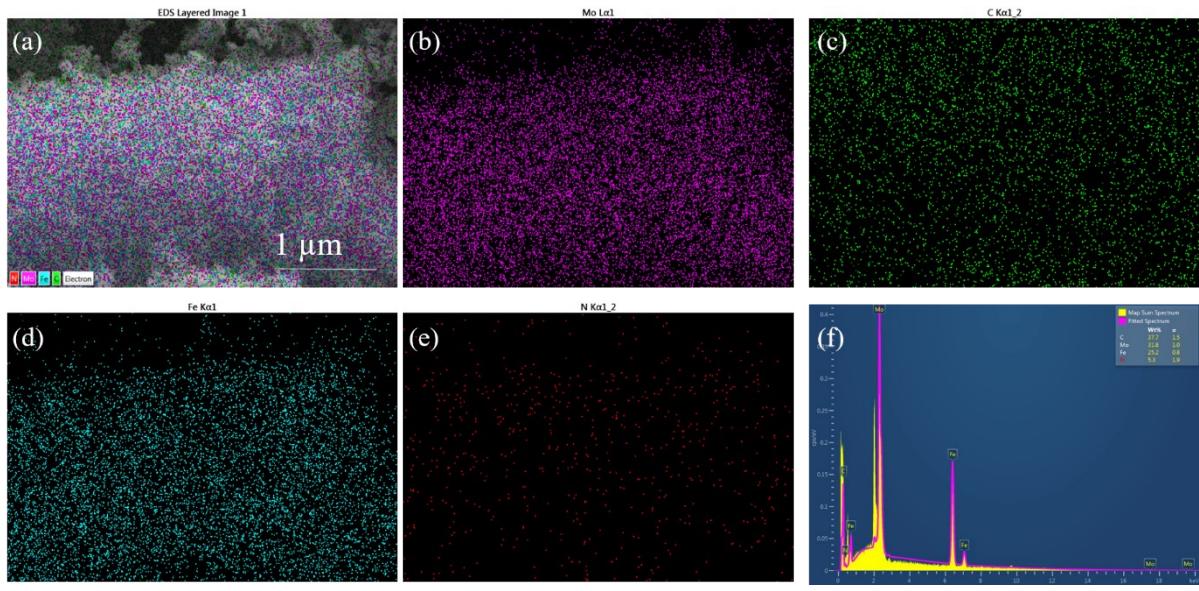
The structural details of synthesized composites were evaluated using X-ray powder diffraction (XRD) measurements on X'pert pro diffractometer, PANalytical using CuK $\alpha$  line ( $\lambda = 1.5406 \text{ \AA}$ ) at 40 kV, 40 mA in the  $2\theta$  range of  $10^\circ - 80^\circ$  with scan rate  $2^\circ/\text{min}$ . Raman spectroscopy (HORIBA, LABRAM HR Evolution) was studied using 633 nm laser excitation for the prepared composites. X-ray photoelectron spectroscopy (XPS) measurements were carried out using Shimadzu ESCA 3400 instrument with AlK $\alpha$  source (Physical Electronics system; 1486.6 eV monochromatic beam) operated at 15 kV, and the XPSPEAK41 software was used for curve fitting and data analysis. A linear-type background was used for data processing. Further, a detailed SEM study was carried out using ‘Quanta 200 FEG FE-SEM’. Transmission electron microscopy (TEM) images were taken on a JEOL Japan, JEM-2100 Plus microscope operated at 100 kV.



