

Supporting Information

The impact of ligand chain length on HER performance of atomically precise Pt₆(SR)₁₂ nanoclusters

Lipipuspa Sahoo¹, Supriti Dutta², Aarti Devi¹, Rashi¹, Swapna K Pati² and Amitava Patra*^{1,3}

¹Institute of Nano Science and Technology, Knowledge City, Sector 81, Mohali 140306, India.

²Theoretical Sciences Unit, School of Advanced Materials, Jawaharlal Nehru Centre for Advanced Scientific Research (JNCASR), Bangalore 560064, India.

³School of Materials Sciences, Indian Association for the Cultivation of Science, Jadavpur, Kolkata-700032, India

*Author to whom correspondence should be addressed; E-mail: msap@iacs.res.in, Phone: (91)-33-2473-4971, Fax: (91)-33-2473-280

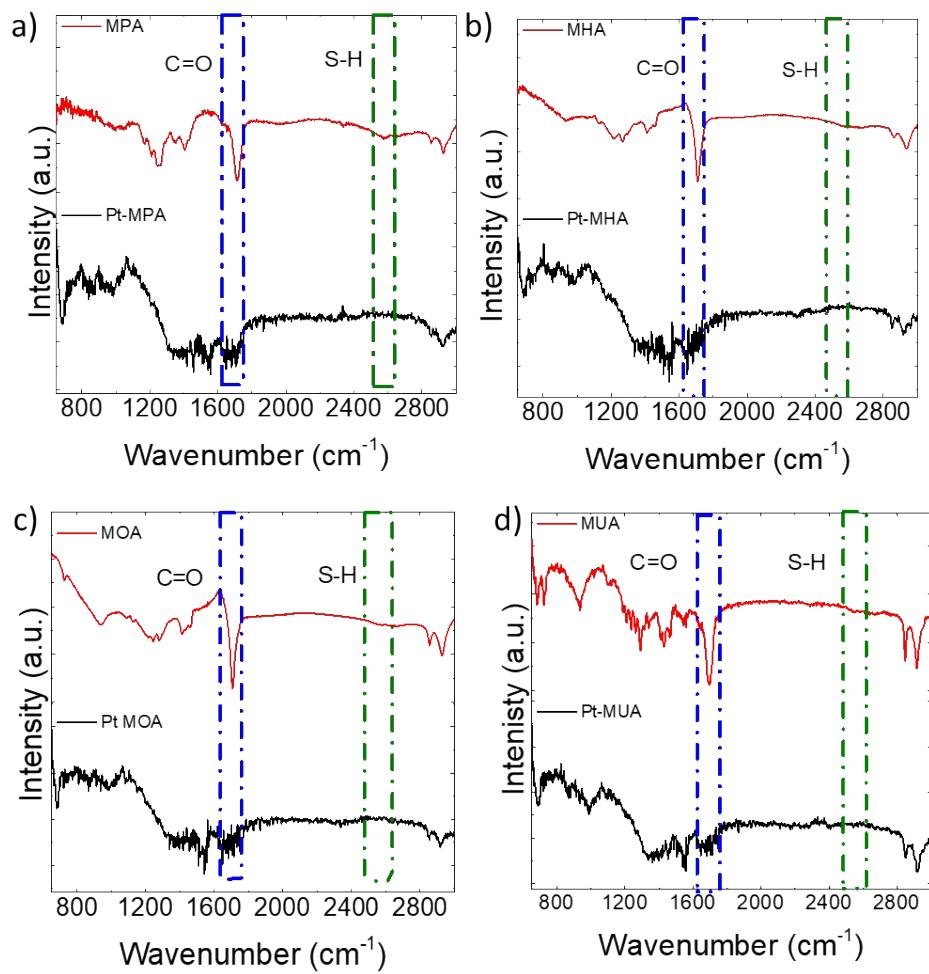


Fig. 1 FT-IR spectra of a) MPA and $\text{Pt}_6(\text{MPA})_{12}$ NCs, (b) MHA and $\text{Pt}_6(\text{MHA})_{12}$ NCs, (c) MOA and $\text{Pt}_6(\text{MOA})_{12}$ NCs, (d) MUA and $\text{Pt}_6(\text{MUA})_{12}$ NCs.

