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# **Supporting Information for**

# Iodine/Oxone<sup>®</sup> oxidative system for the Synthesis of Selenylindoles Bearing Benzenesulfonamide Moiety as Carbonic Anhydrase I, II, IX, and XII Inhibitors

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# **Table of Contents**

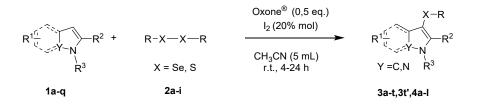
1.	General information and materials	S2
2.	General Procedure	S2
3.	<sup>1</sup> H, <sup>13</sup> C, <sup>77</sup> Se NMR spectra of compounds <b>3</b> and <b>4</b>	S3
4.	HMRS Spectra of new compounds	S26
5.	Carbonic anhydrase inhibition	S34
6.	Human Carbonic Anhydrase activity	S34
7.	References	S35

# 1. General Information and materials

Reagents and solvents were purchased from Sigma Aldrich, Alfa Aesar and VWR International and employed without further purification, except for the starting products indole carboxamides 1f-1g and diselenides 2bg,i that are prepared according to the literature procedures.<sup>1-7</sup> Thin layer chromatography (TLC) was performed in 60 F254 (Merck, KGaA, Darmstadt, Germany) silica gel supported on aluminium sheets. Reaction products were purified by column chromatography on Merck 60 (70-230 mesh) silica gel. Yields correspond to isolated compounds. Purity is estimated to be  $\geq$  95% based on <sup>1</sup>H NMR spectroscopic analysis. NMR experiments were carried out at 25 °C on a Bruker Avance NEO 400 MHz spectrometer equipped with Sample Case operating at 400 MHz for <sup>1</sup>H and 100.62 MHz for <sup>13</sup>C or a Bruker Avance NEO 600 MHz spectrometer equipped with a Prodigy<sup>TM</sup>BBO-Cryoprobe operating at 600.13 MHz for <sup>1</sup>H, 150.90 MHz for <sup>13</sup>C in CDCl<sub>3</sub>, CD<sub>3</sub>OD, DMSO-d<sub>6</sub> and CD<sub>3</sub>COCD<sub>3</sub>. <sup>77</sup>Se NMR spectra are referenced to diphenyl diselenide external standard (PhSe)<sub>2</sub> and were recorded at 114 MHz.<sup>8</sup> Chemical shifts ( $\delta$ ) are reported in ppm. The <sup>1</sup>H NMR spectra registered in CD<sub>3</sub>OD did not show signals for the exchangeable protons of indole NH, SO<sub>2</sub>NH<sub>2</sub>, such protons are visible in DMSO-d<sub>6</sub>. High-resolution mass spectrometry (HRMS) measurements were performed using Synapt G2-Si mass spectrometer (Waters) equipped with an APCI/ ESI source and quadrupole-Time-of-Flight mass analyzer. The mass spectrometer operated in the positive ion detection mode. To ensure accurate mass measurements, data were collected in centroid mode and mass was corrected during acquisition using leucine enkephalin solution as an external reference. The results of the measurements were processed using the MassLynx 4.1 software (Waters) incorporated with the instrument. Melting points were determined in Kofler melting apparatus and value are uncorrected.

# 2. General Procedure

The indole **1a-k**, **m-q** (1.0 eq., 0.5 mmol) or the pyrazole **1l** (1.0 eq. 0.5 mmol), the diselenide **2a-g**, **2i** (0.5 eq., 0.25 mmol) or disulfide **2h** (0.5 eq., 0.25 mmol), Oxone<sup>®</sup> triple salt (0.5 eq., 0.25 mmol) and the iodine (20 mol%) were dissolved in 5 mL of CH<sub>3</sub>CN. The reaction mixture was vigorously stirred at room temperature for 4 to 24 hours. Then, the reaction was quenched with a saturated solution of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and extracted with ethyl acetate (3 x 10 mL), dried with Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated under reduced pressure. The crude mixture was purified by column chromatography on silica gel to afford the 3-selenylindoles **3a-q**, 4-selenylpyrazoles **3r-s**, mono- and bis- sulfenylindoles **3t-3t'**, and 3-selenylindoles containing benzensulfonamide moiety **4a-l**.



3. <sup>1</sup>H NMR, <sup>13</sup>C and <sup>77</sup>Se NMR spectra of compounds 3 and 4

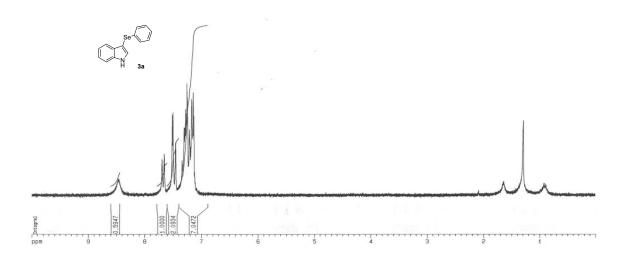


Figure 1. <sup>1</sup>H NMR spectrum of compound 3a<sup>9</sup> in CDCl<sub>3</sub>

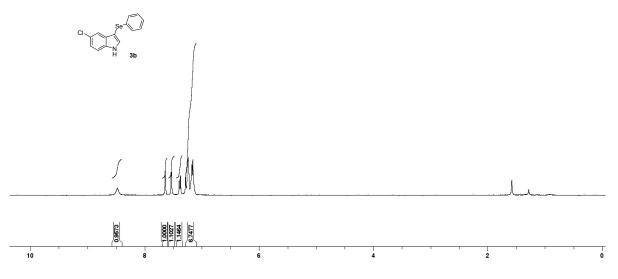


Figure 2. <sup>1</sup>H NMR spectrum of compound **3b**<sup>9</sup> in CDCl<sub>3</sub>

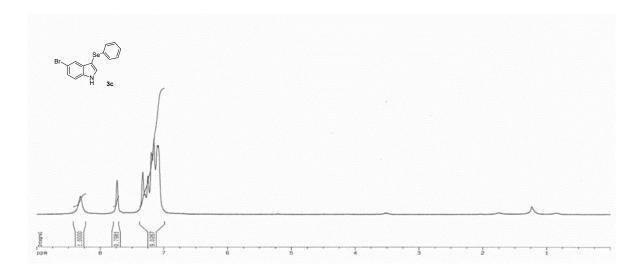


Figure 3. <sup>1</sup>H NMR spectrum of compound 3c<sup>9</sup> in CDCl<sub>3</sub>

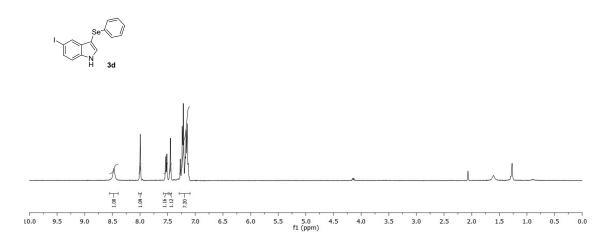


Figure 4. <sup>1</sup>H NMR spectrum of compound 3d<sup>9</sup> in CDCl<sub>3</sub>

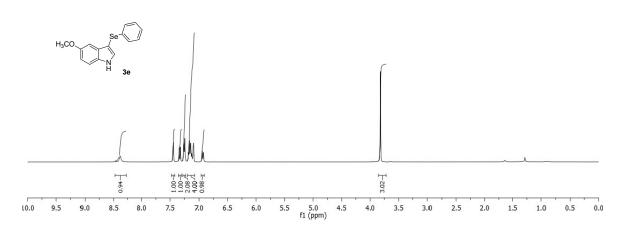


Figure 5. <sup>1</sup>H NMR spectrum of compound **3e**<sup>9</sup> in CDCl<sub>3</sub>

