Supporting Information for

HFIP Mediated Oxime Ether Synthesis: A Metal, Base and Additive Free Approach

Dibyashree Dolakasharia^a and Utpal Bora^{a*}

^aDepartment of Chemical Sciences, Tezpur University, Napaam, Tezpur, Assam, India, Pin 784028 Email: ubora@tezu.ernet.in

Table of Contents

1.	General Information	S2
2.	Preparation of Starting Materials	S2
3.	General Procedure	S2
4.	Characterization Data of Products	S4
5.	Scaled-up Reaction	S 8
6.	Experiments for Mechanism Investigation	S 8
7.	Copies of ¹ H and ¹³ C NMR Spectra	S 11

1. General Information

All reactions were carried out in Tarsons spinot digital magnetic stirrers under standard conditions. All chemicals except oximes (excluding 1i, 1v, 1w and 1x) were purchased commercially from Sigma Aldrich, Merck, Alfa Aesar or BLDPharm and used without further purification. The oximes were synthesized according to literature procedures. Solvents used for extraction and chromatographic separations were distilled prior use. All reactions involving heating are carried out in an oil bath.

Chromatography: Analytical thin layer chromatography (TLC) was performed using Merck silica gel 60F254 plates using short wave (254 nm) UV light. Column chromatography purifications were performed over silica gel (60-120 mesh).

¹**H** NMR and ¹³**C** NMR spectra were recorded on a JEOL JNM ECS NMR spectrometer (400 MHz and 101 MHz respectively) using CDCl₃ as solvent and TMS as internal standard. Chemical shifts (δ) are reported in parts per million (ppm) relative to the central peak of the solvent. And multiplicities are indicated as: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet) and dd (doublet of doublet). Coupling constants (*J* values) are given in hertz (Hz).

Melting points (MP) were determined by JSGW Digital Melting Point Apparatus.

CHN content: The organic content (wt % C, H, N) in the synthesized compounds was determined by combustion analysis using a PERKIN ELMER 2400 SERIES 2 CHN analyzer. FTIR spectra were measured on a Perkin Elmer Frontier MIR-FIR FT-IR spectrophotometer with the range 4000–400 cm–1 (resolution: 4 cm⁻¹) using KBr as the reference.

UV-Visible experiments were carried out in Shimadzu, UV-2550 spectrophotometer.

2. Preparation of Starting Materials

All oximes (except 1i, 1v, 1w, 1x) were prepared according to the literature procedure.^{1,2}

$$\begin{array}{c} 0 \\ R_1 \\ \hline R_2 \end{array}^{+} NH_2OH.HCI \\ \hline CH_3OH, reflux, 6 h \end{array} \xrightarrow{HO} N \\ \hline R_1 \\ \hline R_2 \end{array}$$

For water soluble ketones or aldehydes, the mixture of ketones or aldehydes, hydroxylamine hydrochloride (1.6 eq.) and sodium acetate (2 eq.) in CH_3OH was stirred under reflux and detected by TLC. After the reaction was completed, the mixture was poured into ice water. Precipitate is formed, filtered off and washed with little water and dried.

$$R_1 + NH_2OH.HCI$$
 $HO N R_1 + NH_2OH.HCI$ $HO N R_1 + R_2 + NH_2OH.HCI$

For water insoluble ketones or aldehydes, the mixture of ketones or aldehydes, hydroxylamine hydrochloride and pyridine in ethanol was stirred under reflux and detected by TLC. After the reaction was completed, the mixture was stirred in an ice bath until the oxime crystallizes. It is then filtered off, washed with little water and dried.



For benzophenone derivatives, the mixture of benzophenone derivative, hydroxylamine hydrochloride and sodium hydroxide in ethanol was stirred under reflux for 5 minutes and detected by TLC. After the reaction was completed, the mixture was poured into ice water. Precipitate is formed, filtered off and washed with little water and dried.

References

- 1. I. Vogel, Practical organic chemistry. 1974.
- 2. A. Lachman, Benzophenone oxime. Organic Syntheses. 2003, 70-70.
- 3. H. Zhuang, Q. Hou, F. Han, H. Lv, and C. Miao, Green Chem. 2023, 25, 310-317.