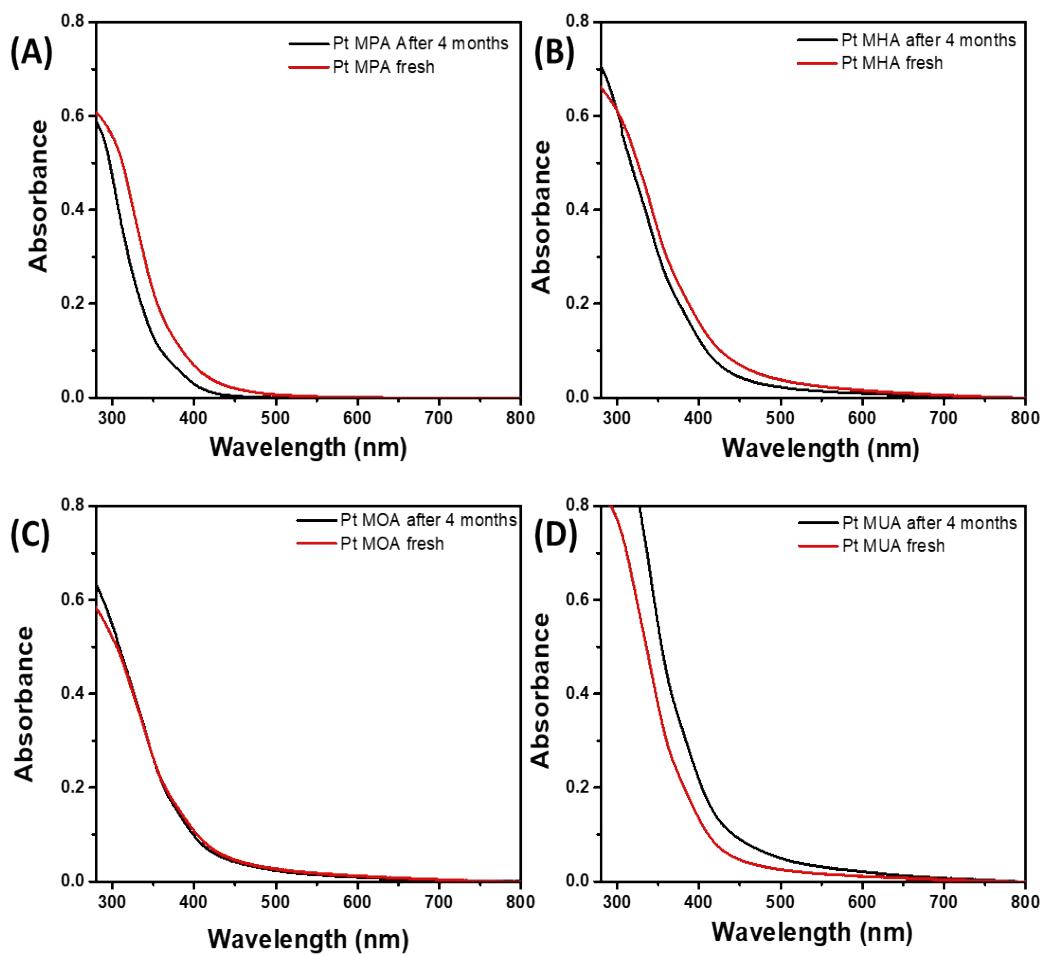


Fig. S2 UV-Vis spectra of as-prepared Pt NCs were recorded after fresh synthesis and after four months.

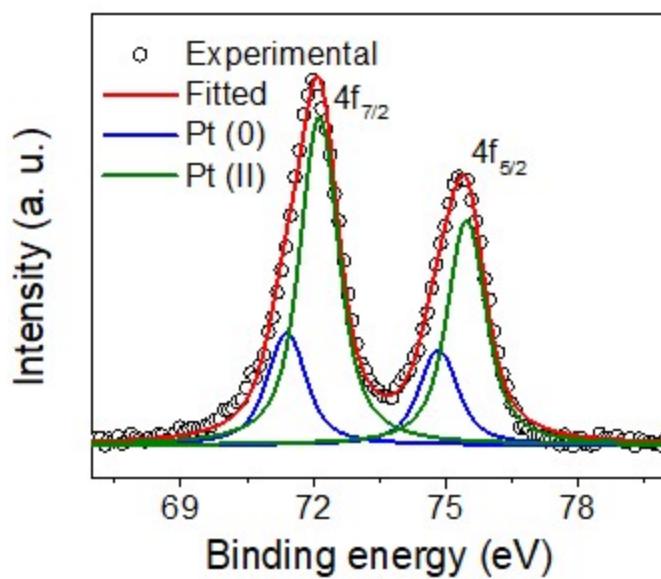


Fig. S3 The deconvolution of Pt 4f XPS spectrum of $\text{Pt}_6(\text{MPA})_{12}$ NCs.