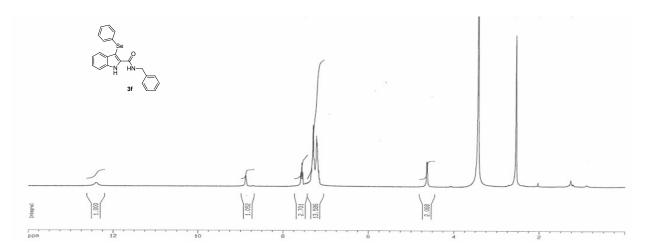


Figure 6. <sup>1</sup>H NMR spectrum of compound 3f in DMSO-d<sub>6</sub>

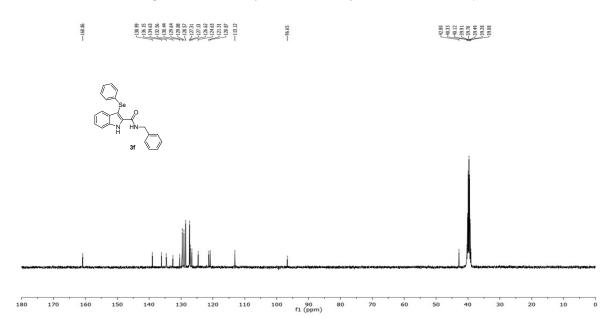


Figure 7. <sup>13</sup>C NMR spectrum of compound **3f** in DMSO-d<sub>6</sub>

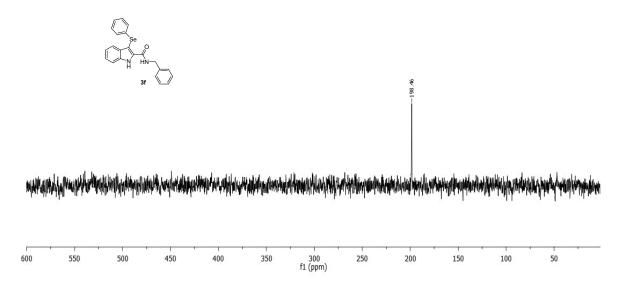


Figure 8. <sup>77</sup>Se NMR spectrum of compound 3f in DMSO-d<sub>6</sub>

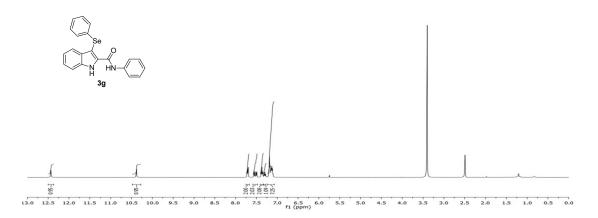


Figure 9. <sup>1</sup>H NMR spectrum of compound 3g in DMSO-d<sub>6</sub>

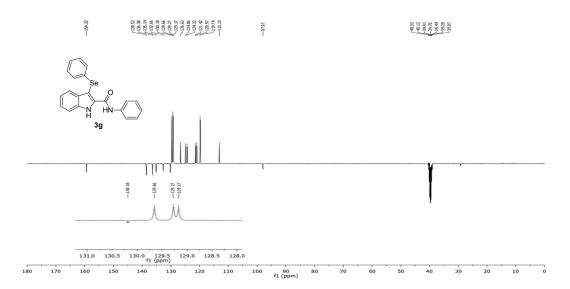


Figure 10. <sup>13</sup>C NMR spectrum of compound 3g in DMSO-d<sub>6</sub>

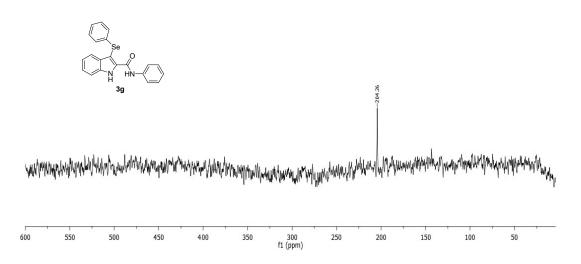


Figure 11. <sup>77</sup>Se NMR spectrum of compound 3g in DMSO-d<sub>6</sub>

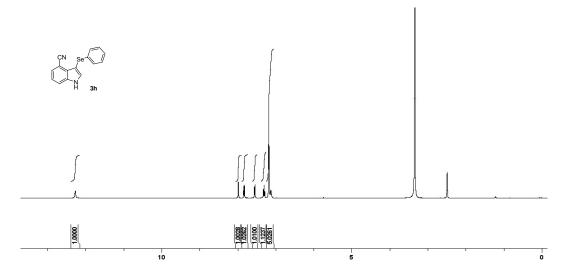


Figure 12. <sup>1</sup>H NMR spectrum of compound **3h**<sup>9</sup> in DMSO-d<sub>6</sub>

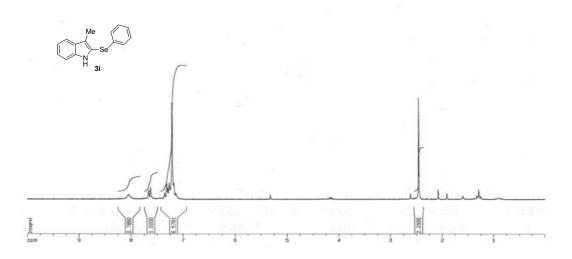


Figure 13. <sup>1</sup>H NMR spectrum of compound 3i<sup>10</sup> in CDCl<sub>3</sub>

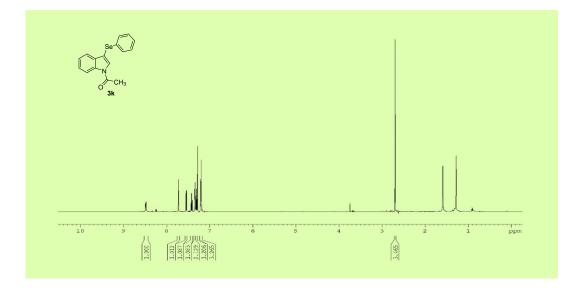


Figure 14. <sup>1</sup>H NMR spectrum of compound 3k in CDCl<sub>3</sub>

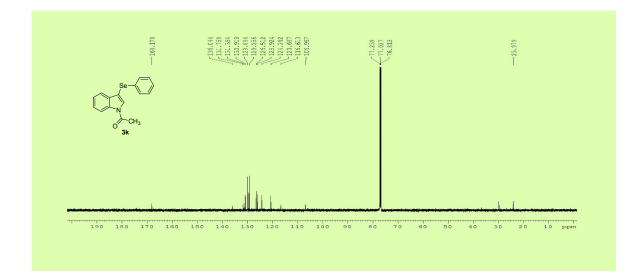
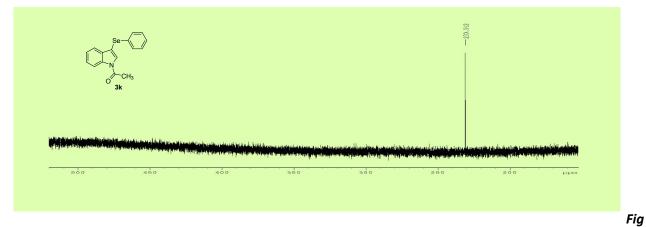


Figure 15. <sup>13</sup>C NMR spectrum of compound **3k** in CDCl<sub>3</sub>



ure 16. <sup>77</sup>Se NMR spectrum of compound 3k in CDCl<sub>3</sub>

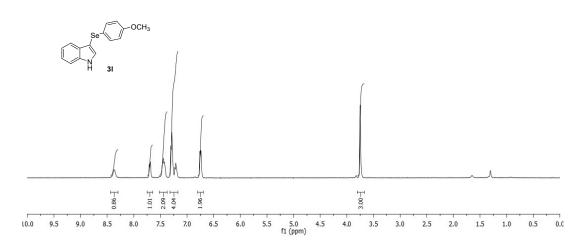


Figure 17. <sup>1</sup>H NMR spectrum of compound 3l<sup>9</sup> in CDCl<sub>3</sub>

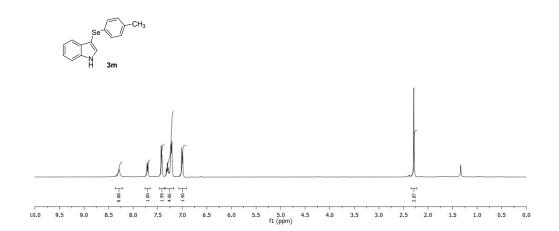


Figure 18. <sup>1</sup>H NMR spectrum of compound 3m<sup>9</sup> in CDCl<sub>3</sub>