3. General Procedure for the synthesis of oxime ether

A 50 mL round bottomed flask was charged with oxime **1** (0.25 mmol) and triphenylmethanol **2** (0.5 mmol) in presence of 1.5 mL HFIP at 30 °C for 4 h. After completion of reaction (confirmed by TLC), reaction mixture was extracted with ethyl acetate and water. The organic layer was dried over Na_2SO_4 and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel, hexane/ethyl acetate) to give the desired products **3a-3v**. **3w** was obtained through simple filtration after work up.

4. Characterization Data of Products

Diphenylmethanone O-trityl oxime (3a)



White solid, 109.6 mg, 99% yield; m.p. = 155-157 °C (lit.³ 155.4-155.7 °C); ¹H NMR (500 MHz, CDCl₃) δ 7.49-7.38 (m, 5H), 7.34-7.17 (m, 20H). ¹³C NMR (101 MHz, CDCl₃) δ 156.7 (s), 144.7 (s), 136.8 (s), 133.9 (s), 129.3 (s), 129.2 (s), 128.8 (s), 128.1 (s), 128.0 (s), 127.6 (s), 127.1 (s), 91.4 (s).

(Z)-(4-chlorophenyl)(phenyl)methanone O-trityl oxime (3b)



White solid, 240.5 mg, 96% yield; m.p. = 130-132 °C (lit.³ 130.7-131.5 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.50 (d, J = 9.1 Hz, 2H), 7.44 (d, J = 6.6 Hz, 2H), 7.37-7.21 (m, 18H). ¹³C NMR (101 MHz, CDCl₃) δ 155.8 (d, J = 15.0 Hz), 144.6 (d, J = 3.3 Hz), 136.5 (s), 135.3 (d, J = 5.0 Hz), 134.8 (s), 133.4 (s), 132.2 (s), 131.0 (s), 129.3 (d, J = 1.3 Hz), 128.4 (dd, J = 16.5, 6.3 Hz), 128.1 (s), 127.7 (d, J = 3.0 Hz), 127.2 (d, J = 3.4 Hz), 91.7 (d, J = 12.6 Hz).

(Z)-(4-bromophenyl)(phenyl)methanone O-trityl oxime (3c)



White solid, 111.4 mg, 86% yield; m.p. = 156-158 °C (lit.³ 155.2-156.0 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.65-7.60 (m, 1H), 7.60-7.08 (m, 23H). ¹³C NMR (101 MHz, CDCl₃) δ 155.8 (d, *J* = 18.5 Hz), 144.5 (d, *J* = 3.2 Hz), 136.3 (s), 135.7 (s), 133.3 (s), 132.7 (s), 131.5-131.0 (m), 129.7-128.9 (m), 128.3 (d, *J* = 3.4 Hz), 128.0 (s), 127.6 (d, *J* = 3.2 Hz), 127.2 (d, *J* = 3.2 Hz), 123.3 (d, *J* = 51.8 Hz), 91.7 (d, *J* = 11.3 Hz).

(Z)-1-phenylethan-1-one O-trityl oxime (3d)



White solid, 92.3 mg, 96% yield; m.p. = 139-141 °C (lit.³ 139.8-140.7 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.47-7.39 (m, 8H), 7.33-7.23 (m,

13H), 2.41 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 154.1(s), 144.9 (s), 137.0 (s), 129.3 (s), 128.9 (s), 128.3 (s), 127.6 (s), 127.1 (s), 126.2 (s), 91.0 (s), 13.2 (s).

(Z)-1-(4-chlorophenyl)ethan-1-one O-trityl oxime (3e)



White solid, 91.6 mg, 89% yield; m.p. = 174-176 °C (lit.³ 173.3-173.6 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.39-7.35 (m, 8H), 7.31-7.22 (m, 12H), 2.37 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 153.1 (s), 144.7 (s), 143.0 (s), 135.1 (d, J = 53.2 Hz), 129.3 (d, J = 12.1 Hz), 128.4 (s), 128.0 (d, J = 1.9 Hz), 127.5 (d, J = 17.1 Hz), 127.2 (s), 91.2 (s), 13.0 (s).

(Z)-1-(4-bromophenyl)ethan-1-one O-trityl oxime (3f)



White solid, 104.0 mg, 91% yield; m.p. = 122-124 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.40-7.36 (m, 8H), 7.33-7.24 (m, 12H), 2.38 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) NMR (101 MHz,) δ 153.2 (s), 144.7 (s), 135.9 (s), 131.4 (s), 129.3 (s), 128.0 (d, *J* = 1.4 Hz), 127.7 (d, *J* = 10.8 Hz), 127.2 (s), 123.2 (s), 91.2 (s), 13.0 (s). Anal. calcd. for C₂₇H₂₂BrNO: C, 71.06; H, 4.86; N, 3.07; found: C, 71.14; H, 4.62; N, 3.17.