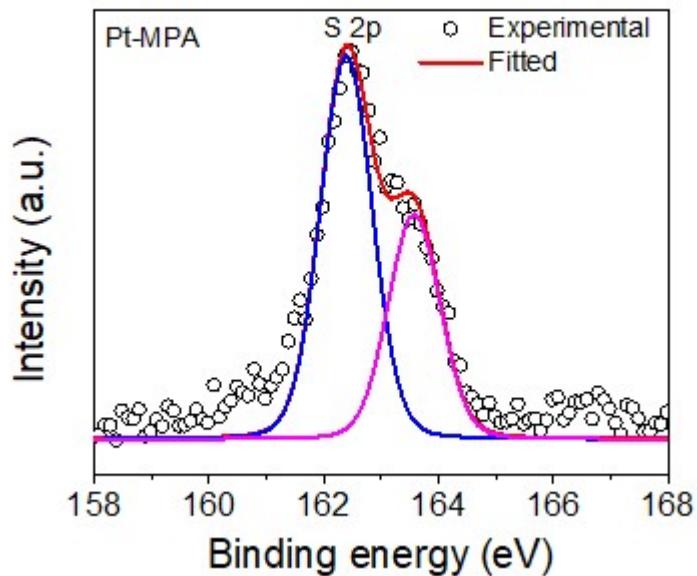


Fig. S4 The deconvolution of S 2p XPS spectrum of $\text{Pt}_6(\text{MPA})_{12}$ NCs.

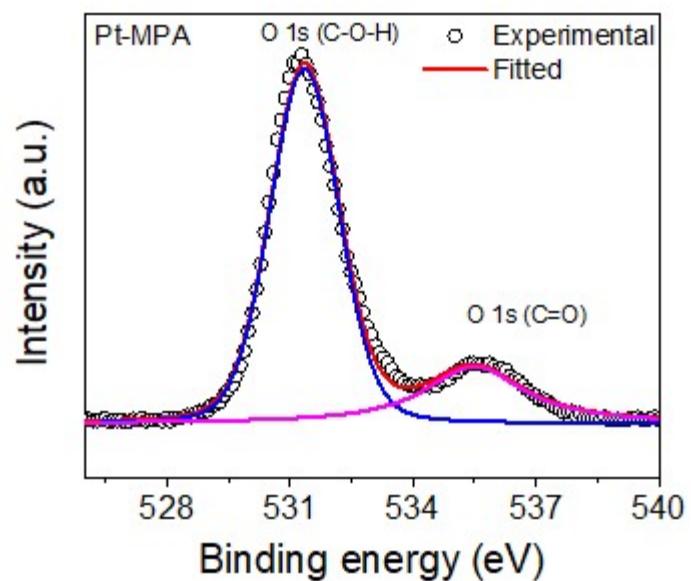


Fig. S5 The deconvolution of O 1s XPS spectrum of $\text{Pt}_6(\text{MPA})_{12}$ NCs.

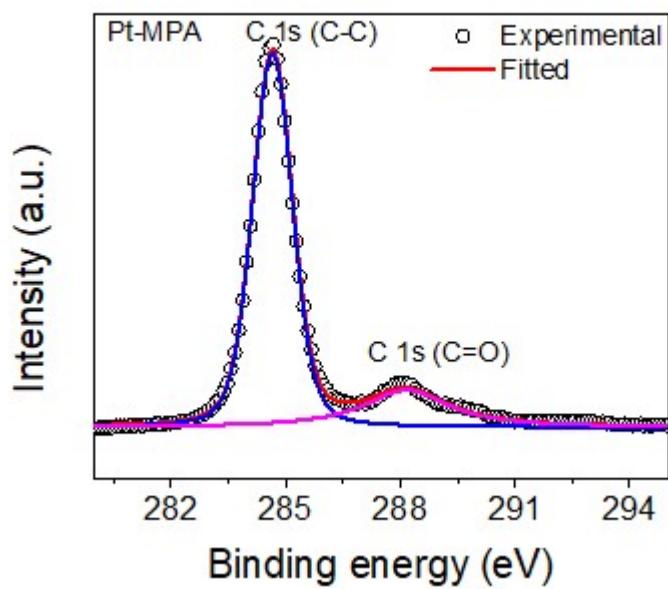


Fig. S6 The deconvolution of C1s XPS spectrum of Pt₆(MPA)₁₂ NCs.