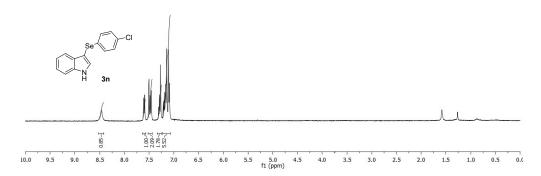


Figure 19. <sup>1</sup>H NMR spectrum of compound 3n<sup>9</sup> in CDCl<sub>3</sub>

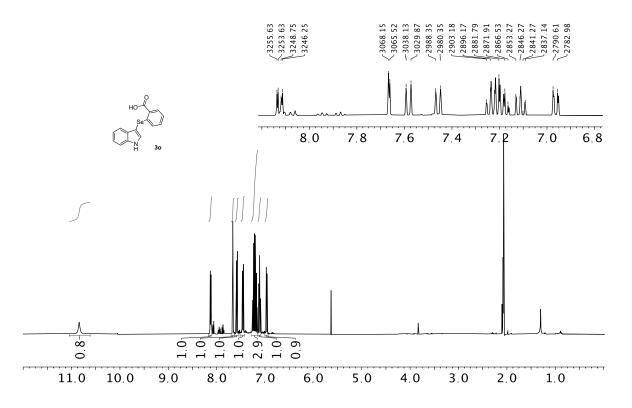
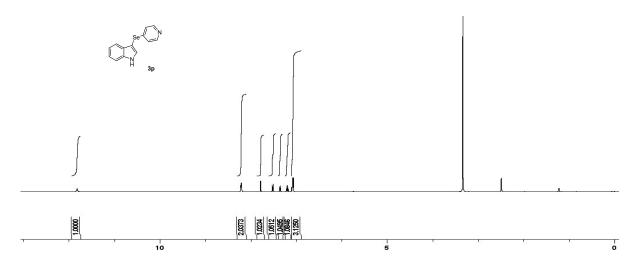
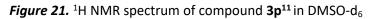
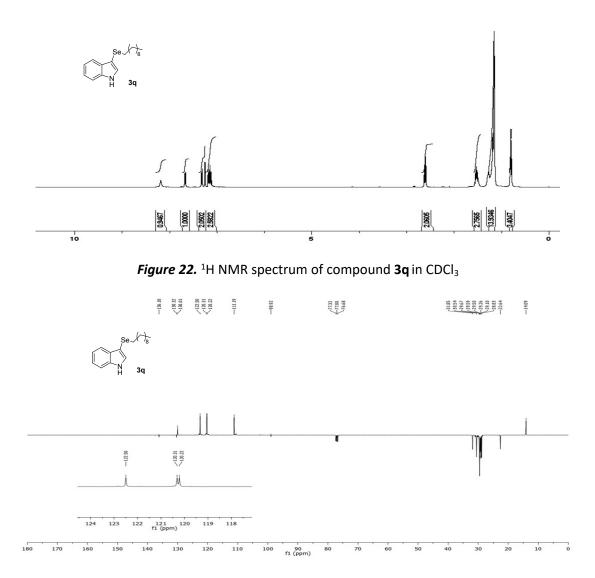
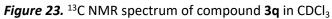


Figure 20. <sup>1</sup>H NMR spectrum of compound 30<sup>9</sup> in CD<sub>3</sub>COCD<sub>3</sub>









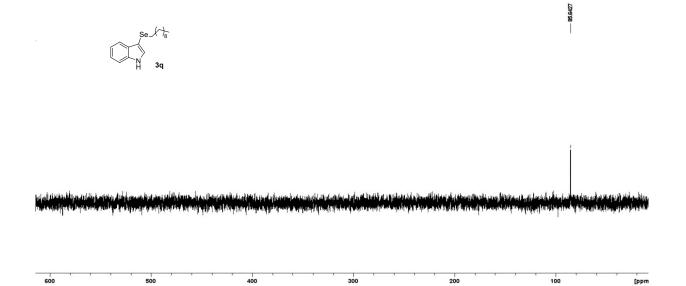


Figure 24. 77Se NMR spectrum of compound 3q in CDCl<sub>3</sub>

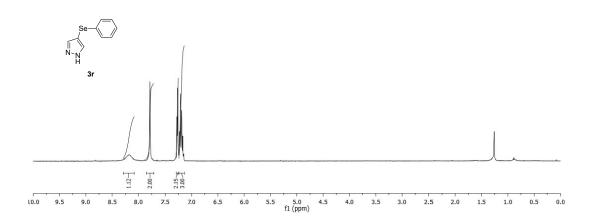


Figure 25. <sup>1</sup>H NMR spectrum of compound  $3r^{12}$  in CDCl<sub>3</sub>

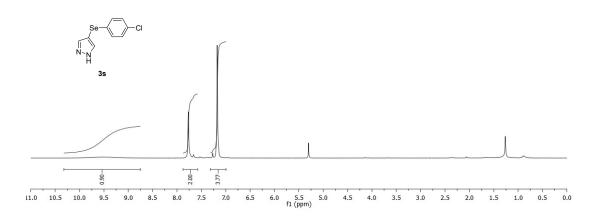
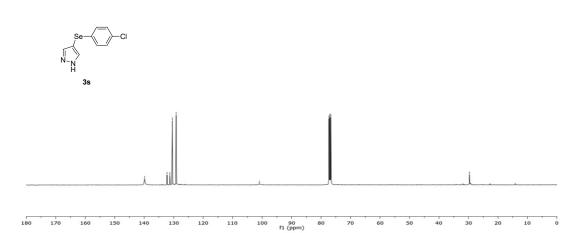


Figure 26. <sup>1</sup>H NMR spectrum of compound 3s in CDCl<sub>3</sub>



*Figure 27.* <sup>13</sup>C NMR spectrum of compound **3s** in CDCl<sub>3</sub>

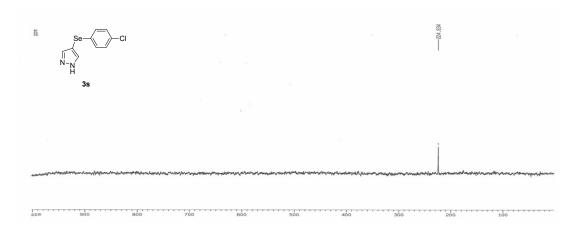
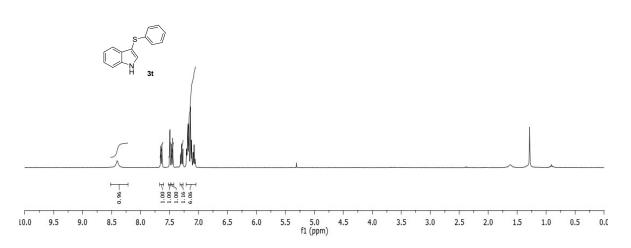


Figure 28. <sup>77</sup>Se NMR spectrum of compound 3s in CDCl<sub>3</sub>





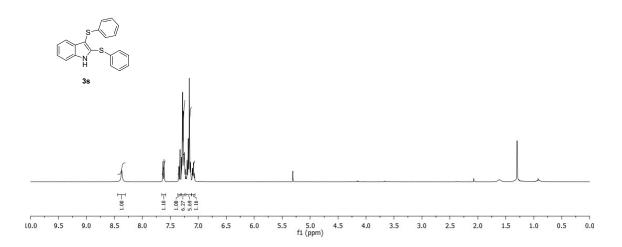


Figure 30. <sup>1</sup>H NMR spectrum of compound 3t'<sup>14</sup> in CDCl<sub>3</sub>