(Z)-1-(4-iodophenyl)ethan-1-one O-trityl oxime (3g)



White solid, 111.3 mg, 91% yield; m.p. = 120-122 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.60-7.56 (m, 2H), 7.39-7.35 (m, 6H), 7.32-7.23 (m, 10H), 7.19-7.15 (m, 2H), 2.36 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) NMR (101 MHz) δ 153.3 (s), 144.7 (s), 137.4 (s), 136.5 (s), 129.3 (s), 127.9 (s), 127.6 (s), 127.2 (s), 95.0 (s), 91.2 (s), 12.9 (s). Anal. calcd. for C₂₇H₂₂INO: C, 64.42; H, 4.41; N, 2.78; found: C, 64.24; H, 4.53; N, 2.66.

(Z)-1-(p-tolyl)ethan-1-one O-trityl oxime (3h)



White solid, 84.5 mg, 86% yield; m.p. = 148-150 °C (lit.³ 149.9-150.7 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.44-7.40 (m, 6H), 7.39-7.35 (m, 2H), 7.33-7.24 (m, 10H), 7.08 (dd, *J* = 8.5, 0.6 Hz, 2H), 2.40 (s, 3H), 2.32 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 154.0 (s), 145.0 (s), 138.9 (s), 134.3 (s), 129.4 (s), 129.0 (s), 127.6 (s), 127.1 (s), 126.1 (s), 90.9 (s), 21.3 (s), 13.2 (s).

(Z)-benzaldehyde O-trityl oxime (3i)



White solid, 59.9 mg, 66% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.28 (s, 1H), 7.46-7.40 (m, 8H), 7.34-7.24 (m, 13C). ¹³C NMR (101 MHz, CDCl₃) δ 148.7 (s), 144.5 (s), 132.8 (s), 129.7 (s), 129.4 (s), 128.6 (s), 127.6 (s), 127.3 (s), 127.2 (s), 91.3 (s); (lit.³).

(Z)-4-chlorobenzaldehyde O-trityl oxime (3j)



White solid, 79.5 mg, 80% yield; m.p. = 122-124 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.23 (s, 1H), 7.42-7.26 (m, 20H). ¹³C NMR (101 MHz, CDCl₃) δ 147.6 (s), 144.4 (s), 135.5 (s), 131.3 (s), 129.4 (s), 128.9 (s), 128.4 (s), 127.7 (s), 127.3 (s), 91.5 (s). Anal. calcd. for C₂₆H₂₀ClNO: C, 78.48; H, 5.07; N, 3.52; found: C, 78.35; H, 5.22; N, 3.45.

(Z)-4-bromobenzaldehyde O-trityl oxime (3l)



White solid, 101.3 mg, 92% yield; m.p. = 139.8-140.7 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.22 (s, 1H), 7.43-7.39 (m, 8H), 7.35-7.24 (m, 12H). ¹³C NMR (101 MHz, CDCl₃) δ 147.7 (s), 144.4 (s), 131.9 (s), 131.7 (s), 129.4 (s), 128.6 (s), 127.7 (s), 127.3 (s), 123.9 (s), 91.6 (s). Anal. calcd. for C₂₆H₂₀BrNO: C, 70.60; H, 4.56; N, 3.17; found: C, 70.78; H, 4.62; N, 3.06.

(Z)-3-bromobenzaldehyde O-trityl oxime (3m)



White solid, 84.0 mg, 76% yield; m.p. = 142-144 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.19 (s, 1H), 7.55 (t, J = 1.7 Hz, 1H), 7.43-7.40 (m, 1H), 7.39-7.36 (m, 6H), 7.33-7.24 (m, 11H), 7.14 (t, J = 7.9 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 147.4 (s), 144.3 (s), 134.8 (s), 132.6 (s), 130.1 (s), 129.9 (s), 129.4 (s), 127.7 (s), 127.3 (s), 125.9 (s), 122.8 (s), 91.6 (s). Anal. calcd. for C₂₆H₂₀BrNO: C, 70.60; H, 4.56; N, 3.17; found: C, 70.72; H, 4.71; N, 3.19.