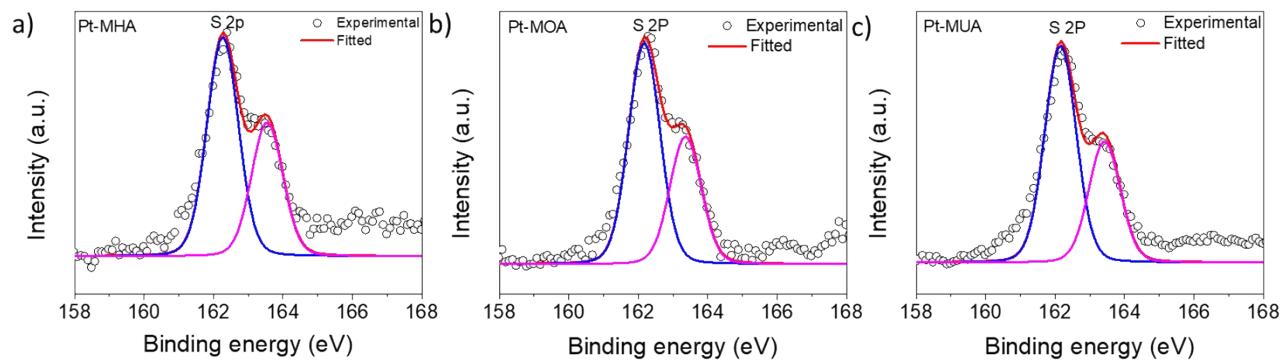


Fig. S7 The deconvolution of S 2p XPS spectrum of (a) Pt₆(MHA)₁₂ NCs, (b) Pt₆(MOA)₁₂ NCs, (c) Pt₆(MUA)₁₂ NCs.

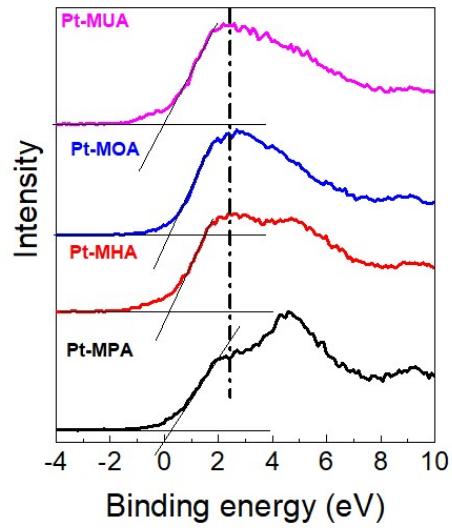


Fig. S8 Valence band spectra of Pt₆(MPA)₁₂ NCs, Pt₆(MHA)₁₂ NCs, Pt₆(MOA)₁₂ NCs, and Pt₆(MUA)₁₂ NCs depicting the method to locate VBM.

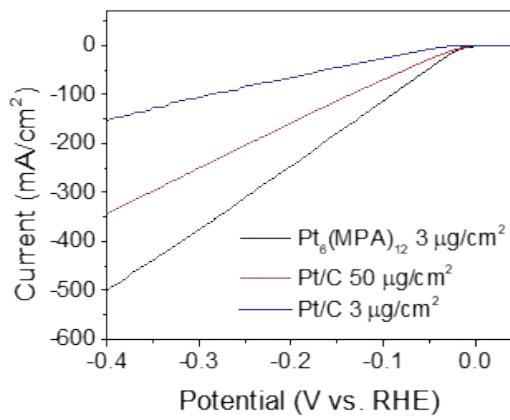


Fig. S9 HER polarization curve of $\text{Pt}_6(\text{MPA})_{12}$ NCs and commercial Pt/C at different mass loading.

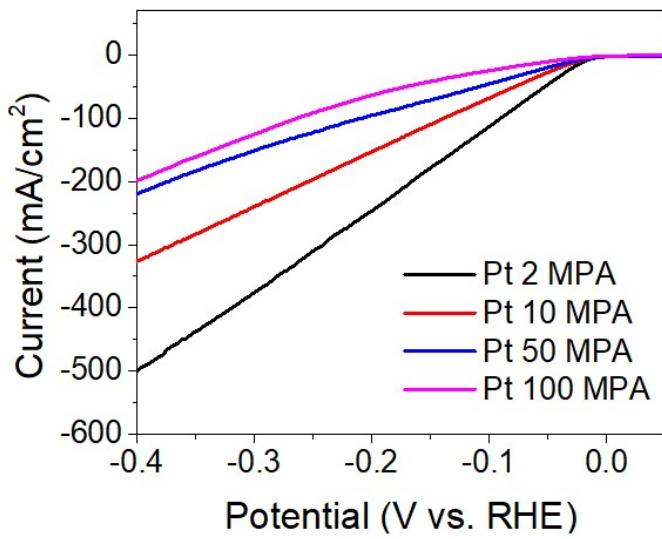


Fig. S10 HER polarization curves of Pt NCs synthesized using 2, 10, 50, and 100 μl of MPA ligand.