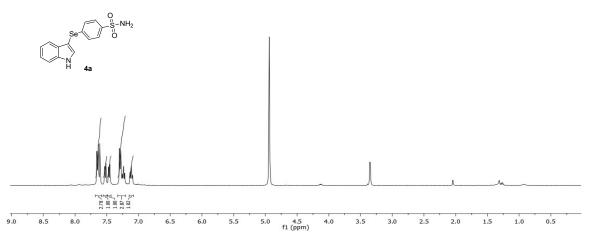


Figure 31. <sup>1</sup>H NMR spectrum of compound 4a in CD<sub>3</sub>OD

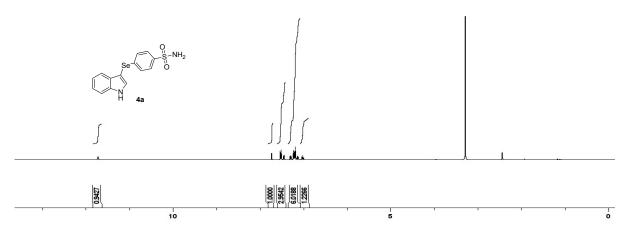
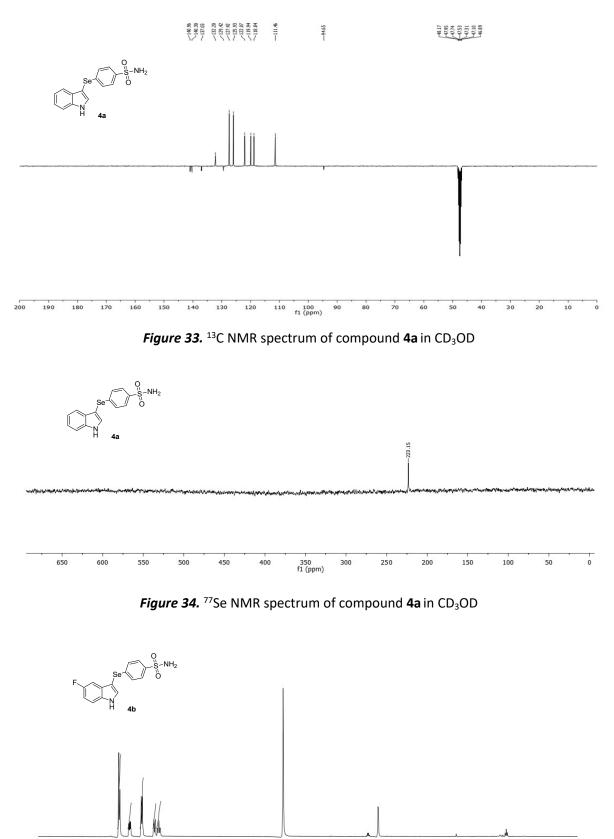


Figure 32. <sup>1</sup>H NMR spectrum of compound 4a in DMSO-d<sub>6</sub>



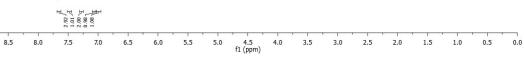


Figure 35. <sup>1</sup>H NMR spectrum of compound 4b in CD<sub>3</sub>OD

9.0

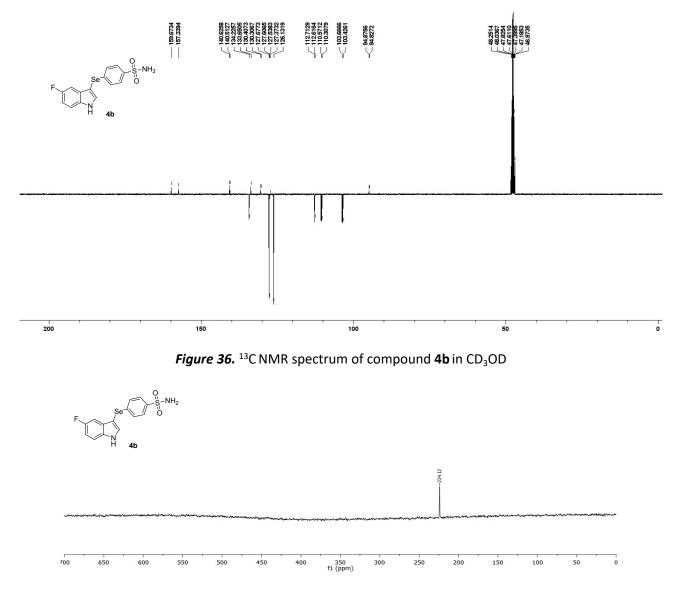


Figure 37. <sup>77</sup>Se NMR spectrum of compound 4b in CD<sub>3</sub>OD

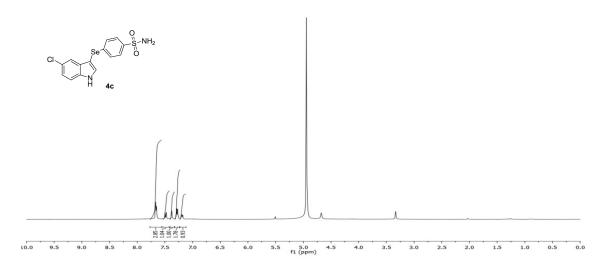
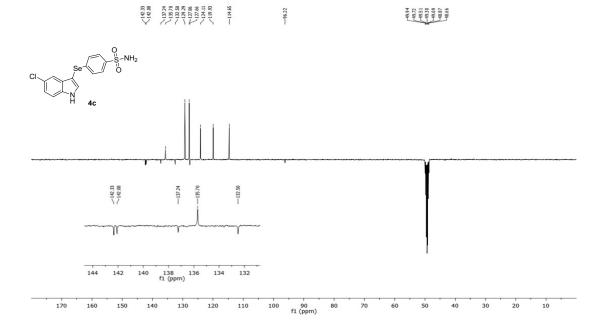


Figure 38. <sup>1</sup>H NMR spectrum of compound 4c in CD<sub>3</sub>OD





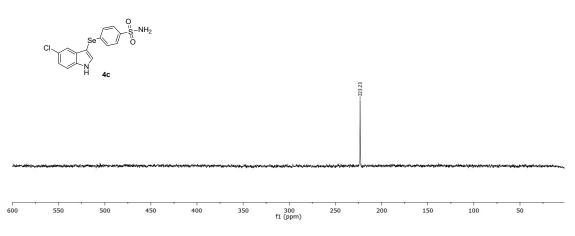
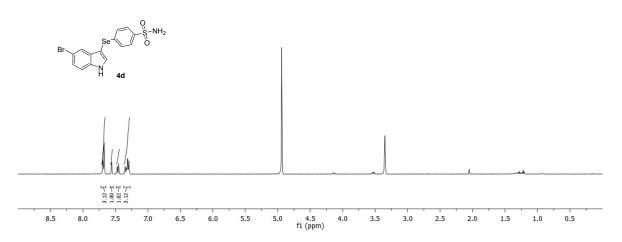
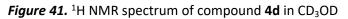