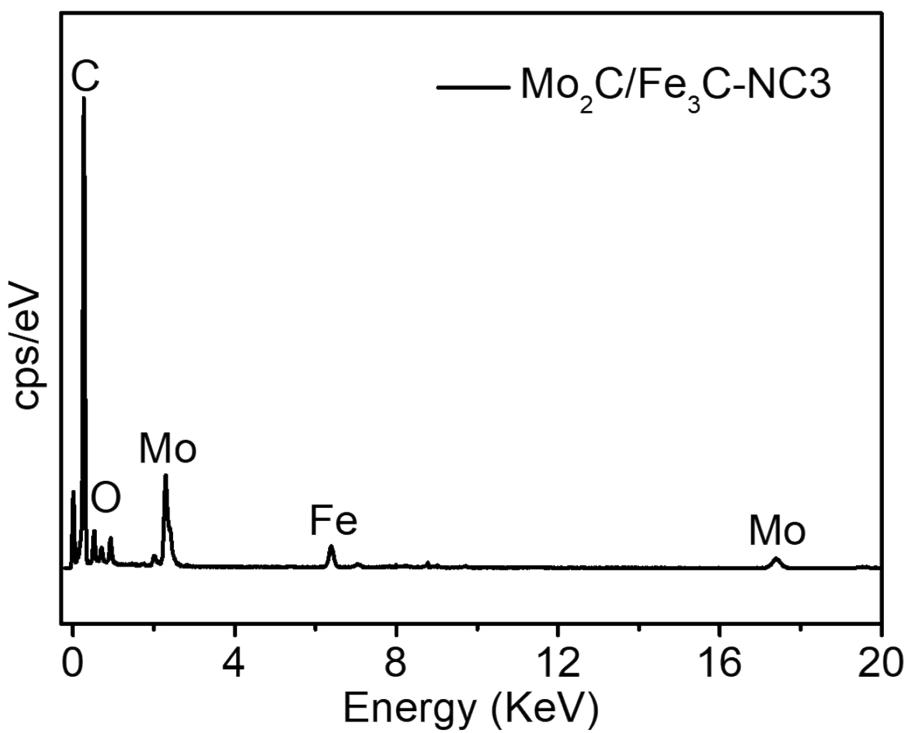
**Figure S1.** SEM images: (a) KBC, and (b) the Mo<sub>2</sub>C/Fe<sub>3</sub>C-NC3 catalyst.



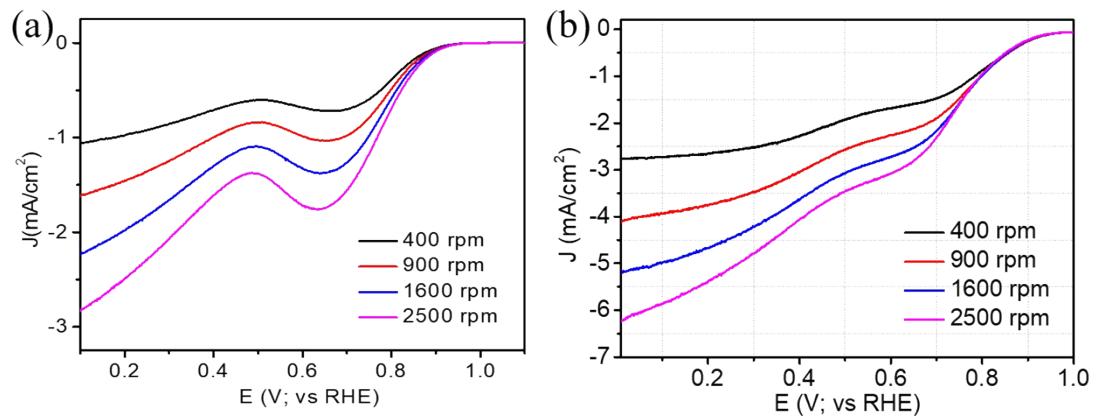
**Figure S2.** SEM images: (a, b) Mo<sub>2</sub>C/Fe<sub>3</sub>C catalyst, (c, d) MoO<sub>2</sub>-C, and (e, f) Fe<sub>2</sub>O<sub>3</sub>-C composites.