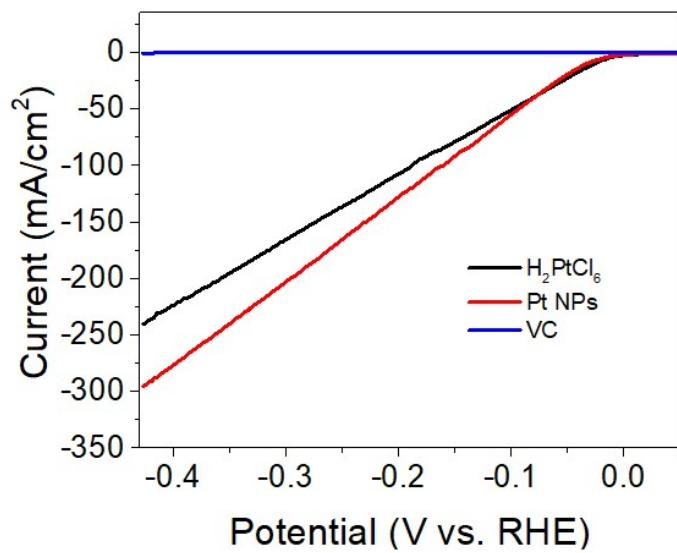


Fig. S11 HER polarization curves of Pt samples synthesized without MPA ligand (Pt NPs) and using only H_2PtCl_6 on Vulcan carbon (Pt content kept same).

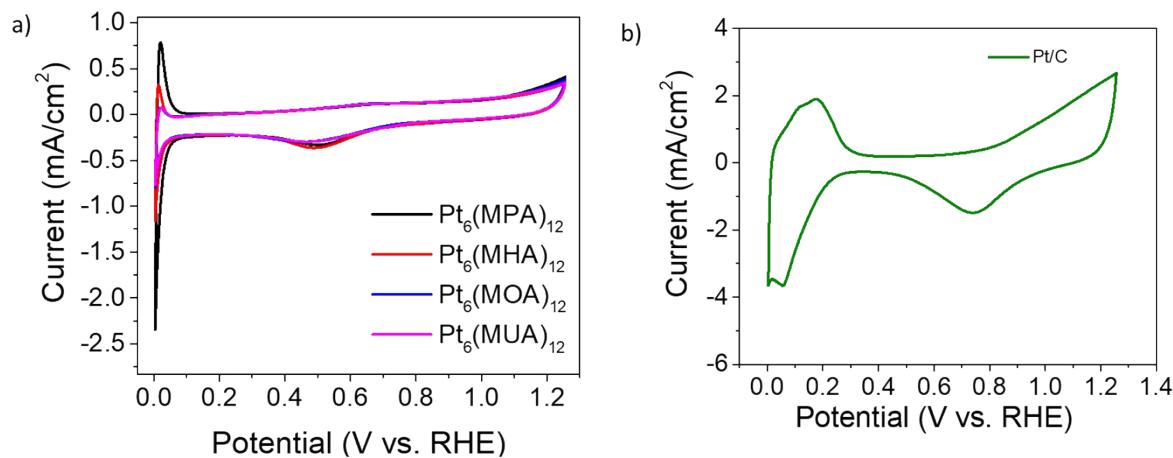


Fig. S12 Cyclic voltammograms of (a) $\text{Pt}_6(\text{SR})_{12}$ NCs (b) Pt/C recorded in Ar saturated 0.5 M H_2SO_4 at a sweep rate of 50 mV/s.

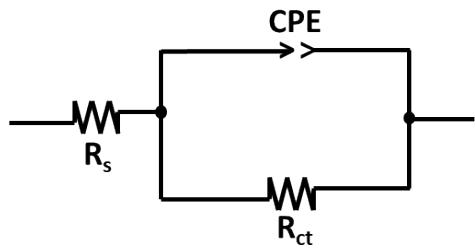


Fig. S13 Standard Randles's equivalent circuit is used to fit the Nyquist plot.