Figure 40. <sup>77</sup>Se NMR spectrum of compound 4c in CD<sub>3</sub>OD





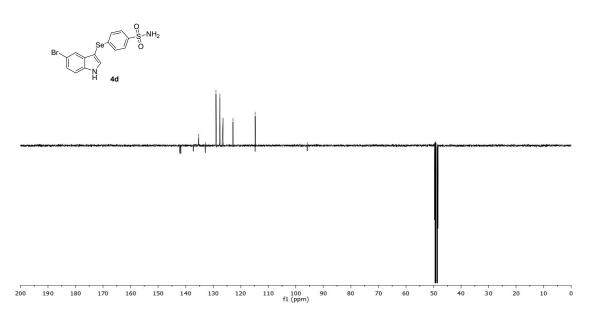


Figure 42. <sup>13</sup>C NMR spectrum of compound 4d in CD<sub>3</sub>OD

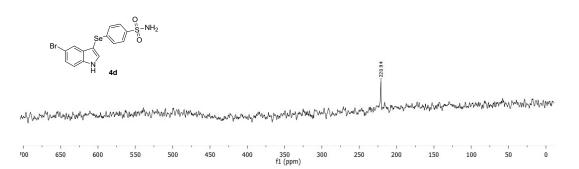


Figure 43. <sup>77</sup>Se NMR spectrum of compound 4d in CD<sub>3</sub>OD

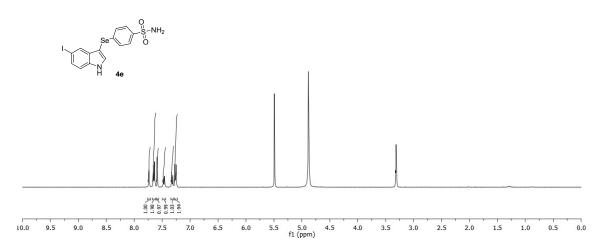
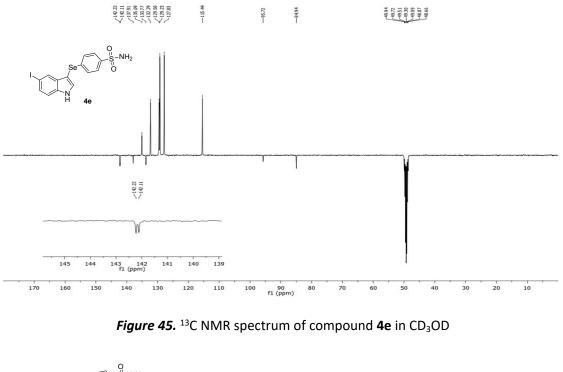
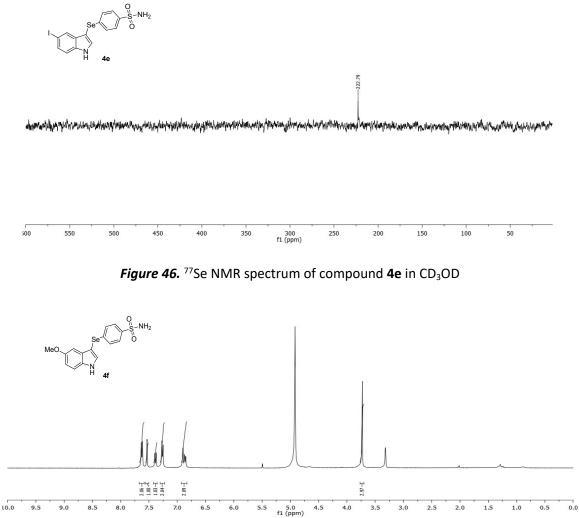
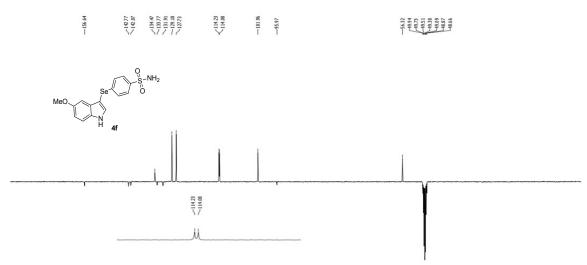


Figure 44. <sup>1</sup>H NMR spectrum of compound 4e in CD<sub>3</sub>OD

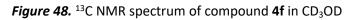








114 113 f1 (ppm) 112 111 110 



f1 (ppm)

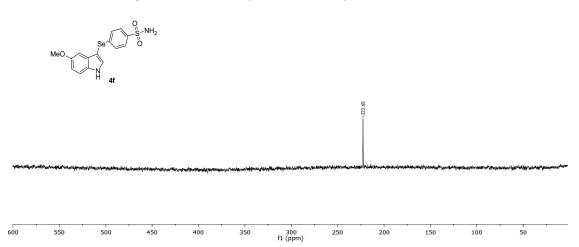
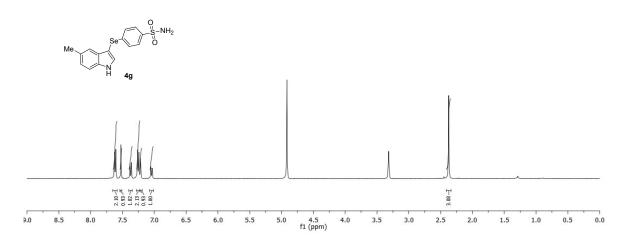
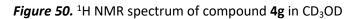
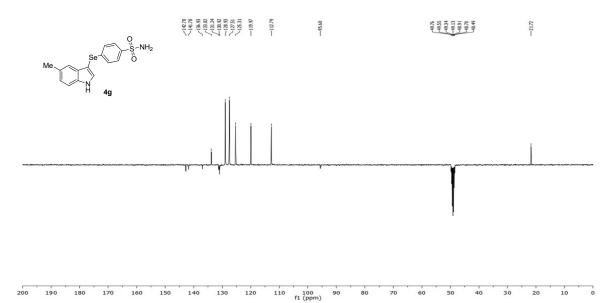
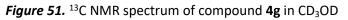