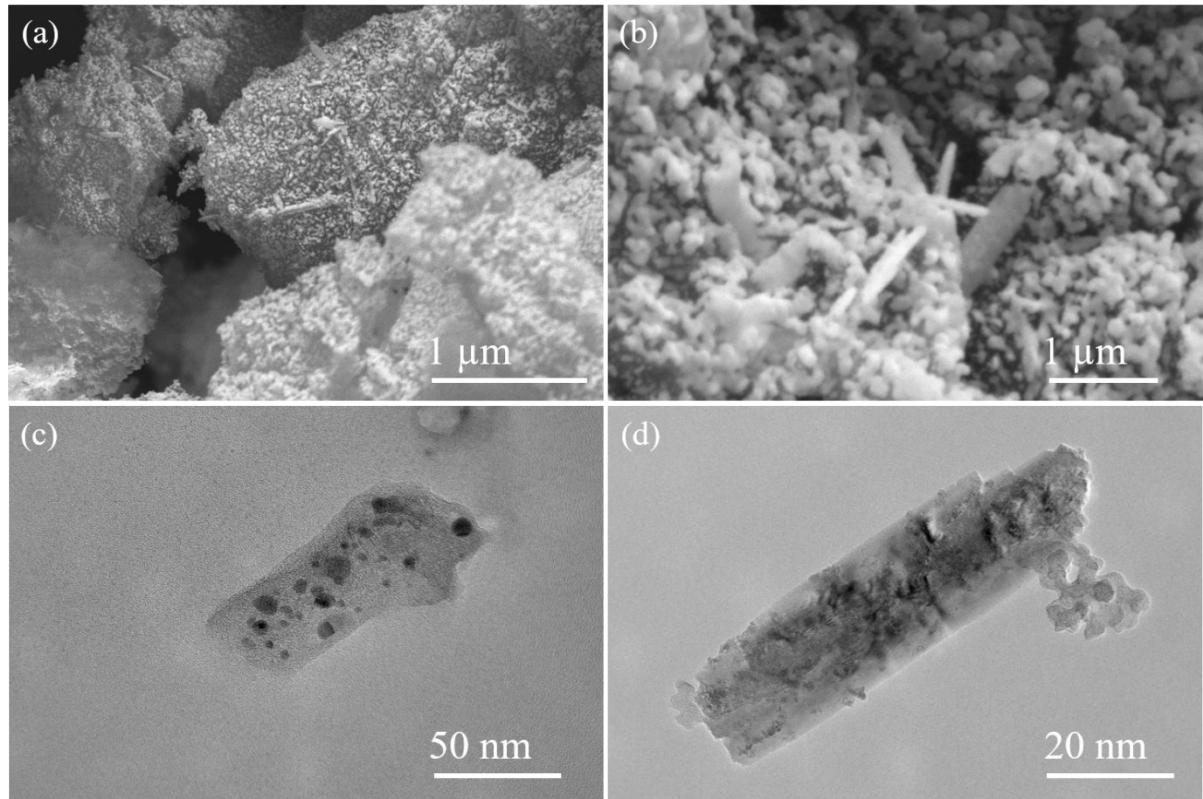
**Figure S3.** HR-SEM element mapping corresponds to (a) EDS layered image, (b) Mo, (c) C, (d) Fe, (e) N, and (f) EDS of  $\text{Mo}_2\text{C}/\text{Fe}_3\text{C}$ -NC3 composite.



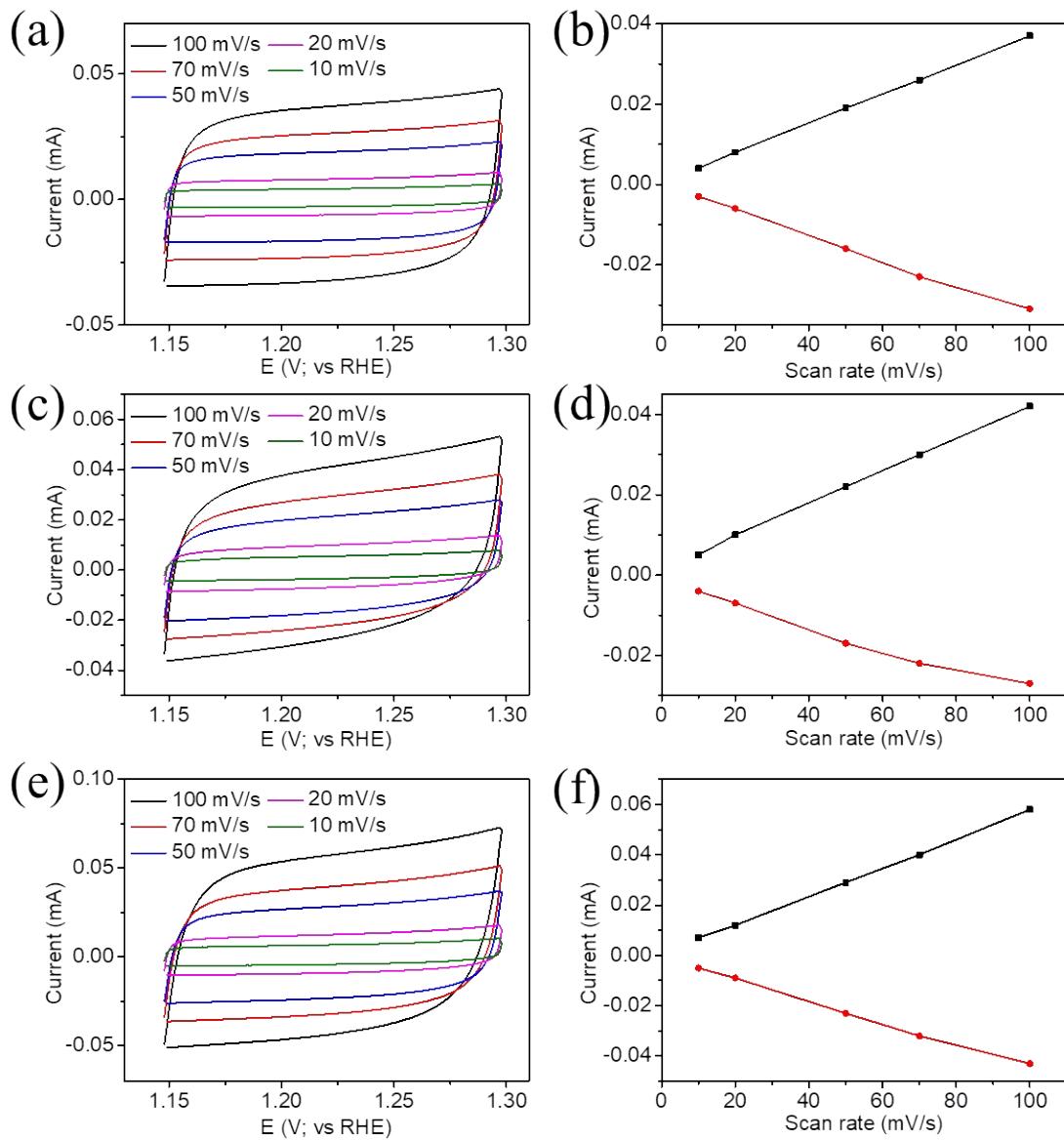
**Figure S4.** HR-TEM EDS spectrum of the prepared  $\text{Mo}_2\text{C}/\text{Fe}_3\text{C}-\text{NC}3$  composite.