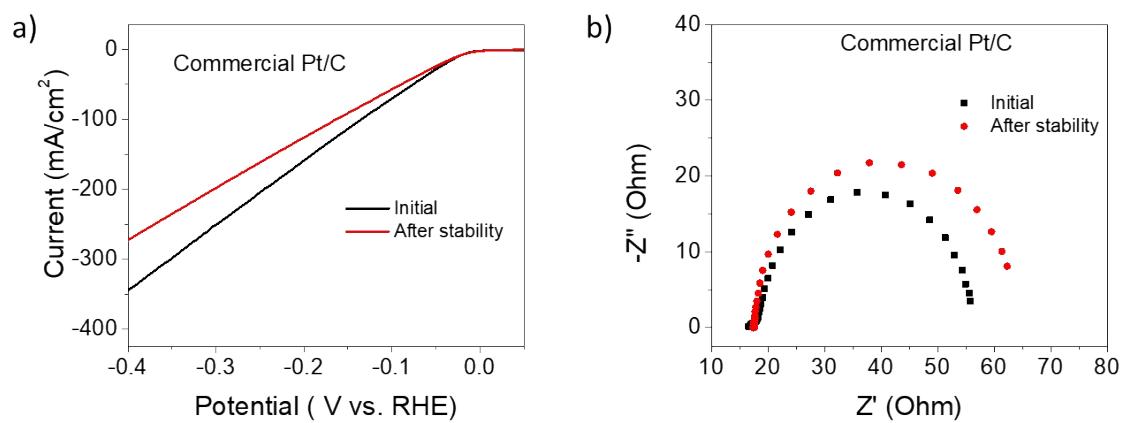


Fig. S14 (a) HER polarization curve of Pt/C was recorded before and after the catalysis, and (b) Nyquist plot of Pt/C was recorded before and after the catalysis.

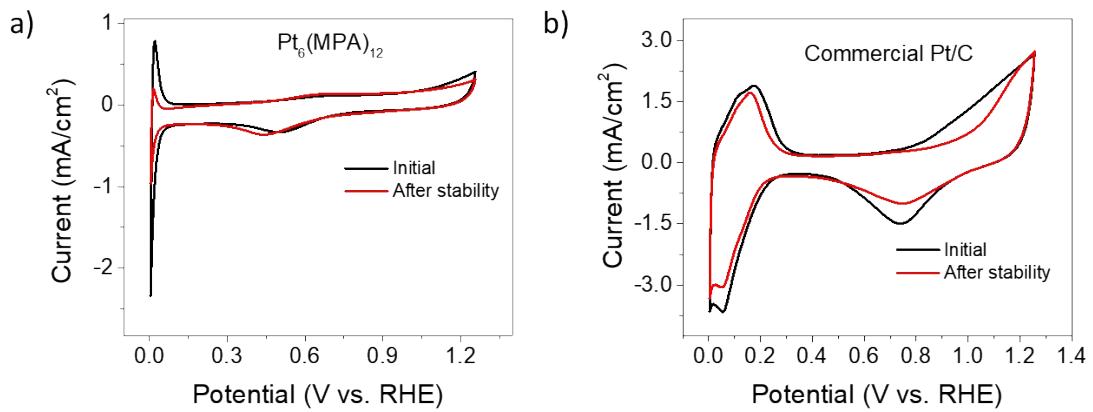


Fig. S15 Cyclic voltammograms of (a) $\text{Pt}_6(\text{MPA})_{12}$ NCs recorded before and after the catalysis, (b) Pt/C recorded before and after the catalysis.

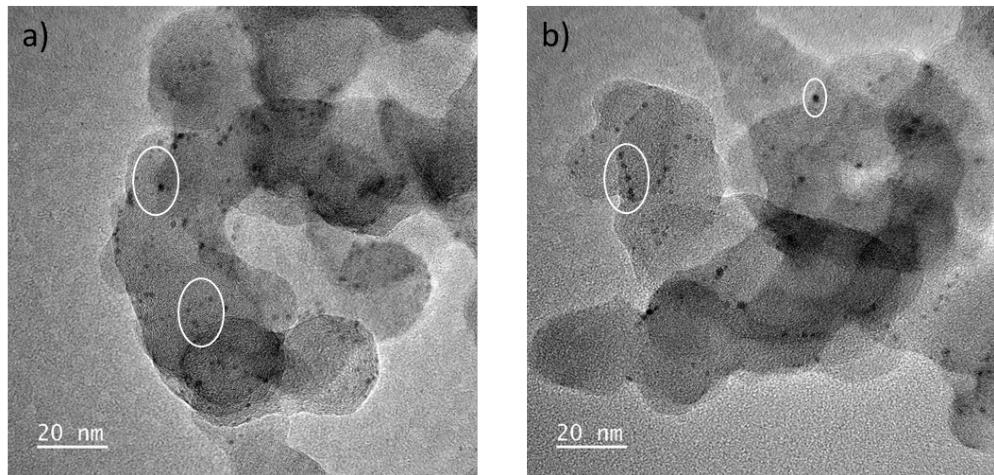


Fig. S16 TEM images of $\text{Pt}_6(\text{MPA})_{12}$ NCs were recorded after the catalysis. The highlighted region shows that the $\text{Pt}_6(\text{MPA})_{12}$ NCs remains strongly intact with the Vulcan Carbon after catalysis.