Figure 49. <sup>77</sup>Se NMR spectrum of compound 4f in CD<sub>3</sub>OD









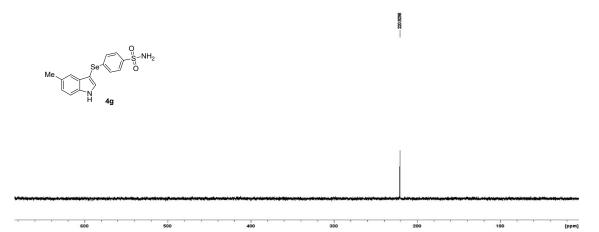


Figure 52. <sup>77</sup>Se NMR spectrum of compound 4g in CD<sub>3</sub>OD

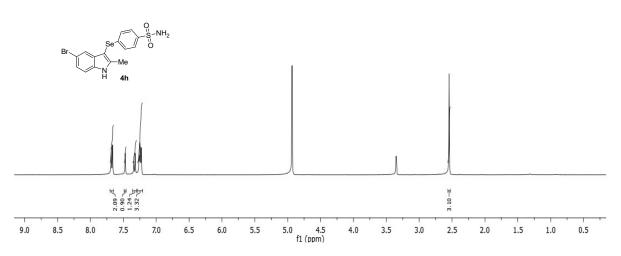


Figure 53. <sup>1</sup>H NMR spectrum of compound 4h in CD<sub>3</sub>OD

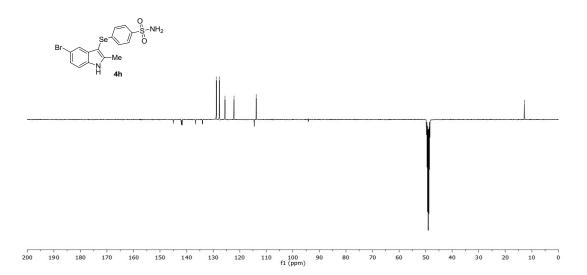


Figure 54. <sup>13</sup>C NMR spectrum of compound 4h in CD<sub>3</sub>OD

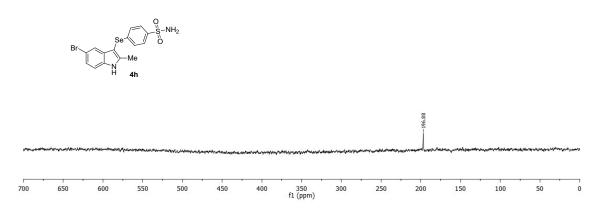


Figure 55. <sup>77</sup>Se NMR spectrum of compound 4h in CD<sub>3</sub>OD

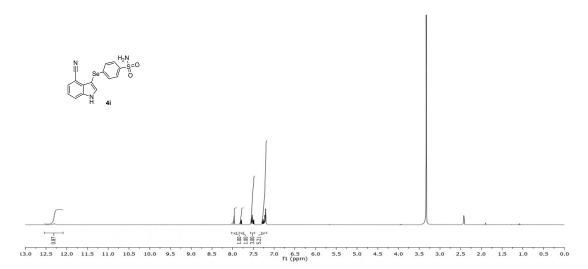
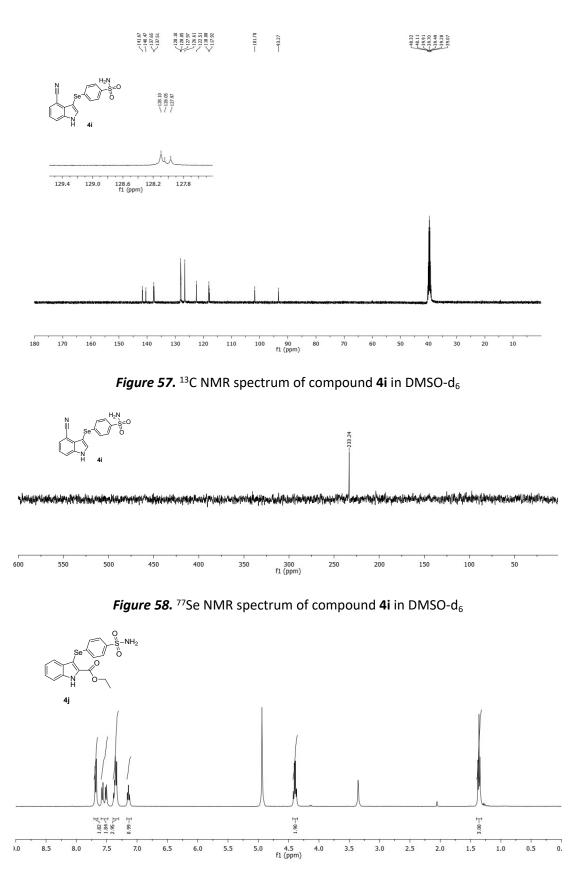
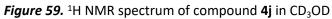
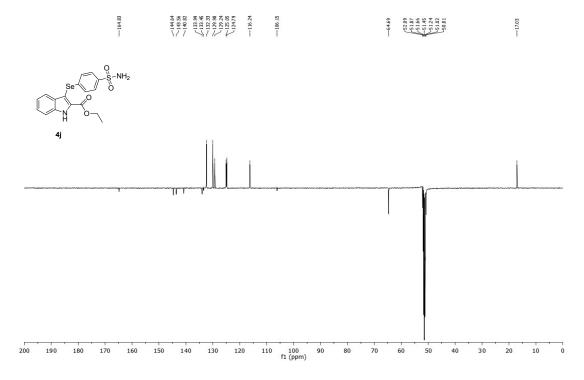
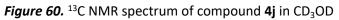


Figure 56. <sup>1</sup>H NMR spectrum of compound 4i in DMSO-d<sub>6</sub>









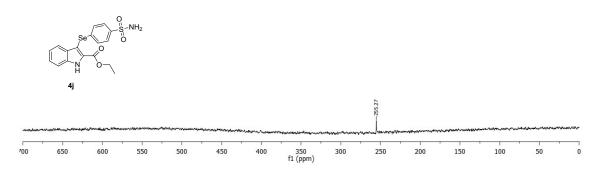


Figure 61. <sup>77</sup>Se NMR spectrum of compound 4j in CD<sub>3</sub>OD

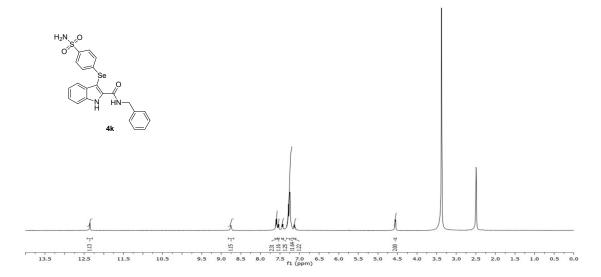


Figure 62. <sup>1</sup>H NMR spectrum of compound 4k in DMSO-d<sub>6</sub>

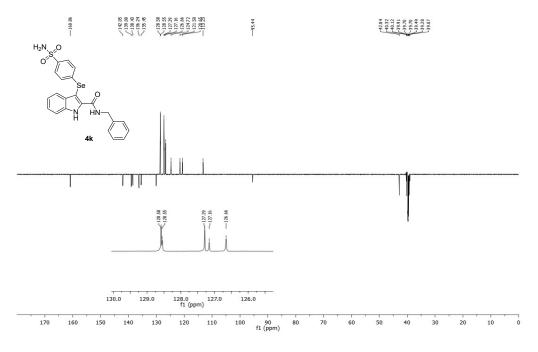


Figure 63. <sup>13</sup>C NMR spectrum of compound 4k in DMSO-d<sub>6</sub>

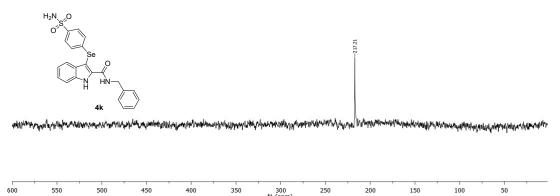


Figure 64. <sup>77</sup>Se NMR spectrum of compound 4k in DMSO-d<sub>6</sub>

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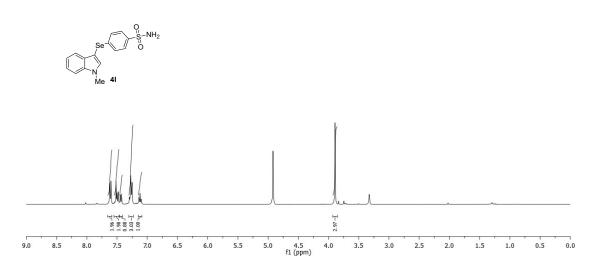


Figure 65. <sup>1</sup>H NMR spectrum of compound 4I in CD<sub>3</sub>OD

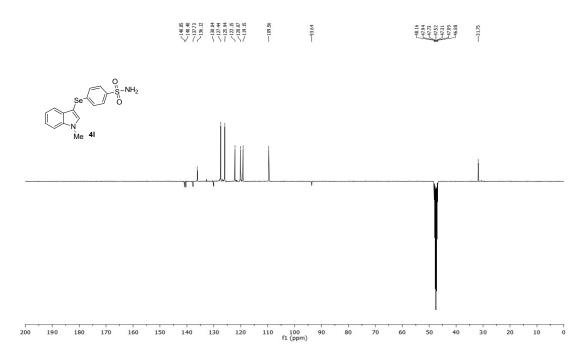


Figure 66. <sup>13</sup>C NMR spectrum of compound 4I in CD<sub>3</sub>OD

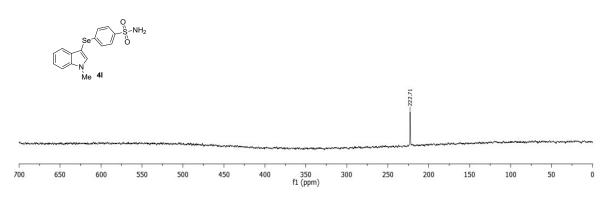
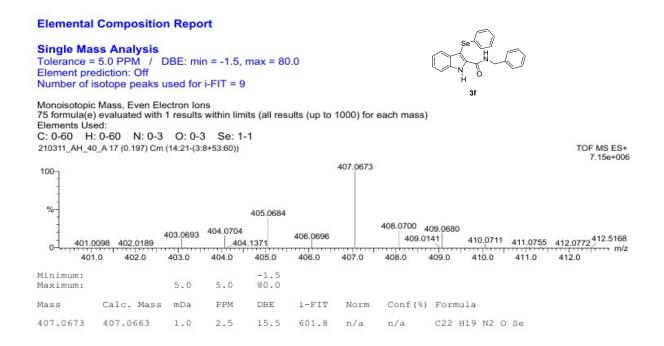
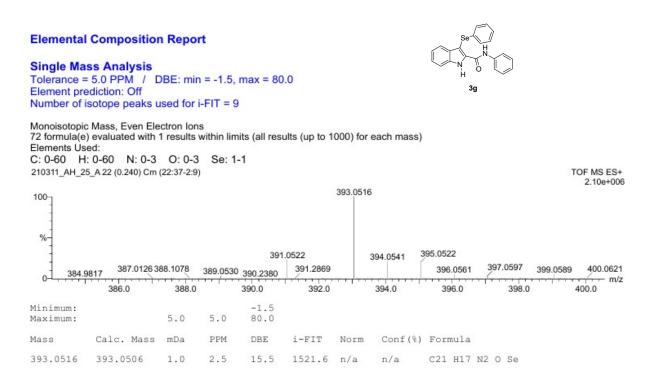