(Z)-4-nitrobenzaldehyde O-trityl oxime (3n)



Pale yellow solid, 101.2 mg, 99% yield; m.p. = 157-159 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 1H), 8.13 (d, *J* = 9.1 Hz, 2H), 7.56 (d, *J* = 9.2 Hz, 2H), 7.52-7.08 (m, 16H). ¹³C NMR (101 MHz, CDCl₃) δ 148.3 (s), 146.7 (s), 144.0 (s), 138.8 (s), 129.3 (s), 127.7 (s), 127.7 (s), 127.5 (s), 124.0 (s), 92.3 (s). Anal. calcd. for C₂₆H₂₀N₂O₃: C, 76.46; H, 4.94; N, 6.86; found: C, 76.54; H, 5.04; N, 6.82.

(Z)-3-nitrobenzaldehyde O-trityl oxime (30)



White sticky solid, 90.9 mg, 89% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 1H), 8.24-8.20 (m, 1H), 8.14 (m, 1H), 7.74 (dt, *J* = 7.8, 1.3 Hz, 1H), 7.44 (t, *J* = 8.0 Hz, 1H), 7.42-7.25 (m, 16H). ¹³C NMR (101 MHz, CDCl₃) δ 148.5 (s), 146.6 (s), 144.0 (s), 134.5 (s), 132.5 (s), 129.7 (s), 129.3 (s), 127.8 (s), 127.5 (s), 124.2 (s), 122.1 (s), 92.0 (s), 77.4 (s), 77.1 (s), 76.8 (s). Anal. calcd. for C₂₆H₂₀N₂O₃: C, 76.46; H, 4.94; N, 6.86; found: C, 76.68; H, 4.80; N, 6.74.

(Z)-4-methylbenzaldehyde O-trityl oxime (3p)



Off white solid, 73.9 mg, 78% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.22 (s, 1H), 7.76-6.77 (m, 19H), 2.31 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 148.6 (s), 144.6 (s), 139.8 (s), 130.0 (s), 129.4 (s), 129.3 (s), 127.6 (s), 127.2 (s), 127.1 (s), 91.1 (s), 21.5 (s). Anal. calcd. for C₂₇H₂₃NO: C, 85.91; H, 6.14; N, 3.71; found: C, 85.86; H, 6.32; N, 3.62.

Diphenylmethanone O-(diphenyl(p-tolyl)methyl) oxime (3q)



White solid, 102.9 mg, 91% yield; m.p. = 95-97 °C (lit.³ 94.8-95.0 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.52-7.43 (m, 5H), 7.36-7.21 (m, 18H), 7.10 (dd, J = 8.6, 0.6 Hz, 2H), 2.34 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 156.7 (s), 144.9 (s), 141.8 (s), 136.8 (s), 136.7 (s), 134.0 (s), 129.4 (s), 129.30 (s), 129.26 (s), 129.1 (s), 128.8 (s), 128.3 (s), 128.2 () 120.12 () 120.06 (c) 127.6 (c) 127.0 (c) 01.2 (c) 21.2 (c)

(s), 128.13 (s), 128.06 (s), 127.6 (s), 127.0 (s), 91.3 (s), 21.2 (s).

Diphenylmethanone O-(1,1-diphenylethyl) oxime (3r)



White solid, 56.6 mg, 60% yield; m.p. = 75-77 °C (lit.³ 74.6-75.3 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.50-7.41 (m, 5H), 7.34-7.32 (m, 2H), 7.28-7.26 (m, 1H), 7.25-7.22 (m, 11H), 7.21-7.17 (m, 2H), 2.13 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 156.8 (s), 146.5 (s), 136.9 (s), 134.0 (s), 129.4 (s), 129.1 (s), 128.7 (s), 128.1 (s), 127.94 (s), 127. 90 (s), 126.8 (s), 126.7 (s), 85.9 (s), 27.4 (s).

Diphenylmethanone O-(2-phenylpropan-2-yl) oxime (3s)



Colourless oily liquid, 42.5 mg, 54% yield; ¹H NMR (400 MHz, CDCl₃) δ 7.47-7.39 (m, 7H), 7.36-7.18 (m, 9H), 1.69 (s, 6H).¹³C NMR

(101 MHz, CDCl₃) δ 155.7 (s), 147.5 (s), 137.3 (s), 133.9 (s), 129.6
(s), 129.0 (s), 128.6 (s), 128.1 (s), 128.02 (s), 128.01 (s), 128.00 (s), 126.5 (s), 125.4 (s), 82.4 (s), 28.3 (s); (lit.³).