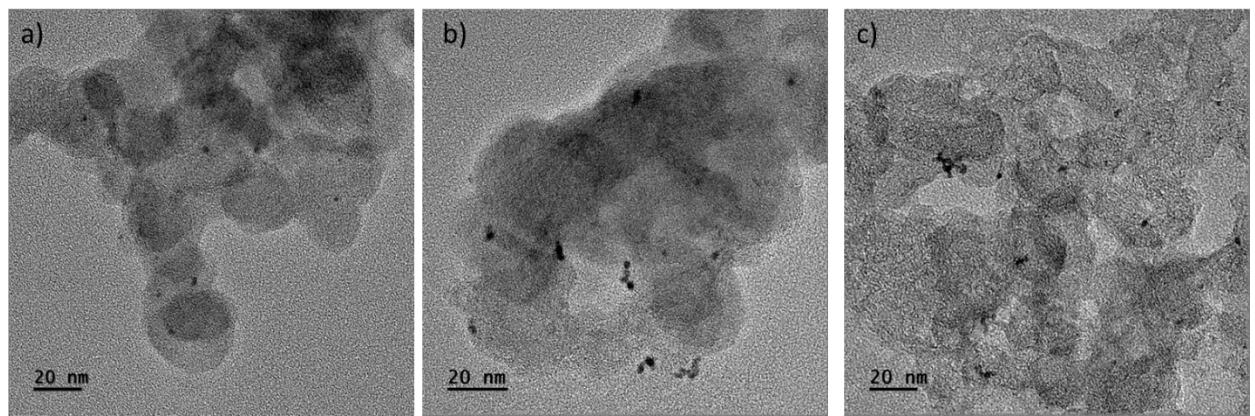


Fig. S17 TEM images of a) Pt₆(MHA)₁₂, b) Pt₆(MOA)₁₂, b) Pt₆(MUA)₁₂ recorded after catalysis.