Figure 67. <sup>77</sup>Se NMR spectrum of compound 4I in CD<sub>3</sub>OD

## 4. HMRS Spectra of new compounds



# Figure 68. HRMS spectrum of compound 3f



gure 69. HRMS spectrum of compound 3g

Fi

#### **Elemental Composition Report** Single Mass Analysis Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 3q Element prediction: Off Number of isotope peaks used for i-FIT = 9 Monoisotopic Mass, Even Electron Ions 16 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-3 Se: 1-1 210311 AH 22A 16 (0.177) Cm (13:25-(3:8+52:60)) TOF MS ES+ 6.39e+006 338.1397 100 336.1413 % 334.1406 335,1396 337,1363 340.1397 339.1414 329.2156 329.8672 332.1443 342,1476 344.1199 m/z 333.1346 341.1429 0 330.0 332.0 344.0 328.0 334.0 336.0 338.0 340.0 342.0 -1.5 Minimum: 5.0 5.0 80.0 Maximum: Mass Calc. Mass mDa PPM DBE i-FIT Norm Conf(%) Formula



n/a

n/a

C18 H28 N Se

688.5

338.1397

338.1387

1.0

3.0

6.5

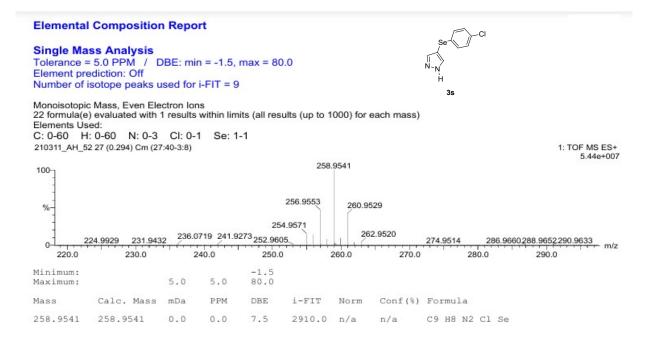


Figure 71. HRMS spectrum of compound 3s

Single Mass Analysis Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9 Monoisotopic Mass, Even Electron lons 155 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used:									
	H: 0-60 N: 0-2 02 A 19 (0.214) Cm								TOF MS ES+
100-1				35	52.9868				6.46e+005
%	336.9450 338.3417		348	350.9878		9870	58,9954 360.3238		
0-1			347.9424	350.0				364.9503 365.0	369.9681 m/z 370.0
Minimum: Maximum:		5.0 5.0	-1.5	000.0	55		555.5	000.0	0.0.0
Mass	Calc. Mass	mDa PPM	I DBE	i-FIT	Norm	Conf(%)	Formula		
352.9868	352.9863 352.9869	0.5 1.4 -0.1 -0.		606.5 609.2	0.067 2.736	93.52 6.48	C14 H13 N2 C22 H9 Se	O2 S Se	



## **Elemental Composition Report**

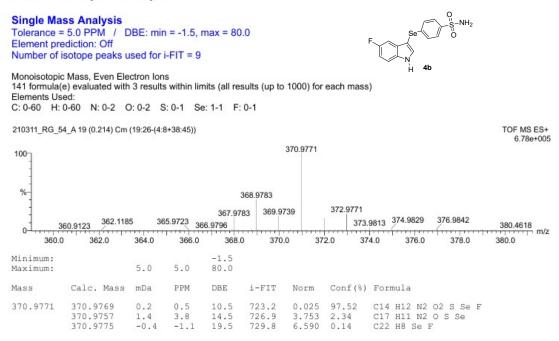


Figure 73. HRMS spectrum of compound 4b

### **Single Mass Analysis**

Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9

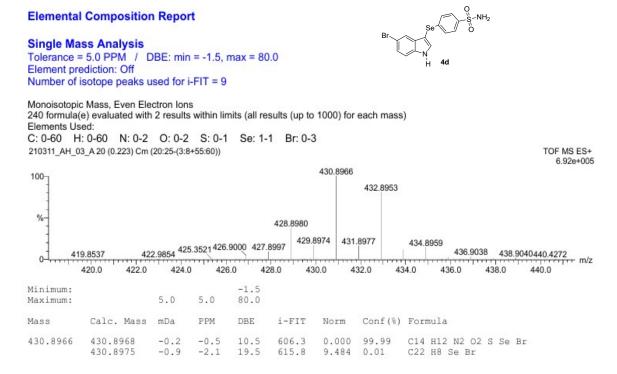
Monoisotopic Mass, Even Electron Ions 147 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-2 O: 0-2 S: 0-1 CI: 0-1 Se: 1-1 210311\_AH\_30\_A 17 (0.197) Cm (14:25-(3:8+57:60))

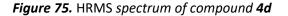
TOF MS ES+ 2.05e+006

100					300.947	0			
%-				384.948	35	388.9	460		
377.22	244 380.9527	382.949 381.9431	93 383.94	74	385.9433	387.9458	390.	9463 391.9507	395.9405 206 0252
			384.		386.0	388.0	390.0	7	390.9352 m/z
Minimum: Maximum:		5.0	5.0	-1.5 80.0					
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula	
386.9475	386.9473 386.9480 386.9461	-0.5	0.5 -1.3 3.6	10.5 19.5 24.5	1257.0 1254.0 1261.8	3.007 0.051 7.805	4.94 95.02 0.04	C14 H12 N2 O2 S C22 H8 Cl Se C23 H3 N2 Se	Cl Se

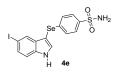
396 0475

# Figure 74. HRMS spectrum of compound 4c





### **Single Mass Analysis**



Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9

### Monoisotopic Mass, Even Electron Ions 162 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-2 O: 0-2 S: 0-1 Se: 1-1 I: 0-1

210311\_AH\_33\_A 25 (0.277) Cm (25:37-(1:8+57:60))

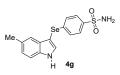
TOF MS ES+ 1.38e+006

100				478.	.8835						1.566	000
%			476	.8846								
	470.9431 473.3077	474.8860	475.8868	477.8846	T.		.8878 482.	8867	485.8565	486.8506	489.9178	
470		474.0	476.0	478.0	480.	.0 4	82.0	484.0	486.0	488.0	490.0	m/z
Minimum: Maximum:		5.0		-1.5 80.0								
Mass	Calc. Mass	mDa	PPM	DBE i-	FIT	Norm	Conf(%)	Formul	a			
478.8835	478.8829 478.8836	0.6 -0.1				0.026 3.678	97.47 2.53	C14 H1 C22 H8	2 N2 O2 Se I	S Se I		

# Figure 76. HRMS spectrum of compound 4e

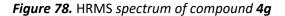
#### **Elemental Composition Report** -NH<sub>2</sub> Single Mass Analysis MeC Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9 Monoisotopic Mass, Even Electron lons 124 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-2 O: 0-4 S: 0-1 Se: 1-1 210311\_AH\_12\_A 17 (0.197) Cm (14:24-(5:9+42:60)) TOF MS ES+ 2.26e+006 382,9972 100-% 380.9980 379.9988 384.9967 374.3056 386.0007\_387.0018389.0050 393.9849395.9965 377.0026\_377.9928 368.0086\_368.9742 0-– m/z 367.5 370.0 372.5 375.0 377.5 380.0 382.5 385.0 387.5 390.0 392.5 395.0 -1.5 Minimum: Maximum: 5.0 5.0 80.0 Calc. Mass mDa DBE Conf(%) Formula Mass PPM i-FIT Norm 382.9972 382.9969 0.3 0.8 10.5 907.9 0.003 99.67 C15 H15 N2 O3 S Se C23 H11 O Se 382.9975 -0.3 -0.8 19.5 913.6 5.709 0.33