**Figure S5.** Polarization curves at different rotation: (a) KBC, and (b)  $\text{Mo}_2\text{C}/\text{Fe}_3\text{C}$  catalysts at scan rate  $10 \text{ mV s}^{-1}$ , in  $\text{O}_2$  saturated  $0.1 \text{ M KOH}$ .



**Figure S6.** (a-b) SEM picture, and (c-d) low resolution TEM pictures for the  $\text{Mo}_2\text{C}/\text{Fe}_3\text{C}$ -NC3 composite after stability analysis.



**Figure S7.** Comparative CVs and ECSA values: (a, b)  $\text{MoO}_2\text{-C}$ , (c, d)  $\text{Fe}_2\text{O}_3\text{-C}$ , and (e, f)  $\text{Mo}_2\text{C}/\text{Fe}_3\text{C}\text{-NC}3$  catalysts.

**Table S1. XPS analysis.**

No.	Name of the element	Core binding energy (eV)	FWHM (eV)
1.	C 1s	283.7	0.9
		284.5	0.9
		285.1	1.6
		286.8	2.0
2.	O 1s	530.5	1.6
		532.0	2.3
		533.7	1.7
3.	Mo 3d	228.6	1.0
		231.9	1.3
		232.6	1.3
		235.7	1.3
4.	N 1s	398.4	1.6
		399.1	1.9
		400.7	2.1
5.	Fe 2p	707.5	1.8
		710.5	2.6
		721	2.9
		725.4	1.26

**Table S2. Comparison of catalytic activity for the prepared catalysts.**

No.	Catalyst	E <sub>onset</sub> (V)	E <sub>1/2</sub> (V)	J <sub>L</sub> (mA/cm <sup>2</sup> )	Mass activity (mA/mg)
1.	MoO <sub>2</sub> -C	0.89	0.76	4.2	38
2.	Fe <sub>2</sub> O <sub>3</sub> -C	0.95	0.8	5.6	83
3.	Mo <sub>2</sub> C/Fe <sub>3</sub> C-NC1	1.00	0.84	6.0	69
4.	Mo <sub>2</sub> C/Fe <sub>3</sub> C-NC2	0.97	0.86	5.9	139
<b>5.</b>	<b>Mo<sub>2</sub>C/Fe<sub>3</sub>C-NC3</b>	<b>1.00</b>	<b>0.89</b>	<b>6.2</b>	<b>221</b>
6.	Mo <sub>2</sub> C/Fe <sub>3</sub> C-NC4	1.00	0.88	5.5	190