Diphenylmethanone O-benzhydryl oxime (3t)



White solid, 53.9 mg, 59% yield; m.p. = 102-104 °C (lit.³ 101.8-102.5 °C); ¹H NMR (400 MHz, CDCl₃) δ 7.44-7.38 (m, 7H), 7.32-7.24 (m, 14H), 6.38 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 157.5 (s), 141.8 (s), 136.6 (s), 133.6 (s), 129.4 (s), 128.8 (s), 128.5 (s), 128.3 (s), 128.2 (s), 128.1 (s), 128.0 (s), 127.4 (s), 127.3 (s), 87.3 (s).

Diphenylmethanone O-(bis(4-methoxyphenyl)methyl) oxime (3u)



White solid, 56.3 mg, 53% yield; ¹H NMR (400 MHz, CDCl₃) δ 7.43-7.36 (m, 7H), 7.31-7.24 (m, 4H), 7.19-7.15 (m, 4H), 6.83-6.79 (m, 4H), 6.29 (s, 1H), 3.76 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 158.9 (s), 157.2 (s), 136.7 (s), 134.3 (s), 133.6 (s), 129.5 (s), 129.3 (s), 128.8 (s), 128.6 (s), 128.2 (s), 128.1 (s), 128.0 (s), 113.6 (s), 86.5 (s), 55.3 (s). Anal. calcd. for C₂₈H₂₅NO₃: C, 79.41; H, 5.95; N, 3.31; found: C, 79.53; H, 6.04; N, 3.25.

(Z)-1-(pyridin-3-yl)ethan-1-one O-trityl oxime (3v)



Off white solid, 76.8 mg, 81% yield; m.p. = 160-162 °C (lit.³ 160.8-161.2 °C); ¹H NMR (400 MHz, CDCl₃) δ 8.70 (dd, J = 2.3, 0.8 Hz, 1H), 8.50 (dd, J = 4.8, 1.7 Hz, 1H), 7.73-7.70 (m, 1H), 7.42-7.39 (m, 6H), 7.34-7.24 (m, 9H), 7.19-7.15 (m, 1H), 2.43 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 152.0 (s), 149.8 (s), 147.6 (s), 144.6 (s), 133.4 (s), 132.7 (s), 129.3 (s), 127.7 (s), 127.3 (s), 123.2 (s), 91.5 (s), 12.9 (s).

9*H*-Fluoren-9-one *O*-trityl oxime (3w)



Yellow solid, 55.7 mg, 51% yield; m.p. = 152-154 °C (lit.³ 152.5-153.0 °C); ¹H NMR (400 MHz, CDCl₃) δ 8.55-8.52 (m, 1H), 7.68-7.64 (m, 1H), 7.58 (dt, *J* = 7.6, 0.9 Hz, 1H), 7.52-7.50 (m, 1H), 7.48-7.45 (m, 6H), 7.43 (dd, *J* = 7.5, 1.2 Hz, 1H), 7.37-7.28 (m, 12H), 7.17 (td, *J* = 7.5, 1.0 Hz, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 152.0 (s), 144.3 (s), 141.8 (s), 140.0 (s), 136.2 (s), 131.0 (s), 130.4 (s), 129.7 (s), 129.6 (s),

129.0 (s), 128.4 (s), 127.8 (s), 127.7 (s), 127.4 (s), 122.0 (s), 120.0 (s), 119.8 (s), 93.4 (s).

(1E,2E)-1,2-diphenylethane-1,2-dione O,O-ditrityl dioxime (3x)



White solid, 127.6 mg, 70% yield; m.p. > 200 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.30-6.99 (m, 40H). ¹³C NMR (101 MHz, CDCl₃) δ 155.2 (s), 144.0 (s), 131.6 (s), 129.5 (s), 129.3 (s), 128.2 (s), 127.4 (s), 127.2 (s), 127.0 (s), 91.8 (s). Anal. calcd. for C₅₂H₄₀N₂O₂: C, 86.16; H, 5.56; N, 3.86; found: C, 86.25; H, 5.68; N, 3.74.

5. Scaled-up Reaction



To a 100 mL round bottomed flask, diphenylmethanone oxime (0.5916 g, 3 mmol, 1.0 equiv.), triphenylmethanol (1.5618 g, 6 mmol, 2.0 equiv.) and HFIP (6 mL) was added. The resulting mixture was stirred at 30 °C for 4 hours. The reaction mixture was extracted with ethyl acetate and water. The organic layer was dried over Na_2SO_4 and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel, hexane/ethyl acetate) to give the desired product in 91% yield.