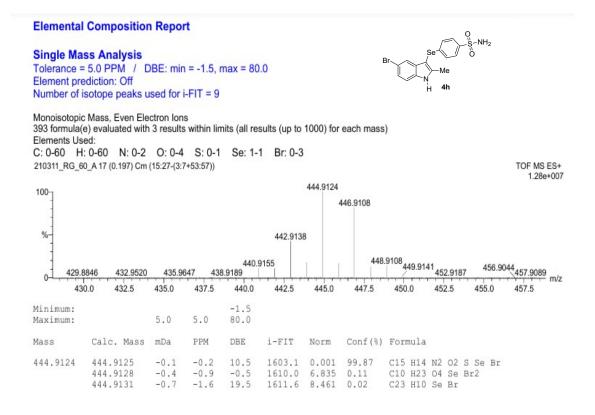
Figure 77. HRMS spectrum of compound 4f

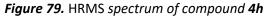


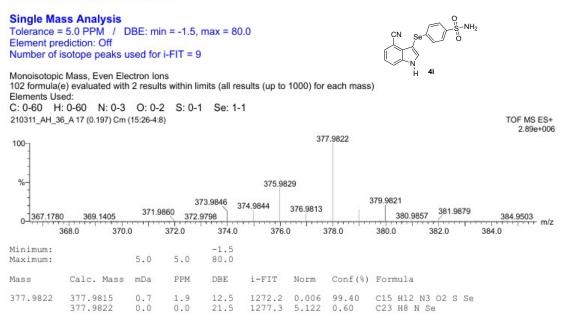
Single Mass Analysis Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9

#### Monoisotopic Mass, Even Electron lons 119 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-2 O: 0-4 S: 0-1 Se: 1-1 210311\_RG\_71\_A 17 (0.197) Cm (16:26-(3:9+52:60)) TOF MS ES+ 3.91e+006 367.0020 100-365.0032 % 365.9983 363.0040 364.0026 369 0022 370.0051 371.0051 376,1812 377,1845 373.0103 359,1973 361.0081 361 9972 0 --- m/z 376.0 358.0 360.0 362.0 370.0 372.0 374.0 364.0 366.0 368.0 Minimum: -1.5 5.0 5.0 Maximum: Mass Calc. Mass mDa PPM DBE i-FIT Conf(%) Formula Norm 367.0020 367.0019 1089.9 0.030 97.02 C15 H15 N2 O2 S Se 0.1 0.3 10.5 367.0026 -0.6 -1.6 19.5 1093.4 3.514 2.98 C23 H11 Se

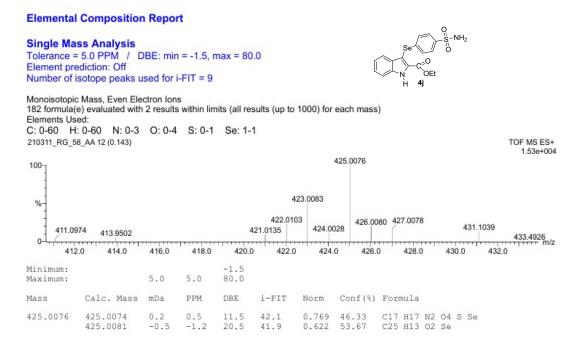








# Figure 80. HRMS spectrum of compound 4i





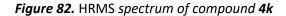
### Single Mass Analysis

Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9

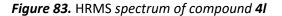
Monoisotopic Mass, Even Electron Ions 161 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-3 O: 0-3 S: 0-1 Se: 1-1 210311\_AH\_46\_A 16 (0.177) Cm (15:21-(4:7+43:60))

TOF MS ES+ 8.68e+006

100						486	0396		0.000+000
%-					484.0	0408			
473.8801	475.8777	479.2538	480.045	482.0424 4	483.0435	485.0430		8.0398 489.0427 490.0425	492.0474 494.0469
474.0	476.0	478.0	480.0	482					492.0 494.0
Minimum: Maximum:		5.0	5.0	-1.5 80.0					
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula	
486.0396	486.0391 486.0397		1.0 -0.2	15.5 24.5	654.3 661.2	0.001 6.927	99.90 0.10	C22 H20 N3 O3 S C30 H16 N O Se	Se



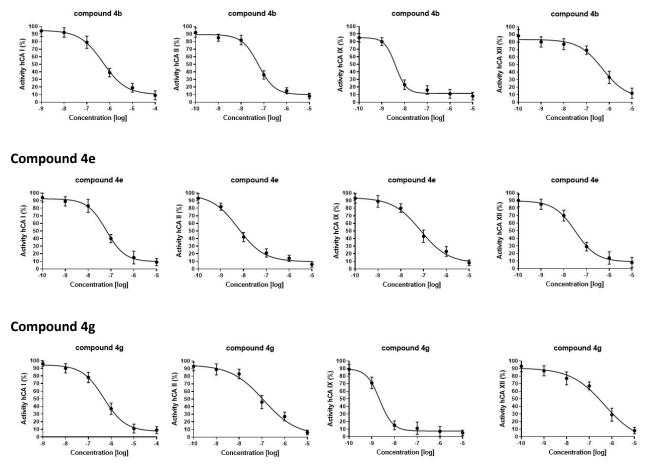
#### **Elemental Composition Report** 0 \_S-NH2 Single Mass Analysis Tolerance = 5.0 PPM / DBE: min = -1.5, max = 80.0 Element prediction: Off Number of isotope peaks used for i-FIT = 9 Monoisotopic Mass, Even Electron Ions 73 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass) Elements Used: C: 0-60 H: 0-60 N: 0-2 O: 0-2 S: 0-1 Se: 1-1 210311\_AH\_35\_A 28 (0.303) Cm (28:43-(4:8+56:60)) TOF MS ES+ 2.47e+006 367.0028 100-% 365.0036 363.0045 364.0045 366.0010 369.0024 361.0080 361.9987 370.0059 373.0103 375.0099 375.9984 355.3018.355.9503 0 - m/z 354.0 356.0 358.0 360.0 362.0 364.0 366.0 368.0 370.0 372.0 374.0 376.0 378.0 Minimum: -1.5 5.0 5.0 80.0 Maximum: Mass Calc. Mass mDa PPM DBE i-FIT Norm Conf(%) Formula 367.0019 1695.9 367.0028 0.9 2.5 10.5 0.003 99.72 C15 H15 N2 O2 S Se 367.0026 0.2 0.5 19.5 1701.8 5.866 0.28 C23 H11 Se



# 5. Carbonic anhydrase inhibition

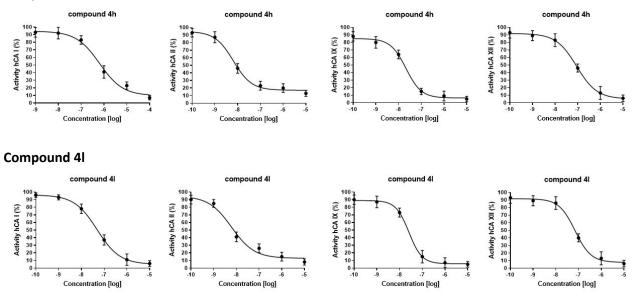
An Applied Photophysics stopped-flow instrument was used to assay the CA catalyzed  $CO_2$  hydration activity.<sup>15</sup> Phenol red (at a concentration of 0.2 mM) was used as an indicator, working at the absorbance maximum of 557 nm, with 20 mM Hepes (pH 7.4) as a buffer, and 20 mM Na<sub>2</sub>SO<sub>4</sub> (to maintain constant ionic strength), following the initial rates of the CA-catalyzed  $CO_2$  hydration reaction for a period of 10–100 s. The  $CO_2$  concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters and inhibition constants.<sup>16</sup> Enzyme concentrations ranged between 5-12 nM. For each inhibitor, at least six traces of the initial 5–10% of the reaction were used to determine the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of the inhibitor (0.1 mM) were prepared in distilled–deionized water and dilutions up to 0.01 nM were done thereafter with the assay buffer. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature prior to the assay, to allow for the formation of the E–I complex. The inhibition constants were obtained by non-linear least-squares methods using PRISM 3 and the Cheng-Prusoff equation as reported earlier and represent the mean from at least three different determinations. All CA isoforms were recombinant proteins obtained in house, as reported earlier.<sup>17-19</sup>

# 6. Human Carbonic Anhydrase activity



# **Compound 4b**

# Compound 4h



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