**Table S3. Comparison of ORR catalytic performances of Mo<sub>2</sub>C/Fe<sub>3</sub>C-NC3 with those of other reported electro-catalysts in alkaline media**

No.	Catalyst	E <sub>onset</sub> (V)	E <sub>1/2</sub> (V)	J <sub>L</sub> (mA/cm <sup>2</sup> )	Reference
1	Mo <sub>2</sub> C/Fe <sub>3</sub> C-NC3	1.00	0.89	6.2	This work
2	Mo <sub>2</sub> C@NC/Co@NG-900	0.922	0.867	5.5	<sup>1</sup>
3	Mo <sub>2</sub> C-GNR	0.93	0.8	4.6	<sup>2</sup>
4	Mo <sub>2</sub> C/NCNT-30	0.85	0.62	4.22	<sup>3</sup>
5	MoC/NGr-3	0.93	0.80	3.091	<sup>4</sup>
6	Mo <sub>2</sub> C/CXG	0.89	0.71	4.1	<sup>5</sup>
7	Mo <sub>2</sub> C/NPCNFs	0.9	0.77	4.8	<sup>6</sup>
8	Fe-PANI@NP	0.85	0.72	4.5	<sup>7</sup>
9	Fe <sub>3</sub> C/N,S-CNS	0.98	0.86	5.8	<sup>8</sup>
10	Fe-SAs/Fe <sub>3</sub> C-Fe@NC	0.98	0.925	5.6	<sup>9</sup>
11	Fe <sub>3</sub> C@N-CNTs	0.98	0.85	5.0	<sup>10</sup>
12	Fe <sub>3</sub> C-Co-NC	1.02	0.89	4.5	<sup>11</sup>

## References

- Y. Wang, K. Li, F. Yan, C. Li, C. Zhu, X. Zhang and Y. Chen, *Nanoscale*, 2019, **11**, 12563-12572, 10.1039/C9NR02981H.
- X. Fan, Y. Liu, Z. Peng, Z. Zhang, H. Zhou, X. Zhang, B. I. Yakobson, W. A. Goddard, III, X. Guo, R. H. Hauge and J. M. Tour, *ACS Nano*, 2017, **11**, 384-394, 10.1021/acsnano.6b06089.
- Y.-J. Song, J.-T. Ren, G. Yuan, Y. Yao, X. Liu and Z.-Y. Yuan, *Journal of Energy Chemistry*, 2019, **38**, 68-77, 10.1016/j.jec.2019.01.002.
- H. Huang, C. Du, S. Wu and W. Song, *The Journal of Physical Chemistry C*, 2016, **120**, 15707-15713, 10.1021/acs.jpcc.5b10341.
- D. Mladenović, M. Vujković, S. Mentus, D. M. F. Santos, R. P. Rocha, C. A. C. Sequeira, J. L. Figueiredo and B. Šljukić, *Nanomaterials*, 2020, **10**, 1805, 10.3390/nano10091805.

6. H. Wang, C. Sun, Y. Cao, J. Zhu, Y. Chen, J. Guo, J. Zhao, Y. Sun and G. Zou, *Carbon*, 2017, **114**, 628-634, 10.1016/j.carbon.2016.12.081.
7. R. Venegas, C. Zúñiga, J. H. Zagal, A. Toro-Labbé, J. F. Marco, N. Menéndez, K. Muñoz-Becerra and F. J. Recio, *ChemElectroChem*, 2022, **9**, e202200115, 10.1002/celc.202200115.
8. Q.-D. Ruan, R. Feng, J.-J. Feng, Y.-J. Gao, L. Zhang and A.-J. Wang, *Small*, 2023, **19**, 2300136, 10.1002/smll.202300136.
9. X. Sun, P. Wei, S. Gu, J. Zhang, Z. Jiang, J. Wan, Z. Chen, L. Huang, Y. Xu, C. Fang, Q. Li, J. Han and Y. Huang, *Small*, 2020, **16**, 1906057, 10.1002/smll.201906057.
10. L. Cui, Q. Zhang and X. He, *J. Electroanal. Chem.*, 2020, **871**, 114316, 10.1016/j.jelechem.2020.114316.
11. H. Wang, C. Sun, E. Zhu, C. Shi, J. Yu and M. Xu, *J. Alloys Compd.*, 2023, **948**, 169728, 10.1016/j.jallcom.2023.169728.