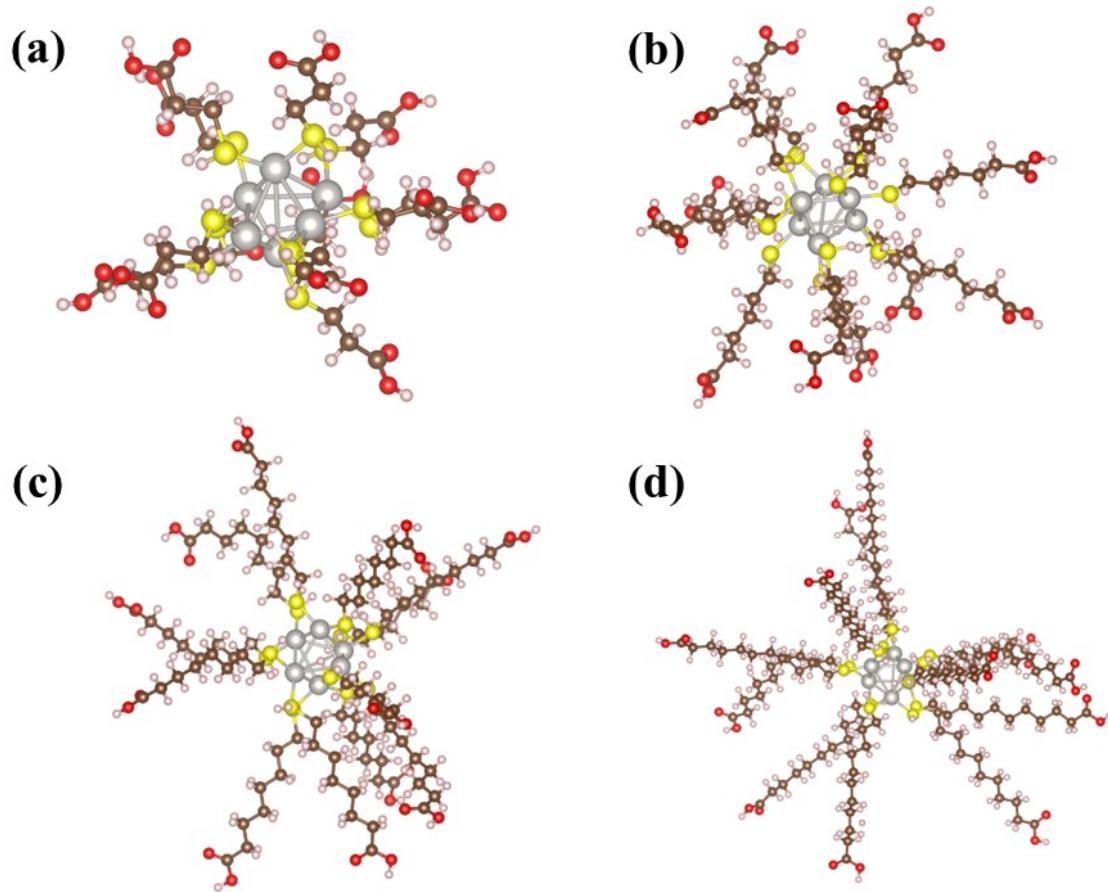


Fig. S18 The optimized geometries of the catalysts (a) Pt₆(MPA)₁₂ NCs, (b) Pt₆(MHA)₁₂ NCs, (c) Pt₆(MOA)₁₂ NCs, (d) Pt₆(MUA)₁₂ NCs. The grey, yellow, brown, red, and white balls represent Pt, S, C, O, and H atoms, respectively.

Table S1. Comparison of electrocatalytic HER performance of the Pt₆(MPA)₁₂ with other Pt-based *state-of-the-art* electrocatalysts developed recently.

Catalyst	Electrolyte	Pt loading amount	Overpotential at 10 mA/cm ² (mV)	Tafel slope (mV/dec)	Reference
Pt ₆ (MPA) ₁₂	0.5 M H ₂ SO ₄	2.55	19	30	This work
Pt ₅ /HMCS	0.5 M H ₂ SO ₄	7.6	20.7	28.3	¹
Pt ₁ /Mesoporous C	0.5 M H ₂ SO ₄	10	26	NA	²
Pt ₁ /OLC	0.5 M H ₂ SO ₄	1.4	38	36	³
Mo ₂ TiC ₂ Tx–PtSA	0.5 M H ₂ SO ₄	12	30	30	⁴
Pt-SAs/WS ₂	0.5 M H ₂ SO ₄	10	32	28	⁵
Pt ₁ /N-C	0.5 M H ₂ SO ₄	6.25	19	14.2	⁶
Pt ₁ O ₁ /Ti _{1-x} O ₂	0.5 M H ₂ SO ₄	1.27	22.2	31	⁷

Pt-Ni ASs	0.5 M H ₂ SO ₄	17	27.7	27	8
Pt ₃ Co@NCNT	0.5 M H ₂ SO ₄	100	42	27.2	9
Pt-SA/ML-WO ₃	0.5 M H ₂ SO ₄	1.1	22	27	10
EG-Pt/ CoP-1.5	0.5 M H ₂ SO ₄	1.53	21	42.5	11
Pt-PVP/ TNR@GC	0.5 M H ₂ SO ₄	21.8	21	27	12
Pt1/NMHCS	0.5 M H ₂ SO ₄	6.55	41	56	13
PtW NPs/C	0.5 M H ₂ SO ₄	20.8	19.4	27.8	14
PtW ₆ O ₂₄ /C	0.5 M H ₂ SO ₄	NA	22	29.8	15
Pt SAs/AG	0.5 M H ₂ SO ₄	31.1	12	29	16
Pt/NGNs	0.5 M H ₂ SO ₄	1.6	38	29	17
Pt ₃ Ni ₂ NWs-S/C	0.5 M H ₂ SO ₄	15.3	28	NA	18

Pt@DNA-GC	0.5 M H ₂ SO ₄	15	26	30	¹⁹
Pt1/MC	0.5 M H ₂ SO ₄	10	25	26	²⁰
Pt1/NPC	0.5 M H ₂ SO ₄	3.8	25	28	²¹
Pt1/NMC	0.5 M H ₂ SO ₄	10	29	26	²²
PtCoFe@CN	0.5 M H ₂ SO ₄	13.11	45	32	²³
400-SWNT/Pt	0.5 M H ₂ SO ₄	19.4	27	38	²⁴
200-SWNT/Pt	0.5 M H ₂ SO ₄	5.6	40	38	²⁴
Pt-MoS ₂	0.5 M H ₂ SO ₄	7.28	38	25	²⁵
Ti-based	0.5 M H ₂ SO ₄	4.5	19	25.9	²⁶
SAL/SAB Pt					
ALD50Pt/NGNs	0.5 M H ₂ SO ₄	1.6	39	29	¹⁷
Pt/NiS@Al ₂ O ₃	0.5 M H ₂ SO ₄	16	34	35	²⁷ .

Pt-Au ₂₅ NCs	0.1 M H ₂ SO ₄	0.8	117	86	28
Pt NCs/rgo	0.5 M H ₂ SO ₄	0.8 wt%	67	46	29

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