6. The Experiments for Mechanism Investigation

(1) The Reaction Phenomenon of Triphenylmethanol and HFIP



Figure S1. (a) 0.5 mmol 2a, (b) 1.5 mL HFIP, colorless and transparent; (c) 2a was slowly added to (b), the color was changed into transparent yellow.

(2) UV-Vis Spectra Analysis

Preparation of diphenylmethanone oxime solution:

Diphenylmethanone oxime (0.0197 g, 0.1 mmol) was dissolved in 3 mL CH₃CN, and the UV-Vis absorption spectra was measured in a 4 mL quartz cuvette as shown in Figure S2, (a).

Preparation of triphenylmethanol solution:

Triphenylmethanol (0.0260 g, 0.1 mmol) was dissolved in 3 mL CH₃CN, and the UV-Vis absorption spectra was measured in a 4 mL quartz cuvette as shown in Figure S2, (b).

Preparation of HFIP solution:

HFIP (10.5 μ L, 0.1 mmol) was dissolved in 3 mL CH₃CN, and the UV-Vis absorption spectra was measured in a 4 mL quartz cuvette as shown in Figure S2, (c).

Preparation of 3a (oxime ether) solution:

Oxime ether **3a** (0.0439 g, 0.1 mmol) was dissolved in 3 mL CH₃CN, and the UV-Vis absorption spectra was measured in a 4 mL quartz cuvette as shown in Figure S2, (d).

The procedure of drawing the UV-Vis absorption spectra of reaction:

Triphenylmethanol (0.5 mmol, 0.1302 g) and HFIP (1.5 mL) were added to a round bottom bottomed flask and stirred. Then, diphenylmethanone oxime was added into the system batch by batch and measured after reacting for 5 min and standing.



Figure S2. The UV-Vis absorption spectra of (a) diphenylmethanone oxime; (b) triphenylmethanol; (c) HFIP; (d) Oxime ether **3a**.

FT-IR Analysis



Figure S3. The FT-IR absorption spectra of (a) triphenylmethanol in CH₃CN; (b) HFIP in CH₃CN; (c) triphenylmethanol + HFIP in CH₃CN.

7. Copies of ¹H NMR and ¹³C NMR Spectra ¹H NMR (500 MHz) Spectrum of 3a in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3a in CDCl₃



190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 Chemical shift (ppm)

¹H NMR (400 MHz) Spectrum of 3b in CDCl₃



¹H NMR (400 MHz) Spectrum of 3c in CDCl₃



¹H NMR (400 MHz) Spectrum of 3d in CDCl₃



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 Chemical shift (ppm)

¹H NMR (400 MHz) Spectrum of 3e in CDCl₃



¹H NMR (400 MHz) Spectrum of 3f in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3f in CDCl₃





~ ~ ~

S18

¹H NMR (400 MHz) Spectrum of 3h in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3h in CDCl₃



¹H NMR (400 MHz) Spectrum of 3i in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3i in CDCl₃



¹H NMR (400 MHz) Spectrum of 3j in CDCl₃





¹H NMR (400 MHz) Spectrum of 3l in CDCl₃



250 230 210 190 170 150 130 110 90 80 70 60 50 40 30 20 10 0 -20 -40 Chemical shift (ppm)

¹H NMR (400 MHz) Spectrum of 3m in CDCl₃



¹H NMR (400 MHz) Spectrum of 3n in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3n in CDCl₃



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 Chemical shift (ppm)

¹H NMR (400 MHz) Spectrum of 30 in CDCl₃



¹³C NMR (101 MHz) Spectrum of 30 in CDCl₃



¹H NMR (400 MHz) Spectrum of 3p in CDCl₃



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 Chemical shift (ppm)

¹H NMR (400 MHz) Spectrum of 3q in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3q in CDCl₃







¹H NMR (400 MHz) Spectrum of 3s in CDCl₃





¹H NMR (400 MHz) Spectrum of 3t in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3t in CDCl₃



¹H NMR (400 MHz) Spectrum of 3u in CDCl₃



¹H NMR (400 MHz) Spectrum of 3v in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3v in CDCl₃



¹H NMR (400 MHz) Spectrum of 3w in CDCl₃



¹³C NMR (101 MHz) Spectrum of 3w in CDCl₃





¹³C NMR (101 MHz) Spectrum of 3x in CDCl₃

